

## Laboratory tests of Draeger Polytron 8000 with FL- 6813260 sensor for NH<sub>3</sub> measurement

R.W. Melse, J.P.M. Ploegaert, N.W.M. Ogink



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# Laboratory tests of Draeger Polytron 8000 with FL-6813260 sensor for NH<sub>3</sub> measurement

Roland W. Melse, Johan P.M. Ploegaert, Nico W. M. Ogink

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**Samenvatting NL:** Twee Draeger Polytron 8000 instrumenten met een FL-6813260 NH<sub>3</sub> sensor zijn getest in kortdurende laboratoriumexperimenten in het bereik van 0 tot 11.0 ppm. De meetonzekerheid bleek lager te zijn dan door de fabrikant werd opgegeven. In het gebied van 1 - 11 ppm is de nauwkeurigheid van de sensor voldoende hoog om deze toe te passen ten behoeve van emissie monitoring. In het gebied van 0 - 1 ppm is echter een grotere nauwkeurigheid nodig om de sensor te kunnen toepassen in natuurlijk geventileerde melkveestallen. De conclusie is dat de sensor een veelbelovende apparaat is voor ammoniakmetingen in en rond stallen, en geschikt is om nader te evalueren onder praktijkomstandigheden. Om de werking van de sensor en de stabiliteit van de kalibratielijnen vast te stellen op de lange termijn en onder praktijkomstandigheden, wordt aanbevolen om de Draeger sensor in een veldstudie nader te testen.

**Summary UK:** Two Draeger Polytron 8000 instruments with FL-6813260 sensor for NH<sub>3</sub> measurement were tested in short-term laboratory experiments in the range from 0 to 11.0 ppm. The measurement uncertainty found in the calibration tests was lower than specified by the manufacturer. In the 1-11 ppm range the measurement accuracy is sufficient to meet requirements for emission monitoring. However higher accuracy levels are needed in the 0-1 ppm range to be of interest for application in naturally ventilated cattle barns. It is concluded that the sensor is a promising device for ammonia measurements in and around animal houses, and is suitable for further evaluation under practical conditions. In order to monitor the sensor performance and the stability of calibration lines on long term under barn conditions, it is recommended to test the Draeger sensor in a field study.

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The ISO 9001 certification by DNV underscores our quality level. All our research commissions are in line with the Terms and Conditions of the Animal Sciences Group. These are filed with the District Court of Zwolle.

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# 1 Introduction

In naturally ventilated cattle houses, ammonia (NH<sub>3</sub>) concentrations in a range of 0.1 - 3 ppm<sup>1</sup> can be expected under summer conditions and about 1 - 6 ppm under winter conditions. In mechanically ventilated pig houses, concentrations can be tenfold. At many pig houses air scrubbers are installed that reduce the ammonia concentrations in the air down to a range of 0.1 - 3 ppm, which is similar to the concentrations that are found in dairy cattle houses in summer.

In the Netherlands a regulatory system is in place which applies an NH<sub>3</sub> emission factor to each animal house based on prescribed design and management practices. This means that the actual emissions from the farms are not monitored, although a large variation in emission levels may exist between farms. A similar system with design-based emission factors is used for air scrubbers and in addition several secondary parameters are measured (pH and EC of trickling water, electricity use, pressure drop over packing, fresh water use and discharge water amount). These parameters are stored electronically using a so-called obligatory "electronic monitoring system". The values and development of these parameters are used for verifying proper scrubber operation and as a consequence, are used for assessing sufficient ammonia removal performance of the scrubber.

Currently no robust and reliable sensor is available for long-term continuous monitoring of NH<sub>3</sub> concentrations in animal houses or air scrubbers at low costs. Especially low concentrations (< 1 ppm) and high humidity (outlet scrubber air) make it difficult to operate sensors for a long period (sensor deteriorates fast and frequent zero-calibration is necessary).

