



Inventory of potential new anti-fouling strategies inspired by nature

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Summary

As part of the Dutch Top consortia for Knowledge and Innovation (TKI) project 'Re-Drag with Nature' an inventory was made of the strategies that sea weeds apply to reduce the settlement and development of fouling organisms on their surface. This inventory was made based upon available literature and the results were assessed for their potency to protect man-made structures against biofouling. The reason for the focus on seaweeds was initiated by the observations that fouling pressure on seaweeds is in general low, and further supported by the knowledge that a lot of effort is currently being made to cultivate seaweeds and to utilise the compounds that can be extracted from it. However, the inventory was not fully limited to seaweeds. Where it inspired the thinking of new alternatives these were also briefly assessed.

Seaweeds make use of physical and chemical strategies to keep their surface free of biofouling organisms. The physical strategies are based on the continuous renewal of surface material and are, therefore, not considered applicable to protect man-made structures without continuous maintenance. Self-polishing coatings are available for ship hulls, but these are only effective in combination with biocides.

An alternative physical process that is not applied by seaweeds, but that may have the potency to reduce fouling is a combination of flow velocity and surface structure that could prevent early stages of fouling organisms from attaching to a surface. As this requires a controlled environment, it is not applicable to ship hulls or maritime infrastructure. It could work in industrial cooling water systems, but as the current methods that are applied there seem to fulfil the needs of the industry, this option was not further investigated.

The chemical anti-fouling strategies of seaweeds are based on the production of a broad range of metabolites with the potential to reduce settlement and/or development of fouling organisms. Although the effectiveness of several of these metabolites has been shown on experimental scale, commercial application of bio-based compounds for anti-fouling purposes does not seem attractive, due to the high costs that are related with legislation/registration procedures for new chemicals.

Apart from producing anti-fouling metabolites themselves, seaweeds can also support specific microbes that have little or no negative impact on their condition, but that prevent the settlement of more harmful species. A similar strategy, where a fouling species with low negative impact is favoured and supported, can have potential to manage fouling in an environmental friendly way on man-made surfaces where some drag is acceptable. As far as we are aware this is a completely new approach to deal with biofouling. When writing this report a proposal for a joint industry project (JIP) is being prepared that aims to investigate the potency of this approach to manage bio-fouling on offshore infrastructure.

1 Introduction

1.1 Re-Drag with Nature

Re-Drag with Nature is a project initiated by MARIN and Wageningen Marine Research (WMR) and in part financed by the Dutch TKI-program. Endures and Wagenborg Shipping are involved as external partners. The overall aim of the project was to investigate whether it is possible to prevent fouling in an environmental friendly way and to predict the effect of fouling on the ship resistance and flow. This serves two overall goals: the development of a prediction model that enables the ship owner to find the optimal balance between fuel costs/emission and maintenance/cleaning costs. Secondly, the development of a new coating (principle) that prevents fouling in an environmental friendly manner.

The specific topics of the project were:

1. To investigate the influence of the structure of biofilms on resistance and the way to calculate/determine this.
2. To investigate the relation between local and climatological circumstances that influence the biofilm development on ship hulls and the effect on the fuel consumption.
3. To investigate whether the mechanisms of seaweed to protect itself from fouling can be used on ship hulls.
4. To set up a JIP and/or NWO-research programme to elaborate on the abovementioned three topics and come to significant steps in the maritime sector.

This report describes the outcome of topic 3 and indicates the related progress on topic 4. The reason for the focus of topic 3 on seaweeds was initiated by the observations that fouling pressure on seaweeds is in general low, and further supported by the knowledge that a lot of effort is currently being made to cultivate seaweed and to utilise the compounds that can be extracted from it.

1.2 Biofouling

Every hard substrate submerged in water will become subjected to biofouling. Biofouling is the consecutive accumulation of organic chemical compounds and bacteria which make up the primary film, primary colonizers (bacteria and diatoms), secondary colonizers (spores of seaweeds and protozoa) and tertiary colonizers (invertebrates like barnacles, mussels and tunicates). The primary and secondary colonizers roughly fall in the process called microfouling and the tertiary colonizers in the process called macrofouling, although micro- and macrofouling overlap slightly. See *Figure 1* for a general overview of the biofouling process (Abarzua & Jakubowski, 1995; Armstrong *et al.*, 2000; Wahl, 1989).

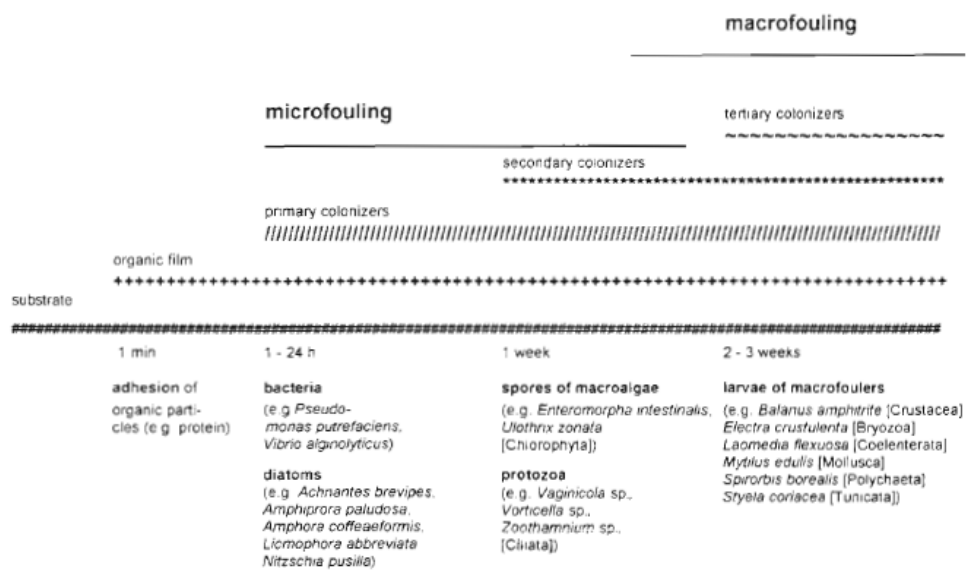


Figure 1 Successive accumulation during biofouling, from Abarzua & Jakubowski, 1995.

