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Structure and Biosynthesis of Benzoxazinoids: Plant Defence Metabolites with Potential as Antimicrobial Scaffolds

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Abstract

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2 Benzoxazinoids, comprising the classes of benzoxazinones and benzoxazolinones, are a set of 3 specialised metabolites produced by the plant family Poaceae (formerly Gramineae), and some 4 dicots. The family Poaceae in particular contains several important crops like maize and wheat. 5 Benzoxazinoids play a role in allelopathy and as defence compounds against (micro)biological 6 threats. The effectivity of benzoxazinones in these functionalities is largely imposed by the 7 subclasses (determined by N substituent). In this review, we provide an overview of all currently 8 known natural benzoxazinoids and a summary of the current state of knowledge of their 9 biosynthesis. We also evaluated their antimicrobial activity based on minimum inhibitory 10 concentration (MIC) values reported in literature. Monomeric natural benzoxazinoids seem to 11 lack potency as antimicrobial agents. The 1,4-benzoxazin-3-one backbone, however, has been shown to be a potential scaffold for designing new antimicrobial compounds. This has been 12 13 demonstrated by a number of studies that report potent activity of synthetic derivatives of 1,4benzoxazin-3-one, which possess MIC values down to 6.25 µg mL⁻¹ against pathogenic fungi 14 (e.g. C. albicans) and 16 µg mL⁻¹ against bacteria (e.g. S. aureus and E. coli). Observations on 15 16 the structural requirements for allelopathy, insecticidal, and antimicrobial activity suggest that 17 they are not necessarily conferred by similar mechanisms.

Highlights

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- Comprehensive overview of natural benzoxazinoids.
- Benzoxazinone subclass (*N* substituent) influences allelopathic and defence potency.
- Monomeric natural benzoxazinoids lack antimicrobial potency.
- Chemical modification of 1,4-benzoxazin-3-ones can enhance antimicrobial activity.
- Potential of 1,4-benzoxazin-3-one as a scaffold for design of antimicrobial drugs.

24 **Keywords**

- 25 Poaceae, cereal, plant defence, antibacterial, antifungal, classification, benzoxazinone,
- benzoxazolinone, synthetic, modification

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1. Introduction

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49 Benzoxazinoids are a set of specialised metabolites that are prominently described in species 50 of the monocot plant family Poaceae. Benzoxazinoids have also been reported to be present in 51 dicots of the families Acanthaceae, Calceolariaceae, Lamiaceae, Plantaginaceae, and 52 Ranunculaceae (Adhikari et al., 2015; Frey et al., 2009; Niemeyer, 2009). Due to the 53 agricultural importance of the crops in Poaceae such as wheat (Triticum aestivum) and maize 54 (Zea mays), benzoxazinoids of this plant family will be our main focus. These specialised 55 metabolites play a role in allelopathic plant-plant interactions, as defence compounds against (micro)biological threats, and as defence regulatory signals (Ahmad et al., 2011; Neal et al., 56 57 2012; Niemeyer, 2009; Sicker et al., 2000). Their allelopathic and defence functionalities, 58 which are discussed in more detail in Section 5, suggest that benzoxazinoids could be 59 potentially interesting leads for antimicrobial compounds. Promising lead structures might also 60 be further modified to optimize their antimicrobial activity, as has been done for allelopathic 61 activity (Macías et al., 2010; Macías et al., 2006b). In this review we aim to provide a comprehensive overview of the structures of natural 62 63 benzoxazinoids and to investigate their biosynthesis and biofunctionality with an emphasis on 64 their antimicrobial potential.

2. Structure of natural benzoxazinoids

2.1. Classification and subclassification

Benzoxazinoids are divided into two classes: benzoxazinones (1,4-benzoxazin-3-one) and benzoxazolinones (1,3-benzoxazol-2-one) (**Figure 1**). Benzoxazolinones were first reported in rye in the 1950s (Virtanen and Hietala, 1955) and benzoxazinones shortly thereafter in the early 1960s, also in rye (Hietala and Virtanen, 1960; Honkanen and Virtanen, 1960). Within these classes, several subclasses are formed by compounds with various *N* substituents (Cambier et al., 1999; Niemeyer, 2009). The class of benzoxazinones is divided into three subclasses: lactams (N-hydro), hydroxamic acids (N-hydroxy), and methyl derivatives (N-methoxy) (**Table** 1). Within the class of benzoxazolinones, N-hydro derivatives are by far the most common due to loss of the N substituent in the transformation from benzoxazinone to benzoxazolinone (Section 4.2). Nevertheless, benzoxazolinones with other N substituents have been reported: N-hydroxy and N-methoxy derivatives were found in Scoparia dulcis (Wu et al., 2012). Possibly these compounds are formed enzymatically from N-hydro benzoxazolinones or by another yet to be discovered biosynthetic pathway. Additionally, N-glucosyl derivatives can be formed as detoxification products of benzoxazinoids (Hofmann et al., 2006; Maag et al., 2014; Sicker et al., 2001) (**Table 1**). Various properties of benzoxazinoids, such as stability and reactivity (Grambow et al., 1986; Wouters et al., 2016), plant defence functionality (Niemeyer, 2009; Oikawa et al., 2004), effects on human health (Adhikari et al., 2015), and fragmentation in mass spectrometry (de Bruijn et al., 2016) were shown to be affected by N substitution.

2.2. Abbreviations and structural overview

The first benzoxazinoids were discovered over 50 years ago. Their systematic IUPAC names are long and inconvenient to use which led to the formulation of abbreviations based on the three letter code BOA. This code was complemented with prefixes of the letters H and M for hydroxy and methoxy substituents, respectively. BOA can refer to compounds with either a benzoxazinone or a benzoxazolinone backbone (**Figure 1**). Because this three letter code does not distinguish classes or subclasses, identification of compounds based on their abbreviation can be challenging, especially for newcomers in the benzoxazinoid field. To facilitate the correct use of these abbreviations and avoid inconsistency, this paper includes an overview of the presently known natural benzoxazinoids of different (sub)classes with their abbreviations (**Table 1**). This overview does not include benzoxazinone *C2 O*-glycosides, which will be

discussed in the following section. In addition to the benzoxazinoids in **Table 1**, zeaoxazolinone, a 7-methoxy-1,3-benzoxazol-2-one dimer from maize (Mohamed et al., 2014), is displayed in **Figure 2A**. Another dimeric benzoxazolinone, 5,5'-bis-benzoxazoline-2,2'-dione (1,3-benzoxazol-2-one dimer) was found in *Acanthus ilicifolius* (D'Souza et al., 1997) (**Figure 2B**).