However, a suitable sensor that could continuously measure NH<sub>3</sub> would enable farmers to real-time monitor the ammonia concentrations in the animal house (e.g. for welfare reasons), the actual emissions from the animal house and the performance of the air scrubber. For monitoring the NH<sub>3</sub> emission from an animal house, the NH<sub>3</sub> concentration measurement could be combined with a CO<sub>2</sub> concentration measurement, the latter enabling an estimation of the air flow rate by means of the CO<sub>2</sub>-balance method. For monitoring the performance of an air scrubber the NH<sub>3</sub> concentration of both inlet and outlet air could be measured.

The German company Draeger has developed a new NH<sub>3</sub> sensor that is suitable for a range of 0.5 - 100 ppm according to the specifications. Draeger states that this sensor is expected to be suitable for long-term use without the need of regular recalibration (low zero-drift) or other maintenance.

The aim of this study is to carry out a short-term laboratory test of this sensor in order to find out how this sensors performs with regard to detection limit and measurement accuracy, especially at the lower concentrations ( $\leq 11$  ppm). Also the aim is to test whether or not the measurement is influenced by the relative humidity of the air.

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<sup>1</sup> 1 ppm NH<sub>3</sub> = 0.70 mg/m<sup>3</sup> at 25°C and 1 atm.



## 2 Material and Methods

### 2.1 Sensors and data processing

In all experiments two NH<sub>3</sub> sensors were used that were connected parallel, generating in duplo measurements.

Sensor name and type: Draeger Polytron 8000 with FL-6813260 sensor for NH<sub>3</sub>

Main specifications according to manufacturer:

- Working principle: electrochemical diffusion sensor;
- Detection limit: 1 ppm;
- Measurement range limit (max): 100 ppm;
- Calibration: the sensor is factory-calibrated with NH<sub>3</sub> and zero gas (2-point calibration);
- Measurement uncertainty:  $\leq \pm 5\%$  of measured value with a minimum of  $\leq \pm 1.5$  ppm (the greater value applies);
- Operation temperature and relative humidity: -40/65°C and 15/95% (although the measurement principle does not require water in the air, Draeger recommends to have an humidity of > 10% when the sensor is purged constantly, i.e. > 5 days);
- Air pressure: ambient (1 bar)  $\pm 1\%$ ;
- Cross references: in Table 1 the known cross-references are given as supplied by the manufacturer;
- Output signal: 4...20 mA.

Table 1

*Known cross-references for Draeger NH<sub>3</sub> sensor FL-6813260.*

Test gas	Concentration (ppm)	NH <sub>3</sub> sensor reading (ppm)
CO	100	< 0.5
H <sub>2</sub> S	20	< 0.8
NO <sub>2</sub>	10	< 0.7
NO	20	< 0.5
CO <sub>2</sub>	16,000	< 0.5
SO <sub>2</sub>	20	-12
H <sub>2</sub>	1,000	< 0.5
Cl <sub>2</sub>	10	-6
C <sub>2</sub> H <sub>4</sub>	1,000	< 0.5
HCN	50	< 0.5
HCl	5	< 0.5
O <sub>3</sub>	0.5	< 0.5
C <sub>2</sub> H <sub>5</sub> OH	250	< 0.5
H <sub>2</sub> O <sub>2</sub>	10	< 0.5
Limonene <sup>a</sup>	saturated	< 0.5
Peracetic acid	saturated	< 0.5
N <sub>2</sub> O	10,000	< 0.5

<sup>a</sup> 1-Methyl-4-(1-methylethenyl)-cyclohexene; <sup>b</sup> CH<sub>3</sub>CO<sub>3</sub>H.



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## 2.2 Experimental setup

Two NH<sub>3</sub> gas cylinders were used, containing 11.0 ppm and 101 ppm in N<sub>2</sub> (Scott Specialty Gases, see certificates in Appendix A and B). The gas from the cylinder was diluted with compressor air using two adjustable gas diluters and led to a measurement chamber where the two NH<sub>3</sub> sensors are placed. In Figure 1 a schematic of the setup is given; in Figure 2 both NH<sub>3</sub> sensors are shown.

