

# Renewable carbon products by using CO<sub>2</sub>

Project deliverable D2 of KB Circular & Climate Neutral: Exploring CO<sub>2</sub> options (KB-34-003-007)

Marieke E. Bruins, Marc P. Lanting, Alexander T. Boedijn, Anna Dörper, Juliën A. Voogt,  
Jan Peter Nap, Daan van Es, Martijn Bekker.

PUBLIC





# Renewable carbon products by using CO<sub>2</sub>

Project deliverable D2 of KB Circular & Climate Neutral: Exploring CO<sub>2</sub> options (KB-34-003-007)

Authors: Marieke E. Bruins, Marc P. Lanting, Alexander T. Boedijn, Anna Dörper, Juliën A. Voogt, Jan Peter Nap, Daan van Es, Martijn Bekker.

Institute: Wageningen Food & Biobased Research

This study was carried out by Wageningen Food & Biobased Research.

Wageningen Food & Biobased Research  
Wageningen, November 2025

---

Public

Report 2767  
DOI: 10.18174/704688

---

WFBR Project number: 6224140000

Version: Final

Reviewer: H.L. Bos

Approved by: J.M. Jetten

Carried out by: Wageningen Food & Biobased Research

Subsidised and funded by: KB Circular & Climate Neutral (KB-34-003-007) and KB Biobased & Circular Society (KB-51-000-004) that is supported by financing from the Dutch Ministry of Agriculture, Fisheries, Food security and Nature.

This report is: Public

The client is entitled to disclose this report in full and make it available to third parties for review. Without prior written consent from Wageningen Food & Biobased Research, it is not permitted to:

- a. partially publish this report created by Wageningen Food & Biobased Research or partially disclose it in any other way;
- b. use this report for the purposes of making claims, conducting legal procedures, for (negative) publicity, and for recruitment in a more general sense;
- c. use the name of Wageningen Food & Biobased Research in a different sense than as the author of this report.

The research that is documented in this report was conducted in an objective way by researchers who act impartial with respect to the client(s) and sponsor(s). This report can be downloaded for free at <https://doi.org/10.18174/704688> or at [www.wur.eu/wfbr](http://www.wur.eu/wfbr) (under publications).

© 2025 Wageningen Food & Biobased Research, institute within the legal entity Stichting Wageningen Research.

PO box 17, 6700 AA Wageningen, The Netherlands, T + 31 (0)317 48 00 84, E [info.wfbr@wur.nl](mailto:info.wfbr@wur.nl), [www.wur.eu/wfbr](http://www.wur.eu/wfbr).

All rights reserved. No part of this publication may be reproduced, stored in a retrieval system of any nature, or transmitted, in any form or by any means, electronic, mechanical, photocopying, recording or otherwise, without the prior permission of the publisher. The publisher does not accept any liability for inaccuracies in this report.

---

# Contents

<b>Summary</b>	<b>4</b>
<b>1 Setting the scene</b>	<b>5</b>
<b>2 Future Carbon sources</b>	<b>6</b>
2.1 Conventional: CO <sub>2</sub> from fossil sources	6
2.2 Direct Air Capture	6
2.3 Industrial CO <sub>2</sub> from biobased processes	7
2.4 CO <sub>2</sub> from (residual) biomass	7
<b>3 Carbon routes to fuels, chemicals, and materials</b>	<b>8</b>
3.1 Carbon from air captured CO <sub>2</sub>	8
3.1.1 Direct use	9
3.1.2 Direct use for plant growth	9
3.1.3 Indirect use	9
3.1.4 Indirect use through chemical conversion	9
3.1.5 Indirect use through microbial conversion	9
<b>4 How to select routes for carbon use</b>	<b>10</b>
<b>5 Methanol and ethanol production from CO<sub>2</sub></b>	<b>11</b>
5.1 Scenario's for methanol and ethanol production from CO <sub>2</sub>	11
5.2 2050 techno-economic outlook for ethanol and methanol production	13
5.2.1 Conventional methanol and ethanol production routes will be the cheapest methods to produce methanol and ethanol	13
5.2.2 Alternative methanol and ethanol routes require more energy than conventional routes	13
5.2.3 Incentives, such as CO <sub>2</sub> tax, play an important role in enabling alternative fossil free production routes	13
5.2.4 CO <sub>2</sub> from point sources instead of DAC is cheaper, a good starting point for cost reduction	14
5.2.5 Decreasing electricity costs would make alternative routes more attractive	14
5.3 Trade-off between costs, energy efficiency, and land-use	14
5.4 Product selection	15
<b>6 Conclusions</b>	<b>16</b>
<b>7 A way forward</b>	<b>17</b>
<b>8 References</b>	<b>18</b>

---

# Summary

A sustainable future will require a phase out of fossil-based resources if we want to reduce greenhouse gas emissions. Carbon dioxide (CO<sub>2</sub>) can be an alternative to fossil-based carbon as a renewable carbon source. Future CO<sub>2</sub> sources include direct air capture (DAC), CO<sub>2</sub> point sources from industrial biobased processes, and biogas from residual biomass. DAC is promising but costly and energy-intensive. Biomass residues and point sources from biobased processes offer more practical short-term solutions due to lower costs and higher efficiency.

Captured CO<sub>2</sub> can be used directly in e.g., carbonated drinks or for greenhouse enrichment, but also indirectly through chemical or microbial conversion. Chemical routes often require the conversion of CO<sub>2</sub> to the more reactive, hydrogen enriched, syngas, which is very energy-intensive, while microbial routes leverage organisms to produce complex molecules but face scalability and cost challenges.

A techno-economic analysis on conventional and alternative production routes for methanol and ethanol show that conventional methods remain the cheapest and most energy-efficient, while CO<sub>2</sub>-based routes demand significant energy and infrastructural investments. In optimisation, trade-offs occur between costs, energy use, CO<sub>2</sub>-emissions and land-use.

Using CO<sub>2</sub> from biogas is the most promising alternative route, as identified in the studied cases for methanol and ethanol production, while other CO<sub>2</sub>-based routes require major technological progress and extensive policy support. A systems approach integrating technical, economic, and environmental considerations is essential for developing sustainable, circular solutions. Redesigning production systems, fostering collaboration among stakeholders, and implementing measures like carbon taxes and reduced fossil subsidies are needed to enable a transition toward renewable carbon use.

