

Synthesis and SAR of phenylazoles, active against Staphylococcus aureus Newman

Chemistry of Heterocyclic Compounds

Solomin, Vitalii V.; Ciruelos, Blanca Fernandez; Velikova, Nadya; Wells, Jerry; Albanese, Marco et al

https://doi.org/10.1007/s10593-023-03151-9

This publication is made publicly available in the institutional repository of Wageningen University and Research, under the terms of article 25fa of the Dutch Copyright Act, also known as the Amendment Taverne.

Article 25fa states that the author of a short scientific work funded either wholly or partially by Dutch public funds is entitled to make that work publicly available for no consideration following a reasonable period of time after the work was first published, provided that clear reference is made to the source of the first publication of the work.

This publication is distributed using the principles as determined in the Association of Universities in the Netherlands (VSNU) 'Article 25fa implementation' project. According to these principles research outputs of researchers employed by Dutch Universities that comply with the legal requirements of Article 25fa of the Dutch Copyright Act are distributed online and free of cost or other barriers in institutional repositories. Research outputs are distributed six months after their first online publication in the original published version and with proper attribution to the source of the original publication.

You are permitted to download and use the publication for personal purposes. All rights remain with the author(s) and / or copyright owner(s) of this work. Any use of the publication or parts of it other than authorised under article 25fa of the Dutch Copyright act is prohibited. Wageningen University & Research and the author(s) of this publication shall not be held responsible or liable for any damages resulting from your (re)use of this publication.

For questions regarding the public availability of this publication please contact $\frac{openaccess.library@wur.nl}{openaccess.library@wur.nl}$

Synthesis and SAR of phenylazoles, active against *Staphylococcus aureus* Newman

Vitalii V. Solomin^{1,2}*, Blanca Fernandez Ciruelos³, Nadya Velikova³, Jerry Wells³, Marco Albanese⁴, Anmol Adhav⁵, Aigars Jirgensons^{1,2}

- ¹ Latvian Institute of Organic Synthesis, 21 Aizkraukles St., Riga LV-1006, Latvia; e-mail: vitalijs.solomins@osi.lv
- ² Riga Technical University, Faculty of Materials Science and Applied Chemistry, 3/7 Paula Valdena St., Riga LV-1048, Latvia
- ³ Wageningen University & Research, Department of Animal Sciences, Droevendaalsesteeg 4, 6708 PB Wageningen, The Netherlands
- ⁴ Oxford Drug Design, Oxford Centre for Innovation, New Road, Oxford OX1 1BY, United Kingdom
- ⁵ Instituto de Biomedicina de Valencia CSIC, 11 Jaime Roig, Valencia 46010, Spain

Published in Khimiya Geterotsiklicheskikh Soedinenii, 2022, 58(12), 737–748

Submitted July 8, 2022 Accepted after revision October 27, 2022

Series of new potent inhibitors of growth of *Staphylococcus aureus* Newman, based on 3,4-diphenylpyrazole and 4,5-diphenylisoxazole derivatives were discovered. Structures of interest were selectively modified to check their structure–activity relationship. Studies revealed the most essential groups in the molecule for the antimicrobial activity retention. Active compounds with good MIC range should contain both nonpolar aromatic residues and hydrogen bond donating groups. The best MIC results in selected cases were lower than 1 µg/ml.

Keywords: diphenylazole, isoflavone, isoxazole, pyrazole, antimicrobial activity, Staphylococcus aureus Newman.

The antibiotic resistance is one of the greatest health challenges requiring efficient solutions to prevent the increased number of lethal outcomes caused by bacterial infections. The control of antimicrobial infections in hospitals is already complicated due to so-called ESKAPE pathogens (*Enterococcus faecium, Staphylococcus aureus, Klebsiella pneumoniae, Acinetobacter baumannii, Pseudomonas aeruginosa*, and *Enterobacter* spp.) which are resistant to virtually all marketed antibiotics. The new antimicrobial drugs acting by yet unexploited mechanism are therefore urgently needed.

Recently Vo et al. reported 3,4-diarylpyrazole-based antibacterial compound series which was repurposed from compounds with anticancer activity acting as a heat shock protein 90 (HSP90) inhibitors (Fig. 1).³ The antibacterial activity was linked to the inhibition of bacterial histidine

kinases by binding to ATP-binding domain which share high similarity to the ATPase domain of eukaryotic HSP90. The representative compound 1 displayed micromolar inhibition of histidine kinases *C. crescentus* CckA and *Salmonella* PhoQ and medium activity against certain Gram negative and Gram positive bacterial strains. Structurally similar hit 2 with good potency against *S. aureus* was revealed in Wells lab by screening of compound libraries in antibacterial susceptibility tests. In this paper, we describe systematic investigation of SAR of diarylpyrazole-based compounds as well as scaffold hopping studies for the replacement of the pyrazole heterocycle with isoxazole.

The key intermediates for the synthesis of 3,4-diarylpyrazoles 8a-l were isoflavones 7a-k (Scheme 1).⁴ These were synthesized from readily available resorcinol (3) and

MIC (*B. subtilis*) 50–74 μg/ml MIC (*E. coli*) 12–25 μg/ml

MIC (S. aureus) 6.25 μg/ml

Figure 1. 3,4-Diarylpyrazoles with antibacterial activity.

phenylacetic acid derivatives **4a**–**c** in two steps. ^{4b,5} The first step included Friedel–Crafts acylation of resorcinol (**3**), catalyzed by BF₃·Et₂O. The resulting acylresorcinols **5a**–**c** underwent condensation with acid anhydride followed by the cyclization to give isoflavones **6a**–**d**. *O*-Alkylation provided isoflavone derivatives **7a**–**k** which were condensed with hydrazine to provide the novel target compounds **8a**–**l**.

Synthesis of previously reported monoaryl pyrazoles 10a,c and novel pyrazoles 10b,d was achieved by the condensation of diketones 9a-c with hydrazine (Scheme 2). One of the monoarylpyrazoles, compound 10a, was further

Scheme 2. Synthesis 3-aryl- and 3,4-diarylpyrazoles 10a-d, 12

$$F_3C$$
 $\begin{array}{c} O & O \\ \hline O$

9, **10** a R^1 = Ph, b R^1 = 2-HO-4-MeOC₆H₃, c R^1 = 2-HOC₆H₄

brominated to obtain bromo derivative 11 which was subjected to Suzuki–Miyaura coupling to provide the novel diarylpyrazole 12 (Scheme 2).

For 6a

4. $\mathbf{5}$ **a** \mathbf{R}^1 = H, **b** \mathbf{R}^1 = 4-Cl, **c** \mathbf{R}^1 = 4-MeO **6 a** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = Me; **b** \mathbf{R}^1 = H, \mathbf{R}^2 = CF₃; **c** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃; **d** \mathbf{R}^1 = 4-MeO, \mathbf{R}^2 = CF₃ **7 a** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **b** \mathbf{R}^1 = H, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **c** \mathbf{R}^1 = 4-MeO, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **d** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Mo,; **g** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **d** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = 2,6-Cl₂C₆H₃CH₂; **h** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = 4-BrC₆H₄CH₂; **i** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = i-Pr; **j** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me₂CH(CH₂)₂; **k** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = H; **8 a** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = Me, \mathbf{R}^3 = H; **b** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **c** \mathbf{R}^1 = H, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **d** \mathbf{R}^1 = 4-MeO, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **e** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Bn; **g** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Mo; **g** \mathbf{R}^1 = 4-Cl, \mathbf{R}^2 = CF₃, \mathbf{R}^3 = Me; **g** \mathbf{R}^3 = Me; **g** \mathbf{R}^3 = Me; **e** \mathbf{R}^3 =

Scheme 3. Synthesis of 4,5-diarylisoxazoles 13a–e, 14, 15
$$OR^3$$
 ($R^3 = MOM$) OH OH OH O

13 a $R^3 = H$, b $R^3 = Bn$, c $R^3 = MOM$, d $R^3 = i$ -Pr, e $R^3 = (CH_2)_2CHMe_2$

Scheme 4. Synthesis of 4,5-diarylisoxazoles 18 and 20

Novel isoxazole-based analogs 13a–e, 14, and 15 were obtained from isoflavone derivatives 6c, 7d,f,i,j (Scheme 3). Their reaction with hydroxylamine provided isoxazoles 13a–e.⁶ *O*-MOM-protected product 13c was methylated at the free phenolic OH group, and the resulting derivative 14 was subjected to MOM deprotection in acid media to obtain isoxazole 15.

Not previously described deoxygenated diarylisoxazole analogs 18 and 20 were prepared starting from isoflavone derivative 6c (Scheme 4). Triflate formation with subsequent reduction has been tried. First, isoflavone 6c transformed to triflate 16 in which the C–O bond was cleaved under Pdcatalyzed hydrogenolysis conditions using triethylsilane as hydrogen transfer reagent. The resulting isoflavone derivative 17 was converted to isoxazole 18. It was then transformed to triflate 19 which was reduced to give product 20.

Previously reported isoxazole 21a and the novel isoxazole 21b without a substituent in position 4 of heterocycle were prepared starting from diketones 9a,b (Scheme 5).

Scheme 5. Synthesis of 4,5-diarylisoxazoles 21

All synthesized compounds 8a-l, 10a-d, 12, 13a-e, 15, 18, 20, 21a,b were subjected to *in vitro* growth inhibiton tests of *Staphylococcus aureus* Newman.⁷ The results of these tests are summarized in Tables 1-4.

Compound with methyl group as R² substituent (compound **8a**, Table 1) exhibited fourfold lower potency compared to the original hit **2**.

An improvement of the antibacterial potency was achieved by addition of methyl group as R³ substituent (compound **8b**, Table 1). Replacement of 4-chlorophenyl with phenyl group as R¹C₆H₄ substituent (compound **8c**) or 4-methoxyphenyl group (compound **8d**) slightly decreased the antibacterial potency. However, *O*-benzyl group as R³ substituent had a positive effect to antibacterial potency (compounds **8e,f**). Curiously, in the case of compounds **8e,f**, the difference in R¹ substitution did not affect MIC values which were retained around 1.56 µg/ml for both of

Table 1. Antibacterial activity of compounds 8a-l

Compound	\mathbb{R}^1	R^2	\mathbb{R}^3	MIC,* μg/ml
8a	4-C1	Me	Н	25
8b	4-C1	CF ₃	Me	3.12
8c	Н	CF ₃	Me	12.5
8d	4-OMe	CF ₃	Me	12.5
8e	4-C1	CF ₃	Bn	1.56
8f	Н	CF ₃	Bn	1.56
8g	4-C1	CF ₃	MOM	3.12
8h	4-C1	CF ₃	$2,6$ - Cl_2Bn	1.56
8i	4-C1	CF ₃	4-BrBn	1.56
8j	4-C1	CF ₃	$PhSO_2$	1.56
8k	4-C1	CF ₃	<i>i</i> -Pr	0.78
81	4-C1	CF ₃	i-Amyl	< 0.39

^{*} Staphylococcus aureus Newman.