Gas diluter #1 is a Signal Model 821 Gas Divider, which has an adjustable dilution ratio up to 1 : 10. The dilution ratio is set with percentages ranging from 10% (i.e. 10% source gas is mixed with 90% of compressed air, so 10 x dilution) to 100% (i.e. 100% source gas without compressed air, no dilution). If necessary, a second gas diluter (#2) was connected in series. Gas diluter #2 is an EPM Diluting Stack Sampler, diluting unit 797-430, adjustable dilution ratio) which was operated at a dilution ratio of 1 : 18.65 (at a pressure of 6.0 bar). In Appendix C and D the calibration reports from both gas diluters are given. In one experiment a humidifier was connected in order to increase the humidity of the air prior to reaching the NH<sub>3</sub> sensors. The humidifier consisted of a bottle with at the bottom a small amount of water (net air volume of bottle, incl. extra tubing: 380 ml). The bottle is placed in a thermostat bath. Through the lid of the bottle an inlet and outlet tube are inserted; the inlet tube is close to the water level but does not touch it, the outlet tube is close to the top of the bottle. By increasing the temperature of the bath, the humidity of the air increases. Before reaching the sensors, the humidity and temperature of the air are measured (Vaisala, Finland; HMP 110). Then the air reaches a chamber where the NH<sub>3</sub> concentrations are measured with the two Draeger Polytron 8000 FL-6813260 sensors. With a precision resistor of 50  $\Omega$  (Neohm UPW-50,  $\pm 0.1\%$ ,  $3 \times 10^{-6} \Omega/^\circ\text{C}$ ) the 4 - 20 mA output signal of the sensors was converted to a 200 - 1000 mV signal that was recorded every 10 seconds with a data logger (Campbell Scientific Inc., Logan UT, USA; type CR1000). The resolution of the data logger is 0.1 mV (i.e. 0.0125 ppm) for values between 200 and 800 mV (i.e. 0 - 75 ppm) and 1 mV (i.e. 0.125 ppm) for values above 800 mV (i.e. > 75 ppm). Eventually the measurement by the Draeger sensors are converted to ppm's by assuming 4 mA is equal to 0 ppm and 20 mA is equal to 100 ppm, in accordance with the specifications of the sensor.

All tubing was made of FEP in order to prevent absorption of ammonia. The experiments were carried out at room temperature (20-23°C) and at atmospheric pressure (1015-1022 hPa); the gas temperature at the sensors was slightly higher (23-25°C). During the experiments no condensation was observed in the tubing or in the measurement chamber.

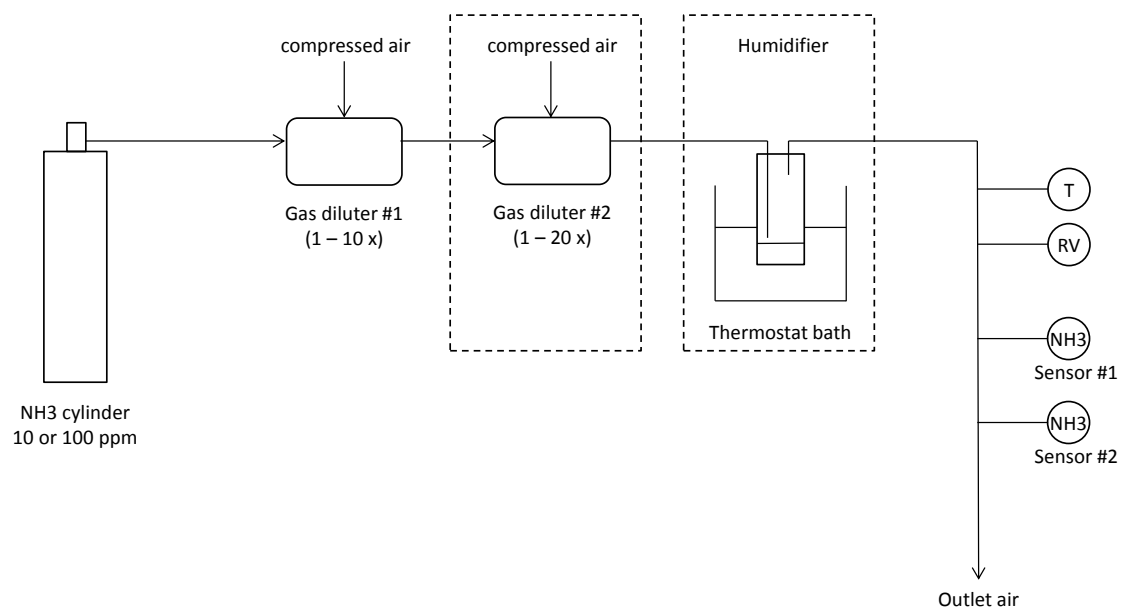
Prior to the experiments the used equipment was checked and calibrated:

- the data logger values were checked and calibrated with a calibrated voltmeter (Fluke, type 87III);
- the value of the precision resistors were checked with a calibrated voltmeter (Fluke, type 87III);
- the gas dilution ratio of both gas diluters was based on available certificates and checked using a chemiluminescent NO<sub>x</sub> analyser (Monitor Labs / API, model 200 A) and a cylinder of NO gas (Scott Specialty Gases, 40 ppm).

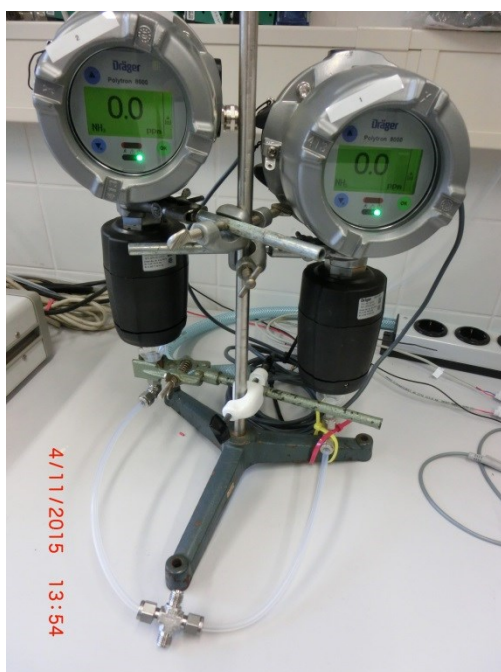
## 2.3 Experiment 1: Detection limit and accuracy at low concentrations

In this experiment a range of NH<sub>3</sub> concentrations between 0 and 11 ppm was supplied to the NH<sub>3</sub> sensors, using different gas cylinders and diluter settings. The humidification unit was switched off. Table 2 explains how the different NH<sub>3</sub> concentrations were generated. Also a zero ppm NH<sub>3</sub> test was done by supplying pure compressed air to the sensors. Each concentration was supplied for about 15 minutes in which period a stable signal could be established. During these experiments the relative humidity was 3-5% and the temperature 23-25°C.

The concentration level that was supplied at the sensors was calculated based on the gas diluter settings and the source gas concentration.



**Figure 1** Schematic of laboratory setup for testing the  $\text{NH}_3$  sensors.



**Figure 2** Picture of the two  $\text{NH}_3$  sensors.

Table 2

Test concentrations of NH<sub>3</sub> - low concentration range (0 - 11 ppm); n.a. = not applicable.

