

THEORY AND APPLICATION OF LANDFARMING TO REMEDIATE PAHS AND MINERAL OIL CONTAMINATED SOILS AND SEDIMENTS

Joop Harmsen

Proland Workshop Pulawy, Poland October 27 28, 2005

WAGENINGENUR

Sediments are not only harbour sediments











<u>Contaminants</u>

Polycyclic Aromatic Hydrocarbons (PAHs)
 biodegradable
 Mineral oil
 biodegradable



Experiment Kreekraksluizen



- 14 years of measurement
- Intensive landfarming (cultivation)
- Passive landfarming (vegetation)



intensive



Results landfarm

Sediment	PAHs (mg/kg d.m.)		Mineral oil (mg/kg d.m.)	
	start	2004	start	2004
Petroleum Harbour 1994	550	22	13,500	1300
Wemeldinge 1994	45	25	2000	500
Geul Harbour 1990	52	2	8100	<200
Zierikzee 1990	65	15	630	<200

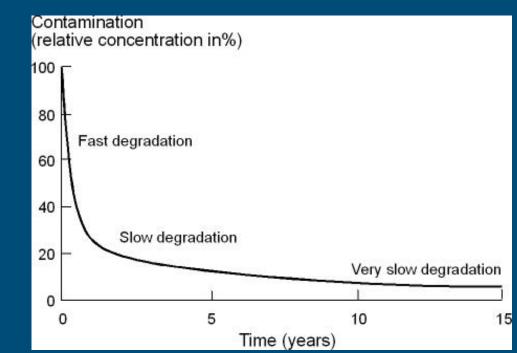


Biological degradation

First step for sediments = dewatering

Three degradable fractions

- fast
- slow
- very slow



3 6 rings





PAHs distribution

degradation 🔶 fast

fast

slow

very slow

= diffusion to water phase

= in equilibrium with water phase

= very slow diffusion to water phase

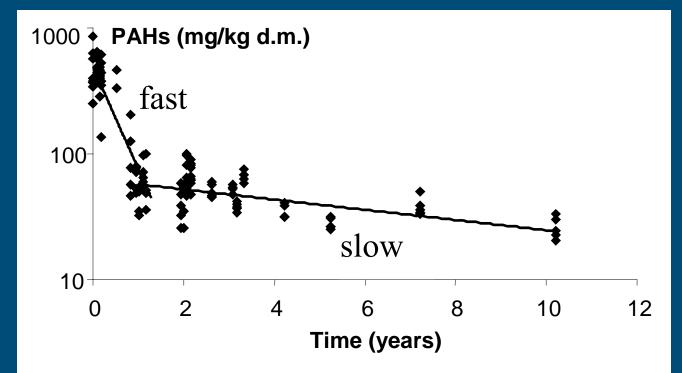
slow

very slow





Degradation of PAHs



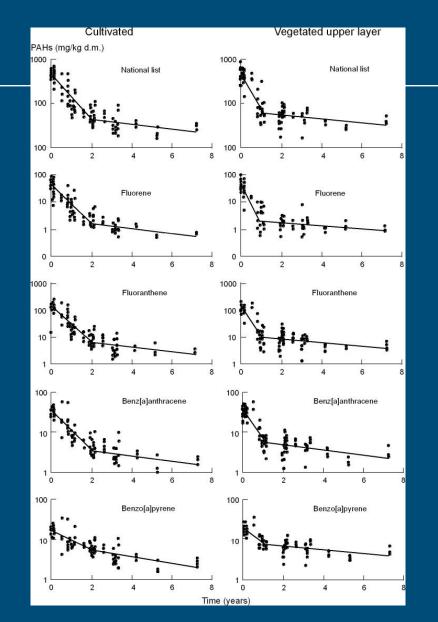
(Dutch National List) in Petroleum Harbour sediment (upper layer of vegetated sediment)