2.3. Glycosylation of benzoxazinones

Benzoxazinones are glycosylated in an early step of the biosynthesis (Section 3.1). Glycosylation is used as a mechanism to store benzoxazinones inside plant cells and to prevent self-toxicity (Cambier et al., 1999; von Rad et al., 2001). Glycosides are unable to undergo ring-opening, limiting their reactivity (Figure 3 and Section 4). So far, glycosides reported in Poaceae are mainly glucosides, whereas in the family Plantaginaceae galactosides were also detected (Wu et al., 2012). Recent studies also indicate the presence of dihexosides in wheat and rye (Secale cereale) (de Bruijn et al., 2016; Hanhineva et al., 2011), tri- and tetrahexosides in wheat and rye beers (Pihlava and Kurtelius, 2016), and acetylhexosides in wheat (de Bruijn et al., 2016). Upon metabolism by mammals, benzoxazinoids may also be conjugated with uronic acids (Adhikari et al., 2012). In addition, reglucosylation of aglycones, with inversion of stereochemistry (from 2R to 2S), is a mechanism of benzoxazinoid detoxification found in some insects (Wouters et al., 2014).

2.4. Substituents on the benzene moiety (R₂ to R₅)

The most common natural substituents at positions R_2 to R_5 are hydroxyl or methylated hydroxyl (methoxyl) groups (Niemeyer, 2009). Other substituents at these positions occurring more rarely in nature are chloro-, acetyl- or glucopyranosyloxy-groups (**Table 1**). Chloroderivatives, for example, can be found in maize (Anai et al., 1996) and *A. ilicifolius* (Acanthaceae) (Kanchanapoom et al., 2001a).

3. Benzoxazinone biosynthesis and distribution in Poaceae

3.1. General biosynthesis of benzoxazinones

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123 The basic biosynthetic pathway of benzoxazinones (up to DIBOA-Glc) was first studied and 124 elucidated in maize (Frey et al., 1997), and later also in wheat (Nomura et al., 2002; Nomura et 125 al., 2003) and rye (Bakera et al., 2015; Rakoczy-Trojanowska et al., 2017). The biosynthesis 126 starts from indole-3-glycerolphosphate (IGP), which is consecutively converted to HBOA, the 127 first benzoxazinoid, in 4 steps by the enzymes BX1 to BX4 (Figure 3). A subsequent 128 glucosylation and hydroxylation leads to DIBOA-Glc, which serves as the starting point for 129 hydroxamic acid biosynthesis. All further downstream benzoxazinoids are synthesized as 130 glucosides and most of the enzymes involved are unable to use the aglycones as substrates 131 (Jonczyk et al., 2008; Oikawa et al., 2002). The formation of these downstream metabolites has 132 mainly been studied in maize and to a lesser extent in wheat and rye (Dutartre et al., 2012; 133 Handrick et al., 2016; Jonczyk et al., 2008; Makowska et al., 2015; Meihls et al., 2013; Tanwir 134 et al., 2017). A summary of the current state of knowledge on the biosynthetic pathways of 135 benzoxazinones is shown in Figure 3. As shown on the right-hand side of this figure, 136 benzoxazinones can be transformed to benzoxazolinones, which is further discussed in **Section** 137 4.2. Next to the Poaceae, benzoxazinoids have been reported in dicots of the families 138 Acanthaceae, Calceolariaceae, Lamiaceae, Plantaginaceae, and Ranunculaceae (Table 1). The 139 biosynthesis in these families might be similar to what has been reported in Poaceae, as 140 indicated by analogues of the benzoxazinoid biosynthetic enzymes that have been found in 141 dicots (Dick et al., 2012; Schullehner et al., 2008). The overview in Figure 3 indicates that 142 several benzoxazinoid biosynthetic pathways require further investigation (dashed arrows). 143 These aspects are discussed in the next section.

3.2. Tentative pathways in benzoxazinone biosynthesis

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According to the generally accepted pathway, as shown in Figure 3, HBOA is converted to DIBOA by BX5, which is then glucosylated by BX8 or BX9. Whereas HBOA aglycons are relatively stable, DIBOA aglycons are reactive (Section 4) and phytotoxic (Section 5). Therefore, the formation of free DIBOA within the plant cell is likely omitted. One possibility could be the stabilisation and rapid glycosylation of DIBOA within a metabolon (i.e. a complex of sequential metabolic enzymes) by metabolic channelling. This would be similar to what has been demonstrated for toxic or labile intermediates in other secondary metabolic pathways (e.g. dhurrin biosynthesis in sorghum) (Jørgensen et al., 2005). As an alternative, we propose a pathway in which HBOA is glycosylated prior to oxidation to form DIBOA-Glc. It was previously proposed that HBOA-Glc and DIBOA-Glc are in fact a redox pair (Hofman and Hofmanová, 1969), which might present a mechanism for the possible interconversion of these compounds. The existence of an enzyme which catalyses the oxidation of HBOA-Glc to DIBOA-Glc has not been thoroughly explored in literature. Maize BX8 and BX9 were both shown to be able to glycosylate HBOA albeit at a much lower conversion rate than DIBOA and DIMBOA (von Rad et al., 2001). The glycosylation of DIMBOA, however, does not seem to serve a function within the biosynthetic pathway, as it is already formed as a glycoside. The formation of HBOA-Glc would be a logical starting point for the biosynthesis of lactams. Further hydroxylation and methylation to produce DHBOA-Glc, HMBOA-Glc, and HM2BOA-Glc might be performed by the same or similar enzymes as those involved in hydroxamic acid biosynthesis (BX6, BX7, and BX13) or by a yet to be discovered part of the BX enzyme-cluster. The compounds TRIBOA-Glc and TRIMBOA-Glc are intermediates of the biosynthesis and are not typically accumulated and detected in maize tissues (Cambier et al., 1999; Handrick et al., 2016). Possibly, DHBOA-Glc serves a similar role as an intermediate in HMBOA-Glc synthesis. An analogue of TRIMBOA-Glc as an intermediate for the biosynthesis of HM₂BOA-Glc is not yet known. The biosynthetic pathway of benzoxazinone lactams presents a gap in our current knowledge. Lactams are the least prevalent subclass of benzoxazinones in maize explaining the lack of research on their biosynthesis, however, lactams are more prominent in other species, such as rye (Tanwir et al., 2013).