Source NH <sub>3</sub> gas (ppm)	Gas diluter #1 (%) SIGNAL	Gas diluter #1 (times diluted) SIGNAL	Gas diluter #2 (times diluted) EPM 6.0 bar	Supplied NH <sub>3</sub> at sensors (ppm)
0	n.a.	n.a.	n.a.	0
11.0	10	9.95	18.65	0.06
11.0	20	5.02	18.65	0.12
11.0	40	2.52	18.65	0.23
11.0	80	1.25	18.65	0.47
11.0	100	1.00	18.65	0.59
11.0	10	9.95	n.a.	1.11
11.0	20	5.02	n.a.	2.19
11.0	30	3.35	n.a.	3.28
11.0	40	2.52	n.a.	4.37
11.0	50	2.00	n.a.	5.49
11.0	80	1.25	n.a.	8.79
11.0	100	1.00	n.a.	11.0

## 2.4 Experiment 2: Influence of relative humidity

In this experiment a selection of NH<sub>3</sub> concentrations was fed to the NH<sub>3</sub> sensors, using the humidification unit to establish different humidity levels of the gas. Gas diluter #2 was switched off. In Table 3 the testing conditions are further explained. Also a zero ppm NH<sub>3</sub> test was done by supplying pure compressed air to the sensors. Each concentration was supplied for about 15 minutes in which period a stable signal could be established. During these experiments the temperature was 23-25°C.

Table 3

Test concentrations of NH<sub>3</sub> at low relative humidity (3-5%), medium (40-50%), and high relative humidity (70-80%); n.a. = not applicable.

Source NH <sub>3</sub> gas (ppm)	Gas diluter #1 (%) SIGNAL	Gas diluter #1 (times diluted) SIGNAL	Supplied NH <sub>3</sub> at sensors (ppm)
0	n.a.	n.a.	0
11.0	10	9.95	1.11
11.0	50	2.00	5.49
101	10	9.95	10.2

## 3 Results and discussion

### 3.1 Experiment 1: Detection limit and accuracy at low concentrations

In Table 4 the results of experiment 1 are shown, after stabilization of the output signal of the sensors for at least 5 minutes. The "supplied NH<sub>3</sub> concentration" is the concentration as calculated from the source gas and the dilution equipment settings. After the concentration was increased, after about 3 minutes a stable output signal was achieved. Within the experimental setup that was chosen it is not possible to distinguish between the time that is needed to reach the desired concentration in the air present at the sensors (as it is no perfectly mixed system), or the time that is needed by the sensors to respond to a step increase or decrease of ammonia concentration of the air supplied to the sensor.

Table 4

*Results of NH<sub>3</sub> measurements with electrochemical diffusion sensors - low concentration range (temperature: 23-25°C; relative humidity: 3-5%); n.d. = not detected.*

Supplied NH <sub>3</sub> concentration (ppm)	Measured NH <sub>3</sub> concentration (ppm)	
	Sensor #1	Sensor #2
0	-0.19	-0.18
0.06	n.d.	n.d.
0.12	n.d.	n.d.
0.23	-0.025	0.068
0.47	0.31	0.41
0.59	0.40	0.49
1.11	1.15	1.24
2.19	2.25	2.35
3.28	3.43	3.53
4.37	4.69	4.76
5.49	5.78	6.04
8.79	9.14	9.40
11.0	11.25	11.68

For the tested concentrations of 0.06 and 0.12 ppm, no increase of the mA output signal of the sensors was detected, therefore these values are reported as "not detected". As can be seen for the 0 ppm value (and for the 0.23 ppm value for sensor #1) slightly negative measured ppm values were found, i.e. the output signal was slightly lower than 4 mA. From Table 4 it follows that both sensors overestimate the supplied NH<sub>3</sub> at values between 1.11 and 11.0 ppm, and underestimate at concentrations of 0.59 ppm and lower.