Degradation of PAHs

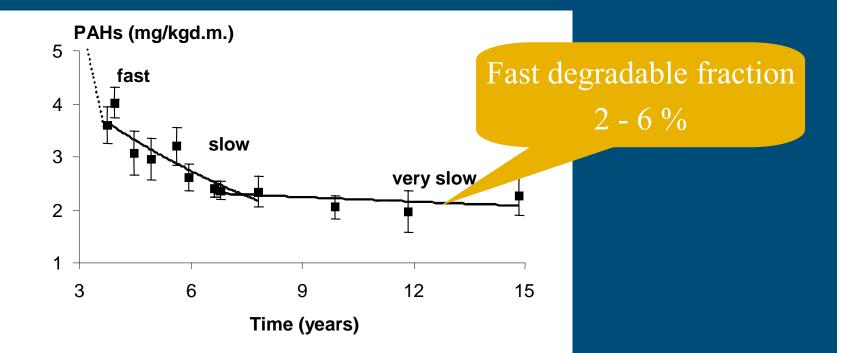
Fluorene

- Fluoranthene
- Benz[a]anthracene
- Benzo[a]pyrene





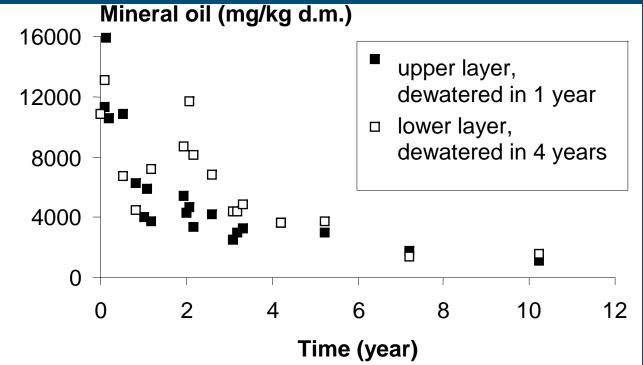
Degradation of PAHs



Geul Harbour sediment. 95% confidence values are given.

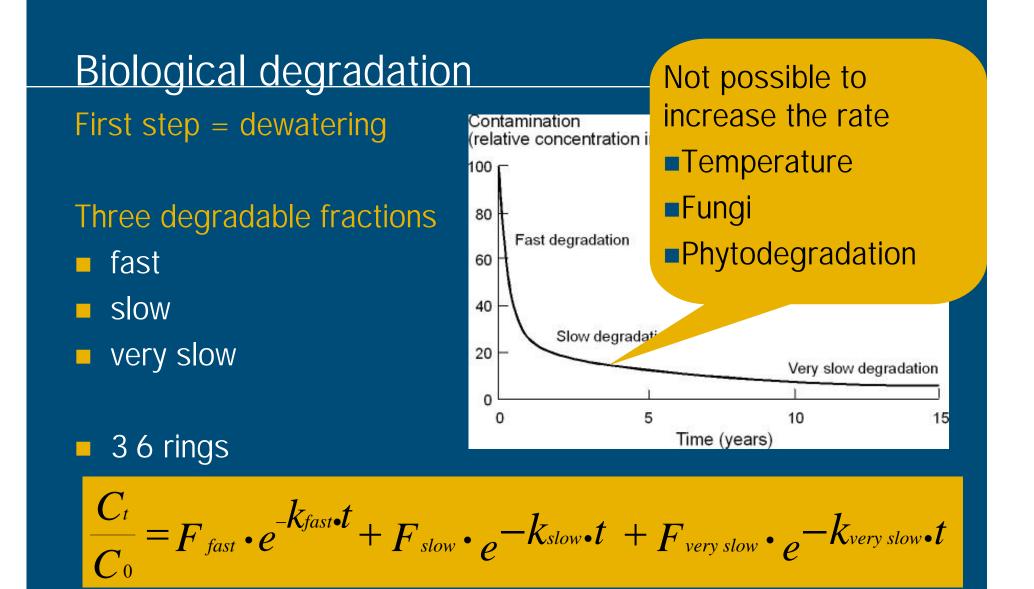


Degradation of mineral oil



Vegetated Petroleum Harbour landfarm (upper layer and lower layer).