Biofouling in the marine environment can have considerable economic and related environmental consequences. In the shipping industry, biofouling on the hull of a ship causes an increase in drag that results in an increase in fuel consumption and CO₂ emission (Abarzua & Jakubowski, 1995; Schultz *et al.*, 2011). Furthermore, biofouling also can cause microbial induced corrosion of the ship's hull, increasing maintenance costs (Abarzua & Jakubowski, 1995).

Apart from these consequences for the ship owners, vessels can also transfer unwanted (invasive) fouling organisms to other regions, which can have economic impact on a wider scale (ICES, 2019). Also for structures that are not moving, biofouling can have a negative impact. Not only by means of microbial induced corrosion (MIC) that can occur under layers of biofouling on steel surfaces (Little & Lee, 2007), but a thick layer of fouling hampers the inspection of the integrity of structures, it adds weight and increases the impact of water currents on for instance the fundamentals of offshore platforms.

1.2.1 Fouling prevention with biocides

Since the beginning of shipping, systems have been developed to prevent biofouling of the submerged parts. The majority of these systems is currently based on the use of paints containing biocides, which are chemical compounds that are released from the paint matrix to induce a constant layer of biocide in the water to prevent biofouling (Amara *et al.*, 2018). Historically, mostly tributyltin (TBT) and copper were used as active substance in anti-fouling paints. TBT was very effective and remained active in the paint for years (Armstrong *et al.*, 2000; Omae, 2003), but turned out to be very toxic to numerous non-targeted marine species. Around 1970 it was already shown that TBT caused shell deformations and reproductive problems in the Pacific oyster *Crassostrea gigas* in France (Alzieu *et al.*, 1986). Furthermore, imposex (male characteristics developing in females) in gastropods occurred at locations of high shipping activities, affecting more than 260 gastropod species (El Ayari *et al.*, 2018). In combination with its persistence in sediments and bioaccumulation (Antizar-Ladislao, 2008), the environmental impact of TBT was substantial. This insight led to local bans on the use of TBT from 1980 onward and a global ban in 2008.

Tributyl-tin paints are so called biocidal anti-fouling paints. These can be classified in three main-types: (1) Conventional; insoluble matrix with microscopic pores to allow the biocide to be released, (2) Erodible; the matrix partially dissolves to release the biocide from the layers underneath, (3) Self-polishing; hydrophobic paint matrix composed of the biocide linked to a polymer from which the biocide is released by hydrolysis followed by erosion of the now hydrophilic polymer residue (Gittens *et al.*, 2013). Tributyl-tin paints fall in the last category.

Stimulated by the TBT-ban alternative anti-fouling compounds were becoming more widely used, often based on zinc and copper compounds (Yebra *et al.*, 2004). Although these compounds are considered less harmful to the environment than TBT, similar problems with toxicity and accumulation seem to be evident from their use (Abarzua & Jakubowski, 1995; Amara *et al.*, 2018; Armstrong *et al.*, 2000; Chen *et al.*, 2017). For instance sea urchin (*Paracentrotus lividus*) embryos exposed to copper developed skeletal and gut abnormalities in their larvae stage (Gittens *et al.*, 2013; Manzo *et al.*, 2008). Zinc pyrithione (ZPT) seemed to be an environmental friendly alternative, because of its rapid photo degradability. But in the absence of light, it persists in the environment (Maraldo & Dahllöf, 2004). Research showed its toxicity to zebrafish (*Danio rerio*) by causing growth retardation, tissue pathological and physiology alternations of organs (Zhao *et al.*, 2018). Alternatives based on organic biocides like Sea-Nine™211 and Irgarol 1051® (2-methylthio-4-t-butylamino-6-cyclopropylamino-s-triazine) showed similar features. In fact Sea-Nine 211 was shown to be more toxic to sea urchin eggs and embryos than tributyltin oxide (TBTO) (Kobayashi & Okamura, 2002).

1.2.2 Environmental friendly alternatives

The above indicates the need for anti-fouling systems that are not based on toxic compounds. One approach was to prevent fouling organisms to settle or adhere by making the surface incompatible to attach to or preventing permanent adhesion (Gittens *et al.*, 2013; Nir & Reches, 2016). Other examples are:

- low surface-energy fouling-release coatings, which cause micro foulers to be released from the hull of a ship by shear stress due to the ships movement, because they cannot attach sufficiently to the ship's hull due to the low surface tension;
- polymer brushes; and
- sol-gel coatings (Gittens *et al.*, 2013).

However, none of these alternatives are as effective as the old tributyltin based coatings were. Therefore, there is still a need for effective environmental friendly methods to prevent biofouling and research on the use of natural compounds with anti-fouling properties is ongoing. These natural anti-fouling compounds can originate from organisms like bacteria, sponges, fungi, microalgae, and seaweeds.

Chemical anti-fouling compounds should preferably work via chemical signalling for the specific target organisms instead of being toxic for non-target species (Almeida & Vasconcelos, 2015; Hellio *et al.*, 2002). To introduce such compounds onto the market, "*In line with the Biocidal Product Regulation (EU) 528/2012, a clear description of the mode of action, biological targets and environmental fate...*" is required (Almeida & Vasconcelos, 2015). This is a costly and time consuming procedure.