The methyl derivative equivalent of DIBOA, 4-*O*-Me-DIBOA-Glc, was annotated in wheat seedlings exposed to fungal stress based on LC-MS analysis (de Bruijn et al., 2016). Thus far, its biosynthesis has not yet been fully elucidated. It was shown that a DIMBOA-Glc 4-*O*-methyltransferase from wheat was also able to convert DIBOA-Glc to 4-*O*-Me-DIBOA-Glc *in vitro*, but the latter compound was not detected *in planta* in that study (Oikawa et al., 2002).

3.3. Distribution of benzoxazinones in Poaceae

Several recent reviews have addressed the genetic background of benzoxazinoid production between the different species within the Poaceae family (Dutartre et al., 2012; Makowska et al., 2015). Three agriculturally important crops that produce benzoxazinoids are maize, wheat, and rye. The profile of benzoxazinones produced between these species varies (**Figure 4**). Overall, maize produces the most diverse profile of benzoxazinones, whereas rye possesses the lowest diversity. Based on the biosynthetic pathways involved (**Figure 3**), it seems like wheat does not possess an active BX14-like enzyme to perform the conversion of DIM₂BOA-Glc into HDM₂BOA-Glc. It is, however, able to produce HDMBOA-Glc which indicates the presence of an active BX10-like enzyme. Rye has not been shown to produce HDMBOA-Glc and other methyl derivatives.

Several other well-known species from the Poaceae family, such as rice (*Oryza sativa*), oat (*Avena sativa*), and barley (*Hordeum vulgare*) do not produce benzoxazinoids. Interestingly, some other members of the genus *Hordeum*, e.g. *Hordeum lecheri*, have been found to produce benzoxazinoids (Grün et al., 2005). There have also been some reports of benzoxazinoid

production in sorghum (Malan et al., 1984; Niemeyer, 1988). Several other less well-known species in the Poaceae (e.g. *Aegilops speltoides*) have been reported to produce benzoxazinoids (Dutartre et al., 2012). As shown by the phylogenetic trees presented by Dutartre and coworkers, the development of the benzoxazinoid biosynthetic cluster does not necessarily follow the general phylogenetic relationships between the different species, as is exemplified by the genus *Hordeum* (Dutartre et al., 2012).

For a more in-depth perspective of benzoxazinoid phylogenomics, we would like to refer readers to the aforementioned reviews (Dutartre et al., 2012; Makowska et al., 2015) (focussed on Poaceae). In addition, there are several studies that provide more information about benzoxazinoids in dicots (Dick et al., 2012; Frey et al., 2009; Schullehner et al., 2008). The main benzoxazinoids produced by dicots are similar to those produced by Poaceae (**Table 1**). In **Section 2** and **Table 1**, several notable molecules unique to dicots are shown such as chloroderivatives of HBOA and DIBOA, and the production of benzoxazinone galactosides rather than glucosides.

3.4. Benzoxazinoid content and induction in planta

The main benzoxazinoid in maize and wheat is DIMBOA (and its glycosides) (Köhler et al., 2015; Villagrasa et al., 2006), whereas in rye this is DIBOA (and its glycosides) (Oikawa et al., 2002; Rakoczy-Trojanowska et al., 2017). Reported values for the total quantities of benzoxazinoids in tissues of different species are highly variable. They range from 4.8 and 95 µg g⁻¹ dry weight in wheat and rye grains (Tanwir et al., 2013), respectively, to over 1900 µg g⁻¹ dry weight in rye shoots (Schulz et al., 2013), and to several mg g⁻¹ fresh weight in maize (Köhler et al., 2015; Meihls et al., 2013). Comparing individual or total benzoxazinoid content from different literature sources is not straightforward. One of the underlying causes is that many papers focus solely on the analysis of a select number of benzoxazinones (usually DIMBOA and DIBOA) whereas in other works total benzoxazinoid content might include

glycosides, other benzoxazinones, and benzoxazolinones. Secondly, the quantitative analysis of these molecules is challenging and has been addressed in multiple publications (Bonnington et al., 2003; Eljarrat and Barceló, 2001; Pedersen et al., 2017). This is mainly due to the activity of glycosidases during sample treatment and the reactivity of the resulting benzoxazinone aglycons (Section 4) (Cambier et al., 1999; Grambow et al., 1986). Furthermore, there are several factors which can have a large effect on the total content and composition of benzoxazinoids reported in plants: (i) plant species and cultivar (Copaja et al., 2006; Eljarrat and Barceló, 2001; Schulz et al., 2013; Zasada et al., 2007), (ii) plant age (Cambier et al., 1999; Köhler et al., 2015; Tanwir et al., 2017), (iii) tissue type (e.g. root, shoot) (Cambier et al., 1999; Copaja et al., 2006; Villagrasa et al., 2006), and (iv) growth conditions (e.g. fertilization) (Walker et al., 2012). Several induction methods have been described to elicit in planta production or diversification of benzoxazinoids. Abiotic elicitation methods include treatment with jasmonic acid (Oikawa et al., 2002), cis-jasmone (Moraes et al., 2008), chitin and chitosan oligomers, or copper chloride (Oikawa et al., 2001). Biotic elicitation methods include exposure to fungi (Ahmad et al., 2011; Oikawa et al., 2004), bacteria (Walker et al., 2012), insects, or insect larvae (Ahmad et al., 2011; Oikawa et al., 2004).

4. Reactivity of benzoxazinones and benzoxazolinones

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As previously described, benzoxazinones are stored as stable glycosides in the plant (**Sections 2.3** and **3.1**, and **Figure 3**). Upon cell damage or when exuded from roots, the glycosides come into contact with glycosidases which leads to the release of the less stable and more reactive benzoxazinone aglycones. The storage of defence compounds as glycosidic precursors which are released under stress conditions by glycosidase action is also observed in plants of other families. In Brassicaceae, for example, glucosinolates (precursors) are deglycosylated leading to the formation of isothiocyanates, which can react with thiols or amines contributing to

244 antimicrobial functionality and other bioactivities (Dufour et al., 2015). Benzoxazinone 245 aglycones can undergo a variety of reactions, as described in the following sections.

4.1. Oxo-cyclo tautomerism of benzoxazinone aglycons

Several mechanisms for the reactivity of ring-closed benzoxazinone aglycones have been suggested, such as metal-complexation by hydroxamic acids (Tipton and Buell, 1970). Another proposed mechanism is based on the electrophilicity of the nitrenium (N^+) ion which is formed when the N substituent acts as a leaving group (Hashimoto and Shudo, 1996; Wouters et al., 2016). Additionally, benzoxazinone aglycones can undergo oxo-cyclo tautomerism in which the heterocycle assumes its ring-opened configuration. In the ring-opened configuration, benzoxazinones possess an α -dicarbonyl moiety which can react with thiols and amines (Wouters et al., 2016). In addition, ring-opening can lead to transformation of benzoxazinones to benzoxazolinones, which is discussed in the next section.