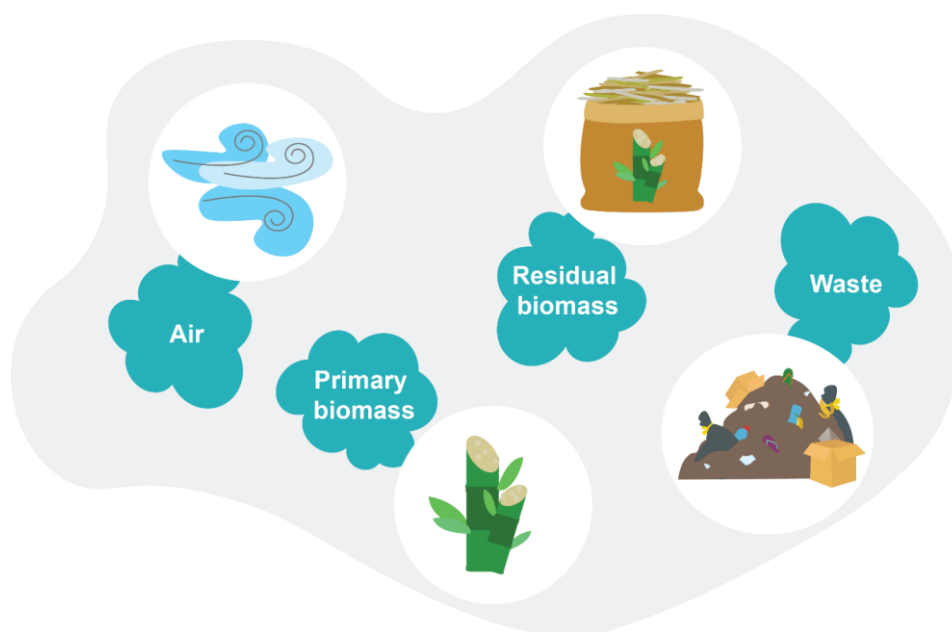
---

# 1 Setting the scene

The intensive use of our fossil raw materials - such as oil, gas, and coal - as energy or carbon sources fuels climate change through the emission of carbon dioxide (CO<sub>2</sub>) during production, use (e.g. fuels), and disposal. CO<sub>2</sub> is the greenhouse gas (GHG) often primarily focused on in terms of climate change as it accounts for 75 % of the GHG emissions worldwide (Ritchie et al., 2020a). In 2022 the annual GHG emission was 53.85 billion tons CO<sub>2</sub> equivalents, of which 65 % was caused by the use of fossil fuels (Ritchie et al., 2020a, 2020b). This is a huge part of the total emissions which urgently requires alternative solutions. Renewable alternatives for energy are found in the use of sun, wind, water, and possibly nuclear energy. These sources produce electricity, but do not provide carbon. However, fossil sources are also used as a carbon source to produce materials and chemicals that we use on a daily basis, from plastics to medicines. To further reduce the use of fossil materials by the industry, we also need to focus on sources of renewable carbon. Which renewable carbon sources to use in future practice is however not yet clear. The solution we are considering here is the use of carbon dioxide (CO<sub>2</sub>) as renewable carbon source. There are many technologies under development to capture CO<sub>2</sub> and convert it into usable materials, these include direct (air capture) or indirect (biomass) capture of CO<sub>2</sub> and conversion via chemical or microbiological routes. These different routes will be discussed in the following parts of this report starting with the capture of CO<sub>2</sub>.

## 2 Future Carbon sources

To produce chemicals and materials from non-fossil CO<sub>2</sub> in the future, an essential first step is to map out the development of available CO<sub>2</sub> sources through to 2050. The technologies we consider must be scalable and applicable well into 2050 and beyond. In this section, we examine the anticipated developments for a selection of renewable CO<sub>2</sub> sources (Figure 1).



**Figure 1** *Proposed renewable sources for carbon such as air, primary biomass (for example sugar cane), residual biomass (for example sugar cane bagasse) and waste streams.*

### 2.1 Conventional: CO<sub>2</sub> from fossil sources

Fossil fuels such as coal, oil, and natural gas have long been the dominant energy sources worldwide. Still, their role is gradually declining as the urgency of climate change increases. Until 2050, dependence on fossil fuels is expected to decrease significantly, partly due to tighter government regulations, growing societal pressure, and increasing competition from renewable energy sources. The reduced or phased-out use of fossil resources will likely have serious regional ramifications for Europe. Investments in developing new fossil energy projects are already under greater scrutiny. In many regions, there is a notable shift towards more sustainable alternatives. Fossil fuels remain a significant source of CO<sub>2</sub>, especially in economies that are slower to switch to green energy. Next to production from energy generation, fossil materials industry such as cement and steel producers emit considerable amounts of CO<sub>2</sub> in their production processes. These could be a "CO<sub>2</sub> point source" from local high production of CO<sub>2</sub>.

### 2.2 Direct Air Capture

Direct Air Capture (DAC) technology is emerging as an approach for reducing CO<sub>2</sub> in the atmosphere. This technology captures CO<sub>2</sub> directly from the air to store it permanently or use it for various purposes, such as producing climate-neutral synthetic fuels. Permanently storing captured CO<sub>2</sub> is known as 'carbon capture and storage' (CCS), while using the captured CO<sub>2</sub> is referred to as 'carbon capture and utilization' (CCU). According to the International Energy Agency (IEA) scenarios for Net Zero Emissions by 2050, DAC technologies are expected to capture more than 75 Mtonnes of CO<sub>2</sub> by 2030 and around 980 Mtonnes by 2050, requiring significant scale-up of current capacity (IEA, 2022).



---

Currently, there are 27 DAC facilities operational in Canada, Europe, and the United States, and there are plans to build many more. These facilities mainly capture CO<sub>2</sub> for storage, not for usage. Governments and industry support DAC with significant financial allocations, such as the United States, which has allocated nearly \$4 billion specifically for DAC development and deployment, and other countries, such as Australia, Canada, Japan, the United Kingdom, and the European Commission, which have also announced major initiatives and funding aimed at boosting DAC technologies (IEA). Also, \$1 billion of private investment has been made, mainly in the private entity ClimeWorks.

Although DAC is currently the most expensive form of carbon capture - due to the low concentration of CO<sub>2</sub> in the atmosphere - costs are expected to fall due to scale-up and innovation. Cost projections for large-scale DAC installations vary considerably. The major challenges are further cost reduction of capital investments, energy efficiency and the right policy support (Young et al., 2023). With optimizations and access to cheap renewable energy, stakeholders claim the costs could fall below \$200 per tonne of CO<sub>2</sub> by 2030. If DAC continues to advance and scale up, it will significantly broaden the range of applications for captured CO<sub>2</sub>. Its flexibility in terms of location, combined with increasing support from governments and the private sector, highlights the potential of this technology.

## 2.3 Industrial CO<sub>2</sub> from biobased processes

Various energy- and food-based processes, such as the production of bioethanol and beer, release large amounts of CO<sub>2</sub>. Europe is not a large producer of bioethanol, but bioethanol production in Brazil is a vast industry producing 80 Mtonnes of CO<sub>2</sub> annually (US Department of Energy, 2022). The annual global beer production is around 190 Mtonnes (Barth-Haas Group, 2023), releasing about 9 Mtonnes of CO<sub>2</sub>. This high local production of CO<sub>2</sub> can be used as a point source. Capturing and using such CO<sub>2</sub> is easier and cheaper than using DAC. Two other biobased processors that produce large amounts of CO<sub>2</sub> are the pulp and paper industry and the sugar production industry. The CO<sub>2</sub> is generated from energy and power production, but also from on-site production of lime that is used during processing.

## 2.4 CO<sub>2</sub> from (residual) biomass

Biomass utilization and waste incineration are often considered renewable CO<sub>2</sub> sources because of their potential to balance the carbon cycle. Biomass, derived from plants and animal waste, can be CO<sub>2</sub>-neutral, provided the CO<sub>2</sub> released during combustion is offset by the CO<sub>2</sub> absorbed by new growth. The development of this sector until 2050 will rely on technological progress, the availability of sustainable residual biomass, and the actual production costs. The costs of CO<sub>2</sub> emissions will largely determine whether it is possible to achieve competitive processes compared to current fossil-based processes.