In Figure 3 the results of Table 4 are plotted, together with a linear regression line for each sensor. This regression line can be considered as the calibration line to be used for each sensor. Furthermore, the line  $Y = X$  is shown. The regression lines can be formulated as ( $R^2$  = coefficient of determination; RSD = Residual Standard Deviation):

**Sensor #1:**

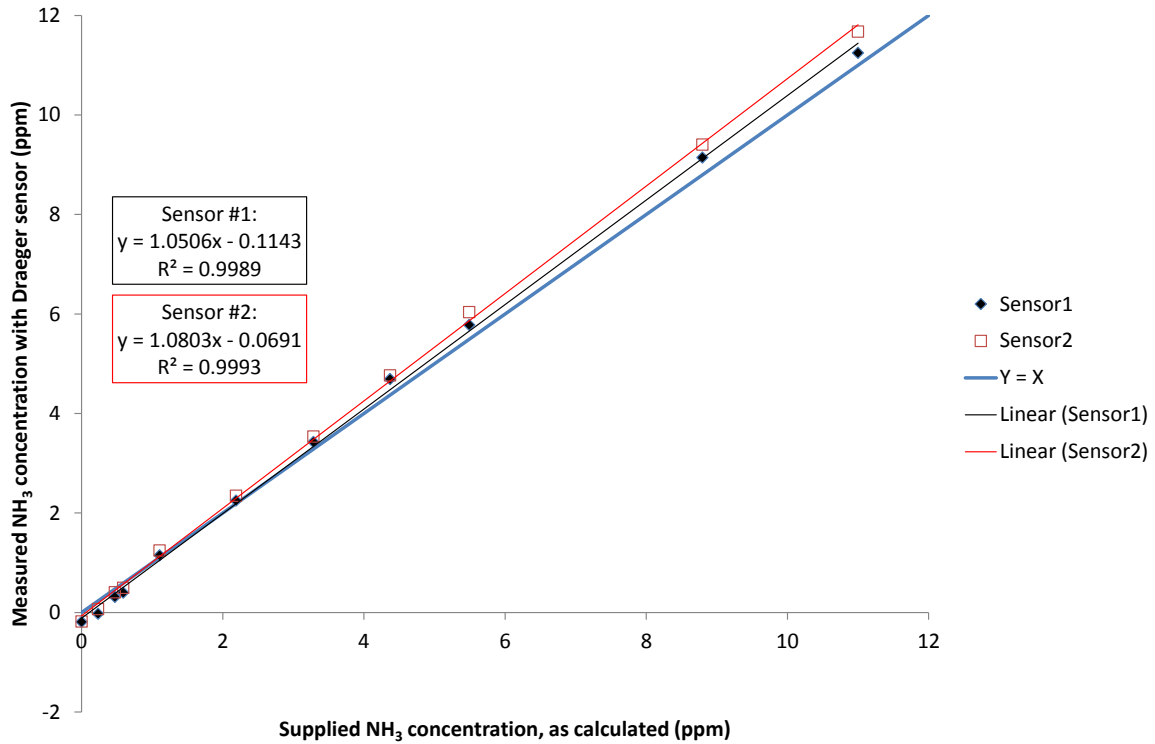
**$\text{Measured [NH}_3\text{] (ppm)} = \text{Supplied [NH}_3\text{] (ppm)} \times 1.0506 - 0.1143$**   
**[ $R^2 = 0.9989$ ; RSD = 0.137]**

**Sensor #2:**

$$\text{Measured [NH}_3\text{] (ppm)} = \text{Supplied [NH}_3\text{] (ppm)} \times 1.0803 - 0.0691$$

$$[R^2 = 0.9993; \text{RSD} = 0.113]$$

The concentration that can be calculated using these calibration lines and the measured values, is called the predicted concentration.



**Figure 3** Measured  $\text{NH}_3$  concentration versus supplied  $\text{NH}_3$  concentration for both Draeger sensors ( $\text{RV} = 3\text{-}5\%$ ).