Table 2. Antibacterial activity of compounds 10a-d, 12

Compound	\mathbb{R}^1	R^2	MIC,* μg/ml
10a	Н	Ph	50
10b	Н	2-(HO)-4-(MeO)C ₆ H ₃	25
10c	Н	$2-(HO)C_6H_4$	125
10d	Н	2,4-di(HO)C ₆ H ₃	250
12	4-ClC ₆ H ₅	Ph	1.56

^{*} Staphylococcus aureus Newman.

the compounds. MOM group as R^3 substituent (compound 8g) only slightly increased activity in comparison with hit compound 2. Substitution of benzyl group with 2,5-dichlorobenzyl (compound 8h), 4-bromobenzyl (compound 8i), and phenylsulfonyl (compound 8j) group did not change the activity of the compounds in comparison with benzyl analog 8e. The best antimicrobial activity in this series was exhibited by the compounds bearing lipophilic R^3 substituents such as isopropyl group (compound 8k) and isoamyl group (compound 8l).

Derivatives **10a**–**d** lacking substituents at position 4 of pyrazole ring showed significantly worse results in comparison with the hit compound **2** (Table 2). However, compound **12** with 4-chlorophenyl group as R¹ substituent and phenyl group as R² substituent exhibited activity four times higher than compound **2** (Table 2). These results point to the importance of the two aryl substituents at the pyrazole ring to ensure high antimicrobial potency. In addition, the high antimicrobial potency of compound **12** implies that hydroxyl groups at the phenyl group as the R¹ substituent are not essential.

Isoxazole analogs 13a-e (Table 3) showed similar potency and SAR to their pyrazole peers 8e,g,k,l, (Table 1). An interesting deviation was observed for isoxazoles 15, 18,

Table 3. Activity of isoxazole-based compounds 13a-e

Compound	\mathbb{R}^3	MIC,* μg/ml
13a	Н	3.12
13b	Bn	0.78
13c	MOM	3.12
13d	<i>i</i> -Pr	0.78
13e	i-Amyl	< 0.39

^{*} Staphylococcus aureus Newman.

Table 4. Activity of simplified isoxazole-based compounds 15, 18, 20, 21a,b

$$R^1$$
 R^2 R_3 R_4 R_5

15, 18, 20, 21a,b

Compound	\mathbb{R}^1	\mathbb{R}^2	MIC,* μg/ml
21a	Н	Ph	Inactive
21b	Н	2-(HO)-4-(MeO)C ₆ H ₃	6.25
15	$4-C1C_6H_5$	2-(MeO)-4-(HO)C ₆ H ₃	3.12
18	4-ClC ₆ H ₅	2-(HO)C ₆ H ₄	3.12
20	4-ClC ₆ H ₅	Ph	Inactive

^{*} Staphylococcus aureus Newman.

20, and **21a**,**b** (Table 4). Compound **21a**, contrary to its pyrazole-based analog **10a**, completely lost activity against *S. aureus*. Compound **21b** possesses increased activity level in comparison with compound **10b**. Surprisingly, methylation of *o*-hydroxy group in R^2 substituent (compound **15**, Table 4) did not affect MIC value – it was retained at 3.12 µg/ml. Finally, compound **20** totally lost the antimicrobial potency (Table 4) in comparison with pyrazole derivative **12** having the same substitution pattern (Table 2).

The most efficient compounds, such as 8k, were exhibiting activity level against *Staphylococcus aureus* Newman comparable with well-known antibiotics such as ampicillin (MIC 1.0 μ g/ml), ciprofloxacin (MIC 0.5 μ g/ml), and vancomycin (MIC 1.0 μ g/ml).

The SAR of the compounds provides the directions for further structural improvements to achieve more potent phenylazole-based antimicrobials. Thus, introduction of the lipophilic groups at position 5 of phenolic ring of the molecule increased the potency of compounds **8e,k**. Further increase of lipophilicity in this position could increase the potency. Additionally, further work should explore another suitable 5-membered cycles such as imidazole, 1,2,3-triazole, or isothiazole as scaffolds to improve the potency of the compounds. Nevertheless, the SAR of the pairs of compounds **12** and **20**, or **18** and **20** implies that at least one NH or OH group should be retained in the inhibitor to preserve its potency.

Our investigation of growth inhibition of *S. aureus* Newman by 3,4-diphenylpyrazole and 4,5-diphenylisoxazole derivatives lead to several very potent antibacterial compounds. The most potent growth inhibitors were pyrazole-based compounds **8k,l** and their isoxazole analogs **13d,e** with MIC <1 µg/ml. The studies revealed the most crucial elements for their antimicrobial activity. The structure should contain at least one hydrogen bond donor either in heterocycle or at aryl groups. Pyrazole replacement with isoxazole in most cases did not affect activity, however, several differences were found. For example, both aryl groups were needed at positions 3 and 4 for pyrazole-based compounds to exhibit high potency, however, relatively potent compound **21b** was found in

isoxazole series lacking aryl group at position 3 of isoxazole. The mechanism of action for pyrazole- and isoxazole-based *S. aureus* growth inhibitors needs further investigation.

Experimental

¹H NMR spectra were recorded on 300, 400, or 600 MHz Bruker spectrometers. ¹³C and ¹⁹F NMR spectra were recorded on 400 (101 and 376 MHz, respectively) or 600 MHz Bruker spectrometers (151 and 564 MHz, respectively) using the residual solvent peak as internal reference (CDCl₃: 7.26 ppm for ¹H nuclei and 77.2 ppm for ¹³C nuclei; DMSO-d₆: 2.50 ppm for ¹H nuclei and 39.5 ppm for ¹³C nuclei; (CD₃)₂CO: 2.05 ppm for ¹H nuclei and 29.8 and 206.3 ppm for ¹³C nuclei). HRMS were determined on a Waters Synapt G2-Si hybrid quadrupole time-of-flight (TOF) mass spectrometer equipped with an electron spray ion source (ESI). Melting points were detected with an OptiMelt MPA100 melting point apparatus, with a heating rate of 3°C/min. When necessary, compounds were purified by crystallization or by column chromatography on silica gel (petroleum ether – EtOAc gradient).

Reagents were purchased from commercial sources and used as received. Reactions requiring anhydrous conditions were performed with the usual precautions for rigorous exclusion of moisture.

Synthesis of 1-(2,4-dihydroxyphenyl)-2-phenylethan- 1-ones 5a-c (General method). Procedure described in literature have been used. BF₃·Et₂O (15.46 ml, 17.47 g, 123 mmol) was added slowly to a solution of resorcinol (**3**) (4.52 g, 41 mmol) and phenylacetic acid **4a-c** (41 mmol) in anhydrous PhMe (120 ml). Resulting solution was heated at 100°C for 3 h and cooled down to room temperature. The mixture was poured into saturated NaOAc solution (300 ml) and then partitioned with EtOAc (300 ml). The EtOAc extract was washed with brine (2×200 ml). The extract was dried over Na₂SO₄ and concentrated *in vacuo*. Thus obtained crude product was triturated with PhMe or purified by column chromatography on silica gel using gradient EtOAc in petroleum ether.

1-(2,4-Dihydroxyphenyl)-2-phenylethan-1-one (5a) was synthesized from phenylacetic acid (**4a**) (1.1 g). Yield 1.23 g (66%), yellowish sticky oil. Spectral data was in accordance with the previously reported.⁹

2-(4-Chlorophenyl)-1-(2,4-dihydroxyphenyl)ethan- 1-one (5b) was synthesized from 4-chlorophenylacetic acid **(4b)** (7.0 g). Yield 6.0 g (56%), slightly pink solid. Spectral data was in accordance with the previously reported. ¹⁰

1-(2,4-Dihydroxyphenyl)-2-(4-methoxyphenyl)ethan-1-one (5c) was synthesized from 4-methoxyphenylacetic acid (4c) (350 mg). Yield 640 mg (78%), yellowish solid. Spectral data was in accordance with the previously reported.¹¹

3-(4-Chlorophenyl)-7-hydroxy-2-methyl-4*H***-chromen-4-one (6a)**. Procedure, described in literature, have been used. ¹² 2-(4-Chlorophenyl)-1-(2,4-dihydroxyphenyl)ethan-1-one (**5b**) (700 mg, 2.7 mmol) and anhydrous NaOAc (437 mg, 5.4 mmol) were dissolved in Ac₂O (4 ml, 42.6 mmol). The mixture was refluxed for 14 h, cooled,

and poured into H₂O. The precipitate was filtered off, dried, and recrystallized from EtOH to obtain acylated intermediate as a slightly yellow solid (594 mg, 1.8 mmol). This material was suspended in EtOH (5 ml), and aqueous NaOH (86.7 mg, 2.2 mmol) was added thereto. After heating at 50°C for 15 min, solvent was distilled off under reduced pressure, residue was dissolved in H₂O and acidified by 1 M HCl. Formed precipitate was filtered off and dried. Yield 496 mg (65% over 2 steps), white-beige solid. Spectral data was in accordance with the previously reported.¹²

Synthesis of 7-hydroxy-3-phenyl-2-(trifluoromethyl)- 4H-chromen-4-ones 6b–d (General method). Procedure, described in literature, have been used. ¹³ Trifluoroacetic acid anhydride (9 ml, 13.6 g, 64.8 mmol) was added dropwise to an ice-cooled solution of deoxybenzoin (16.2 mmol) in pyridine (20 ml). The resulting solution was stirred for 14 h at room temperature. Reaction mixture was diluted with EtOAc (200 ml), washed with 1 M HCl (3×150 ml), brine (150 ml), and dried over anhydrous Na₂SO₄, followed by evaporation *in vacuo*. Crude product was triturated with EtOH–H₂O, 1:1 (for compounds **6b,c**) or EtOAc – petroleum ether, 1:3 (for compound **6d**) mixture.

7-Hydroxy-3-phenyl-2-(trifluoromethyl)-4*H***-chromen-4-one (6b)** was synthesized from compound **5a** (570 mg). Yield 497 mg (65%), white-beige solid. Spectral data was in accordance with the previously reported. ¹⁴

3-(4-Chlorophenyl)-7-hydroxy-2-(trifluoromethyl)- 4H-chromen-4-one (6c) was synthesized from compound **5b** (4.25 g). Yield 3.47 g (63%), white-yellow solid, mp 248–250°C. 1 H NMR spectrum (300 MHz, DMSO- d_6), δ , ppm (J, Hz): 11.18 (1H, s, OH); 7.93 (1H, d, J = 8.8, H-5); 7.52 (2H, d, J = 8.5 C_6 H₄Cl); 7.31 (2H, d, J = 8.5, C_6 H₄Cl); 7.02 (1H, dd, J = 8.8, J = 2.2, H-8); 6.96 (1H, d, J = 2.2, H-6). 13 C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm (J, Hz): 175.1; 164.1; 156.6; 146.8 (q, J = 35.5); 133.6; 131.9; 128.6; 128.2; 127.6; 124.0; 119.4 (q, J = 276.4); 116.6; 115.5; 102.4. 19 F NMR spectrum (376 MHz, DMSO- d_6), δ , ppm: -62.86. Found, m/z: 341.0198 [M+H] $^+$. C_{16} H₉ClF₃O₃. Calculated, m/z: 341.0192.