Europe is strongly committed to biogas production from biomass residues, intending to produce 35 billion m<sup>3</sup> of biomethane from biogas by 2030 (European Biogas Association, 2023). The methane fraction is commercially valuable as "green gas", leading to the separation and on-site venting of the CO<sub>2</sub>. This CO<sub>2</sub> stream has a high concentration, typically over 98% CO<sub>2</sub>. In Europe, this activity is estimated to release around 35 Mtonnes of CO<sub>2</sub> annually by 2030, which could be used for further valorization. In some cases this CO<sub>2</sub> is already sold to greenhouses to stimulate crop growth.

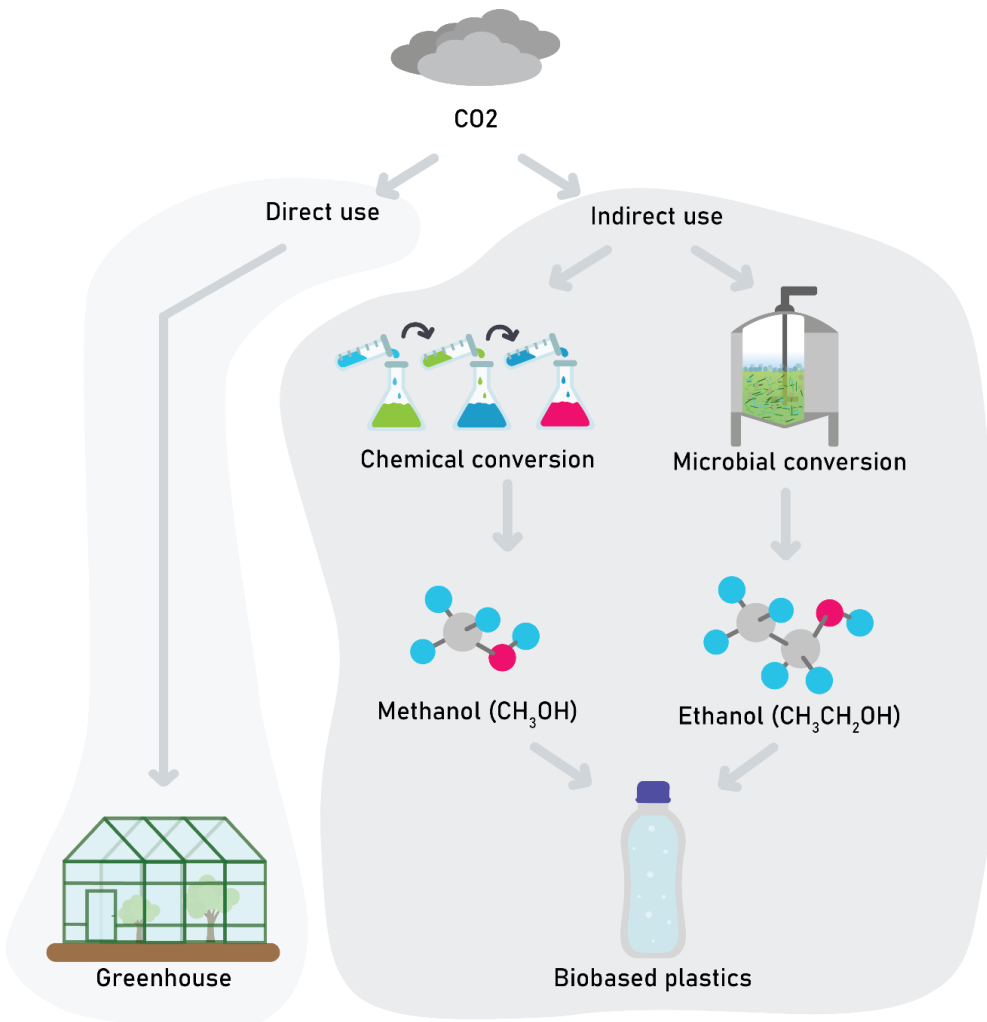
Recycling of biobased material is never 100% efficient due to practical limits in the maximum conversion. Therefore, some biobased material will still have to be incinerated. Such incineration reduces the waste volume going to landfills and simultaneously generates energy. In the Netherlands, 911 ktonne of CO<sub>2</sub> emissions (AVR, 2021) are released annually through waste incineration. Equivalent European figures are not available to the best of our knowledge. Innovations in waste sorting and processing will reduce the amount of incinerated recyclable material.

# 3 Carbon routes to fuels, chemicals, and materials

The previous chapter discussed the different sources for carbon, while this chapter will focus on the different use and conversion pathways of carbon captured from different sources.

## 3.1 Carbon from air captured CO<sub>2</sub>

Despite being despised as a greenhouse gas, CO<sub>2</sub> has a variety of practical applications. Many technologies that convert CO<sub>2</sub> to a higher-value commodity exist. Some technologies are already applied at industrial scale, in other cases, they are only demonstrated at the laboratory scale (Bos et al., 2023; Hepburn et al., 2019; IEA, 2019). Figure 2 shows different conversion pathways of captured CO<sub>2</sub> to produce different materials. The pathways shown are the direct use of CO<sub>2</sub>, and indirect use of CO<sub>2</sub> via chemical or microbial conversion.



**Figure 2** *Schematic overview of three conversion pathways of air captured CO<sub>2</sub>. For each pathway a potential chemical building block is shown as intermediate product. These may be used to produce chemicals and materials.*

---

### 3.1.1 Direct use

After air capture of CO<sub>2</sub> the molecule can be directly used in several products. Examples of direct CO<sub>2</sub> utilization in various industries are: the production of carbonated drinks such as sparkling water, by infusing CO<sub>2</sub> to beverages at high pressure; carbon dioxide-enhanced oil recovery, where CO<sub>2</sub> is pumped into oil reservoirs to pump up more oil; and the production of supercritical CO<sub>2</sub> as a solvent, which is obtained by maintaining CO<sub>2</sub> at specific pressure and temperature conditions.

### 3.1.2 Direct use for plant growth

Also greenhouses can be direct users of CO<sub>2</sub> through OCAP (Organic CO<sub>2</sub> for Assimilation by Plants), which yearly delivers 300 ktonne of pure CO<sub>2</sub> to greenhouse horticulture in the Westland, Lansingerland, Delfgauw and Wilgenlei. This saves around 115 million m<sup>3</sup> of natural gas per year, which would otherwise been used to generate heat and CO<sub>2</sub> for the greenhouses, reducing CO<sub>2</sub> emissions by 205 ktonne per year. They get this CO<sub>2</sub> from Shell Pernis that produces it as a side-product from hydrogen production. The CO<sub>2</sub> is transported through a system of pipelines that were previously in use for oil transportation.

### 3.1.3 Indirect use

Alternatively, besides the direct use of CO<sub>2</sub> in industrial processes or greenhouses, CO<sub>2</sub> can also be indirectly used. CO<sub>2</sub> functions then as a carbon source to produce new materials and chemicals, which involves the conversion by chemical or microbiological routes.