2 Seaweeds and anti-fouling

It is clear that marine organisms are not completely free of biofouling. For instance, certain species of barnacles are capable of attaching to whales (Seilacher, 2005) and the blue mussel *Mytilus edulis* often settles on filamentous algae (Dobretsov, 1999). However, the degree of biofouling on marine organisms differs from that on non-living surfaces. The later stages of biofouling, such as macro fouling, often do not occur and the degree of biofouling differs between organisms (Armstrong *et al.*, 2000). Certain species of marine seaweeds are very effective in protecting themselves from biofouling (Armstrong *et al.*, 2000; Bhadury & Wright, 2004; Brock *et al.*, 2007; Dahms & Dobretsov, 2017; Walters *et al.*, 1996; Wikström & Pavia, 2004). They need to be, since coverage by fouling organisms may decrease photosynthesis and increase dragforces causing the algae to be dislodged from the bottom (Davis *et al.*, 1989).

Seaweeds can use physical and chemical strategies, alone or in combination to reduce the impact of biofouling, as summarised below.

2.1 Physical anti-fouling strategies

Some seaweed species are very effective in protecting themselves from biofouling by physical strategies. They can continuously shed the outer layer of cells and mucilaginous cover to release fouling organisms from their leaves and prevent accumulation of these organisms (Armstrong *et al.*, 2000; Halat *et al.*, 2015; Nylund & Pavia, 2005). Up to 25% of the frond epidermis can be shed per week (Halat *et al.*, 2015).

Furthermore, *Laminaria* species and seagrass can shed the distal ends of their blades to remove fouling organisms (Armstrong *et al.*, 2000; Mann, 1973). Since the meristem is located at the base of the blade, they can continue growing even if the distal blade is lost (Rolin *et al.*, 2017).

2.2 Chemical anti-fouling strategies

Another method applied against biofouling is chemical in nature. Marine organisms can secrete so-called biogenic compounds in the water to defend themselves against a variety of fouling organisms like bacteria, algae, fungi, protozoa or macrofoulers (Abarzua & Jakubowski, 1995; Bhadury & Wright, 2004). These biogenic compounds are secondary metabolites produced for protection and thus not essential for life (Abarzua & Jakubowski, 1995). Up to now, already a wide range of these biogenic compounds has been identified and purified, such as lactones, furanes, peptides, phenols, carotenoids, alkaloids and terpenoids (Almeida & Vasconcelos, 2015; Bhadury & Wright, 2004). Seaweeds also make use of this strategy and secrete secondary metabolites in order to, among other things, prevent biofouling on their surfaces. These secondary metabolites are often seasonally produced and can be produced at different live stages of the seaweeds (Khfaji & Boney, 1979; Rickert *et al.*, 2016; Saha & Wahl, 2013). Furthermore, these compounds can work on different levels of the biofouling process (Hellio *et al.*, 2001a). The active metabolites can either be produced and excreted by the seaweed itself or by symbiotic micro-organisms.

2.2.1 Production of secondary metabolites by seaweeds

The secondary metabolites produced by seaweeds can, on a microbial level, provide protection against pathogens, prevent microbe-induced premature decomposition and inhibit the formation of a bacterial biofilm (Dobretsov *et al.*, 2011; Engel *et al.*, 2006; Puglisi *et al.*, 2007). These secondary metabolites can be produced continuously or only when the seaweeds comes into contact with the target organisms and its chemical signals (Amsler & Fairhead, 2005). An overview of compounds with antimicrobial and anti-fouling properties found in microalgae is presented in *Table 1* (Goecke *et al.*, 2010).

An example is the algae *Bonnemaisonia hamifera* whose extracts have been shown to inhibit growth of bacteria in a laboratory scale study, at compound concentrations naturally produced on the surface of the algae. This indicates that *B. hamifera* can control the bacterial growth on its surface by producing growth-inhibiting biogenic compounds (Nylund *et al.*, 2005).

Another, extensively researched, example is the benthic marine macro algae *Delisea pulchra* from the class Rhodophyceae, that secretes a class of secondary metabolites called halogenated furanones or fimbrolides. This class of secondary metabolites acts as an antagonist of the acylated homoserine lactone (AHL) regulatory system inhibiting the quorum sensing (QS) of bacteria. Quorum sensing is "a cell-cell communication and gene regulatory mechanism that allows bacteria to coordinate swarming, biofilm formation, stress resistance, and production of toxins and secondary metabolites in response to threshold concentrations of QS signals that accumulate within a diffusion-limited environment." (Dobretsov *et al.*, 2009). By inhibiting QS the halogenated furanones inhibits bacterial colonization and thus biofilm formation on the surface of the seaweeds (Defoirdt *et al.*, 2004; Dworjanyn *et al.*, 1999; Hentzer *et al.*, 2002; Manefield *et al.*, 2002). One example of the effect of these halogenated furanones is the research by Ren *et al.* (2002) where they showed that a furanone can prevent the bacterium *Bacillus subtilis* from forming a biofilm and inhibits its swarming motility (Ren *et al.*, 2002). Besides a number of other benefits achieved by preventing the accumulation of microbes (no increase of hydrodynamic drag, no reduced buoyancy and elasticity of tissue, no grazers attracted, and no nutrient loss to bacteria (Goecke *et al.*, 2010)), it works as an anti-fouling method by preventing the successive settlement of micro- and macrofoulers (Denys *et al.*, 1995; Hellio *et al.*, 2001a; Kupper *et al.*, 2001; Manefield *et al.*, 2002; Nylund *et al.*, 2005; Othmani *et al.*, 2016).