7-Hydroxy-3-(4-methoxyphenyl)-2-(trifluoromethyl)-4*H*-chromen-4-one (6d) was synthesized from compound 5c (640 mg). Yield 258 mg (31%), light-brown solid. Spectral data was in accordance with the previously reported.¹³

Synthesis of 7-(alkyloxy)-3-(4-chlorophenyl)-2-(trifluoromethyl)-4H-chromen-4-ones 7a–j (General method). Alkyl chloride, iodide, or bromide (0.6 mmol) was added to a stirred solution of compound 6b–d (0.45 mmol) and K_2CO_3 (1 mmol) in DMF (3 ml). The resulting solution was stirred for 14 h at room temperature. Reaction mixture was diluted with EtOAc (50 ml), washed with brine (3×50 ml), dried over anhydrous Na_2SO_4 , and evaporated *in vacuo*. Crude product was crystallized from EtOH.

3-(4-Chlorophenyl)-7-methoxy-2-(trifluoromethyl)-4*H*-chromen-4-one (7a) was synthesized from compound 6c (300 mg), using MeI as alkylating agent. Yield 258 mg (83%), white solid, mp $162-164^{\circ}$ C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (*J*, Hz): 8.12 (1H, d, J = 8.9,

H-5); 7.42 (2H, d, J = 8.5, C₆H₄Cl); 7.20 (2H, d, J = 8.4, C₆H₄Cl); 7.05 (1H, dd, J = 8.9, J = 2.4, H-8); 6.95 (1H, d, J = 2.3, H-6); 3.95 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (J, Hz): 175.9; 165.3; 157.1; 148.3 (q, J = 36.5); 135.1; 131.4; 128.7; 128.0; 127.7; 124.6; 119.4 (q, J = 276.7); 117.1; 116.2; 100.3; 56.2. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ, ppm: –63.56. Found, m/z: 355.0360 [M+H]⁺. C₁₇H₁₁ClF₃O₃. Calculated, m/z: 355.0349.

7-Methoxy-3-phenyl-2-(trifluoromethyl)-4*H***-chromen-4-one (7b)** was synthesized from compound **6b** (200 mg) using MeI as alkylating agent. Yield 98 mg (47%), yellowish solid. Spectral data was in accordance with the previously reported. ¹⁵

7-Methoxy-3-(4-methoxyphenyl)-2-(trifluoromethyl)- 4*H***-chromen-4-one** (**7c**) was synthesized from compound **6d** (138 mg) using MeI as alkylating agent. Yield 114 mg (79%), beige solid, mp 135–138°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ, ppm (*J*, Hz): 8.13 (1H, d, J = 8.9, H-5); 7.19 (2H, d, J = 8.7, C₆H₄OMe); 7.03 (1H, dd, J = 8.9, J = 2.4, H-8); 6.97 (2H, d, J = 8.8, H Ar, C₆H₄OMe); 6.94 (1H, d, J = 2.4, H-6); 3.95 (3H, s, OCH₃); 3.85 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (*J*, Hz): 176.5; 165.1; 160.1; 157.1; 148.1 (q, J = 35.6); 131.2; 128.0; 125.5; 121.2; 119.6 (q, J = 275.8); 117.3; 116.0, 113.9, 100.2, 56.2, 55.4. ¹⁹F NMR spectrum (376 MHz, DMSO-*d*₆), δ, ppm: –62.58. Found, m/z: 351.0857 [M+H][†]. C₁₈H₁₄F₃O₄. Calculated, m/z: 351.0844.

7-(Benzyloxy)-3-(4-chlorophenyl)-2-(trifluoromethyl)- 4*H***-chromen-4-one** (**7d**) was synthesized from compound **6c** (500 mg) using BnBr as alkylating agent. Yield 585 mg (92%), white solid, mp 160–162°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ, ppm (*J*, Hz): 8.14 (1H, d, J = 8.9, H-5); 7.49–7.35 (7H, m, C₆H₄Cl, OCH₂C₆H₅); 7.20 (2H, d, J = 8.4, C₆H₄Cl); 7.13 (1H, dd, J = 8.9, J = 2.4, H-8); 7.03 (1H, d, J = 2.3, H-6); 5.20 (2H, s, OCH₂Ph). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (*J*, Hz): 175.9; 164.3; 156.9; 148.3 (q, J = 36.2); 135.4; 135.1; 131.4; 129.0; 128.7 (2C); 128.1; 127.7; 127.6; 124.6; 119.4 (q, J = 276.8); 117.3; 116.7; 101.4; 70.9. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: -62.74. Found, m/z: 431.0672 [M+H][†]. C₂₃H₁₅ClF₃O₃. Calculated, m/z: 431.0662.

7-(Benzyloxy)-3-phenyl-2-(trifluoromethyl)-4*H*-chromen-4-one (7e) was synthesized from compound 6b (180 mg) using BnBr as alkylating agent. Yield 60 mg (26%), white solid, mp 123–125°C. ¹H NMR spectrum (400 MHz, (CD₃)₂CO), δ, ppm (*J*, Hz): 8.02 (1H, d, J = 9.1, H-5); 7.53–7.49 (2H, m, H Ph); 7.44–7.37 (5H, m, H Ph); 7.37–7.32 (1H, m, H Ph); 7.32–7.27 (2H, m, H Ph); 7.26 (1H, d, J = 2.2, H-8); 7.20 (1H, dd, J = 8.9, J = 2.4, H-6); 5.34 (2H, s, OCH₂C₆H₅). ¹³C NMR spectrum (101 MHz, (CD₃)₂CO), δ, ppm (*J*, Hz): 176.1; 165.1; 157.8; 148.3 (q, J = 35.7); 137.1; 130.9 (q, J = 1.4); 130.7; 129.5; 129.3; 129.1; 128.7; 128.6; 128.2; 126.6; 120.6 (q, J = 275.6); 118.1; 117.3; 102.3; 71.5. ¹⁹F NMR spectrum (376 MHz, (CD₃)₂CO), δ, ppm: –64.31. Found, m/z: 397.1057 [M+H]⁺. C₂₃H₁₆F₃O₃. Calculated, m/z: 397.1052.

3-(4-Chlorophenyl)-7-(methoxymethoxy)-2-(trifluoromethyl)-4*H***-chromen-4-one** (7f) was synthesized from compound **6c** (400 mg) using MOMCl as alkylating agent.

Yield 262 mg (58%), off-white solid, mp 125–127°C.
¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (J, Hz): 8.14 (1H, d, J = 8.9, H-5); 7.42 (2H, d, J = 8.5, C₆H₄Cl); 7.23–7.17 (3H, m, H Ar); 7.13 (1H, dd, J = 8.9, J = 2.3, H-6); 5.31 (2H, s, OCH₂OCH₃), 3.52 (3H, s, OCH₂OCH₃).
¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (J, Hz): 176.0; 162.7; 156.7; 148.5 (q, J = 36.2); 135.2; 131.4; 128.7; 128.0; 127.7; 124.6; 119.4 (q, J = 276.4); 117.9; 117.0; 103.3; 94.6; 56.7.
¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: –63.57. Found, m/z: 385.0465 [M+H]⁺. C₁₈H₁₃ClF₃O₄. Calculated, m/z: 385.0454.

3-(4-Chlorophenyl)-7-[(2,6-dichlorobenzyl)oxy]-2-(trifluoromethyl)-4*H***-chromen-4-one (7g) was synthesized from compound 6c** (150 mg) using 2-(bromomethyl)-1,3-dichlorobenzene as alkylating agent. Yield 172 mg (78%), light-yellow solid, mp 196–197°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (*J*, Hz): 8.16 (1H, d, *J* = 9.4, H-5); 7.46–7.39 (4H, m, H Ar); 7.31 (1H, dd, *J* = 8.8, *J* = 7.2, H Ar); 7.21 (2H, d, *J* = 8.4, C₆H₄Cl); 7.16–7.10 (2H, m, H Ar); 5.42 (2H, s, OC $\underline{\text{H}}_2\text{C}_6\text{H}_3\text{Cl}_2$). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (*J*, Hz): 175.9; 164.3; 156.9; 137.2; 135.2; 131.4; 131.2; 130.9; 128.8; 128.7; 128.1; 127.7; 124.7; 119.4 (d, *J* = 276.6) 117.5; 116.6; 101.3; 100.1; 65.9. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: –63.55. Found, m/z: 498.9877 [M+H]⁺. C₂₃H₁₃Cl₃F₃O₃. Calculated, m/z: 498.9882.

7-[(4-Bromobenzyl)oxy]-3-(4-chlorophenyl)-2-(trifluoromethyl)-4H-chromen-4-one (7h) was synthesized from compound 6c (150 mg) using 1-bromo-4-(bromomethyl)benzene as alkylating agent. Yield 114 mg (51%), pink solid, mp 153–155°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (J, Hz): 8.14 (1H, d, J = 8.9, H-5); 7.56 (2H, d, J = 8.4, C_6H_4Br); 7.42 (2H, d, J = 8.5, C_6H_4Cl); 7.33 (2H, d, J = 8.4, C₆H₄Br); 7.20 (2H, d, J = 8.4, C₆H₄Cl); 7.11 (1H, dd, J = 8.9, J = 2.4, H-8); 7.00 (1H, d, J = 2.3, H-6); 5.15 (2H, s, OCH₂C₆H₄Br). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (J, Hz): 175.9; 164.0; 156.9; 148.4 (q, J = 36.0); 135.2; 134.4; 132.2; 131.3; 129.2; 128.7; 128.2; 127.6; 124.7; 122.7; 120.8; 119.4 (q, J = 276.7); 116.6; 101.4; 70.1. 19 F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -63.53. Found, m/z: 508.9775 [M+H]⁺. $C_{23}H_{14}BrClF_3O_3$. Calculated, *m/z*: 508.9767.

3-(4-Chlorophenyl)-7-isopropoxy-2-(trifluoromethyl)- 4H-chromen-4-one (7i) was synthesized from compound **6c** (250 mg) using *i*-PrI as alkylating agent. Yield 275 mg (98%), yellowish solid, mp 128–130°C. ¹H NMR spectrum (300 MHz, CDCl₃), δ , ppm (*J*, Hz): 8.11 (1H, d, J = 8.9, H-5); 7.42 (2H, d, J = 8.6, C₆H₄Cl); 7.20 (2H, d, J = 8.4, C₆H₄Cl); 7.00 (1H, dd, J = 8.9, J = 2.3, H-8); 6.91 (1H, d, J = 2.3, H-6); 4.70 (1H, hept, J = 5.8, OCH(CH₃)₂), 1.43 (6H, d, J = 6.1, OCH(CH₃)₂). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (J, Hz): 175.9; 163.8; 157.1; 148.2 (q, J = 36.3); 135.1; 131.4; 128.7; 128.0; 127.8; 124.6; 119.5 (q, J = 276.7); 117.2; 116.8; 101.5; 71.3; 21.9. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -63.58. Found, m/z: 383.0656 [M+H]⁺. C₁₉H₁₅CIF₃O₃. Calculated, m/z: 383.0662.