### 3.1.4 Indirect use through chemical conversion

There are several pathways possible to convert CO<sub>2</sub> into products of greater value, the first step however often involves the breakdown of CO<sub>2</sub> into carbon monoxide (CO), an even smaller molecule. This reaction is not available yet on an industrial scale and very energy intensive. It is done because CO is more reactive and therefore easier to convert into other molecules such as methanol, which can be used as an alternative fuel for ships. These small, single-carbon molecules are called C1 molecules because they contain only one carbon atom. From these C1 molecules, larger molecular structures are built, involving a large number of chemical reaction steps.

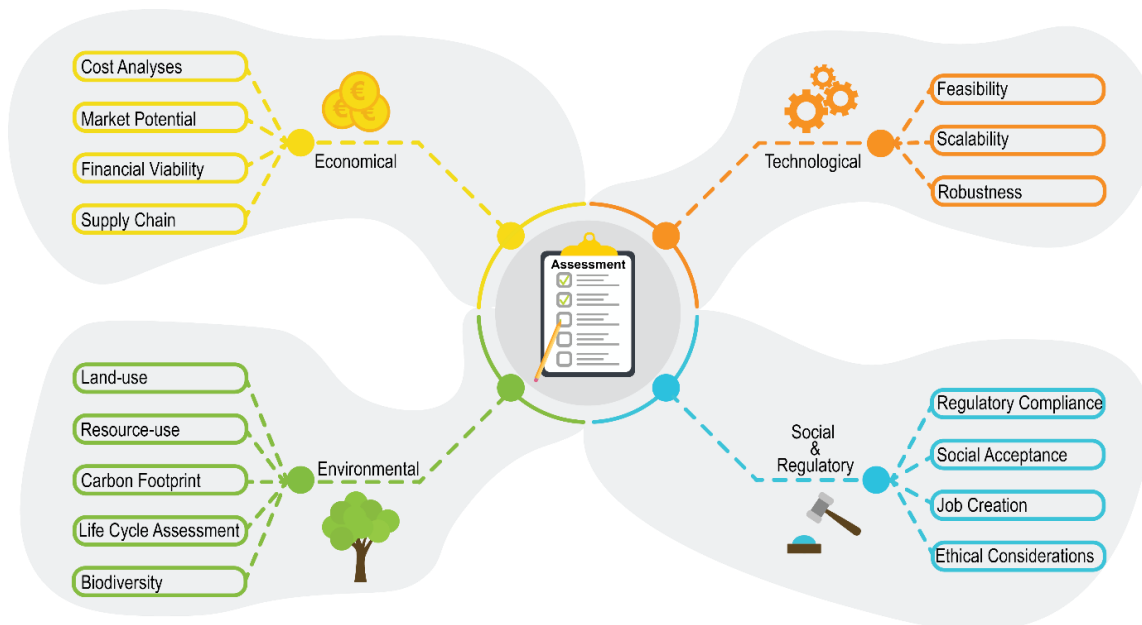
### 3.1.5 Indirect use through microbial conversion

Many organisms can utilize CO<sub>2</sub> through a large variety of biosynthetic pathways. Such pathways always require a significant energy input to catalyse the reactions from CO<sub>2</sub> to biochemicals. This energy can be supplied by an external electron donor (H<sub>2</sub> or CO being the most commonly tested) or (sun)light. Photosynthetic organisms such as algae and cyanobacteria use energy from the sun(light) through photosynthesis and utilize CO<sub>2</sub> to produce sugars and other useful compounds. Those are either stored in the microbial biomass or excreted. Algae for providing polyunsaturated fatty acids in fish feed can be grown photosynthetically, which has been done already for a long time. Other attempt to directly produce chemicals such as ethylene, isoprene and even polylactic acid (Guerrero et al., 2012, Zhou, 2010, Angermayr et al., 2014). While these processes theoretically offer high yields per acre, achieving such efficiency in practice remains challenging due to biological and technological constraints (Anemaet et al., 2010, Liu et al., 2021). Furthermore, the high costs associated with algae production facilities significantly limit the scalability and economic viability of these systems.

Non-photosynthetic bacteria, like *Clostridium autoethanogenum* can metabolize H<sub>2</sub> and CO<sub>2</sub> to produce ethanol and a variety of other products (Wan et al., 2023, Karim et al., 2020, da Souza et al., 2019). Such products are in general excreted by the bacterium allowing straightforward isolation from the fermentation broth, although some such as PHB are produced intracellularly (da Souza et al., 2019). The microbial route presents unique advantages compared to chemical processes, notably the ability to efficiently convert CO<sub>2</sub> directly into complex molecules with minimal side-product formation, due to the high specificity of enzymes (Fackler et al., 2021). Solubility of the hydrogen is a challenge leading to the requirement of recycling of hydrogen from the off-gas. Also, the conversion of gasses to products occurs at rates that are very low compared to current industrial fermentation processes, necessitating large facilities and therefore high investments that in general make the economics on the overall processes challenging.

## 4 How to select routes for carbon use

There are many options for using CO<sub>2</sub> in novel processes for biobased products and their contribution to - or promise for - the future material transition away from fossil resources. Also, there are many ways to evaluate and compare all options. Many parameters must be considered, including the technology itself, its economy, its environmental impact, and any social, ethical, or regulatory issues associated with using and disposing of a new biobased product (Figure 3).



**Figure 3** Overview of the key elements of an assessment framework, including examples of economic, environmental, technical and social & regulatory considerations.

As a first step in our assessment of CO<sub>2</sub> use for renewable chemicals, we performed an analysis that focussed on the technical feasibility and energy use, as well as on the economic aspects resulting in a comparative assessment of costs and investments. Land use was included as an additional important sustainability parameter. Methanol and ethanol were selected as examples of molecules that are used as a starting point for other products, also called chemical building blocks.

---

## 5 Methanol and ethanol production from CO<sub>2</sub>

Methanol and ethanol are important chemicals used in a wide range of applications, including fuels and the manufacturing of medicines, perfumes, and cosmetics. Their supply is essential for many consumer goods, which is why we decided to focus on these bulk chemicals in this project. We compared new, renewable pathways for producing ethanol and methanol from captured carbon with conventional methods for producing these bulk chemicals (Lanting et al. 2025).