The secondary metabolites produced by seaweeds can also directly work as anti-fouling compounds against secondary and tertiary colonizers. They can be lethal, inhibit settlement or inhibit the growth of the settling organism (Brock *et al.*, 2007; Dworjanyn *et al.*, 2006; Hellio *et al.*, 2002; Walters *et al.*, 1996; Wikström & Pavia, 2004). An example is the secretion of waterborne metabolites, primarily phlorotannins, by the brown seaweed *Fucus vesiculosus*, which inhibits the settlement of the barnacle *Balanus improvisus* (Brock *et al.*, 2007; Wikström & Pavia, 2004). It was shown in the research by Brock *et al.* (2007) that the phlorotannin levels of *F. vesiculosus* could, under natural conditions, reach concentrations high enough to inhibit barnacle larvae settlement (Brock *et al.*, 2007). Another example are meroditerpenoids produced by the macro alga *Cystoseira baccata* that inhibit the enzyme phenoloxidase. This enzyme is necessary in the production of mussel byssal thread plaques. The meroditerpenoids, therefore, inhibit the adhesion of mussel larvae to the algal surface (Mokrini *et al.*, 2008). Some species like the red alga *Delisea pulchra* produce compounds that are effective against a whole range of fouling organisms.

Table 1 List of antimicrobial and anti-fouling compounds found in seaweeds; Table 5 from Goecke et al., 2010. AV = antiviral, AE = anti-fouling, AF = antifungal activity, GNI = antibiotic activity against Gram-negative bacteria, GPI = antibiotic activity against Gram-positive bacteria.

Macroalga	Compounds	Activity	Source
Chlorophyta			
<i>Avrainvillea nigricans</i>	5'-hydroxy isoavrainvilleol	GPI	Colon et al. (1987)
<i>Caulerpa</i> spp.	Sesquiterpenoids	GPI, GNI	Paul et al. (1987)
<i>Codium iyengarii</i>	Iyengaroside-A, clerosterol galactoside	GPI, GNI	Ali et al. (2002)
<i>Penicillium capitatus</i>	Capisterones A, B	AF	Puglisi et al. (2004)
<i>Tydemania expeditionis</i>	Sulphated triterpenoids	AF	Jiang et al. (2008)
<i>Ulva fasciata</i>	Labdane diterpenoids	GNI	Chakraborty et al. (2010)
Heterokontophyta, Phaeophyceae			
<i>Canistrocarpus cervicornis</i>	Diterpenes	AE	Bianco et al. (2009)
<i>Cystoseira spinosa</i> var. <i>squarrosa</i>	Tetraprenyltoluquinol	GPI, GNI	Amico et al. (1988)
<i>Cystoseira tamariscifolia</i>	Methoxybifurcarenone	GNI	Bennamara et al. (1999)
Dictyotaceae	Dolabellane derivatives	AF	Tringali et al. (1986)
<i>Dictyopteris zonarioides</i>	Zonarol & isozonarol	AF	Fenical et al. (1973)
<i>Dictyota menstrialis</i>	Dictyol D, pachydictyol A	AE	Schmitt et al. (1995)
<i>Dilophus guineensis</i>	Dilophic acid	GPI	Schlenk & Gerwick (1987)
<i>Dilophus okamurai</i>	Spatane-type diterpenes	AE	Kurata et al. (1988)
<i>Fucus vesiculosus</i>	Polyhydroxylated fucophlorethol	GPI, GNI	Sandsdalen et al. (2003)
<i>Landsburgia quercifolia</i>	1,4-naphthoquinone	GPI, AF	Perry et al. (1991)
<i>Lobophora variegata</i>	Lobophorolide	AF	Kubanek et al. (2003)
<i>Sargassum</i> spp.	Polyphenols	AE, GNI	Sieburth & Conover (1965)
<i>Stoechospermum marginatum</i>	Spatane diterpenoids	GPI	De Silva et al. (1982)
<i>Stoechospermum marginatum</i>	Sulfated fucan	AV	Adhikari et al. (2006)
Rhodophyta			
<i>Asparagopsis armata</i>	Halomethanes, haloether, haloacetals	GPI, GNI	Paul et al. (2006)
<i>Bonnemaisonia hamifera</i>	Poly-brominated 2- heptanone	GNI	Nylund et al. (2008)
<i>Callophycus serratus</i>	Bromophycolides	AF	Lane et al. (2009)
<i>Dasya pedicellata</i> var. <i>stanfordiana</i>	P-hydroxybenzaldehyde	GPI, GNI	Fenical & McConnell (1976)
Delesseriaceae	Almazole D	GNI	N'Diaye et al. (1996)
<i>Delisea pulchra</i>	Halogenated furanones	AE	Maximilien et al. (1998)
<i>Delisea pulchra</i>	Halogenated furanones	GPI, GNI	Wright et al. (2006)
<i>Grateloupia indica</i>	Galactan sulphate	AV	Chattopadhyay et al. (2007)
<i>Laurencia chilensis</i>	3-hydroxi-4-methyl acetophenone	GPI, GNI	Valdebenito et al. (1982)
<i>Laurencia majuscula</i>	Brominated sesquiterpenes	GPI, GNI	Vairappan et al. (2010)
<i>Laurencia pannosa</i>	Pannosanol, pannosane	GNI	Suzuki et al. (2001a)
<i>Laurencia</i> spp.	Laurinterol, isolaurinterol	GNI	Vairappan et al. (2001b)
<i>Laurencia</i> spp.	Brominated sesquiterpenes	GPI, GNI	Bansemir et al. (2004)
<i>Osmundaria serrata</i>	Lanosol ethyl ether	GPI, GNI, AF	Barreto & Meyer (2006)
<i>Rhodomela confervoides</i>	Bromophenols	GPI, GNI	Xu et al. (2003)
<i>Sphaerococcus coronopifolius</i>	Bromosphaerone, 12S-hydroxybromosphaerodiol	GPI	Etahiri et al. (2001)