3-(4-Chlorophenyl)-7-(isopentyloxy)-2-(trifluoromethyl)-4*H***-chromen-4-one (7j) was synthesized from compound 6c** (250 mg) using isoamyl bromide as alkylating agent. Yield 297 mg (98%), yellow solid, mp 80–82°C. ¹H NMR spectrum (300 MHz, CDCl₃), δ, ppm (J, Hz): 8.11 (1H, d, J = 8.9, H-5); 7.42 (2H, d, J = 8.5, C₆H₄Cl); 7.20 (2H, d, J = 8.4, C₆H₄Cl); 7.03 (1H, dd, J = 8.9, J = 2.3, H-8); 6.94 (1H, d, J = 2.3, H-6); 4.12 (2H, t, J = 6.6, OCH₂CH₂CH(CH₃)₂); 1.95–1.80 (1H, m, OCH₂CH₂CH(CH₃)₂); 1.77 (2H, t, J = 6.5, OCH₂CH₂CH(CH₃)₂); 1.00 (6H, d, J = 6.5, OCH₂CH₂CH(CH₃)₂). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (J, Hz): 175.9; 164.8; 157.1; 148.3 (q, J = 36.3); 135.1; 131.4; 128.6; 127.9; 124.6; 119.5 (q, J = 276.7); 116.9; 116.6; 110.2; 100.7; 67.6; 37.7; 25.2; 22.7. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ, ppm: -63.59. Found, m/z: 411.0981 [M+H]⁺. C₂₁H₁₉ClF₃O₃. Calculated, m/z: 411.0975.

3-(4-Chlorophenyl)-4-oxo-2-(trifluoromethyl)-4Hchromen-7-yl benzenesulfonate (7k). Phenylsulfonyl chloride (233 mg, 1.33 mmol) was added to a stirred solution of compound 6c (300 mg, 0.88 mmol) and Et₃N (245 μl, 1.76 mmol) in CH₂Cl₂ (5 ml). Resulting solution was stirred for 14 h at room temperature. The reaction mixture was diluted with CH₂Cl₂ (15 ml), washed with brine (3×50 ml), dried over anhydrous Na₂SO₄, and evaporated in vacuo. Crude product was purified by silica gel column chromatography, using gradient from 5 to 20% EtOAc in petroleum ether. Yield 390 mg (92%), brown oil. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (*J*, Hz): 8.16 (1H, d, J = 8.8, H-5); 7.91 (2H, d, J = 7.3, H Ar); 7.74 (1H, t, J = 7.5, H Ar); 7.60(2H, t, J = 7.9, H Ar); 7.49-7.35 (3H, m, H Ar); 7.18 (2H, m, Hd, J = 8.4, C₆H₄Cl); 7.07 (1H, dd, J = 8.8, J = 2.2, H-6). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (J, Hz): 175.9; 155.4; 154.1; 149.1 (q, J = 36.6); 135.5; 135.1; 135.0; 131.2; 129.7; 128.8; 128.6; 128.5; 127.0; 125.0; 121.9; 121.0; 119.2 (q, J = 277.2); 112.4. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -63.56. Found, m/z: 481.0128 [M+H]⁺. $C_{22}H_{13}ClF_3O_5S$. Calculated, m/z: 481.0124.

Synthesis of pyrazoles 2, 8a–l (General method). Hydrazine hydrate (1.3 ml, 27.6 mmol) was added to a stirred solution of compound **6a,c**, **7a–k** (0.67 mmol) in EtOH (5 ml). Resulting solution was stirred for 2 h at reflux. Reaction mixture was evaporated to dryness and triturated with cold H₂O.

4-[4-(4-Chlorophenyl)-5-(trifluoromethyl)-1*H*-pyrazol-**3-yl|benzene-1,3-diol (2)** was synthesized from compound **6c** (258 mg). Yield 208 mg (77%), white solid. Spectral data was in accordance with the previously reported. ¹⁶

4-[4-(4-Chlorophenyl)-5-methyl-1*H*-pyrazol-3-yl]-benzene-1,3-diol (8a) was synthesized from compound 6a (183 mg). Yield 158 mg (94%), white solid, mp 231–234°C. ¹H NMR spectrum (300 MHz, DMSO- d_6), δ , ppm (J, Hz): 12.52 (1H, s, NH); 10.58 (1H, s, OH), 9.38 (1H, s, OH); 7.37–7.20 (4H, m, C₆H₄Cl); 6.76 (1H, d, J = 7.8, H Ar); 6.29–6.11 (2H, m, H Ar); 2.19 (3H, s, CH₃). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm: 157.9; 157.0; 147.0; 137.6; 131.4; 130.3; 128.8; 128.5; 128.1; 115.1; 109.7; 106.3; 102.8; 12.9. Found, m/z: 301.0752 [M+H][†]. $C_{16}H_{14}ClN_2O_2$. Calculated, m/z: 301.0744.

2-[4-(4-Chlorophenyl)-5-(trifluoromethyl)-1*H*-pyrazol-**3-yl]-5-methoxyphenol (8b)** was synthesized from compound **7a** (600 mg). Yield 600 mg (96%), white-beige solid, mp 178–180°C. ¹H NMR spectrum (300 MHz,

DMSO- d_6), δ, ppm (J, Hz): 7.39 (2H, d, J = 8.4, C₆H₄Cl); 7.19 (2H, d, J = 8.4, C₆H₄Cl); 6.85 (1H, d, J = 8.5, H-5); 6.42 (1H, d, J = 2.4, H-2); 6.29 (1H, dd, J = 8.5, J = 2.4, H-6); 3.68 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ, ppm (J, Hz): 161.2; 156.8; 139.8; 138.1 (q, J = 36.0); 132.2; 131.9; 131.4; 130.6; 128.4; 122.2 (q, J = 269.2); 116.9; 107.8; 105.0; 101.5; 55.2. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: –57.88. Found, m/z: 369.0628 [M+H]⁺. C₁₇H₁₃ClF₃N₂O₂. Calculated, m/z: 369.0618.

5-Methoxy-2-[4-phenyl-5-(trifluoromethyl)-1*H***-pyrazol-3-yl]phenol (8c)** was synthesized from compound 7b (91 mg). Yield 82 mg (86%), white solid, mp 145–147°C. ¹H NMR spectrum (400 MHz, DMSO- d_6), δ, ppm (J, Hz): 7.35–7.24 (3H, m, H Ar); 7.21–7.14 (2H, m, H Ar); 6.85 (1H, d, J = 8.5, H-3); 6.45 (1H, d, J = 2.5, H-6); 6.31 (1H, dd, J = 8.6, J = 2.5, H-4); 3.68 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ, ppm (J, Hz): 160.9; 156.7; 139.3; 138.1 (q, J = 34.8); 131.8; 131.5; 129.6; 128.1; 127.1; 122.2 (q, J = 269.2); 118.0; 108.0; 104.7; 101.3; 55.0. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: –57.80. Found, m/z: 335.1021 [M+H]⁺. $C_{17}H_{14}F_3N_2O_2$. Calculated, m/z: 335.1007.

5-Methoxy-2-[4-(4-methoxyphenyl)-5-(trifluoromethyl)- 1*H***-pyrazol-3-yl]phenol (8d)** was synthesized from compound **7c** (102 mg). Yield 77 mg (73%), beige powder, mp 75–79°C. 1 H NMR spectrum (400 MHz, DMSO- d_6), δ, ppm (J, Hz): 7.09 (2H, d, J = 8.2, C₆H₄OMe); 6.87 (3H, t, J = 8.2, H Ar); 6.45 (1H, s, H-6); 6.32 (1H, d, J = 9.9, H-4); 3.73 (3H, s, OCH₃); 3.68 (3H, s, OCH₃). 13 C NMR spectrum (101 MHz, DMSO- d_6), δ, ppm (J, Hz): 160.7; 158.3; 156.7; 139.2; 138.1 (q, J = 36.0); 131.7; 130.7; 123.5; 122.3 (q, J = 270.1); 117.6; 113.6; 108.2; 104.7; 101.3; 55.0 (q, J = 4.4). 19 F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: –57.84. Found, m/z: 365.1121 [M+H]⁺. C₁₈H₁₆F₃N₂O₃. Calculated, m/z: 365.1113.

5-(Benzyloxy)-2-[4-(4-chlorophenyl)-5-(trifluoromethyl)- 1*H***-pyrazol-3-yl]phenol (8e)** was synthesized from compound **7d** (289 mg). Yield 274 mg (92%), white solid, mp 78–81°C. ¹H NMR spectrum (300 MHz, DMSO- d_6), δ , ppm (J, Hz): 7.46–7.28 (7H, m, C_6H_4Cl , $OCH_2C_6\underline{H}_5$); 7.18 (2H, d, J = 8.4, C_6H_4Cl); 6.89 (1H, d, J = 8.5, H-3); 6.50 (1H, d, J = 2.3, H-6); 6.43 (1H, dd, J = 8.5, J = 2.3, H-4); 5.03 (2H, s, $OC\underline{H}_2C_6H_5$). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm (J, Hz): 160.2; 156.7; 139.8; 138.2 (q, J = 34.7); 136.9; 132.2; 131.9; 131.4; 130.6; 128.6; 128.4; 128.1; 127.9; 122.2 (q, J = 269.2); 117.0; 108.0; 105.8; 102.4; 69.3. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ , ppm: –57.59. Found, m/z: 445.0934 [M+H]⁺. $C_{23}H_{17}ClF_3N_2O_2$. Calculated, m/z: 445.0931.

5-(Benzyloxy)-2-[4-phenyl-5-(trifluoromethyl)-1*H***-pyrazol-3-yl]phenol (8f)** was synthesized from compound **7e** (48 mg). Yield 46 mg (92%), white solid, mp 84–88°C.

¹H NMR spectrum (400 MHz, (CD₃)₂CO), δ, ppm (*J*, Hz): 7.46–7.26 (9H, m, H Ar, OCH₂C₆H₅); 6.90 (1H, d, J = 8.6, H-3); 6.63 (1H, d, J = 2.4, H-6); 6.40 (1H, dd, J = 8.6, J = 2.4, H-4), 5.06 (2H, s, OCH₂C₆H₅).

¹³C NMR spectrum (101 MHz, (CD₃)₂CO), δ, ppm (*J*, Hz): 161.4; 157.3; 140.3; 138.1; 132.9; 132.2; 131.1; 129.3; 129.1; 128.7; 128.5; 128.2; 122.3 (q, J = 268.8); 119.0; 109.4; 107.0; 103.5; 70.5.

¹⁹F NMR spectrum (376 MHz, (CD₃)₂CO), δ, ppm: –59.66.

Found, m/z: 411.1344 [M+H]⁺. C₂₃H₁₈F₃N₂O₂. Calculated, m/z: 411.1320.