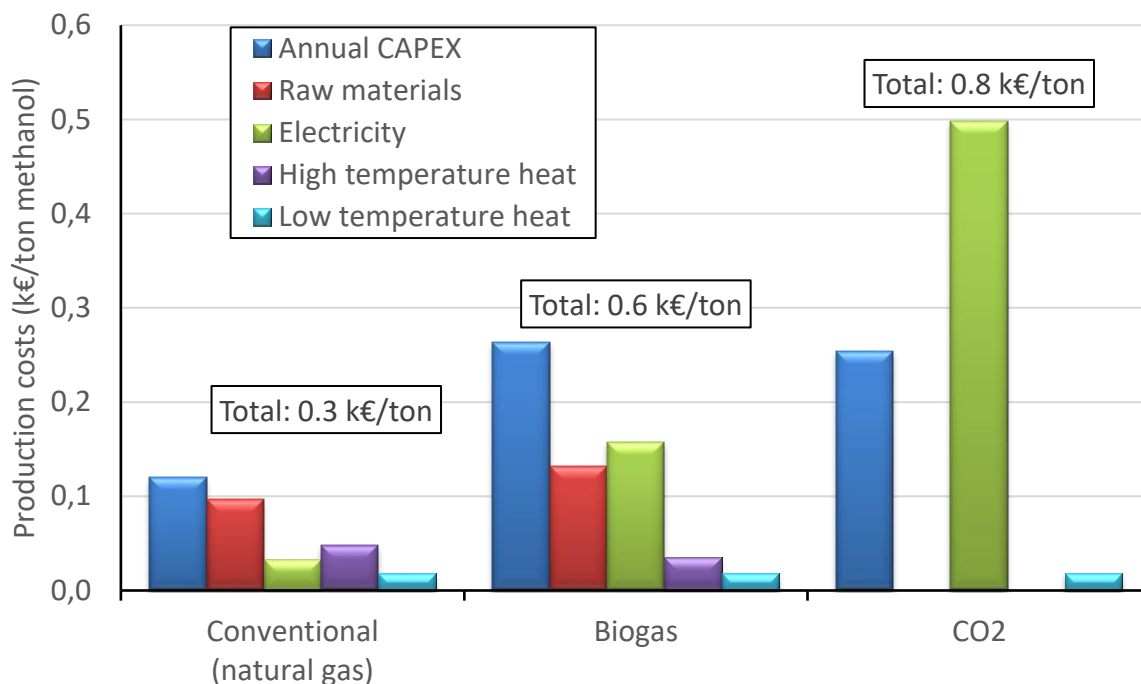
### 5.1 Scenario's for methanol and ethanol production from CO<sub>2</sub>

The conventional method for producing methanol involves using natural gas. Natural gas is a fossil fuel and therefore not a sustainable source for methanol due to its finite nature and release of GHGs during processing. Renewable routes that were compared to this conventional production process were the use of biogas from anaerobic digestion of biomass and air captured CO<sub>2</sub>.

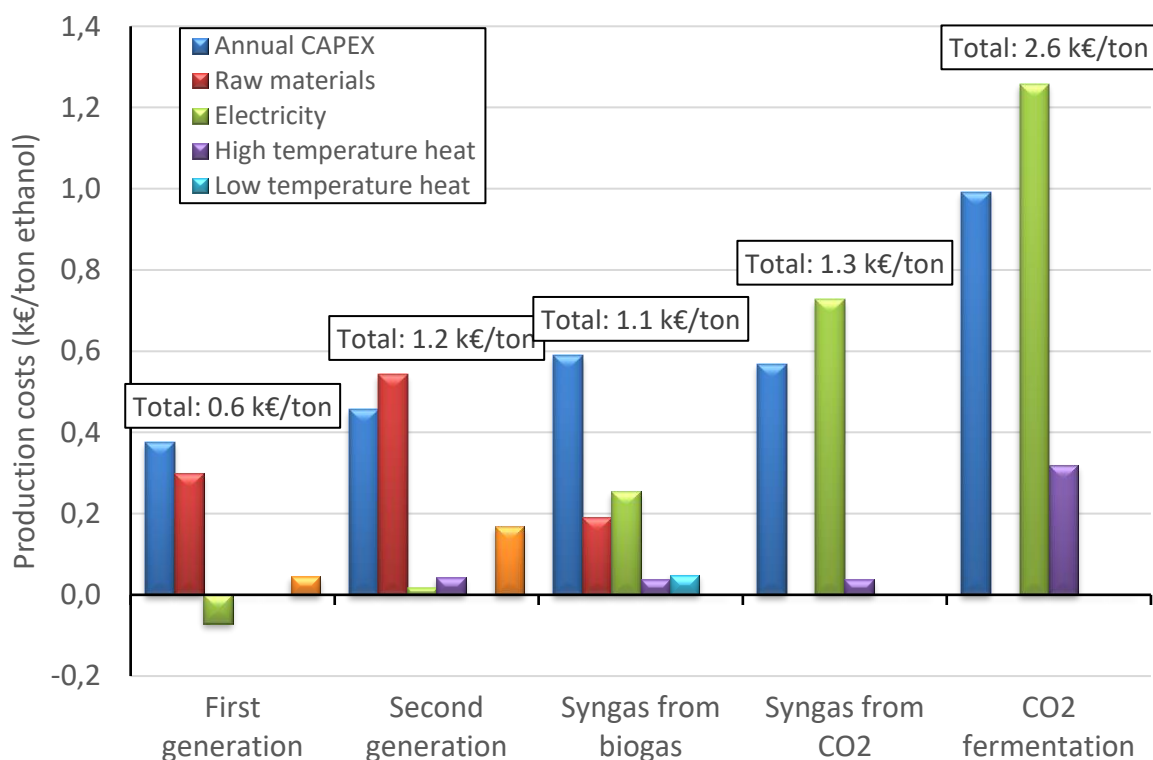
Ethanol on the other hand is conventionally already produced by fermenting biomass with microorganisms. This route does not require the use of fossil fuels, however, ethanol production competes with other sectors for biomass such as the feed and food sector. This is because biomass used for ethanol production, such as sugar cane, can also be used for animal feed and the land used to grow sugar cane could be used to produce other food and feed crops. Alternative routes explored in this project include the fermentation of biomass residues such as corn stover for second generation fermentation, biogas from anaerobic digestion, and again utilizing air captured CO<sub>2</sub>. The routes were compared to each other based on the techno-economic feasibility, energy-use, and for ethanol production also land-use.

The techno-economic results of the three methanol production routes are shown in Figure 4 (adapted from Lanting et al., 2025). CAPEX, raw materials, and High Temperature (HT) heat show the highest costs for the conventional route. For both alternative routes, CAPEX and electricity (assumed 50 €/MWh), especially for the route based on CO<sub>2</sub>, show high contributions to the overall costs.

The ethanol results are shown in Figure 5 (adapted from Lanting et al., 2025). Again, the ethanol conventional production route, first generation fermentation, show lowest costs. In this route, main costs are associated with CAPEX and raw material costs, i.e. sugar cane. The highest costs in the second generation route are also CAPEX and raw material costs. For the alternative routes, electricity costs are increased and show high contribution to the overall costs.



**Figure 4** *Techno-Economic Assessment results of the evaluated methanol production routes (adapted from Lanting et al., 2025). The production costs (€/ton ethanol) are the sum of annualized CAPEX, raw material, electricity, heat (Low Temperature), heat (High Temperature) and consumables.*



**Figure 5** *Techno-Economic Assessment results of the evaluated ethanol production routes (adapted from Lanting et al., 2025). The production costs (€/ton ethanol) are the sum of annualized CAPEX, raw material, electricity, heat (Low Temperature), heat (High Temperature) and consumables.*

## 5.2 2050 techno-economic outlook for ethanol and methanol production

### 5.2.1 Conventional methanol and ethanol production routes will be the cheapest methods to produce methanol and ethanol

The techno-economic analysis of our current situation clearly shows that the conventional routes are the cheapest. Also when technology readiness of the new routes increases, the conventional methanol and ethanol production methods will still be the cheapest routes to produce these bulk chemicals when nothing else changes. We therefore also studied the influence of other factors, such as CO<sub>2</sub> source, energy pricing, CO<sub>2</sub> tax, and land use allocation.