To prove the anti-fouling properties of the produced secondary metabolites, it is necessary to show that the compound(s) is (are) naturally found in sufficient concentrations on the surface of the organism. This method was developed by De Nys et al. (1998) who were able to extract non-polar (hexane-soluble) compounds from the surface of algae without destroying the algal surface cells, thereby accurately measuring the compound concentrations on the surface of the plant. The furanones are typically present at about 100 ng cm⁻² on the surface of *D. pulchra* (Dworjanyn et al., 2006). They are produced in vesicles in specialised gland cells, mostly found amongst the cortical cells of the plant (Dworjanyn et al., 1999). It has been shown that the natural occurring concentrations of the furanones can inhibit the settlement of the green seaweed *Ulva sp.*, red algae *Ceramium sp.*, red algae *Polysiphonia sp.* and the brown alga *Ectocarpus siliculosus* (Denys et al., 1995; Dworjanyn et al., 2006). The function of other compounds on the surface of *D. pulchra*, like lipids, is unclear. It is hypothesized that these lipids help with the even spread of the furanones over the plants surface. A lipid matrix on the plant's surface may also prevent rapid degradation or dissolution of the furanones in the water (Dworjanyn et al., 2006)

2.2.2 Microbe symbiosis

Instead of producing secondary metabolites to prevent biofouling itself, some seaweeds form symbiotic relationships with microbes (Goecke *et al.*, 2010; Satheesh *et al.*, 2016). Seaweeds in nature are generally colonized by an array of microbes (Wahl *et al.*, 2012). The distribution of these microbes on the surface of seaweeds indicates that it is not due to fouling but mediated by the seaweeds itself (Armstrong *et al.*, 2000; Armstrong *et al.*, 2001; Campbell *et al.*, 2015; Dobretsov *et al.*, 2006). The bacterial communities of seaweeds are often highly specific and their composition differs from that occurring in the surrounding seawater (Goecke *et al.*, 2010). Although the benefits for either partner in such a symbiosis is not always known, it is assumed that the benefits for microbes are nutritional in nature (Bonar *et al.*, 1986) and give them an advantageous position over other microbes (Armstrong *et al.*, 2000), while the benefit for the seaweeds is protection against micro- and macrofoulers (Satheesh *et al.*, 2016; Wahl *et al.*, 2012). It has been found that microbes associated with microalgae produce secondary metabolites that can inhibit the settlement of other marine organisms. This has been studied extensively and for many marine organisms, like sponges, corals, seaweed. Satheesh *et al.* (2016) provide an extensive overview of the research done into the macro organisms-microbe relationship. An example is the finding of a new species of the marine bacterial genus *Pseudoalteromonas*, associated with the marine alga *Ulva lactuca*, which exhibits anti-fouling properties (Egan *et al.*, 2001a). A similar finding was the production of an anti-algal peptide by the bacterium *Pseudoalteromonas tunicata*, found on the alga *Ulva australis*, which inhibits spore germination (Egan *et al.*, 2001b). The actual defence against biofouling is thus chemical in nature.

2.3 Application of seaweed strategies

2.3.1 Physical anti-fouling strategies

The constant renewal and shedding of the surface area that seaweeds apply as strategy to remove fouling organisms is considered not applicable in practice to protect man-made objects. Although technically possible, it would require too thick coating layers and/or too frequent maintenance to be protective for longer periods.

2.3.2 Examples of seaweeds anti-fouling compounds

To evaluate the anti-fouling effectiveness of a new compound versus its toxicity, the LC₅₀/EC₅₀ quotient, or therapeutic ratio, is used. The LC₅₀ being the 50% lethal concentration for the test organisms and the EC₅₀ the effective concentration that inhibits 50% of the biological activities of the test organism (Almeida & Vasconcelos, 2015; Gittens *et al.*, 2013). Values of the LC₅₀/EC₅₀ quotient higher than 15 indicate a possible non-toxic anti-fouling compound, while much higher values indicate suitable anti-fouling compounds for commercial use (Almeida & Vasconcelos, 2015; Qian *et al.*, 2010).

Different researchers have shown the anti-fouling properties of compounds extracted from a wide variety of seaweeds. Salama *et al.* (2018) studied the anti-fouling activity of extracts of three macro algal species, *Chaetomorpha linum*, *Turbinaria ornata*, and *Sargassum polycystum*, against barnacle larvae in both a laboratory assay using Petri dishes and in a field study by mixing the crude seaweeds extracts with varnish, coating nylon net panels and hanging them for 3 months submerged in the Red sea. All three methanol extracts inhibited the settlement of cypris larvae in the laboratory study. In the field, only the extracts of *T. ornata* and *S. polycystum* significantly reduced biofouling. Analyses showed that the crude extracts contained fatty acids, their derivatives phytosterols and terpenoids, and some other compounds. The results from this study indicate that crude algal extracts could be used as natural anti-fouling compounds in anti-fouling paints, although toxicity testing of these crude extracts is necessary (Salama *et al.*, 2018).