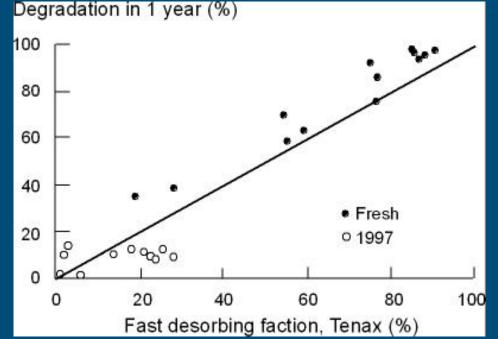






Measured availability for predicting degradation

- Prediction of fast degradation = fast available fraction
 - Solid phase (Tenax)
 - Mild extract (Hac)
 - Mild oxidation (Cuypers)
 - XAD (Northcot Jones, Lie)
 - SFE (Hawthorne, Loibner)
 - Dextrine (Reid, Doick)
 - Mild extract (Thiele and Brunner)



Prediction of very slow available fraction is possible



Theoretical considerations

Pollutants biodegradable at aerobic conditions

- Passive landfarming
 - Minimal interventions
 - Dewatering for sediments
 - Biodegradation
- Potential limiting factors
 - Availability of appropriate micro organisms No problem
 - Bioavailability of the pollutants to the micro organisms
 - Supply of oxygen for the biodegradation process



Bioavailability of pollutants: non equilibrium approach

Mass transfer resistance in the sediment aggregate

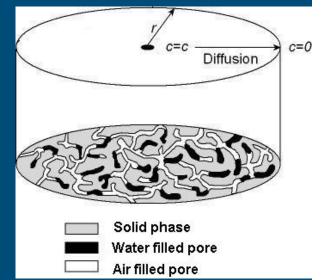
Simplified transport model

$$\frac{\mathrm{dc}}{\mathrm{dt}} = -\mathrm{kc}$$

k = first order kinetic rate constant

c = average contaminant concentration

$$\frac{c_t}{c_0} = e^{-k \cdot t}$$





Rate constant k for spherical particles

$$k = \frac{6 D_s^w}{[\theta + \rho_s (1 - \varepsilon) K_d]r^2}$$

- θ = internal aggregate moisture content
- ε = volume fraction of total pore space
 - r = distance from the surface to the centre
 - ρ_s = specific weight of the wet sediment
- D_s^W = average contaminant diffusion coefficient in sediment pore water

$$K_d$$
 = partition coefficient

$$D_s^w = \eta \ \theta \ D^w$$

 η = tortuosity factor = function of θ

 D_w = diffusion coefficient of contaminant in water



Simplified expression for the rate constant

$$k \sim \frac{\theta^2}{(1 - \epsilon)r^2}$$

- Comparison slurry reactor with fresh sediment and unsaturated landfarm
 - slurry ($\epsilon = 0.7$, pores 100% filled with water)
 - landfarm (ϵ = 0.55, pores 20 to 50 % filled with water

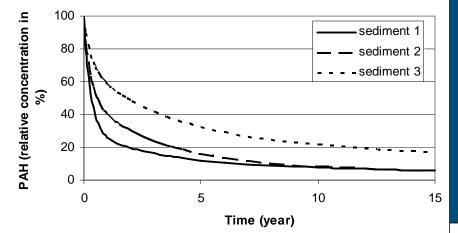
• r_{slurry} < r_{landfarm} Factor 2?

$$k_{slurry}/k_{landfarm} = 12$$
 76

Found for slow degradation k_{slurry (tenax)}/k_{landfarm} = 49
 To increase rate r < 0.1 1mm



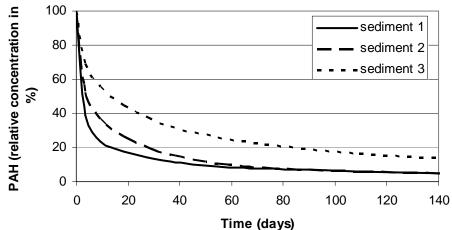
Landfarm versus bioreactor



Available fractions

	fast	slow	very slow
Sed 1	0.7	0.2	0.1
Sed 2	0.5	0.4	0.1
Sed 3	0.3	0.4	0.3

Landfarm



Bioreactor (slurry)