2-[4-(4-Chlorophenyl)-5-(trifluoromethyl)-1*H*-pyrazol-3-yl]-5-(methoxymethoxy)phenol (8g) was synthesized from compound 7f (136 mg). Yield 117 mg (83%), whitegray powder, mp 190–193°C. ¹H NMR spectrum (400 MHz, DMSO- d_6), δ , ppm (J, Hz): 10.00 (2H, br. s, NH, OH); 7.39 (2H, d, J = 8.3, C_6H_4Cl); 7.19 (2H, d, J = 8.3, C_6H_4Cl); 6.91 (1H, d, J = 8.5, H-3); 6.57 (1H, d, J = 2.0, H-6); 6.43 (1H, dd, J = 8.5, J = 2.0, H-4); 5.13 (2H, s, OCH₂OCH₃); 3.35 (3H, s, OCH₂OCH₃). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm (J, Hz): 158.5; 156.6; 139.5; 138.1 (q, J = 34.3); 132.0; 131.7; 131.3; 130.5; 128.3; 122.1 (q, J = 269.0); 116.9; 108.6; 106.9; 103.3; 93.6; 55.6. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ , ppm: -57.84. Found, m/z: 399.0735 [M+H]⁺. $C_{18}H_{15}ClF_3N_2O_3$. Calculated, m/z: 399.0723.

2-[4-(4-Chlorophenyl)-5-(trifluoromethyl)-1*H*-pyrazol-3-yl]-5-[(2,6-dichlorobenzyl)oxy]phenol (8h) was synthesized from compound 7g (161 mg). Yield 138 mg (83%), white solid, mp 92–94°C. ¹H NMR spectrum (400 MHz, DMSO- d_6), δ , ppm (J, Hz): 7.60–7.52 (2H, m, H Ar); 7.46 (1H, dd, J = 8.9, J = 7.1, H Ar); 7.40 (2H, d, J = 8.5, C_6H_4Cl); 7.21 (2H, d, J = 8.4, C_6H_4Cl); 6.95 (1H, d, J = 8.3, H-3); 6.59–6.47 (2H, m, H Ar); 5.16 (2H, s, OC $\underline{H}_2C_6H_3Cl_2$). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm (J, Hz): 160.3; 156.8; 139.8; 138.3 (q, J = 34.4); 136.2; 132.2; 132.1; 131.8; 131.6; 131.5; 130.6; 128.9; 128.4; 122.2 (q, J = 269.1); 117.0; 108.6; 105.3; 102.3; 65.0. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ , ppm: –57.83. Found, m/z: 513.0157 [M+H]⁺. $C_{23}H_{15}Cl_3F_3N_2O_2$. Calculated, m/z: 513.0151.

5-[(4-Bromobenzyl)oxy]-2-[4-(4-chlorophenyl)-5-(trifluoromethyl)-1*H*-pyrazol-3-yl]phenol (8i) was synthesized from compound 7h (114 mg). Yield 93 mg (79%), beige solid, mp 83–85°C. ¹H NMR spectrum (300 MHz, DMSO- d_6), δ, ppm (J, Hz): 7.58 (2H, d, J = 8.4, C₆H₄Br); 7.44–7.30 (4H, m, C₆H₄Cl, C₆H₄Br); 7.18 (2H, d, J = 8.4, C₆H₄Cl); 6.86 (1H, d, J = 8.5, H-3); 6.47 (1H, d, J = 2.4, H-6); 6.37 (1H, dd, J = 8.6, J = 2.4, H-4); 5.01 (2H, s, OCH₂C₆H₄Br). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ, ppm (J, Hz): 160.0; 156.8; 139.7; 138.2 (q, J = 33.3); 136.4; 132.2; 131.9; 131.5; 131.4; 130.6; 130.0; 128.4; 122.2 (q, J = 269.1); 121.1; 117.0; 108.2; 105.7; 102.4; 68.5. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: –57.84. Found, m/z: 525.0023 [M+H][†]. C₂₃H₁₆BrClF₃N₂O₂. Calculated, m/z: 525.0040.

4-[4-(4-Chlorophenyl)-5-(trifluoromethyl)-1*H***-pyrazol-3-yl]-3-hydroxyphenyl benzenesulfonate (8j) was synthesized from compound 7k (250 mg). Yield 228 mg (89%), beige solid, mp 193–196°C. ^{1}H NMR spectrum (400 MHz, DMSO-d_6), \delta, ppm (J, Hz): 7.87–7.79 (3H, m, H Ar); 7.66 (2H, t, J = 7.8, H Ar); 7.40 (2H, d, J = 8.5, C₆H₄Cl); 7.15 (2H, d, J = 8.4, C₆H₄Cl); 6.90 (1H, d, J = 8.5, H-2); 6.51 (1H, d, J = 2.4, H-5); 6.28 (1H, dd, J = 8.5, J = 2.4, H-6). ^{13}C NMR spectrum (101 MHz, DMSO-d_6), \delta, ppm (J, Hz): 158.0 (d, J = 16.5); 149.4 (d, J = 4.1); 140.3 (d, J = 14.2); 138.2 (d, J = 34.3); 134.9; 134.4; 132.0; 131.4; 131.0 (d, J = 8.6); 130.1 (d, J = 13.6); 129.8; 128.3; 128.1; 122.4 (d, J = 272.6);**

116.3 (d, J = 9.0); 115.3 (d, J = 4.6); 111.2 (d, J = 9.9); 109.6. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ , ppm: -57.30. Found, m/z: 495.0407 [M+H]⁺. $C_{22}H_{15}ClF_3N_2O_4S$. Calculated, m/z: 495.0393

2-[4-(4-Chlorophenyl)-5-(trifluoromethyl)-1*H***-pyrazol-3-yl]-5-isopropoxyphenol (8k)** was synthesized from compound 7i (130 mg). Yield 129 mg (95%), white solid, mp 176–178°C. ¹H NMR spectrum (400 MHz, DMSO- d_6), δ, ppm (J, Hz): 7.38 (2H, d, J = 8.1, C₆H₄Cl); 7.18 (2H, d, J = 8.1, C₆H₄Cl); 6.86 (1H, d, J = 8.5, H-3); 6.41 (1H, s, H-6); 6.32 (1H, d, J = 8.4, H-4); 4.49 (1H, sept, J = 5.7, OCH(CH₃)₂); 1.22 (6H, d, J = 5.9, OCH(CH₃)₂). ¹³C NMR spectrum (151 MHz, CDCl₃), δ, ppm (J, Hz): 159.2; 156.8; 139.8; 138.0 (q, J = 34.6); 131.9; 131.6; 131.3; 130.7; 128.2; 122.2 (q, J = 269.0); 116.6; 107.4; 106.1; 102.8; 69.2; 21.8. ¹⁹F NMR spectrum (564 MHz, CDCl₃), δ, ppm: –53.02. Found, m/z: 397.0932 [M+H]⁺. C₁₉H₁₇ClF₃N₂O₂. Calculated, m/z: 397.0931.

2-[4-(4-Chlorophenyl)-5-(trifluoromethyl)-1H-pyrazol-3-vl]-5-(isopentyloxy)phenol (81) was synthesized from compound 7j (140 mg). Yield 141 mg (97%), light-yellow solid, mp 73–75°C. ¹H NMR spectrum (600 MHz, CDCl₃), δ, ppm (J, Hz): 7.38 $(2H, d, J = 8.4, C_6H_4C1)$; 7.24 (2H, d, J)J = 8.3, C₆H₄Cl); 6.82 (1H, d, J = 8.8, H-3); 6.49 (1H, s, H-6); 6.27 (1H, dd, J = 8.8, J = 1.9, H-4); 3.92 (2H, t, J = 6.7, OCH₂CH₂CH(CH₃)₂); 1.77 (1H, sept, J = 6.7, $OCH_2CH_2CH_1(CH_3)_2$; 1.63 (2H, q, J = 6.7, $OCH_2CH_2CH_1(CH_3)_2$); 0.93 (6H, d, J = 6.6, OCH₂CH₂CH(CH₃)₂). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm (J, Hz): 160.6; 156.7; 139.8; 138.3 (q, J = 32.9); 132.2; 131.9; 131.5; 130.7; 128.4; 122.2 (q, J = 269.1); 116.9; 107.7; 105.5; 102.0; 66.0; 37.5; 24.7; 22.5. ¹⁹F NMR spectrum (376 MHz. CDCl₃), δ , ppm: -59.35. Found, m/z: 425.1240 [M+H]⁺. $C_{21}H_{21}ClF_3N_2O_2$. Calculated, m/z: 425.1244.

Synthesis of pyrazoles 10a–c (General method). Hydrazine hydrate (1.86 ml, 38 mmol) was added to a stirred solution of diketone **9a–c** (1.9 mmol) in EtOH (15 ml). The resulting solution was stirred for 14 h at reflux. Reaction mixture was evaporated to dryness and triturated with cold H₂O.

3-Phenyl-5-(trifluoromethyl)-1*H***-pyrazole (10a)** was synthesized from 4,4,4-trifluoro-1-phenylbutane-1,3-dione (9a) (1.7 g). Yield 1.52 g (91%), white solid. Spectral data was in accordance with the previously reported. ¹⁷

5-Methoxy-2-[5-(trifluoromethyl)-1*H***-pyrazol-3-yl]-phenol (10b)** was synthesized from 4,4,4-trifluoro-1-(2-hydroxy-4-methoxyphenyl)butane-1,3-dione (**9b**) (500 mg) (synthesized analogously to literature-described procedure¹⁸ from 1-(2-hydroxy-4-methoxyphenyl)ethan-1-one). Yield 397 mg (80%), white solid, mp 149–151°C. ¹H NMR spectrum (400 MHz, DMSO- d_6), δ, ppm (*J*, Hz): 7.59 (1H, d, J = 8.5, H-3); 6.97 (1H, s, H-6); 6.61–6.44 (2H, m, H-4 pyrazol, H-5); 3.74 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ, ppm (*J*, Hz): 160.6; 155.6; 141.4; 140.9 (q, J = 39.0); 128.6; 122.1 (q, J = 268.0); 108.1; 105.5; 101.6; 101.4; 55.1. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: –60.28. Found, m/z: 259.0698 [M+H]⁺. C₁₁H₁₀F₃N₂O₂. Calculated, m/z: 259.0694.