### 5.2.2 Alternative methanol and ethanol routes require more energy than conventional routes

The conventional methanol production route consumes energy rich fossil methane as feedstock. Conventional ethanol production route uses sugar cane as feedstock. Besides, biomass residues are co-generated to produce electricity, resulting as 'negative' electricity consumption in this study, since the produced electricity is regarded as a by-product. These two conventional routes start thus with energy rich feedstocks. This means that the electricity consumption of these routes is less than the alternative routes, which can also be seen from Figure 4 and 5.

Regarding the alternative routes, the biobased routes are more energy efficient than the CO<sub>2</sub> based routes. This is obvious, as energy is already stored in the biomass/feedstock. Conversion of energy poor CO<sub>2</sub> to energy rich syngas, which is used as intermediate for both methanol and ethanol production, requires much energy (Figure 4 and 5).

It can be concluded that transition towards a CO<sub>2</sub> based society requires much extra energy, whether in the form of hydrogen or not. This leads to the necessary investment in infrastructure.

### 5.2.3 Incentives, such as CO<sub>2</sub> tax, play an important role in enabling alternative fossil free production routes

In the study of Lanting et al. (2025), different scenarios were outlined. Including CO<sub>2</sub> taxes in the TEA, which would increase the conventional methanol production costs, was one of these scenarios. The alternative route based on biogas would become cost-competitive with the conventional route when CO<sub>2</sub> taxes will reach 145 €/ton CO<sub>2</sub>. The route based on CO<sub>2</sub> will become cost-competitive when CO<sub>2</sub> taxes increases to 182 €/ton CO<sub>2</sub>. The current carbon tax in Europe is 74 €/ton CO<sub>2</sub> (Trading Economics, 2025). When we adjust the TEA by only considering CO<sub>2</sub> taxes, these routes could become realistic before 2050, as CO<sub>2</sub> taxes are estimated to increase to roughly 750 €/ton CO<sub>2</sub> in 2050 (Matthey and Bünger, 2019, Quinet, 2019).

**Table 1** *Required CO<sub>2</sub> taxes to reach cost parity for renewable methanol production to the fossil route at different energy prices.*

Methanol route via	Required CO <sub>2</sub> tax at 25 €/MWh (€/ton CO <sub>2</sub> )	Required CO <sub>2</sub> tax at 50 €/MWh (€/ton CO <sub>2</sub> )	Required CO <sub>2</sub> tax at 100 €/MWh (€/ton CO <sub>2</sub> )
Biogas	117	145	200
CO <sub>2</sub>	101	182	344

---

#### 5.2.4 CO<sub>2</sub> from point sources instead of DAC is cheaper, a good starting point for cost reduction

Results from the TEA show that CO<sub>2</sub> costs using DAC will be about 125 €/ton CO<sub>2</sub> in 2050. When CO<sub>2</sub> would be obtained from point sources instead, the CO<sub>2</sub> costs could drop to about 60 €/ton CO<sub>2</sub> (Hong, 2022). This possible reduction in CO<sub>2</sub> feedstock costs would lead to a decrease in the overall costs for the CO<sub>2</sub> based methanol route of about 90€/ton methanol. For the CO<sub>2</sub> based ethanol production route, the overall production costs would decrease with 230 €/ton ethanol.

#### 5.2.5 Decreasing electricity costs would make alternative routes more attractive

Another scenario that was outlined, was by adjusting electricity costs. Some people hypothesize that the amount of renewable electricity production will increase in short-term, resulting in a decrease in electricity costs. However, this will require serious improvement of and investments in the current energy infrastructure. Lower energy costs would have a significant impact on overall costs of the alternative production routes, since these routes require large amounts of electricity. These energy requirements are mainly linked to obtaining CO<sub>2</sub> from air and converting energy poor CO<sub>2</sub> into energy rich syngas.

Regarding the alternative methanol production routes, only the methanol route based on CO<sub>2</sub> will become cost-competitive when lowering electricity prices. The required electricity price is then 6 €/MWh. The only alternative ethanol production route that could become competitive to the conventional route under this scenario is the CO<sub>2</sub> based syngas fermentation route. This requires electricity prices to fall to 5 €/MWh. However, when combined with CO<sub>2</sub> taxes, cost parity already be achieved earlier as illustrated in table 1.

### 5.3 Trade-off between costs, energy efficiency, and land-use

Required land-use is another important factor besides costs and energy efficiency for ethanol production routes. Although the conventional ethanol production route is cheap and energy efficient, this route requires (arable) land to grow sugar cane, i.e. 0.2 ha·y/ton ethanol. The same amount of land is needed for second generation route when using corn stover. The alternative routes based on gas fermentation require 0.02-0.06 ha·y/ton ethanol when energy is generated via solar panels and/or for producing biogas from a maize silage/manure slurry. This leads thus to a trade-off between costs and land-use for ethanol production. When we move toward a fossil free society, carbon should be obtained from biomass, CO<sub>2</sub>, or via recycling, increasing the pressure on land-use. As land is already used for the current food and feed production, this will only be feasible when there is e.g. a reduction on land-use for feed enabled by a reduction in the use of animal products (the protein transition). Predicted land-use for materials is maximal 5% of the available land, but biofuel production will require much more land (Bos et al., in preparation). This will make the alternative routes via CO<sub>2</sub> even more important in the future.

Across the different alternative pathways investigated, the most promising were the once using biomass residues, regardless of whether methanol or ethanol was produced as the end product. This is because production cost, energy efficiency, and land use were best using residual biomass as a starting source compare to air captured CO<sub>2</sub>. One reason for that is that CO<sub>2</sub> air capture and production of syngas from CO<sub>2</sub> are currently still at an early technical developmental stage with a low TRL and require not just great investments, but also great amounts of energy. We calculated using the most positive scenario approximately 59 GJ of electricity is required for 1 ton of ethanol produced. Practically this would mean that to replace current global ethanol production (29.5 billion gallons) with CO<sub>2</sub> produced ethanol about 1450 TWh of electricity would be required (5% of global electricity production). We therefore foresee the greatest short term potential in using alternative pathways based on biogas from residual biomass (e.g. manure and a co-substrate) as these have both limited land use and electricity demand and only minimally affects overall costs.



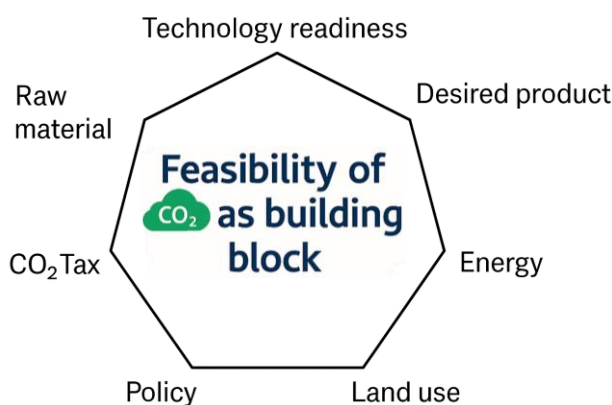
---

## 5.4 Product selection

In this case study, ethanol and methanol were selected to illustrate the potential use of CO<sub>2</sub> for renewable products. However, the selection of the required product will also influence feasibility of a CO<sub>2</sub>-based route. An important aspect to consider is the oxidation state of the product. CO<sub>2</sub> contains two oxygen molecules for every carbon molecule. Most products, just like ethanol and methanol, contain relatively less oxygen. the amount of oxygen that needs to be removed to get to the product is important as this requires a lot of energy. (van Vliet et al., 2025) Making chemicals that contain relatively high amounts of oxygen, or even equal amount to CO<sub>2</sub>, such as oxalic acid will lead to processes that require much less energy.