Concentration of pollutants in water phase

Pollutants in water phase are available for biodegradation

Partition coefficient between sediment and water phase

$$K_{d} = \frac{1000 \text{ fr}_{oc} K_{oc}}{1 + \text{fr}_{DOC} K_{oc}^{DOC}}$$

- = mass fraction of organic carbon in kg/kg
- = mass fraction of dissolved organic carbon
- = partition coefficient organic carbon and water
 - = partition coefficient between dissolved organic matter and water

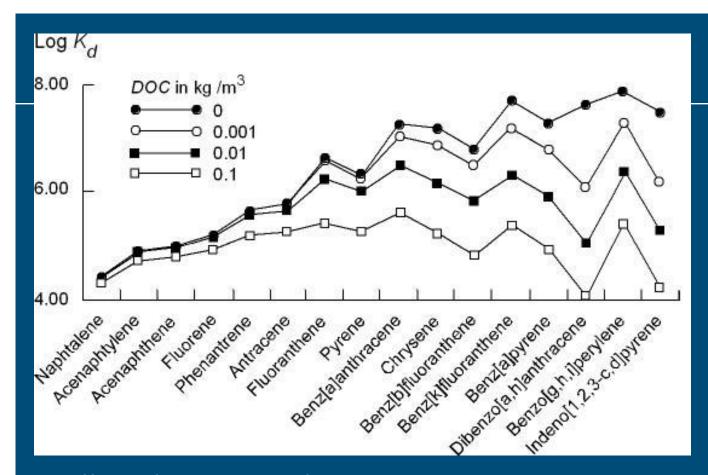


• fr_{oc}

• fr_{DOC}

• K_{oc}

• K_{oc}

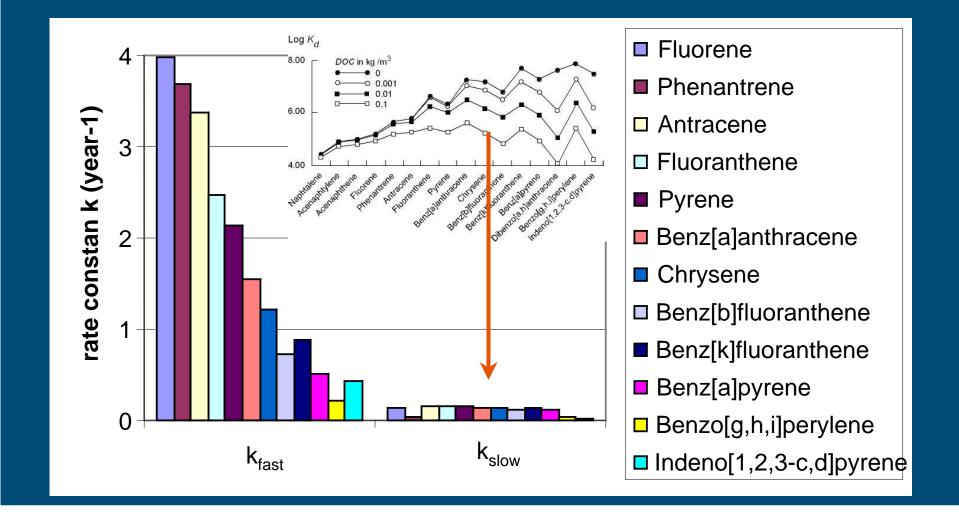


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$$k = \frac{6 D_s^w}{[\theta + \rho_s (1 - \varepsilon) K_d]r^2}$$



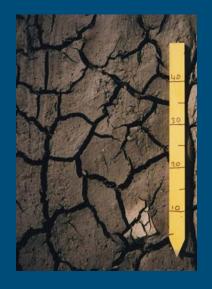
Degradation coefficients





Oxygen transport

- Landfarm of sediment: dewatering and ripening
 - Compaction of the sediment layer,
 - Development of cracks between aggregates,
 - Aeration of smaller aggregates,
 - Finally a soil with vegetation.