2-[5-(Trifluoromethyl)-1*H***-pyrazol-3-yl]phenol (10c)** was synthesized from 4,4,4-trifluoro-1-(2-hydroxyphenyl)-

butane-1,3-dione (**9c**) (390 mg) according to the literaturedescribed procedure. ¹⁹ Yield 271 mg (71%), white-beige solid. Spectral data was in accordance with the previously reported. ²⁰

4-[5-(Trifluoromethyl)-1*H*-pyrazol-3-yl]benzene-**1,3-diol (10d)**. Boron tribromide (1 M in CH₂Cl₂, 3.65 ml, 3.65 mmol) was added dropwise to an ice-cooled solution of compound **10b** (236 mg, 0.91 mmol) in CH₂Cl₂ (5 ml). After stirring for 2 h at 25°C, the reaction mixture was poured into ice water, and neutralized with addition of saturated aqueous NaHCO₃, followed by extraction with CH₂Cl₂ (2×15 ml). Organic extracts were combined, dried over Na₂SO₄, and evaporated in vacuo. Pure product was obtained by trituration with EtOAc - petroleum ether. Yield 31 mg (14%), white solid, mp >220°C (decomp.). ¹H NMR spectrum (400 MHz, DMSO- d_6), δ , ppm (J, Hz): 13.32 (1H, br. s, NH); 10.22 (1H, br. s, OH); 9.65 (1H, br. s, OH); 7.45 (1H, d, J = 8.5, H-5); 6.90 (1H, s, H-4 pyrazol); 6.44 (1H, d, J = 2.2, H-6); 6.31 (1H, dd, J = 8.5, J = 2.3, H-2). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm: 159.0; 155.6; 141.7; 128.6; 107.2; 106.6; 102.9; 100.9 (CF₃ and C-CF₃ carbons can not be observed due to proton exchange process). 19 F NMR spectrum (376 MHz, DMSO- d_6), δ , ppm: -60.24. Found, m/z: 245.0550 [M+H]⁺. $C_{10}H_8F_3N_2O_2$. Calculated, *m/z*: 245.0538.

4-(4-Chlorophenyl)-3-phenyl-5-(trifluoromethyl)-1H**pyrazole** (12). 4-Bromo-3-phenyl-5-(trifluoromethyl)-1*H*pyrazole 11 (405 mg, 1.39 mmol) (synthesized by literaturedescribed procedure²¹) was dissolved in dioxane-H₂O, 4:1 mixture (20 ml) under argon, followed by addition of 4-chlorophenylboronic acid (282 mg, 1.8 mmol), caesium carbonate (1.18 g, 3.62 mmol), and PdCl₂(PPh₃)₂ (59 mg, 0.08 mmol). Resulting mixture was heated for 14 h at 100°C. After cooling to room temperature, the reaction mixture was diluted with EtOAc (60 ml), washed with brine (3×60 ml), dried over anhydrous Na₂SO₄, and evaporated in vacuo. Crude product was purified by silica gel column chromatography, using gradient from 5 to 20% EtOAc in petroleum ether to obtain product. Yield 118 mg (26%), white solid, mp 187-190°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ, ppm (*J*, Hz): 11.02 (1H, br. s, NH); 7.45–7.37 (4H, m, H Ar); 7.35–7.21 (5H, m, H Ar). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm (J, Hz): 142.0; 139.6 (q, J = 35.0); 132.9; 132.1; 129.8; 129.2; 129.0; 128.8; 127.9; 125.7; 121.9 (q, J = 269.4); 116.3. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ , ppm: -58.19. Found, m/z: 323.0571 [M+H]^+ . $C_{16}H_{11}ClF_3N_2$. Calculated, m/z: 323.0563.

Synthesis of isoxazoles 13a–e (General method). Hydroxylamine hydrochloride (42 mg, 0.6 mmol) was added to a stirred solution of compound 6c, 7e,f,i,j (0.3 mmol) in pyridine (2 ml). The resulting solution was stirred for 14 h at 90°C. After cooling to room temperature, the reaction mixture was poured into H₂O (70 ml) which was acidified with 1 M HCl. Product further was extracted with EtOAc (40 ml), organic layer was washed with brine (3×30 ml), dried over anhydrous Na₂SO₄, and evaporated *in vacuo*. Obtained isoxazoles 13a,b,d,e were analytically pure and no further purification needed. Isoxazole 13c was purified by column chromatography on silica gel, using 5 to 30% gradient of EtOAc in petroleum ether.

4-[4-(4-Chlorophenyl)-3-(trifluoromethyl)isoxazol-5-yl]-benzene-1,3-diol (13a) was synthesized from compound **6c** (130 mg). Yield 115 mg (85%), slightly gray solid, mp 165–168°C. 1 H NMR spectrum (400 MHz, DMSO- d_6), δ, ppm (J, Hz): 9.95 (2H, d, J = 3.7, 2OH); 7.51–7.45 (2H, m, C₆H₄Cl); 7.30–7.25 (2H, m, C₆H₄Cl); 7.09 (1H, d, J = 8.5, H-5); 6.33 (1H, d, J = 2.2, H-2); 6.28 (1H, dd, J = 8.5, J = 2.3, H-6). 13 C NMR spectrum (101 MHz, DMSO- d_6), δ, ppm (J, Hz): 169.3; 161.4; 157.1; 152.6 (d, J = 35.2); 133.3; 131.6; 131.1; 128.7; 127.0; 120.0 (q, J = 271.8); 113.4; 107.4; 103.6; 102.8. 19 F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: –60.58. Found, m/z: 356.0302 [M+H]⁺. C_{16} H₁₀ClF₃NO₃. Calculated, m/z: 356.0301.

5-(Benzyloxy)-2-[4-(4-chlorophenyl)-3-(trifluoromethyl)isoxazol-5-yl]phenol (13b) was synthesized from compound **7d** (130 mg). Yield 129 mg (96%), white solid, mp 141–143°C. ¹H NMR spectrum (600 MHz, DMSO- d_6), δ, ppm (J, Hz): 10.26 (1H, s, OH); 8.64 (1H, s, H Ar); 7.54–7.27 (9H, m, C₆H₄Cl, OCH₂C₆H₅); 6.62 (1H, dd, J = 8.6, J = 2.2, H-6); 6.55 (1H, d, J = 2.1, H-4); 5.12 (2H, s, OCH₂C₆H₅). ¹³C NMR spectrum (151 MHz, DMSO- d_6), δ, ppm (J, Hz): 168.8; 161.9; 157.1; 152.7 (q, J = 35.2); 149.4; 136.5; 136.4; 133.4; 131.7; 131.2; 128.7; 128.5; 128.0; 127.8; 126.7; 124.0; 119.9 (q, J = 271.8); 114.0; 106.4; 105.4; 102.3; 69.3. ¹⁹F NMR spectrum (376 MHz, DMSO- d_6), δ, ppm: -60.58. Found, m/z: 446.0767 [M+H][†]. C₂₃H₁₆ClF₃NO₃. Calculated, m/z: 446.0771.

2-[4-(4-Chlorophenyl)-3-(trifluoromethyl)isoxazol-5-yl]-5-(methoxymethoxy)phenol (13c) was synthesized from compound **7f** (190 mg). Yield 124 mg (63%), white solid, mp 119–121°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (J, Hz): 7.40 (2H, d, J = 8.6, C₆H₄Cl); 7.25 (2H, d, J = 8.5, C₆H₄Cl); 7.00 (1H, d, J = 8.8, H-3); 6.65 (1H, d, J = 2.4, H-6); 6.53 (1H, dd, J = 8.8, J = 2.4, H-4); 6.23 (1H, br. s, OH); 5.16 (2H, s, OCH₂OCH₃); 3.47 (3H, s, OCH₂OCH₃). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (J, Hz): 167.5; 161.1; 155.8; 154.6 (q, J = 36.4); 135.4; 131.3; 130.5; 129.6; 126.0; 119.9 (q, J = 272.4); 113.4; 109.7; 106.4; 105.1; 94.3; 56.5. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: –61.51. Found, m/z: 400.0550 [M+H]⁺. C₁₈H₁₄ClF₃NO₄. Calculated, m/z: 400.0563.

2-[4-(4-Chlorophenyl)-3-(trifluoromethyl)isoxazol-5-yl]- 5-isopropoxyphenol (13d) was synthesized from compound **7i** (124 mg). Yield 125 mg (97%), white solid, mp 111–113°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (J, Hz): 7.40 (2H, d, J = 8.6, C₆H₄Cl); 7.25 (2H, d, J = 8.4, C₆H₄Cl); 6.95 (1H, d, J = 8.8, H-3); 6.46 (1H, d, J = 2.4, H-6); 6.36 (1H, dd, J = 8.8, J = 2.4, H-4); 6.31 (1H, br. s, OH); 4.53 (1H, septet, J = 6.0, OCH(CH₃)₂); 1.33 (6H, d, J = 6.1, OCH(CH₃)₂). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (J, Hz): 167.8; 162.0; 156.1; 154.6 (q, J = 36.2); 135.4; 131.3; 130.4; 129.6; 126.1; 119.9 (q, J = 272.3); 112.3; 109.4; 104.9; 104.0; 70.4; 22.1. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -61.54. Found, m/z: 398.0768 [M+H]⁺, C₁₉H₁₆ClF₃NO₃, Calculated, m/z: 398.0771.

2-[4-(4-Chlorophenyl)-3-(trifluoromethyl)isoxazol-5-yl]-5-(isopentyloxy)phenol (13e) was synthesized from compound **7j** (140 mg). Yield 132 mg (91%), white-yellow solid, mp 112–114°C. ¹H NMR spectrum (600 MHz,

CDCl₃), δ , ppm (*J*, Hz): 7.40 (2H, d, J = 8.5, C₆H₄Cl); 7.25 (2H, d, J = 8.6, C₆H₄Cl); 6.96 (1H, d, J = 8.8, H-3); 6.48 (1H, d, J = 2.4, H-6); 6.39 (1H, dd, J = 8.8, J = 2.4, H-4); 6.32 (1H, br. s, OH); 3.97 (2H, t, J = 6.7, OCH₂CH₂CH₂CH(CH₃)₂); 1.80 (1H, dsept, J = 13.4, J = 6.7, OCH₂CH₂CH₂CH(CH₃)₂); 1.66 (2H, q, J = 6.7, OCH₂CH₂CH(CH₃)₂); 0.95 (6H, d, J = 6.7, OCH₂CH₂CH(CH₃)₂); 1.67 (NMR spectrum (151 MHz, CDCl₃), δ , ppm (*J*, Hz): 167.8; 163.1; 156.0; 154.6 (q, J = 36.1); 135.4; 131.3; 130.3; 129.6; 126.1; 119.9 (q, J = 272.4); 112.9; 108.7; 105.0; 103.1; 66.9; 37.8; 25.2; 22.7. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -61.54. Found, m/z: 426.1085 [M+H]⁺. C₂₁H₂₀ClF₃NO₃. Calculated. m/z: 426.1084.

4-(4-Chlorophenyl)-5-[2-methoxy-4-(methoxymethoxy)phenyl]-3-(trifluoromethyl)isoxazole (14). Methyl iodide (53 µl, 0.85 mmol) was added to a stirred solution of compound 13c (114 mg, 0.285 mmol) with K₂CO₃ (98 mg, 0.71 mmol) in DMF (2 ml) at room temperature. The reaction mixture was stirred for 14 h and then partitioned between EtOAc (20 ml) and brine (10 ml). Organic layer was separated, washed with brine (2×20 ml), dried over anhydrous Na₂SO₄ to obtain pure product. Yield 117 mg (99%), colorless oil. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (*J*, Hz): 7.36 (1H, d, J = 8.6, H-6); 7.32 $(2H, d, J = 8.6, C_6H_4C1); 7.17 (2H, d, J = 8.4, C_6H_4C1);$ 6.69 (1H, dd, J = 8.6, J = 2.2, H-5); 6.53 (1H, d, J = 2.2, H-3); 5.20 (2H, s, OCH₂OCH₃); 3.49 (3H, s, OCH₂OCH₃); 3.40 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (*J*, Hz): 167.7; 161.3; 158.1; 153.8 (q, J = 36.1); 134.4; 131.8; 130.6; 128.8; 127.6; 120.2 (q, J = 272.2); 114.6; 109.1; 108.1; 100.5; 94.5; 56.5; 55.1. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -61.32 Found, m/z: 414.0734 [M+H]⁺. C₁₉H₁₆ClF₃NO₄. Calculated, *m/z*: 414.0720.