## 6 Conclusions

A techno-economic analysis on conventional and alternative production routes for methanol and ethanol was performed to illustrate the feasibility of using CO<sub>2</sub> as a renewable carbon source. The production processes show trade-offs between costs, energy efficiency, CO<sub>2</sub>-emissions and land use. For new routes to be attractive, a combination of CO<sub>2</sub> taxes, CO<sub>2</sub> from point sources, increased fossil raw material prices, and a decrease in electricity costs will be needed. In many cases, conventional fossil routes still show the lowest production costs when comparing to biobased alternatives.



**Figure 6** *Parameters that interact on the feasibility of using CO<sub>2</sub> as a chemical building block.*

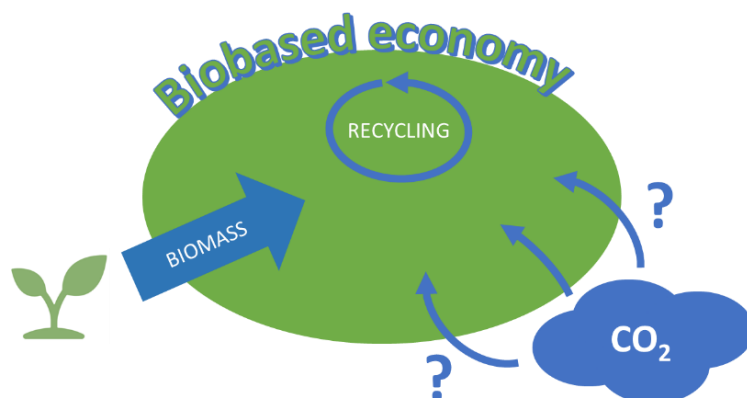
The high costs calculated for the alternative methanol and ethanol routes are associated with the conversion of CO<sub>2</sub> to the more reactive, hydrogen enriched, syngas. CO<sub>2</sub> based routes require large amounts of renewable energy. However, there are ways to reduce production costs by lowering energy prices, obtaining CO<sub>2</sub> from point sources instead of DAC, and by implementing CO<sub>2</sub> taxes. The implementation of a carbon tax, starting from around 100 €/ton CO<sub>2</sub> for methanol production routes can be a way to reach cost parity with fossil-based routes. Difference in raw material pricing can also increase the feasibility of the alternative routes. Methanol production from biogas will become feasible at a combined natural gas price of 500 €/ton and a CO<sub>2</sub> tax of 74 €/ton.

Using CO<sub>2</sub> from biogas is the most promising alternative route as identified in the studied cases for methanol and ethanol production. The 1<sup>st</sup> and 2<sup>nd</sup> generation processes for bioethanol are limited by the amount of available land to grow the biomass. The level of technology development is an important factor for the alternative routes to become implemented. In the end, there are multiple variables that influence the feasibility of the alternative routes. A combination of market price development and governmental measures can allow for cost parity.

Finally, it is important to consider the type of molecule that you want to produce while using CO<sub>2</sub>. The relative amount of oxygen that needs to be removed to get to the product is important as this requires a lot of energy. Making chemicals that contain relatively high amounts of oxygen will lead to processes that require much less energy.

## 7 A way forward

The intensive use of our fossil raw materials as energy or carbon source fuels climate change through the emission of carbon dioxide (CO<sub>2</sub>) during production processes leading to global climate change. Considering CO<sub>2</sub> as renewable carbon source should be part of a sustainable, biobased future, in addition the current focus on biomass use and recycling (figure 7). Biomass is an important attractive starting material as it is a dense C-source. However, the availability will never be enough to fulfil future biobased demands. In addition carbon re-use is needed and all systems need to be circular. Additional CO<sub>2</sub> capture may not only be needed as a carbon source that neutralises CO<sub>2</sub> emissions, but also to restore the environment (mitigate CO<sub>2</sub> increase in the atmosphere).



**Figure 7** *Biomass is an important starting material for the biobased economy. It needs to be complemented by carbon re-use. CO<sub>2</sub> capture may be needed as additional carbon source to meet future demands.*

As there is not just one magic bullet, we need to integrate information on availability, technical possibilities, specific demand and economic factors. Potential feedstock use (primary, secondary, tertiary) and their synergies and trade-offs need to be taken into account. A high-over systems-approach should be combined with detailed calculations on techno-economics, sustainability and circularity to get to new insights in chain development and better choices in routes for CO<sub>2</sub> capture and use (CCU). These new systems should not just be improved versions of the old ones, but we need to redesign systems to develop future materials and chemicals that are sustainable and recyclable. End-of-life solutions should become an integral part of production and use.

The new developments can have many unintended side-effects. Increased competition for cheap biomass and waste will drive the price for biomass. The actual allocation of land-use will have to be reconsidered in LCA models to compensate for the better use of side-streams. The new systems are also likely to operate on a smaller scale than the current fossil based system. Sources from the agri-food domain and their processing are diverse and often require a local approach, considering where to install new systems. Point sources within agri-food can e.g. be a starting point for the design of new chemical or microbial routes and chains for CCU starting from digesters or crop processing.

Next to the technological improvements that may be provided by science and industry, other stakeholders are at least equally important. The motivation from consumers and the commitment of NGOs are needed to direct policy makers and industry towards more sustainable alternatives. Current subsidies on fossil fuels are hampering a level playing field. Education and communication are important tools to inform all stakeholders and get everyone on the same (sustainability) page.