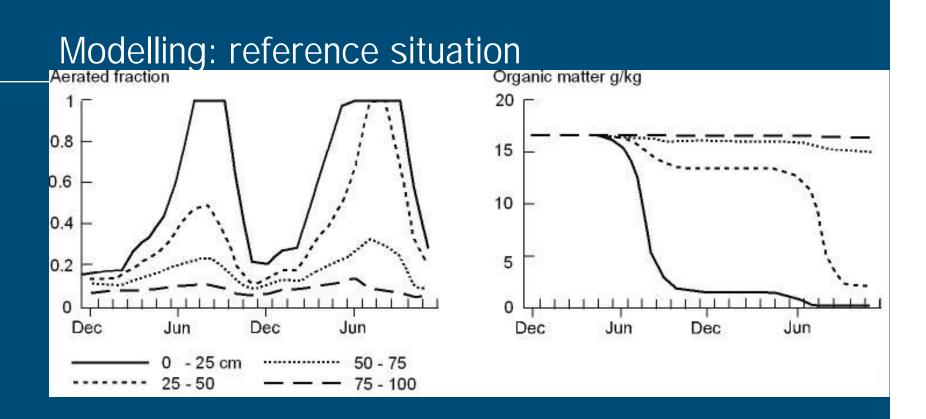


Modelling of aeration in a landfarm

Relevant physical and microbiological processes and parameters

- Formation of cracks and aeration through cracks as a result of shrinking
- Sediment moisture profile as a function of relevant properties of the sediment, presence of vegetation and period of the year
- Size of aggregates as a function of moisture potential
- Oxygen consumption due to degradation of organic matter (fast, slow and very slow degradable fractions)
- Oxygen diffusion in aggregates
- Variation of temperature during the year
- Dependency of the activity of micro organisms on temperature