When the measurement signal (in mA) is used as input variable and the supplied  $\text{NH}_3$  concentration as output, the following equations can be formulated:

**Sensor #1:**

$$\text{Supplied [NH}_3\text{] (ppm)} = \text{Measurement signal (mA)} \times 5.9425 - 23.658$$

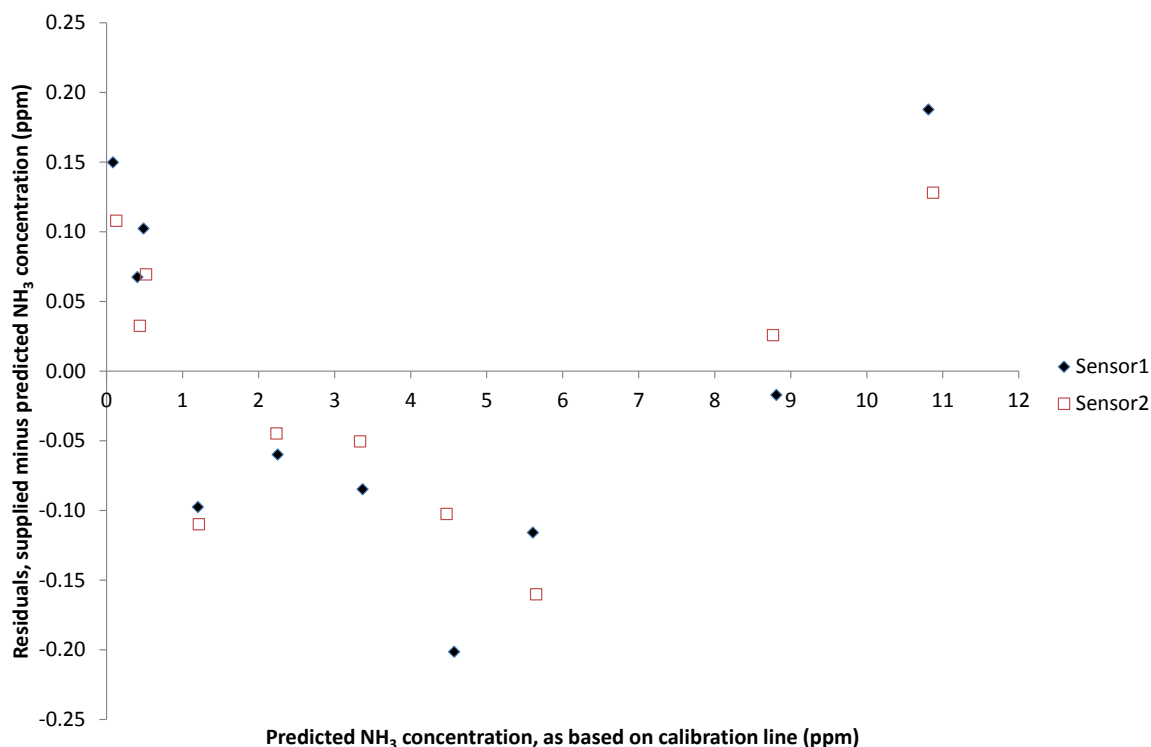
$$[R^2 = 0.9989; \text{RSD} = 0.13]$$

**Sensor #2:**

$$\text{Supplied [NH}_3\text{] (ppm)} = \text{Measurement signal (mA)} \times 5.7813 - 23.059$$

$$[R^2 = 0.9991; \text{RSD} = 0.10]$$

In Figure 4 the residuals are shown, i.e. the deviations between supplied and the predicted concentrations. Figure 4 clearly shows that residuals are not evenly distributed over the whole concentration range. In the range  $< 1$  ppm the predicted concentrations are lower than the supplied concentrations (positive residual values) and in the range  $1 - 6$  ppm the predicted concentrations are higher than the supplied concentrations (negative residual values). The highest measurements then show positive residual values again. The residuals show that the sensors do not respond rectilinearly in the offered concentration range. For this reason the calibration lines and residual values were calculated when the concentration range is split into a  $0 - 1$  ppm and a  $1 - 11$  ppm range; this is shown in Figure 5 and 6 for Sensor #1 and #2, respectively.