4.2. Transformation of benzoxazinones to benzoxazolinones

One of the possible follow-up reactions of ring-opening of benzoxazinone aglycones is the spontaneous transformation into the corresponding benzoxazolinone (**Figure 3**) (Wouters et al., 2016). Several different multi-step mechanisms have been proposed for this transformation, all of which involve a deprotonation step and eventually lead to loss of the *N* substituent (R₁) (Wouters et al., 2016). The reactivity of aglycones towards this transformation is mainly determined by their *N* substituent which acts as the leaving group. Methyl derivatives (*N*-methoxy) are more reactive than hydroxamic acids (*N*-hydroxy), whereas lactams (*N*-hydro) are practically unreactive because H is not a suitable leaving group (Grambow et al., 1986; Wouters et al., 2016). Due to the deprotonation step, the reaction is pH dependent. It proceeds rapidly around neutral pH and even more quickly in alkaline conditions. Below pH 7, the degradation rate decreases, with a rate constant at pH 4 which is approximately 150-fold lower than that at pH 7, as demonstrated for DIBOA (Bredenberg et al., 1962).

Whether alternative pathways exist to produce benzoxazolinones *in planta* is unknown. The detection of benzoxazolinones, especially those with *N*-hydroxy and *N*-methoxy substituents (Wu et al., 2012) (**Section 2.1**, **Table 1**), in some plants might indicate the existence of pathways dedicated to the production of specific benzoxazolinones.

4.3. Transformation products of benzoxazolinones

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274 As demonstrated by the discovery of zeaoxazolinone in maize (Figure 2A) (Mohamed et al., 275 2014) and 5,5'-bis-benzoxazoline-2,2'-dione in A. ilicifolius (Figure 2B) (D'Souza et al., 1997), 276 benzoxazolinones can form dimers in planta. Surprisingly, zeaoxazolinone consists of two 7-277 methoxy-1,3-benzoxazol-2-one subunits, although the monomeric 7-methoxy derivative has 278 not yet been found (Table 1). The exact mechanism of dimer formation and the enzymes 279 involved needs to be further elucidated. 280 Benzoxazolinones can be detoxified by plants or insects using various mechanisms, e.g. 281 hydroxylation, and O or N glycosylation (**Table 1**) (Hofmann et al., 2006; Maag et al., 2014; 282 Schulz et al., 2016; Wieland et al., 1998). Furthermore, in soil, benzoxazolinones can undergo microbial transformations (Fomsgaard et al., 2004; Schulz et al., 2013; Wouters et al., 2016). 283 284 In the presence of bacteria or fungi, benzoxazolinones are converted to aminophenol 285 intermediates (e.g. 2-aminophenol), which can subsequently react to aminophenoxazinones 286 (e.g. 2-amino-phenoxazin-3-one) (Fomsgaard et al., 2004; Macías et al., 2009). Besides 287 aminophenoxazinones, other transformation products formed via aminophenol intermediates 288 are acetamides and malonamic acids (Fomsgaard et al., 2004; Schulz et al., 2013). These transformation products, however, have lost their benzoxazinoid structural motif and are 289 290 therefore not considered in more detail in this work.

5. In planta functionality of benzoxazinoids

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292 Allelopathic functionality of benzoxazinoids has been well-established and is one of the main 293 functions of these compounds. Benzoxazinoid levels are especially high in roots and root 294 exudates of young maize, wheat, and rye, as they compete with other plants during early growth 295 (Niemeyer, 2009; Schulz et al., 2013). General structure-activity relationships observed for 296 allelopathic or phytotoxic activity are that hydroxamic acids (usually represented by DIBOA 297 and DIMBOA) are more potent than lactams (usually represented by HBOA and HMBOA). 298 The benzoxazolinones BOA and MBOA are in between hydroxamic acids and lactams in terms 299 of allelopathic potency. DIBOA and DIBOA-Glc typically seem to be the most potent 300 allelopathic natural benzoxazinoids, exhibiting significant root length inhibition at concentrations between 10⁻³ to 10⁻⁴ M (depending on target species) (Macías et al., 2005, 301 302 2006a). Interestingly, glycosylation of DIBOA does not seem to affect its activity (Macías et 303 al., 2005, 2006a; Wouters et al., 2016). It was also shown to be possible to enhance selectivity 304 and potency of the phytotoxic activity by chemical modification of 4-hydroxy-1,4-benzoxazin-305 3-one (Macías et al., 2010; Macías et al., 2006b). 306 Besides allelopathy, benzoxazinoids also play a role as defence compounds against insects and 307 microorganisms. In general, accumulation of benzoxazinoids is positively correlated with 308 resistance against disease and insects in maize, wheat, and rye (Ahmad et al., 2011; Copaja et 309 al., 2006; Meihls et al., 2013; Niemeyer, 2009; Søltoft et al., 2008). As with allelopathy, 310 hydroxamic acids seem to be more potent than lactams (Wouters et al., 2016). Moreover, 311 benzoxazinones with increased methylation (i.e. N-OCH₃ or C7-OCH₃) were found to be 312 correlated with disease resistance (Søltoft et al., 2008). In another study, it was suggested that 313 methylation of DIMBOA-Glc to HDMBOA-Glc might be a mechanism by which MBOA 314 production is accelerated, as HDMBOA will more quickly transform to MBOA than DIMBOA 315 (Oikawa et al., 2004). In that study, they observed increased concentrations of HDMBOA-Glc

in maize upon fungal infection and larval feeding. Additionally, MBOA was shown to be more effective at inhibiting fungal germination than its precursors (Oikawa et al., 2004).

Wouters and co-workers postulated that the capacity to undergo ring-opening increased activity of benzoxazinones against insects. Consequently, glycosides were found to be less active than aglycons (Wouters et al., 2016). The toxicity of MBOA to insects was lower than that of DIMBOA (Wouters et al., 2016), suggesting that insecticidal activity is probably conferred by benzoxazinones via another mechanism (**Section 4**) than antifungal activity.

These defence functionalities associated with benzoxazinoids indicate that they might be potentially interesting leads for antimicrobial compounds. As an example, the antibiotic ofloxacin shares structural features (1,4-benzoxazin ring system) with natural benzoxazinoids (Hayakawa et al., 1986).