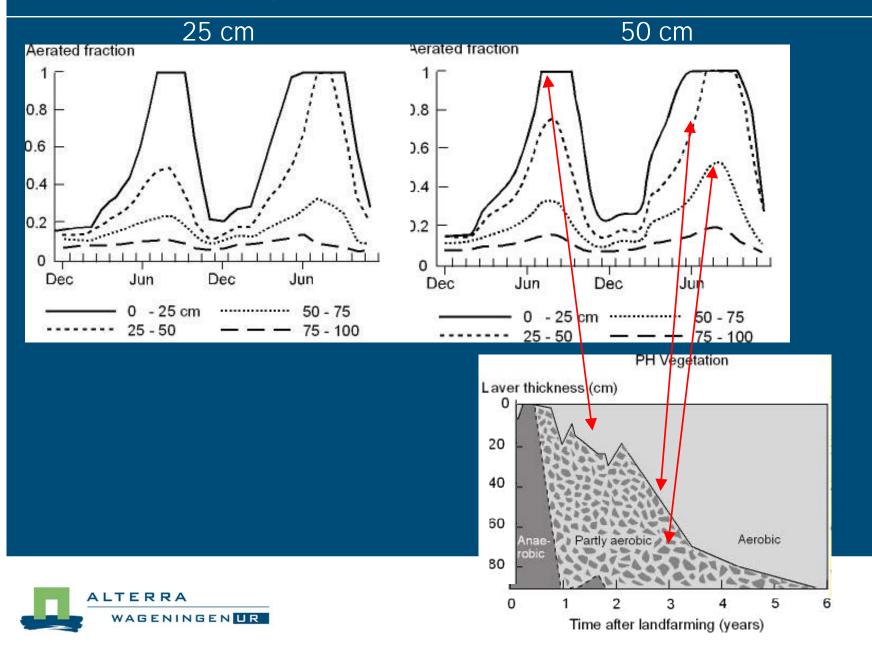




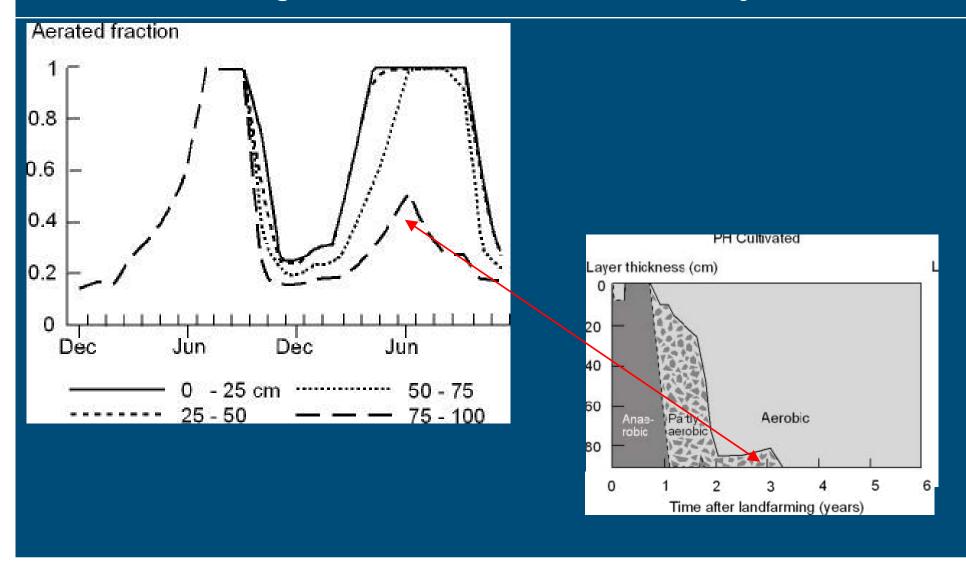
Aeration and degradation of fast degradable organic matter using the landfarm aeration model (reference situation: no cultivation and vegetation in the second year)



Modelling: root depth



Modelling: Intensivation in the second year





Results landfarm aeration model

Amount 'Soft' Organic matter responsible for oxygen consumption

- Increase of temperature during dewatering not effective (Too high oxygen consumption)
- Vegetation effective (Higher air filled pore volume at rooted depth)



Role of vegetation (phytoremediation)

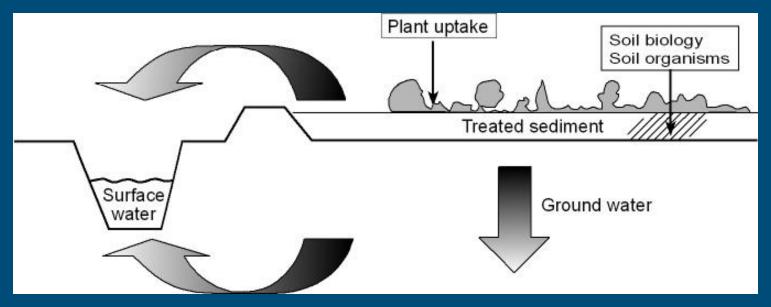
Dewatering rate (root depth)
Improvement of structure (aeration)
Size of aggregate ?

Phytoextraction

- Not for PAHs and Oil
- Only possible at low K_{oc} values (not present in sediment)



Risks on a landfarm



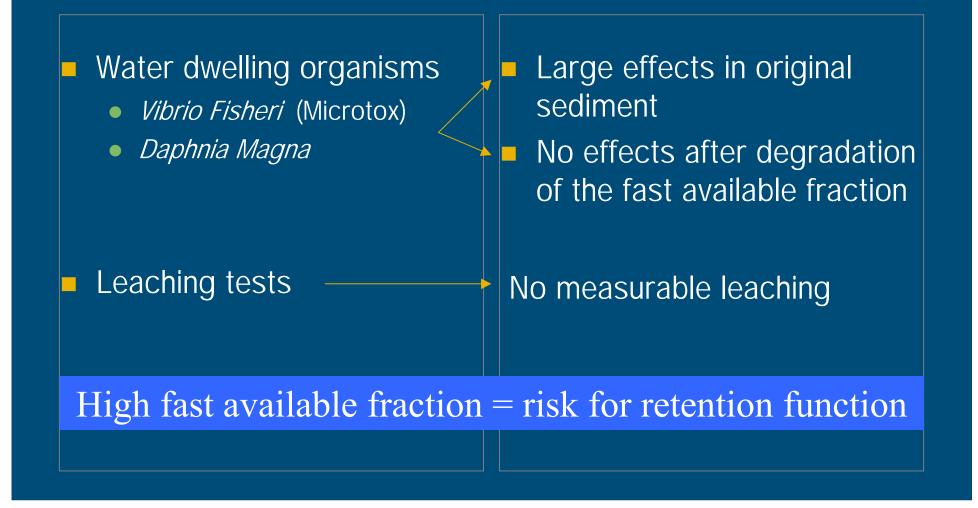
- Retention function
 - Ability to adsorb pollutants to prevent mobilization and translocation in the food chain