4-[4-(4-Chlorophenyl)-3-(trifluoromethyl)-isoxazol-5-yl]-**3-methoxyphenol (15)**. 12 M HCl (30 μ l, 0.97 mmol) was added to a stirred solution of isoxazole 14 (88 mg, 0.21 mmol) in MeOH (3 ml), and the reaction mixture was heated at 60°C for 3 h. After cooling to room temperature, reaction mixture was evaporated to dryness, partitioned between EtOAc (10 ml) and brine (10 ml). Organic layer was separated, dried over anhydrous Na₂SO₄, and evaporated in vacuo to obtain pure product. Yield 76 mg (97%), whiteyellow solid, mp 148–150°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (*J*, Hz): 7.35–7.28 (3H, m, H Ar); 7.16 $(2H, d, J = 8.4, C_6H_4C1)$; 6.47 (1H, dd, J = 8.4, J = 2.3) H-6); 6.37 (1H, d, J = 2.3, H-2); 5.38 (1H, br. s, OH); 3.39 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, CDCl₃), δ , ppm (J, Hz): 167.8; 159.8; 158.4; 153.8 (q, J = 36.1); 134.4; 132.1; 130.6; 128.8; 127.6; 120.1 (q, J = 272.1); 114.5; 108.2; 108.0; 99.7; 55.1. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -61.32. Found, m/z: 370.0457 [M+H]⁺. C₁₇H₁₂ClF₃NO₃. Calculated, *m/z*: 370.0458.

3-(4-Chlorophenyl)-4-oxo-2-(trifluoromethyl)-4*H***-chromen-7-yl trifluoromethanesulfonate (16)**. Trifluoromethanesulfonic anhydride (530 μl, 3.22 mmol) was added dropwise under ice cooling to a stirred solution of compound **6c** (1 g, 2.93 mmol) and Et₃N (818 μl, 5.87 mmol) in dry CH₂Cl₂ (40 ml). After 1 h of stirring at room temperature, reaction mixture was poured into H₂O, oganic layer

was washed with brine (20 ml), separated, dried over anhydrous Na₂SO₄, and evaporated *in vacuo*. Product was purified by silica gel column chromatography using 5% EtOAc in petroleum ether. Yield 1.32 g (95%), lightyellow oil. ¹H NMR spectrum (400 MHz, CDCl₃), δ, ppm (J, Hz): 8.35 (1H, d, J = 8.9, H-5); 7.58 (1H, d, J = 2.3, H-8); 7.51–7.35 (3H, m, H Ar); 7.20 (2H, d, J = 8.4, C₆H₄Cl). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (J, Hz): 175.6; 155.4; 153.2; 149.4 (q, J = 36.5); 135.7; 131.2; 129.4; 128.9; 126.7; 125.4; 122.9; 120.5; 120.4; 120.1; 117.7; 117.2; 112.0. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ, ppm: -63.60; -72.50. Found, m/z: 472.9687 [M+H]⁺. C₁₇H₈ClF₆O₅S. Calculated, m/z: 472.9685.

3-(4-Chlorophenyl)-2-(trifluoromethyl)-4H-chromen-**4-one (17)**. Triethylsilane (1.12 ml, 7 mmol) and PdCl₂(PPh₃)₂ (98 mg, 0.14 mmol) were added to a stirred solution of compound 16 (1.32 g, 2.8 mmol) in DMF (15 ml) at room temperature. The reaction mixture was allowed to heat at 60°C for 1 h. After cooling to room temperature, reaction mixture was partitioned between EtOAc (30 ml) and brine (20 ml). Organic layer was separated, washed with brine (2×20 ml), dried over anhydrous Na₂SO₄, and evaporated in vacuo. Crude residue was crystallized from petroleum ether to obtain pure product. Yield 530 mg (58%), yellow solid, mp 107–109°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (J, Hz): 8.24 (1H, dd, J = 8.0, J = 1.6, H-5); 7.80 (1H, ddd, J = 8.6, J = 7.2, J = 1.7, H-7); 7.59 (1H, d, J = 8.5, d)H-8); 7.50 (1H, t, J = 7.6, H-6); 7.44 (2H, d, J = 8.4, C_6H_4Cl); 7.21 (2H, d, J = 8.4, C_6H_4Cl). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (*J*, Hz): 176.8; 155.2; 148.8 (q, J = 36.3); 135.3; 135.2; 131.4; 128.7; 127.6; 126.6; 126.5; 124.6; 123.3; 119.4 (q, J = 277.0); 118.5. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -63.60. Found, m/z: 325.0249 [M+H]^+ . C₁₆H₉ClF₃O₂. Calculated, m/z: 325.0243.

2-[4-(4-Chlorophenyl)-3-(trifluoromethyl)isoxazol-5-vllphenol (18). Hydroxylamine hydrochloride (227 mg, 3.26 mmol) was added to a stirred solution of compound 17 (530 mg, 1.63 mmol) in pyridine (5 ml). The resulting solution was stirred for 14 h at 90°C. After cooling to room temperature, the reaction mixture was poured into H₂O (100 ml) which was acidified with 1 M HCl. Product further was extracted with EtOAc (60 ml), organic layer was washed with brine (3×30 ml), dried over anhydrous Na₂SO₄, and evaporated in vacuo. Yield 550 mg (99%), pink solid, mp 147-150°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ , ppm (*J*, Hz): 7.40–7.33 (3H, m, C₆H₄Cl, H Ar); 7.27–7.22 (2H, m, C_6H_4Cl); 7.14 (1H, dd, J = 7.9, J = 1.7, H-3); 6.96 (1H, dd, J = 8.3, J = 1.1, H-4); 6.88 (1H, ddd, J = 7.8, J = 7.3, J = 1.1, H-6); 6.04 (1H, br. s, OH). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (*J*, Hz): 167.4; 154.5 (q, J = 36.4); 154.2; 135.5; 133.1; 131.1; 129.8; 129.5; 125.9; 121.1; 119.9 (q, J = 272.4); 117.9; 114.7; 112.7. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ , ppm: -61.42. Found, m/z: 340.0341 [M+H]^+ . C₁₆H₁₀ClF₃NO₂. Calculated, m/z: 340.0352.

2-[4-(4-Chlorophenyl)-3-(trifluoromethyl)isoxazol-5-yl]-phenyl trifluoromethanesulfonate (19). Trifluoromethanesulfonic anhydride (200 μ l, 1.22 mmol) was added dropwise under ice cooling to a stirred solution of compound 18 (295 mg, 0.87 mmol) and triethylamine (242 μ l, 1.74 mmol) in

dry CH₂Cl₂ (10 ml). After 1 h of stirring at room temperature, the reaction mixture was poured into H₂O, oganic layer was washed with brine (15 ml), separated, dried over anhydrous Na₂SO₄, and evaporated *in vacuo*. Product was purified by silica gel column chromatography, using 5% EtOAc in petroleum ether. Yield 399 mg (97%), lightyellow oil. ¹H NMR spectrum (400 MHz, CDCl₃), δ, ppm (*J*, Hz): 7.60 (1H, ddd, J = 8.3, J = 6.8, J = 2.5, H-6); 7.44–7.33 (5H, m, C₆H₄Cl, H Ar); 7.18 (2H, d, J = 8.4, C₆H₄Cl). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (*J*, Hz): 164.3; 154.2 (q, J = 36.7); 146.6; 135.7; 133.2; 131.8; 130.9; 129.5; 129.0; 125.0; 123.2; 120.5; 118.6 (q, J = 320.9); 119.8 (q, J = 272.4); 117.1. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ, ppm: -61.36, -73.40. Found, m/z: 471.9847 [M+H]⁺. C₁₇H₉CIF₆NO₄S. Calculated, m/z: 471.9845.

4-(4-Chlorophenyl)-5-phenyl-3-(trifluoromethyl)isoxazole (20). Triethylsilane (772 µl, 4.83 mmol) and PdCl₂(PPh₃)₂ (28 mg, 0.04 mmol) were added to a stirred solution of compound 19 (380 mg, 0.8 mmol) in DMF (7 ml) at room temperature. Reaction mixture was allowed to heat at 60°C for 1 h. After cooling to room temperature, the reaction mixture was partitioned between EtOAc (20 ml) and brine (20 ml). Organic layer was separated, washed with brine (2×10 ml), dried over anhydrous Na₂SO₄, and evaporated in vacuo. Product was purified by silica gel column chromatography, using gradient from 1 to 5% EtOAc in petroleum ether. Yield 190 mg (73%), white solid, mp 88–90°C. ¹H NMR spectrum (400 MHz, CDCl₃), δ, ppm (J, Hz): 7.54–7.47 (2H, m, H-2,6); 7.47–7.41 (3H, m, H Ar); 7.41-7.34 (2H, m, H Ar); 7.29 (2H, d, J = 8.4, C₆H₄Cl). ¹³C NMR spectrum (101 MHz, CDCl₃), δ, ppm (J, Hz): 168.2; 154.8 (q, J = 36.1); 135.5; 131.6; 131.1; 129.6; 129.2; 127.2; 126.3; 126.1; 119.9 (q, J = 272.2); 113.2. ¹⁹F NMR spectrum (376 MHz, CDCl₃), δ, ppm: -61.64. Found, m/z: 324.0398 [M+H]⁺. C₁₆H₁₀ClF₃NO. Calculated, *m/z*: 324.0403.