6. Antimicrobial potential of natural benzoxazinoids

One of the first reports on the antimicrobial potential of benzoxazinoids described the inhibition of the gram-negative bacterium *Xanthomonas stewartii* (i.e. *Pantoea stewartii*), the causal agent of Stewart's wilt, by MBOA from maize (Whitney and Mortimore, 1961). A more recent study indicated that DIMBOA also possesses antibacterial properties against the pathogenic bacterium *Staphylococcus aureus*, albeit at high concentrations (Gleńsk et al., 2016). The novel benzoxazinoid zeaoxazolinone (**Figure 2A**) showed fungal inhibition comparable to clotriamazole against amongst others *Aspergillus flavus* and *Candida albicans* (Mohamed et al., 2014). Unfortunately, no MIC values were reported for this dimeric benzoxazinoid. The antifungal and possibly antibacterial potential of this promising compound should be further investigated. The antimicrobial activity of 5,5'-bis-benzoxazoline-2,2'-dione (**Figure 2B**) has not yet been evaluated.

Despite the extensive research on benzoxazinoids, only a limited number of papers evaluated

activity of these compounds against human pathogens, commonly represented by S. aureus

(gram-positive bacterium), Escherichia coli (gram-negative bacterium), and C. albicans (yeast, fungus). Three papers by Bravo, Lazo and co-workers (Bravo et al., 1997; Bravo and Lazo, 1993, 1996) and one other paper (Gleńsk et al., 2016) reported minimum inhibitory concentration (MIC) values of benzoxazinoids against these three pathogens (Table 2). The antifungal activity of these compounds has been studied more extensively than their antibacterial activity. MBOA and DIMBOA (MIC 450 and 500 µg mL⁻¹, respectively) are most effective against fungi. As expected, the antifungal activity of benzoxazinoid glucosides (i.e. detoxified derivatives) is very limited (MIC >1000 µg mL⁻¹) (Bravo and Lazo, 1996). MIC values against bacteria have only been reported for the most common natural benzoxazinoids, namely the benzoxazinone aglycons DIMBOA and DIBOA, and the benzoxazolinones BOA and MBOA. Overall, DIMBOA is the natural benzoxazinoid with the lowest reported MIC values (Table 2). Our general observations regarding antimicrobial activity of monomeric natural benzoxazinoids are that: (i) hydroxamic acids are more active than lactams, (ii) hydroxylation at C7 increases activity, and (iii) glycosylation decreases activity. This is in accordance with a previous review (Wouters et al., 2016). Interestingly, the second and third observation are in contrast with findings for allelopathy (phytotoxicity) where these factors do not seem to affect activity (Section 5). This suggests that allelopathy, antimicrobial, and insecticidal activity are not necessarily conferred by the same structural motifs. Nevertheless, it seems that hydroxamic acids are more active than lactams with regard to all three of these functionalities. Even DIMBOA, the most antimicrobial monomeric natural benzoxazinoid, has relatively poor activity (lowest reported MIC 250 µg mL⁻¹ against S. aureus) (Gleńsk et al., 2016) compared to other natural compounds, such as flavonoids (e.g. MIC 3.9-15.6 µg mL⁻¹ of apigenin against S. aureus) (Gibbons, 2004) and prenylated isoflavonoids (e.g. MIC 3.13 µg mL⁻¹ of licoricidin against S. aureus) (Gibbons, 2004).

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366 Interestingly, the benzoxazolinone transformation product 2-amino-phenoxazin-3-one was selectively very potent as an antimicrobial compound against the species *Helicobacter pylori* (MIC 1 μ g mL⁻¹) (Hanawa et al., 2010), some *Mycobacterium* spp. (MIC 2.8 – 11.3 μ g mL⁻¹) 368 (Shimizu et al., 2004), and *Chlamydia pneumoniae* (MIC 2.1 µg mL⁻¹) (Uruma et al., 2005). 369 Activity of this compound against many other bacteria (e.g. E. coli and S. aureus) was limited (Atwal et al., 1992; Shimizu et al., 2004). 372 Overall, the antimicrobial potential of monomeric natural benzoxazinoids seems to be limited, yet several known antifungals such as carbendazim and boscalid contain the NHCO structure 374 (in open chain configuration) coupled to an aromatic system (Śmist et al., 2016). This suggests that structural modification of benzoxazinoids might enhance their antimicrobial activity.

7. Potential of the 1,4-benzoxazin-3-one scaffold

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In addition to their work on natural benzoxazinoids, Bravo and Lazo also evaluated some synthetic derivatives of these compounds. They showed that some synthetic 1,4-benzoxazin-3one hydroxamic acids lacking the C2 hydroxyl group possessed more antifungal potential against C. albicans than DIMBOA (Bravo and Lazo, 1993, 1996). These studies first demonstrated the potential of benzoxazinoids as scaffolds for the design of antimicrobial compounds. The 1,4-benzoxazin-3-one scaffold in particular, typically without the hydroxamic acid motif conferred by the N-OH group, was further utilized in several other studies in which more extensive modifications were applied. One study showed that three synthetic benzoxazinones, with a combination of a C8 chloro and a C2 alkane (n-propyl, n-butyl, n-pentyl) substituent, had antifungal activity comparable to fluconazole (Özden et al., 2000). Another study found two other benzoxazinone derivatives, namely a C2 ethyl and N-acetyl benzoxazinone, with promising antifungal activity against a variety of phytopathogenic fungi (Śmist et al., 2016). Unfortunately, no MIC values were determined in these studies.

Several other studies do, however, report MIC values for synthetic 1,4-benzoxazin-3-ones and these values are up to 40-fold lower than those reported for DIMBOA (Figure 5). For example, a number of compounds with N-alkyl and C7 halogen substituents were found to possess good activity against several pathogenic bacteria but limited antifungal activity (Fang et al., 2011). In addition, two other studies describe multiple compounds with good activity against pathogenic bacteria as well as fungi (Alper-Hayta et al., 2006; Yalcin et al., 2003). An example of an all-round active molecule is 2-(2-ethyloxy-2-oxo-ethyl)-6-chloro-7-nitro-1,4-benzoxazin-3-one with MIC 12.5 µg mL⁻¹ against C. albicans, and MIC 25 µg mL⁻¹ against S. aureus and E. coli (Figure 5) (Alper-Hayta et al., 2006). Most of the antimicrobial synthetic 1,4benzoxazin-3-one derivatives do not possess a hydroxyl-moiety at C2 and can therefore not undergo ring-opening. Neither is the formation of a nitrenium ion or metal complexation very likely, considering their structure. Establishing structure-activity relationships for these antimicrobial compounds might help in elucidating their antimicrobial mode-of-action. Overall, synthetic benzoxazinoids based on a 1,4-benzoxazin-3-one scaffold show promising antimicrobial activity which might be further enhanced by targeted modifications based on previously reported findings. Considering the general observation that hydroxamic acids are more reactive and bioactive than lactams, perhaps its 4-hydroxy derivative would be an even more potent scaffold which was already used to produce synthetic allelopathic compounds (Macías et al., 2010).