Another study tested crude ethanol extracts from the red alga *Chondrus crispus*. The dried extracts had more anti-fouling activity against marine bacteria and other microfoulers, while the fresh extracts

had more anti-fouling activity against macrofoulers. They also tested the anti-fouling properties of the dried extracts in a field study by dissolving 1% dried algal extract in a controlled depletion polymer paint, coating primer-coated steel coupons and submerging them into the sea in Southampton (UK) for 3 months (April-July). It was shown that for a period of 10 weeks the paint with the dried algal extract had lower fouling than the negative control (blank panel) and the positive control (same coating with biocide chlorathalonil) (Chambers *et al.*, 2011).

The brown algae *Bifurcaria bifurcata*, *Halidrys siliquosa*, *Halopteris scoparia*, *Dictyopteris membranacea*, *Bifurcariopsis capensis*, *Cladostephus verticillatus* and *Bifurcaria brassicaeformis* have been shown to have anti-fouling activity against several marine bacteria and fungi, but low toxicity against the larvae of marine invertebrates (Hellio *et al.*, 2000). *Bifurcaria bifurcata* has been shown to have antimicrobial activity against two marine bacteria, *Cobetia marina* and *Pseudoalteromonas haloplanktis*, which showed seasonal variation with the highest level of activity between April and August. Furthermore, concerning the effectivity of *B. bifurcata* extracts towards adhesion and toxicity of the barnacle *B. amphitrite* cypris larvae, the EC₅₀ of settlement was lowest in the period April to July (reduced settlement at >5 µg/ml). However, the *B. amphitrite* nauplii larvae toxicity test showed that the April to August extracts were toxic to nauplii with the extracts from May (LC₅₀=55.6 µg/ml) and June (LC₅₀=38.3 µg/ml) being the most toxic. Even so, for the other months the LC₅₀ is much higher than the EC₅₀. This indicates that for some months the active fractions of *B. bifurcata* extracts are effective against *B. amphitrite* larvae settlement and marine bacteria growth (Marechal *et al.*, 2004).

Hellio *et al.* studied the anti-fouling effectivity of compounds from numerous brown algae, like *Bifurcaria*, against fungi, bacteria, diatoms, seaweeds and the blue mussel. They found numerous compounds, like diterpenoid compounds and some pure molecules, with anti-fouling and antimicrobial activity (Hellio *et al.*, 2001b). They performed a similar experiment in 2002 with extracts from thirty algae, finding twelve promising extracts with high anti-fouling activity and low toxicity against invertebrate larvae (Hellio *et al.*, 2002).

Othmani *et al.* tested eight compounds extracted from the brown seaweed *Taonia atomaria* for their anti-fouling activity against five marine bacteria and two barnacle species. The same two barnacle species were used for toxicity testing. They found one compound, (-)-gleenol, which could be a new anti-fouling compound because of its low EC₅₀ and moderate toxicity (Othmani *et al.*, 2016).

More seaweeds compounds can be found in Bhadury & Wright, 2004; Dahms & Dobretsov, 2017; and Fusetani & Clare, 2006.

2.3.3 Application in anti-fouling systems

Around the year 2000 the BRITE/EURAM 3 Project was conducted within the Fourth Framework Programme of the European Community, with the aim to develop environmentally compatible anti-fouling coatings for the protection of ships, water systems, fish cages, and other immersed structures against aquatic growth. One part of the project consisted of synthesizing 204 compounds that fall into four different classes of biocide compounds. The composition of these compounds was tested using laboratory barnacle bioassays and microbiological assays. Sixteen of the most promising compounds were used for further testing. These tests were: static exposure of panels with model paints in the North Sea and Mediterranean, anti-barnacle properties of paints by laboratory assays, and environmental tests specially developed to seawater conditions. In the end, by combining all these results with physical and economical properties, there were three different (groups of) compounds that were considered promising for further development in coatings and patenting. The compound CAULB13 appeared to be the most promising one and was, therefore, tested on various rafts and ships. This compound was shown to have "...satisfactory anti-fouling properties up to 26 months; relatively low toxic effects on non-target organisms; appropriate water solubility; simple synthesis and cheap, readily available starting material for synthesis; good compatibility with other paint constituents; good thermal stability." (CORDIS BRPR960159, 2001).

3 Potency of natural anti-fouling strategies

3.1 Chemistry based anti-fouling

3.1.1 Seaweed strategies

The literature review shows that seaweeds produce numerous compounds to prevent biofouling of their surface and that these compounds can be extracted and applied in coatings that reduce biofouling in both laboratory and field settings. Despite this, we are not aware of any commercially available anti-fouling coatings that are based on these substances. It is suggested that this is due to insufficient funding for this type of R&D research and (therefore?) a lack of strong incentives for scientists from different disciplines to fully commit themselves to these goals (Qian *et al.*, 2010). If this is really the case, it suggests that until now it has not been found economically feasible to invest in the development of bio-based anti-fouling products.

The large-scale production of the natural anti-fouling compounds for a price or at a scale that is attractive for commercial companies to invest in is a major challenge (Qian *et al.*, 2010). Preferably, the active compounds are produced by direct extraction from marine macro-organisms that are harvested from mariculture farms. The increasing interest in seaweed farming might result in a more steady flow of natural basis material, but the isolation of specific compounds from marine algae is still very expensive and time consuming (Dahms & Dobretsov, 2017). The use of combinatorial genetic or metabolic engineering, or hybrids, might be a solution to this problem in the future (Bhadury & Wright, 2004), but at this moment chemical synthesis still is a better option.

Apart from the challenge of making the production profitable, the registration of the active compound as a new biocide before it is allowed to the market is a serious hurdle to be overcome. This requires the time consuming and costly development of an environmental risk assessment dossier describing the half-life, breakdown, environmental fate, toxicity and other possible negative environmental impacts of the new compound (Qian *et al.*, 2010). Together with the costs that are always involved in R&D this probably holds back investments in the development of bio-based anti-fouling coatings.