Habitat function

• Ability to serve as a habitat for soil living organisms and plants and their interaction



Retention Function





Habitat Function

Sediment assays

- Crassostrea gigas
- Columphium volator
- Chronomus riparius

Soil assays

- Lumbricus rubellus
- Folsomia candidal
- Bioaccumulation
 - Oligochates
- Biodegradation

Biological development

- biomass
- incorporation of ³H Thymidine and ¹⁴C Leucine
- Genetic diversity
- Population of nematodes



Habitat function

- Slower decrease of risks
- Risks are related with fast degradable fraction
- Fast degradable fraction is degraded, but the slow degradable fraction gives a new fast degradable fraction
- No measurable risks if this fraction is small

■ After >5 years risks are low



Reuse contaminated soils and sediments



Biomass production

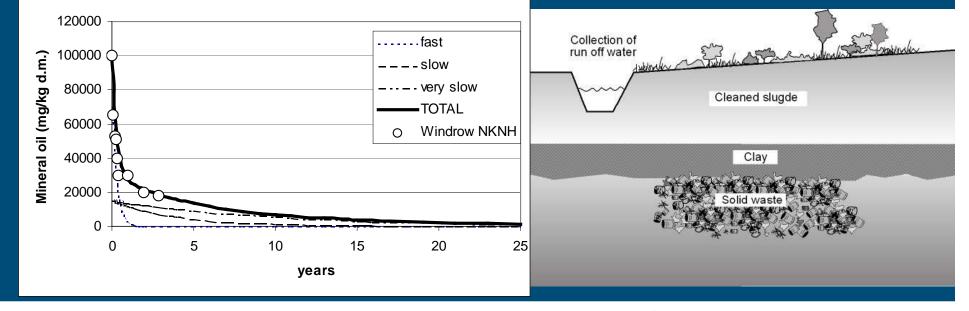
Oostwaardhoeve , The Netherlands
100 ha (1 million m³).
20 Euro/m³ – profit biomass



Contaminated sludge in Nizhnekamsk







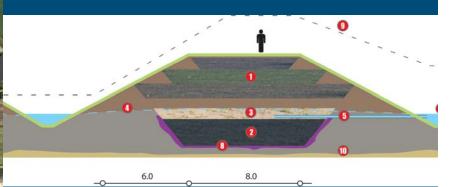






Watervast, sediments and safety (Competition)

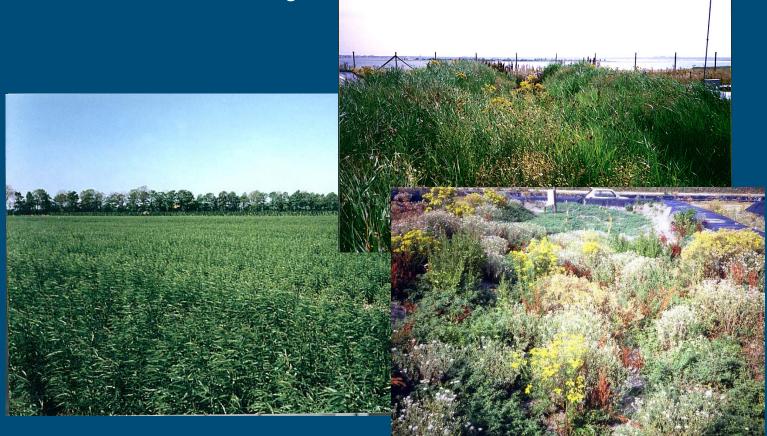




Area Overbetuwe, The Netherlands
162 km dikes = 6.5 million m³
Costs project 65 million Euro
Costs for contamination 2 3 Euro/m³



What about heavy metals?



Biodegradation and phytostabilization/extraction



Landfarming of polycyclic aromatic hydrocarbons and mineral oil contaminated sediments



Joop Harmsen

joop.harmsen@wur.nl