8. Conclusions

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In this review, we provided an overview of all 32 currently known natural benzoxazinoids (excluding *C*2 glycosides), which includes one dimeric benzoxazolinone. The current state of knowledge on the biosynthesis of these compounds in Poaceae was summarised and a gap in the knowledge on the pathways responsible for lactam biosynthesis was identified. Within the family Poaceae, maize, wheat, and rye are the major benzoxazinoid producers, differences in

416	the benzoxazinoid diversity between these species suggests a more advanced development of
417	the biosynthesis in maize than in the other two species. The subclass (N substituent) of
418	benzoxazinones amongst others affects their allelopathy and defence functionality.
419	We evaluated the antimicrobial activity of natural benzoxazinoids based on MIC values
420	reported in literature. Even though these compounds play an important role in plant defence
421	and allelopathy, monomeric natural benzoxazinoids seem to lack potency as antimicrobial
422	agents. Observations on the structural requirements for allelopathy, insecticidal, and
423	antimicrobial activity suggest that they are not necessarily conferred by similar mechanisms.
424	The 1,4-benzoxazin-3-one backbone has been shown to be a potential scaffold for designing
425	new antimicrobial compounds with activity against pathogenic bacteria and fungi. This has
426	been demonstrated by a number of studies that report potent activity (MIC down to 6.25-16 μg
427	mL ⁻¹ against C. albicans, E.coli, and S. aureus) of synthetic derivatives of 1,4-benzoxazin-3-
428	one.

Conflict of interest

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The authors declare no conflict of interest regarding this research.

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712 **Figures and tables**

713 Benzoxazinone Benzoxazolinone

Figure 1. The backbones of the two classes of benzoxazinoids: benzoxazinones (1,4-715 benzoxazin-3-one) and benzoxazolinones (1,3-benzoxazol-2-one). R₁ substituent determines 716 subclass, see **Table 1**. **Table 1.** Overview of monomeric natural benzoxazinoids of the classes benzoxazinones (1,4-benzoxazin-3-one backbone with *C*2 hydroxylation) and benzoxazolinones (1,3-benzoxazol-2-one backbone) reported in literature, with their systematic name and abbreviation. All benzoxazinones are displayed as aglycones. Note that some of these compounds have, so far, only been detected as their 2-*O*-glycosides. n.a., no abbreviation assigned in literature.

Benzoxazinones	R ₂	R ₃	R ₄	R ₅	Systematic name	Abbreviation ^a	Natural source b	Ref
N-hydro (R ₁ = H): lactams								
							Poaceae	(Hofman and Hofmanová, 1969; Tanwir et al., 2013
							Acanthaceae	(Baumeler et al., 2000; Pratt et al., 1995
	Н	Н	Н	Н	2-hydroxy-1,4-benzoxazin-3-one	HBOA	Calceolariaceae	(Bravo et al., 2005
							Plantaginaceae	(Pratt et al., 1995
							Lamiaceae	(Alipieva et al., 2003
R ₅							Poaceae	(de Bruijn et al., 2016; Woodward et al., 1979
R ₄ . 8 10. 2 OH	Н	Н	ОН	Н	2,7-dihydroxy-1,4-benzoxazin-3-one	DHBOA	Acanthaceae	(Bravo et al., 2004
2 011					, , , , , , , , , , , , , , , , , , , ,		Calceolariaceae	(Bravo et al., 2005
1 I I							Lamiaceae	(Alipieva et al., 2003
							Poaceae	(Cambier et al., 1999; Tanwir et al., 2013
R_3 6 $\searrow 5$ 4 N 3 $\searrow 0$	Н	Н	OCH ₃	Н	2-hydroxy-7-methoxy-1,4-benzoxazin-3-one	НМВОА	Acanthaceae	(Baumeler et al., 2000; Pratt et al., 1995
l I							Plantaginaceae	(Pratt et al., 1995
N ₂ N ₁	Н	Н	OCH ₃	OCH₃	2-hydroxy-7,8-dimethoxy-1,4-benzoxazin-3-one	HM₂BOA	Poaceae	(Cambier et al., 1999; Hofman and Masojídková 1973
	Cl	Н	OCH ₃	Н	5-chloro-2-hydroxy-7-methoxy-1,4-benzoxazin-3-one	Cl-HMBOA	Poaceae	(Le-Van and Wratten, 1984
	Н	Н	Cl	Н	7-chloro-2-hydroxy-1,4-benzoxazin-3-one	n.a.	Acanthaceae	(Kanchanapoom et al., 2001a
	ОН	Н	Н	Н	2,5-dihydroxy-1,4-benzoxazin-3-one	n.a.	Acanthaceae	(Kanchanapoom et al., 2001a
	Н	ОН	Н	Н	2,6-dihydroxy-1,4-benzoxazin-3-one	n.a.	Lamiaceae	(Alipieva et al., 2003
N-hydroxy (R1 = OH): hydro	oxamio	acids						
, , , , , , , , , , , , , , , , , , , ,							Poaceae	(Cambier et al., 1999; Tanwir et al., 2013
							Acanthaceae	(Baumeler et al., 2000; Bravo et al., 2004
					2.4.4% days 4.4.6 and 2.2.4.	DIDOA	Ranunculaceae	(Özden et al., 1992
	Н	Н	Н	Н	2,4-dihydroxy-1,4-benzoxazin-3-one	DIBOA	Calceolariaceae	(Bravo et al., 2005
							Plantaginaceae	(Pratt et al., 1995
							Lamiaceae	(Alipieva et al., 2003
	Н	Н	ОН	Н	2,4,7-trihydroxy-1,4-benzoxazin-3-one	TRIBOA	Poaceae	(Woodward et al., 1979
	п	П	ОП	П	2,4,7-tilliyul0xy-1,4-beli20xd2iii-5-0ile	INIBUA	Acanthaceae	(Pratt et al., 1995)
							Poaceae	(Cambier et al., 1999; Tanwir et al., 2013)
	Н	Н	OCH ₃	Н	2,4-dihydroxy-7-methoxy-1,4-benzoxazin-3-one	DIMBOA	Acanthaceae	(Baumeler et al., 2000; Pratt et al., 1995)
							Plantaginaceae	(Pratt et al., 1995
	Н	Н	ОН	OCH ₃	2,4,7-trihydroxy-8-methoxy-1,4-benzoxazin-3-one	TRIMBOA	Poaceae	(Handrick et al., 2016
	Н	Н	OCH₃	OCH ₃	2,4-dihydroxy-7,8-dimethoxy-1,4-benzoxazin-3-one	DIM₂BOA	Poaceae	(Cambier et al., 1999; Niemeyer, 2009
	Н	Н	Cl	Н	7-chloro-2,4-dihydroxy-1,4-benzoxazin-3-one	7-Cl-DIBOA	Acanthaceae	(Kanchanapoom et al., 2001b
N-methoxy (R₁= OCH₃) : me	thyl d	erivative	25					
	Н	Н	Н	Н	2-hydroxy-4-methoxy-1,4-benzoxazin-3-one	4- <i>O</i> -Me-DIBOA	Poaceae	(de Bruijn et al., 2016; Oikawa et al., 2002
	Н	Н	OCH₃	Н	2-hydroxy-4,7-dimethoxy-1,4-benzoxazin-3-one	HDMBOA	Poaceae	(Cambier et al., 1999; Tanwir et al., 2013)