3.1.2 Chemical cues by other organisms

The chemistry based anti-fouling strategy that is applied by seaweeds, the production of secondary metabolites that repel fouling organisms, is also used by many other organisms. The chances and limitations are comparable and, therefore, not further discussed here.

Biochemical signals do, however, also play a role in the development of a fouling community. For instance in the succession from the early colonisers bacteria, diatoms and protozoa that attracts the larvae of the invertebrate tertiary colonizers like barnacles, mussels, tubeworms etc. (Abarzua & Jakubowski, 1995). Settlement of the tubeworm *Hydroides elegans* for instance is strongly induced by the presence of diverse biofilms, including those that consist of monocultures of bacteria. This suggests that the larvae are able to detect the presence of extracellularly polymers that are secreted by the early colonisers (Lam *et al.*, 2003).

Bivalves of various species also respond to trophic cues for the settlement of larvae (Forêt *et al.*, 2018). Larval settlement is influenced by saturated fatty acids, possibly linked to organic detritus and bacterial production deriving from terrestrial inputs. This implies that recruits of bivalves tend to settle in areas with more food availability (Leal *et al.*, 2018). A similar positive cue is described for the blue mussel (*Mytilus edulis*) (Scott *et al.*, 2016).

The presence of other species can also form a negative settlement cue. This was shown in an experimental set-up with larvae of the blue mussel that were allowed to choose between regions that were treated with an extract of crabs, a major predator of juvenile mussels and regions treated with extract of their food (algae). The latter clearly induced settlement of the larvae, while few larvae settled in regions treated with crab extract (Scott *et al.*, 2016).

Settlement cues can also come from individuals of the same species. The larvae of the tubeworm *Hydroides elegans* only metamorphose in presence of chemical cues produced by conspecifics. Without these cues, the larvae remain planktonic and will not settle (Bryan *et al.*, 1997).

A similar mechanism has been described for barnacles, that secrete a so-called settlement-inducing-protein-complex (SCIP). This stimulates a clustered settlement, which increases the reproductive chances of the group. While low concentrations of SCIP stimulate settlement, higher concentrations have the opposite effect to avoid overpopulation of an area (Kotsiri *et al.*, 2018).

It seems obvious that such negative settlement cues are highly species specific. Therefore, these substances probably have less potential as a general anti-fouling compounds. But even if this is not the case the production and registration of such substances will hold the same challenges as described above for the application of seaweed related metabolites for the same purpose.

3.2 Physical based anti-fouling

3.2.1 Seaweed strategies

The only known physical strategy applied by seaweeds to prevent biofouling, the constant renewal fouled material, seems not directly applicable to protect man-made structures without continuous effort and costs.

However, settling of fouling organisms is affected by physical conditions such as local flow velocity and surface and the following paragraphs contain an inventory of these conditions for the main groups of fouling organisms, seaweeds, tubeworms, barnacles and mussels.

3.2.2 Flow velocity

New surfaces are colonized by fouling organisms mainly through settlement of pelagic larvae.

Successful settlement is only possible at relatively low flow velocities that allow the organisms to make contact with the substrate.

Cyprid larvae of the striped barnacle *Amphibalanus amphitrite* showed highest settlement rates in a tube with free stream velocity between 3-15 cm/s (Qian *et al.*, 1999). The researchers noted that the swimming capability of an organism heavily affects their rate of attachment. Recent research looked more into detail and showed that cyprid larvae of the bay barnacle (*Balanus improvises*) can make use of small time-windows of relative lower flow speeds near the substrate surface that are caused by turbulence (Larsson *et al.*, 2016). A local flow velocity of 1.9-2.4 cm/s, that occurred only for 0.14 seconds was sufficient for these larvae to settle. This situation appeared near the boundary layer due to turbulence at a free-stream velocity of 20 cm/s.

Furthermore, they found that the cyprid's swimming speed was approximately 1.8 cm/s, explaining the critical local flow velocity, as they swim against the current (negative rheotaxis) to become almost stationary (Larsson *et al.*, 2016).

It seems obvious that a similar mechanism applies for all fouling animal species.

In addition, the larvae of the blue mussel (*Mytilus edulis*) initially settle at locations with reduced flow velocity. However, after the first growth, they can detach and move to places with higher flow velocity and better food availability (Dobretsov & Wahl, 2007).

For the tubeworm *Hydroides elegans* the highest settlement has been reported in free-flow velocities of 1 to 3 cm/s. At 10 cm/s settlement was only 20% (Qian *et al.*, 1999).

The examples mentioned above indicate that biofouling can be prevented by maintaining a high flow velocity for all times. Logically this strategy is not possible to protect a ship's hull or a static maritime structure. It could in theory be applied to prevent fouling in cooling water circuits if these could be constructed in a way that turbulence will be avoided. In practice, this seems a difficult and probably impossible task.

3.2.3 Surface structure

As the settling larvae have to get a hold on the substrate, a completely smooth surface reduces the attachment strength of fouling organisms. A more or less structured surface facilitates the development of a community of fouling organisms. However, the preferred texture can differ per species.

Cyprids of *Amphibalanus amphitrite* seem to prefer sinusoidal linear textures, with a distance below 32 μm for the sinusoidal shapes (Aldred *et al.*, 2010). They seem to avoid ranges of 64-256 μm , while again preferring larger distances of 512 μm . It is hypothesized that the size of the cyprid and its attachment organs are of critical importance for these size-ranges. At sinusoidal textures with a distance of 32 μm or smaller, the attachment organs can easily connect with the surface. At slightly larger textures, the attachment organs are placed at curved angles, making their attachment weaker, while at even larger distances (256 μm), the gaps between the sinusoidal structures are roughly half that of the size of the cyprid, making it hard for it to angle itself for proper attachment. At very large distances (>512 μm), the cyprid can fit between the sinusoidal shapes, giving it a strong attachment. For this reason, preferred structures of certain substrates will highly differ between species of different sizes.