---

## 8 References

- Angermayr SA, Van der Woude AD, Correddu D, Vreugdenhil A, Verrone V and Hellingwerf KJ. (2014) Exploring metabolic engineering design principles for the photosynthetic production of lactic acid by *Synechocystis* sp. PCC6803. *Biotechnology for Biofuels* 7, 99.
- Anemaet IG, Bekker M and Hellingwerf KJ. (2010) Algal photosynthesis as the primary driver for a sustainable development in energy, feed, and food production. *Mar Biotechnol* (NY). 12(6):619-29.
- AVR, 2021. Afval energie barometer. Retrieved July 22, 2024 from <https://www.afvalvergroeners.nl/upload/docs/afval-energie-barometer.pdf>
- Barth-Haas Group, 2023. Beer production worldwide from 1998 to 2022. Retrieved July 22, 2024 from <https://www.statista.com/statistics/270275/worldwide-beer-production/>
- Bos, H.L., van Es, D.S., Harmsen, P., and Sena, N., 2023. The renewable future of materials : How to produce our everyday products once we phased out fossil oil and gas. Wageningen, the Netherlands. Retrieved July 22, 2024 from <https://edepot.wur.nl/583970>
- Bos, HL, Lanting, MP, Braamhaar, D, Vellinga, TV (in preparation) Biomass for food, feed and non-food in Europe, how to balance future demand and supply?
- European Biogas Association, 2023. European Biogas Association Statistical Report 2023. Brussels, Belgium. Retrieved July 22, 2024 from <https://www.europeanbiogas.eu/wp-content/uploads/2023/12/EBA-Statistical-Report-2023-Excerpt.pdf>
- Fackler N, Heffernan J, Juminaga A, Doser D, Nagaraju S, Gonzalez-Garcia RA, Simpson SD, Marcellin E and Köpke M. (2021) Transcriptional control of *Clostridium autoethanogenum* using CRISPRi. *Synth Biol* (Oxf). 6(1):ysab008
- Guerrero F, Carbonell V, Cossu M, Correddu D and Jones PR. (2012) Ethylene synthesis and regulated expression of recombinant protein in *Synechocystis* sp. PCC 6803. *PLoS ONE* 7, e50470
- Hepburn, C., Adlen, E., Beddington, J., Carter, E.A., Fuss, S., Mac Dowell, N., Minx, J.C., Smith, P., and Williams, C.K., 2019. The technological and economic prospects for CO<sub>2</sub> utilization and removal. *Nature* 575(7781): 87-97.
- Hong, W. Y. (2022). A techno-economic review on carbon capture, utilisation and storage systems for achieving a net-zero CO<sub>2</sub> emissions future. *Carbon Capture Science & Technology*, 3, 100044.
- IEA, 2019. Putting CO<sub>2</sub> to use. Creating value from emissions. Paris, France. Retrieved July 22, 2024 from <https://www.iea.org/reports/putting-CO2-to-use>
- IEA, 2022. Direct Air Capture A key technology for net zero. Paris, France. Retrieved July 22, 2024 from [https://iea.blob.core.windows.net/assets/78633715-15c0-44e1-81df-41123c556d57/DirectAirCapture\\_Akeytechnologyfornetzero.pdf](https://iea.blob.core.windows.net/assets/78633715-15c0-44e1-81df-41123c556d57/DirectAirCapture_Akeytechnologyfornetzero.pdf)
- Karim AS, Dudley QM, Juminaga A, Yuan Y, Crowe SA, Heggestad JT, Garg S, Abdalla T, Grubbe WS, Rasor BJ, Coar DN, Torculas M, Krein M, Liew FE, Quattlebaum A, Jensen RO, Stuart JA, Simpson SD, Köpke M and Jewett MC. (2020) In vitro prototyping and rapid optimization of biosynthetic enzymes for cell design. *Nat Chem Biol*. 16(8):912-919.
- Kwawu, C.R. and Aniagyei, A., 2021. A review on the computational studies of the reaction mechanisms of CO<sub>2</sub> conversion on pure and bimetals of late 3d metals. *Journal of Molecular Modeling* 27(7): 200.
- Lanting, MP, Vogt JA, Meesters KPH, van Es DS, Bekker M and Bruins, M.E. (2025). respective techno-economic assessment of carbon capture & utilization and biobased processes for methanol and ethanol production. *Sustainable Energy Fuels*, 2025,9, 4660-4673
- Liu Y, Cruz-Morales P, Zargar A, Belcher MS, Pang B, Englund E, Dan Q, Yin K and Keasling JD. (2021) Biofuels for a sustainable future. *Cell*. 2021 Mar 18;184(6):1636-1647.
- Matthey, A and Bünger, B (2019) Methodological Convention 3.0 for the Assessment of Environmental Costs. German Environment Agency (UBA)
- Ministerie van Financiën, 2024. Factsheet verhoging tarief CO<sub>2</sub>-heffing industrie. Retrieved July 24, 2024 from <https://open.overheid.nl/documenten/6e58a8d9-d3b2-491e-b627-b3f186eb2b64/file>
- Quinet, A. (2019) What Value Do We Attach to Climate Action? *Economie et Statistique* p.165-79.
- Ritchie, H, Rosado, P, and Roser, M, 2020a. Greenhouse gas emissions. Retrieved July 22, 2024 from <https://ourworldindata.org/greenhouse-gas-emissions>
- Ritchie, H, Rosado, P, and Roser, M, 2020b. CO<sub>2</sub> emissions by fuel. Retrieved July 22, 2024 from <https://ourworldindata.org/emissions-by-fuel>

- 
- de Souza Pinto Lemgruber, R, Valgepea, K, Tappel, R, Behrendorff, JB, Palfreyman, RW, Plan, M, Hodson, MP, Simpson, SD, Nielsen, LK, Köpke, M and Marcellin, E (2019) Systems-Level Engineering and Characterisation of *Clostridium Autoethanogenum* through Heterologous Production of Poly-3-Hydroxybutyrate (PHB). *Metabolic Engineering* 53: 14–23
- Statista (2024) Carbon pricing worldwide. Retrieved July 24, 2024 from <https://www.statista.com/study/117023/global-carbon-pricing/>
- Trading Economics (2025) EU Carbon Permits.
- US Department of Energy, 2022. Alternative Fuels Data Center. Washington, USA. Retrieved June 20, 2024 from <https://afdc.energy.gov/data>
- van Vliet, DM, Verhoeven, M, Maaskant, E and Post W (2025) Routes towards CO<sub>2</sub>-based materials for building, textiles and packaging. Wageningen Food & Biobased Research. Report 2637.
- Wan, S, Lai, M, Gao, X, Zhou, M, Yang, S, Li, Q, Li, F, Xia, L and Tan, Y (2023) Recent progress in engineering *Clostridium autoethanogenum* to synthesize the biochemicals and biocommodities. *Synth Syst Biotechnol.* 9(1):19-25
- Young J, McQueen, N, Charalambous, C, Foteinis, S, Hawrot, O, Ojeda, M, Pilorgé, H, Andresen, J, Psarras, P, Renforth, P, Garcia, S and van der Spek, M (2023) The cost of direct air capture and storage can be reduced via strategic deployment but is unlikely to fall below stated cost targets. *One Earth* 6(7):899-917
- Zhou J, Li Y. (2010) Engineering cyanobacteria for fuels and chemicals production. *Protein & Cell* 1, 207–210

To explore  
the potential  
of nature to  
improve the  
quality of life



---

Wageningen Food & Biobased Research  
Bornse Weilanden 9  
6708 WG Wageningen  
The Netherlands  
E [info.wfbr@wur.nl](mailto:info.wfbr@wur.nl)  
[wur.eu/wfbr](http://wur.eu/wfbr)

Report 2767



---

The mission of Wageningen University & Research is “To explore the potential of nature to improve the quality of life”. Under the banner Wageningen University & Research, Wageningen University and the specialised research institutes of the Wageningen Research Foundation have joined forces in contributing to finding solutions to important questions in the domain of healthy food and living environment. With its roughly 30 branches, 7,700 employees (7,000 fte), 2,500 PhD and EngD candidates, 13,100 students and over 150,000 participants to WUR’s Life Long Learning, Wageningen University & Research is one of the leading organisations in its domain. The unique Wageningen approach lies in its integrated approach to issues and the collaboration between different disciplines.

---