Benzoxazolinones	R ₂	R ₃	R ₄	R ₅	Systematic name	Abbreviation ^a	Natural source b	Ref.
N-hydro (R 1 = H)								
	Н	Н	Н	Н	1,3-benzoxazol-2-one	воа	Poaceae Acanthaceae Ranunculaceae Calceolariaceae Plantaginaceae	(Cambier et al., 1999; Hanhineva et al., 2011) (Bravo et al., 2004; Pratt et al., 1995) (Özden et al., 1992) (Bravo et al., 2005) (Pratt et al., 1995)
D	Н	Н	ОН	Н	6-hydroxy-1,3-benzoxazol-2-one	BOA-6-OH	PDP ^c	(Schulz and Wieland, 1999; Wieland et al., 1998)
κ ₅ Ι	Н	ОН	Н	Н	5-hydroxy-1,3-benzoxazol-2-one	BOA-5-OH	PDP	(Hofmann et al., 2006)
R_4 $\frac{7}{10}$	ОН	Н	Н	Н	4-hydroxy-1,3-benzoxazol-2-one	n.a.	Acanthaceae	(Huo et al., 2005; Zhao et al., 2015)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Н	NO ₂	ОН	Н	5-nitro-6-hydroxy-1,3-benzoxazol-2-one	Nitro-BOA-6-OH	MDP ^c	(Schulz et al., 2017)
R ₃ 5	Н	Н	OCH ₃	Н	6-methoxy-1,3-benzoxazol-2-one	МВОА	Poaceae Acanthaceae Plantaginaceae	(Cambier et al., 1999; Niemeyer, 2009) (Pratt et al., 1995) (Pratt et al., 1995)
R_2	Н	Н	OCH ₃	OCH₃	6,7-dimethoxy-1,3-benzoxazol-2-one	M ₂ BOA	Poaceae	(Anai et al., 1996; Cambier et al., 1999)
	Н	Cl	OCH ₃	Н	5-chloro-6-methoxy-1,3-benzoxazol-2-one	CI-MBOA	Poaceae	(Kato-Noguchi et al., 1998)
	Cl	Н	OCH ₃	OCH₃	4-chloro-6,7-dimethoxy-1,3-benzoxazol-2-one	Cl-M ₂ BOA	Poaceae	(Anai et al., 1996)
	Н	Н	OGlc	Н	6-β-D-glucopyranosyloxy-1,3-benzoxazol-2-one	BOA-6- <i>O</i> -Glc	PDP	(Hofmann et al., 2006; Wieland et al., 1998)
	Н	OGlc	Н	Н	5-β-D-glucopyranosyloxy-1,3-benzoxazol-2-one	BOA-5- <i>O</i> -Glc	PDP	(Hofmann et al., 2006)
	COCH ₃	Н	Н	Н	4-acetyl-1,3-benzoxazol-2-one	4-ABOA	Poaceae	(Fielder et al., 1994)
N-hydroxy (R1 = OH)								
	Н	Н	OCH₃	Н	3-hydroxy-6-methoxy-1,3-benzoxazol-2-one	3-OH-MBOA	Plantaginaceae	(Wu et al., 2012)
N-methoxy (R1 = OCH3)								
	Н	Н	OCH₃	Н	3,6-dimethoxy-1,3-benzoxazol-2-one	DMBOA	Plantaginaceae	(Wu et al., 2012)
N-glucosyl (R1 = Glc)								
	Н	Н	OCH₃	Н	3-β-D-glucopyranosyl-6-methoxy-1,3-benzoxazol-2-one	MBOA- <i>N</i> -Glc	IDP ^c PDP	(Maag et al., 2014) (Hofmann et al., 2006; Wieland et al., 1998)

^a Most commonly used abbreviation reported in literature, for some compounds multiple different abbreviations are in use.

^b Natural source (plant family or detoxification product) in which the benzoxazinoid has been detected. Note that genera Scoparia and Calceolaria, formerly members of Scrophulariaceae, have been moved to Plantaginaceae and Calceolariaceae, respectively.

^c PDP, plant detoxification product; IDP, insect detoxification product; MDP, microbial detoxification product.

724

725 **Figure 2.** Structures of zeaoxazolinone (A), a 7-methoxy-1,3-benzoxazol-2-one dimer from

726 Zea mays (Mohamed et al., 2014), and 5,5'-bis-benzoxazoline-2,2'-dione (**B**), a 1,3-benzoxazol-

727 2-one dimer from *Acanthus ilicifolius* (D'Souza et al., 1997).