5-Phenyl-3-(trifluoromethyl)isoxazole (21a). Hydroxylamine hydrochloride (1.74 g, 25 mmol) was added to a stirred solution of 4,4,4-trifluoro-1-phenyl-1,3-butanedione (9a) (1.8 g, 8.33 mmol) in EtOH (50 ml) at room temperature. The reaction mixture was allowed to heat at 80°C for 3 h. After cooling to room temperature, the reaction mixture was partitioned between CH₂Cl₂ (50 ml) and 1 M HCl (20 ml). Organic layer was separated, washed with brine (2×20 ml), dried over anhydrous Na₂SO₄, and evaporated in vacuo. Obtained residue then was dissolved in glacial AcOH (25 ml), and 98% H₂SO₄ (2 ml, 37.5 mmol) was added dropwise. The resulting mixture was heated at 100°C for 1 h and cooled down to room temperature followed by evaporation in vacuo. After trituration of crude residue with cold H₂O, solid precipitated from the solution. This solid was filtered off, the filter cake was washed several times with distilled H₂O and dried in air. Yield 1.125 g (63%), beige crystals. Spectral data was in accordance with those previously reported.²²

5-Methoxy-2-[3-(trifluoromethyl)isoxazol-5-yl]phenol (21b). Hydroxylamine hydrochloride (341 mg, 4.91 mmol) was added to a stirred solution of 4,4,4-trifluoro-1-(2-hydroxy-4-methoxyphenyl)butane-1,3-dione **(9b)** (600 mg, 2.3 mmol)

in pyridine (7 ml) at room temperature. The reaction mixture was allowed to heat at 100°C for 3 h. After cooling to room temperature, reaction mixture was partitioned between EtOAc (30 ml) and 1 M HCl (30 ml). Organic layer was separated, washed with brine (2×30 ml), dried over anhydrous Na₂SO₄, and evaporated in vacuo. Obtained residue then was dissolved in glacial AcOH (10 ml), and 98% H_2SO_4 (500 µl, 9.38 mmol) was added dropwise. The resulting mixture was heated at 100°C for 1 h and cooled down to room temperature followed by evaporation in vacuo. After trituration of crude residue with cold H₂O, solid precipitated from the solution. This solid was filtered off, filter cake was washed several times with distilled H₂O and dried in air. Crude compound was purified by trituration with hot EtOAc – petroleum ether 1:3 mixture to obtain pure product. Yield 242 mg (38%), beige crystals, mp >150°C (decomp.). ¹H NMR spectrum (400 MHz, DMSO- d_6), δ , ppm (J, Hz): 10.98 (1H, s, OH); 7.76 (1H, d, J = 9.5, H-3); 7.12 (1H, s, H-4 isoxazole); 6.63–6.58 (2H, m, H-4,6); 3.78 (3H, s, OCH₃). ¹³C NMR spectrum (101 MHz, DMSO- d_6), δ , ppm (J, Hz): 169.5; 162.8; 156.9; 155.0 (q, J = 37.0); 128.2; 120.0 (q, J = 270.9); 106.4; 105.8; 101.4; 98.0; 55.3. ¹⁹F NMR spectrum (376 MHz, DMSO-*d*₆), δ, ppm: -62.25. Found, m/z: 260.0530 [M+H]⁺. $C_{11}H_9F_3NO_3$. Calculated, m/z: 260.0535.

MIC assay. All of the compounds for the MIC tests had the purity level not lower than 95%. Staphylococcus aureus strain Newman was cultured overnight at 37°C in Mueller Hinton broth (MHB) (Oxoid). The MIC was determined using the microdilution method according to guidelines of the Clinical Laboratory Standards Institute. In a 96-well plate, a series of twofold dilutions of each compound were added to a 1:100 dilution of an overnight culture of S. aureus in a final volume of 100 µl and incubated overnight at 37°C. The final concentration of the compounds was in a range 50–0.39 μg/ml. MIC was determined as the lowest concentration where no growth was detected by measurement of optical density at 600 nm (OD600). Compounds that did not inhibit growth were retested at higher concentrations (250–1.95 µg/ml). Wells containing bacteria with or without 1% DMSO and medium alone were included as controls in every plate.

Supplementary information file containing ¹H, ¹³C, ¹⁹F NMR spectra and HRMS data of all synthesized compounds is available at the journal website http://link.springer.com/journal/10593.

This project has received funding from the European Union's Horizon 2020 research and innovation programme under the Marie Sklodowska-Curie grant agreement No. 765147.

References

 Murray, C. J. L.; Ikuta, K. S.; Sharara, F.; Swetschinski, L.; Robles Aguilar, G.; Gray, A.; Han, C.; Bisignano, C.; Rao, P.; Wool, E.; Johnson, S. C.; Browne, A. J.; Chipeta, M. G.; Fell, F.; Hackett, S.; Haines-Woodhouse, G.; Kashef Hamadani, B. H.; Kumaran, E. A. P.; McManigal, B.; Agarwal, R.; Akech, S.; Albertson, S.; Amuasi, J.; Andrews, J.; Aravkin, A.; Ashley, E.;

- Bailey, F.; Baker, S.; Basnyat, B.; Bekker, A.; Bender, R.; Bethou, A.; Bielicki, J.; Boonkasidecha, S.; Bukosia, J.; Carvalheiro, C.; Castañeda-Orjuela, C.; Chansamouth, V.; Chaurasia, S.; Chiurchiù, S.; Chowdhury, F.; Cook, A. J.; Cooper, B.; Cressey, T. R.; Criollo-Mora, E.; Cunningham, M.; Darboe, S.; Day, N. P. J.; De Luca, M.; Dokova, K.; Dramowski, A.; Dunachie, S. J.; Eckmanns, T.; Eibach, D.; Emami, A.; Feasey, N.; Fisher-Pearson, N.; Forrest, K.; Garrett, D.; Gastmeier, P.; Giref, A. Z.; Greer, R. C.; Gupta, V.; Haller, S.; Haselbeck, A.; Hay, S. I.; Holm, M.; Hopkins, S.; Iregbu, K. C.; Jacobs, J.; Jarovsky, D.; Javanmardi, F.; Khorana, M.; Kissoon, N.; Kobeissi, E.; Kostyanev, T.; Krapp, F.; Krumkamp, R.; Kumar, A.; Kyu, H. H.; Lim, C.; Limmathurotsakul, D.; Loftus, M. J.; Lunn, M.; Ma, J.; Mturi, N.; Munera-Huertas, T.; Musicha, P.; Mussi-Pinhata, M. M.; Nakamura, T.; Nanavati, R.; Nangia, S.; Newton, P.; Ngoun, C.; Novotney, A.; Nwakanma, D.; Obiero, C. W.; Olivas-Martinez, A.; Olliaro, P.; Ooko, E.; Ortiz-Brizuela, E.; Peleg, A. Y.; Perrone, C.; Plakkal, N.; Ponce-de-Leon, A.; Raad, M.; Ramdin, T.; Riddell, A.; Roberts, T.; Robotham, J. V.; Roca, A.; Rudd, K. E.; Russell, N.; Schnall, J.; Scott, J. A. G.; Shivamallappa, M.; Sifuentes-Osornio, J.; Steenkeste, N.; Stewardson, A. J.; Stoeva, T.; Tasak, N.; Thaiprakong, A.; Thwaites, G.; Turner, C.; Turner, P.; van Doorn, H. R.; Velaphi, S.; Vongpradith, A.; Vu, H.; Walsh, T.; Waner, S.; Wangrangsimakul, T.; Wozniak, T.; Zheng, P.; Sartorius, B.; Lopez, A. D.; Stergachis, A.; Moore, C.; Dolecek, C.; Naghavi, M. Lancet 2022, 399(10325), 629.
- Santajit, S.; Indrawattana, N. Biomed. Res. Int. 2016, 2475067.
- 3. Vo, C. D.; Shebert, H. L.; Zikovich, S.; Dryer, R. A.; Huang, T. P.; Moran, L. J.; Cho, J.; Wassarman, D. R.; Falahee, B. E.; Young, P. D.; Gu, G. H.; Heinl, J. F.; Hammond, J. W.; Jackvony, T. N.; Frederick, T. E.; Blair, J. A. *Bioorg. Med. Chem. Lett.* **2017**, *27*, 5235.
- (a) Kupchevskaya, I. P.; Khilya, V. P. Dopov. Akad. Nauk Ukr. RSR, Ser. B: Geol., Khim. Biol. Nauki 1978, 3, 234.
 (b) Drysdale, M. J.; Dymock, B. W.; Barril-Alonso, X.; Workman, P.; Pearl, L. H.; Prodromou, C.; MacDonald, E. WO Patent 2003055860, 2003.
- (a) Pivovarenko, V. G.; Khilya, V. P.; Vasil'ev, S. A. *Chem. Nat. Compd.* 1989, 25, 542. (b) Schiltz, G. E.; Mishra, R. K.; Han, H.; Abdulkadir, S. A.; Izquierdo-Ferrer, J.; Jain, A. D. WO Patent 2020046382, 2020.

- (a) Moskvina, V. S.; Shilin, S. V.; Khilya, V. P. Chem. Heterocycl. Compd. 2015, 51, 799.
 (b) Drysdale, M. J.; Dymock, B. W.; Finch, H.; Webb, P.; McDonald, E.; James, K. E.; Cheung, K. M.; Mathews, T. P. WO Patent 2004072051, 2004.
- Cockerill, F. R. C. Laboratory Standards, I. Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria that Grow Aerobically: Approved Standard; Clinical and Laboratory Standards Institute: Wayne, 2012.
- Kushner, P. J.; Myles, D. C.; Harmon, C. L.; Hodges Gallagher, L. C. US Patent 20160311805A1, 2016.
- 9. Balasubramanian, S.; Nair, M. G. Synth. Commun. 2000, 30, 469.
- Yeap, G.-Y.; Yam, W.-S.; Takeuchi, D.; Osakada, K.; Gorecka, E.; Mahmood, W. A. K.; Boey, P.-L.; Hamid, S. A. *Liq. Cryst.* 2008, 35, 315.
- Ng, L.-T.; Ko, H.-H.; Lu, T.-M. Bioorg. Med. Chem. 2009, 17, 4360.
- Garazd, M. M.; Frasinyuk, M. S. Chem. Nat. Compd. 2019, 55, 813.
- Frasinyuk, M. S.; Bondarenko, S. P.; Khilya, V. P.; Liu, C.; Watt, D. S.; Sviripa, V. M. Org. Biomol. Chem. 2015, 13, 1053
- Wu, E. S. C.; Loch, J. T.; Toder, B. H.; Borrelli, A. R.; Gawlak, D.; Radov, L. A.; Gensmantel, N. P. *J. Med. Chem.* 1992, 35, 3519.
- Semeniuchenko, V.; Exner, T. E.; Khilya, V.; Groth, U. Appl. Organomet. Chem. 2011, 25, 804.
- Tang, B.; Frasinyuk, M. S.; Chikwana, V. M.; Mahalingan, K. K.; Morgan, C. A.; Segvich, D. M.; Bondarenko, S. P.; Mrug, G. P.; Wyrebek, P.; Watt, D. S.; DePaoli-Roach, A. A.; Roach, P. J.; Hurley, T. D. J. Med. Chem. 2020, 63, 3538.
- Wang, Y.; Han, J.; Chen, J.; Cao, W. Tetrahedron 2015, 71, 8256
- Cotman, A. E.; Cahard, D.; Mohar, B. Angew. Chem., Int. Ed. 2016, 55, 5294.
- Liu, C.; Cui, Z.; Yan, X.; Qi, Z.; Ji, M.; Li, X. Molecules 2016, 21(7), 828.
- Sapegin, A. V.; Kalinin, S. A.; Smirnov, A. V.; Dorogov, M. V.; Krasavin, M. *Tetrahedron* 2014, 70, 1077.
- Jeon, S. L.; Choi, J. H.; Kim, B. T.; Jeong, I. H. J. Fluorine Chem. 2007, 128, 1191.
- Poh, J.-S.; Garcia-Ruiz, C.; Zuniga, A.; Meroni, F.;
 Blakemore, D. C.; Browne, D. L.; Ley, S. V. Org. Biomol. Chem. 2016, 14, 5983.