Zoospores of the species *Ulva linza* and *Ulva compressa* prefer surfaces with grooves that they can squeeze into, meaning roughly their own diameter or slightly smaller (Callow *et al.*, 2002). More precisely, the zoospores of *U. linza*, are 5 μm and prefer grooves of 5 μm . The researchers suggest that in valleys or grooves with a width that is less than twice the radius of the (spheroid) organism, the contact surfaces of adhesion become contact 'points', minimizing adhesion strength. However, the zoospores perform amoeboid-like space-filling movement, which makes the shape of the attached spore being partly determined by the availability of space. The zoospores also clearly choose their location to be most beneficial to them, since non-living beads of a similar shape and size dropped on the same surface did not aggregate in the same way as the zoospores did.

For the mussel *Mytilus galloprovincialis*, it was found that settlement on heterogeneous surfaces was preferred. More specifically, there was an avoidance of a homogenous ridged surface with a uniform distance of 1–2 μm between ridges with a mean depth of 1.5 μm (Scardino *et al.*, 2003).

In addition to the preference of the blue mussel and pearl oyster for heterogeneous surfaces, Kobak (2001) found that zebra mussels preferred dark, shaded substrate.

In contrary to the previous groups, physical surface parameters seem of minor importance to the fouling tubeworm *Hydroides elegans* (Lam *et al.*, 2003).

Based on the above it could thus be possible to produce a specific surface structure to reduce fouling of specific species, although it might not be possible to create a surface structure that prevents settlement of all unwanted species.

3.3 Symbiosis

Seaweeds can live in symbioses with microbes that produce substances that prevent fouling of other organisms.

Symbiotic interactions are by definition not possible with non-living (man-made) structures, however, it might be possible to create conditions at the structure's surface that benefits fouling species with low negative impact, just like seaweeds that facilitate the development of micro-organisms that produce metabolites that prevent the settlement of less wanted fouling species.

This would be a new way to deal with biofouling: Instead of fighting the constant battle against fouling organisms which until now has been tried, one could also try to manage the fouling community in such a way that it is dominated by species with low impact.

The ideal 'low impact species' on man made structures:

- 1) add little water resistance to the surface;
- 2) do not develop high biomass;
- 3) have no negative impact on the integrity or durability of the structure /does not promote microbial induced corrosion;
- 4) prevents settlement of other, less preferred, species;

With these characteristics in mind, inspection videos of the underwater construction from various offshore platforms in the North Sea were analysed. This revealed at least one candidate species that seems to meet the qualifications of a 'low impact species'.

This approach by definition allows the settlement of the preferred fouling organism at the surface. As this will result in increased drag, it may be less useful for application on a ship's hull. In addition it will be very difficult if not impossible to maintain the conditions that favour the preferred fouling species above others on a ship's hull that is traveling between regions. On static infrastructure, however, it could be an alternative approach to reduce the impact of fouling, without the application of toxic agents.

4 Conclusion

Seaweeds make use of physical and chemical strategies to keep their surface free of biofouling organisms that can negatively affect their conditions.

The physical strategies that are applied by seaweeds are not considered applicable to protect man-made structures as it needs continuous renewal of the surface material.

Other physical anti-fouling strategies that were explored are flow velocity and surface structure. Combined these aspects may have the potency to prevent fouling. However, this requires a controlled environment and is, therefore, not applicable for protecting ship hulls or maritime infrastructure against fouling. It could work in industrial cooling water systems, but this option was not further investigated. In general, the current methods to prevent fouling in cooling water systems seem to fulfil the needs of the industry.

The chemical anti-fouling strategies of seaweeds are based on the production of a broad range of metabolites with the potential to reduce settlement and/or development of fouling organisms. Although the effectiveness of several of these metabolites has been shown on experimental scale, none of these substances has made it into a commercial product so far. An important reason is the expensive and long lasting registration process that is required for bringing a new biocide to the market, sometimes in combination with the relatively high production costs of the compounds. The commercial application of bio-based compounds for anti-fouling purposes is, therefore, not attractive.

Apart from producing anti-fouling metabolites themselves, seaweeds can also support specific microbes that have little or no negative impact on their condition, but that prevent the settlement of more harmful species. A similar strategy, where a fouling species with low negative impact is favoured and supported, can have potential to manage fouling in an environmental friendly way on man-made surfaces where some drag is acceptable. This would be a new way to deal with biofouling, as alternative for fighting the constant battle against all fouling organisms which is common practice until now.

As spin-off of the exploring study presented in this report, a proposal for a joint industry project will be prepared that aims to investigate the potency of favouring low negative impact fouling organisms to manage biofouling on offshore infrastructures.

5 Quality Assurance

Wageningen Marine Research utilises an ISO 9001:2015 certified quality management system. This certificate is valid until 15 December 2021. The organisation has been certified since 27 February 2001. The certification was issued by DNV GL.

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Justification

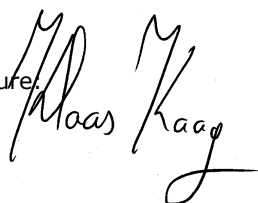
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The scientific quality of this report has been peer reviewed by the a colleague scientist and the head of the department of Wageningen Marine Research.

Approved: Dr. Klaas Kaag
Researcher

Signature:

Handwritten signature of Klaas Kaag in black ink.

Date: 16th of May 2019

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