Benzoxazinones

Benzoxazolinones

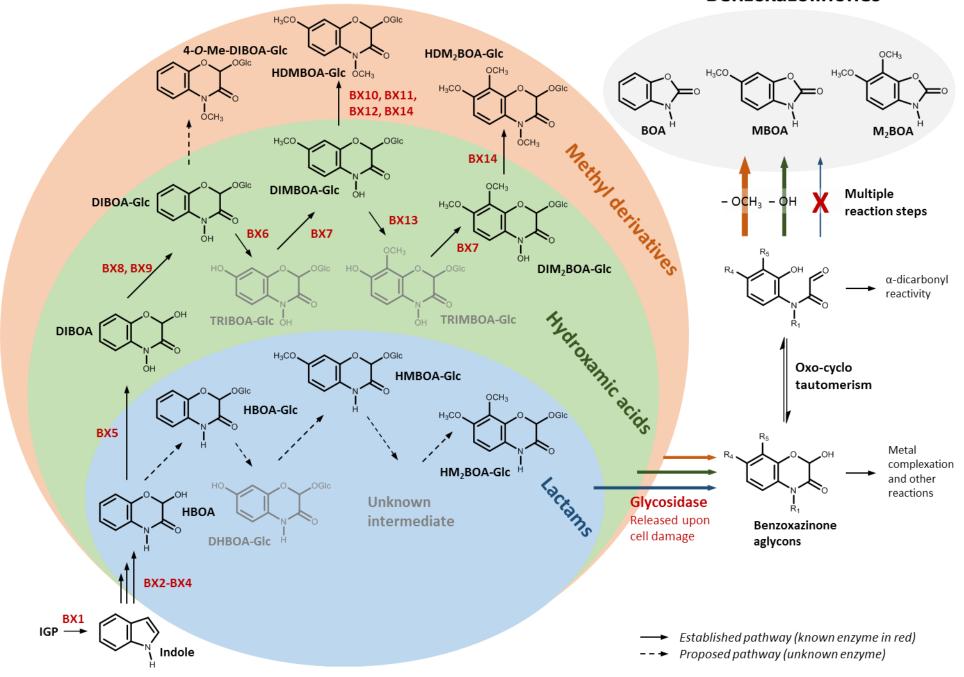


Figure 3. The biosynthetic pathway of the three subclasses of benzoxazinones in plants (Dutartre et al., 2012; Frey et al., 1997; Handrick et al., 2016; Jonczyk et al., 2008; Meihls et al., 2013; Niemeyer, 2009; Oikawa et al., 2002) and their transformation into benzoxazolinones (Wouters et al., 2016). Structures of biosynthetic intermediates are shown in grey. Different mechanisms have been proposed for the multi-step degradation from the ring-opened configuration of benzoxazinones to the corresponding benzoxazolinone (Wouters et al., 2016). IGP, indole-3-glycerol phosphate.

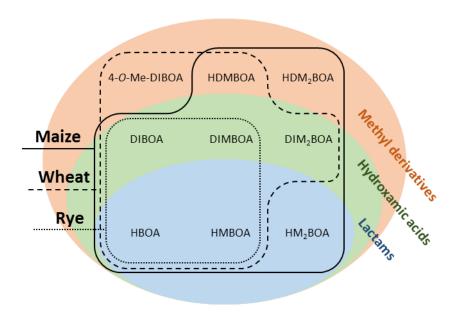


Figure 4. Distribution of individual benzoxazinones amongst the three major benzoxazinoid producing species of the family Poaceae (Cambier et al., 1999; de Bruijn et al., 2016; Le-Van and Wratten, 1984; Søltoft et al., 2008; Tanwir et al., 2013; Woodward et al., 1979). The biosynthetic intermediates DHBOA, TRIBOA, and TRIMBOA are not included.

Table 2. Antimicrobial activity of natural benzoxazinoids against the pathogens *S. aureus* (gram-positive bacterium), *E. coli* (gram-negative bacterium) and *C. albicans* (yeast, fungus).
 Colour of abbreviation indicates benzoxazinone subclass, green is hydroxamic acid whereas
 blue is lactam.

		Antimicro	obial activity,	_			
Abbreviation	Class	S. aureus	E. coli	C. albicans	Ref.		
DIMBOA	Benzoxazinone	250 - 500	666	500	(Bravo and Lazo, 1993, 1996; Gleńsk et al., 2016)		
DIBOA	Benzoxazinone	500	1250	666	(Bravo and Lazo, 1993, 1996)		
MBOA	Benzoxazolinone	>1000	>1000	450	(Bravo et al., 1997)		
BOA	Benzoxazolinone	>1000	>1000	650	(Bravo et al., 1997)		
DIBOA-2-O-Glc	Benzoxazinone	n.a.	n.a.	>1000	(Bravo and Lazo, 1996)		
DIMBOA-2-O-Glc	Benzoxazinone	n.a.	n.a.	>1000	(Bravo and Lazo, 1996)		
HMBOA	Benzoxazinone	n.a.	n.a.	1000	(Bravo and Lazo, 1996)		
HBOA	Benzoxazinone	n.a.	n.a.	>1000	(Bravo and Lazo, 1996)		

744745

746 747 n.a., no data available.

^a Minimum inhibitory concentration, range indicates that different values were reported in literature.

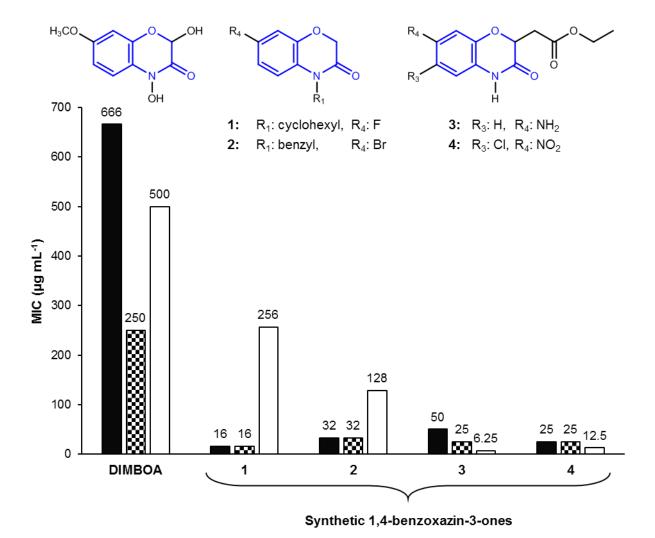


Figure 5. Antimicrobial activity of the most potent monomeric natural benzoxazinone DIMBOA (Bravo and Lazo, 1993; Gleńsk et al., 2016) compared to four synthetic benzoxazinones (Alper-Hayta et al., 2006; Fang et al., 2011). The 1,4-benzoxazin-3-one scaffold is shown in blue. Minimum inhibitory concentration (MIC) displayed in μg mL⁻¹ against gram-negative bacterium *E. coli* (black), gram-positive bacterium *S. aureus* (checkered), and fungus *C. albicans* (white).