**Figure 4** Residual values (i.e. supplied minus predicted NH<sub>3</sub> concentration) for both Draeger sensors (RV = 3-5%).

The regression lines shown in Figure 5a and 6a can be formulated as follows:

**0 - 1 ppm, Sensor #1:**

$$\text{Measured [NH}_3\text{] (ppm)} = \text{Supplied [NH}_3\text{] (ppm)} \times 1.2336 - 0.2636$$

[R<sup>2</sup> = 0.9858; RSD = 0.071]

**0 - 1 ppm, Sensor #2:**

$$\text{Measured [NH}_3\text{] (ppm)} = \text{Supplied [NH}_3\text{] (ppm)} \times 1.2958 - 0.2168$$

[R<sup>2</sup> = 0.9954; RSD = 0.042]

**1 - 11 ppm, Sensor #1:**

$$\text{Measured [NH}_3\text{] (ppm)} = \text{Supplied [NH}_3\text{] (ppm)} \times 1.0247 + 0.0781$$

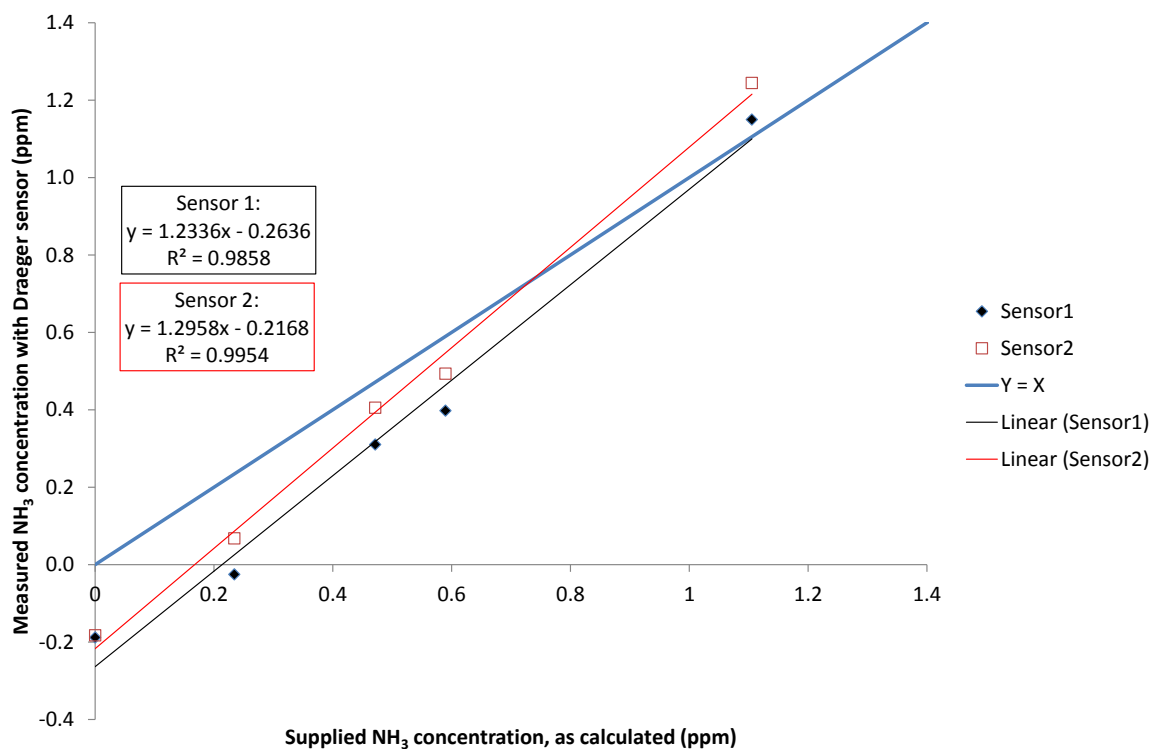
[R<sup>2</sup> = 0.9994; RSD = 0.096]

**1 - 11 ppm, Sensor #2:**

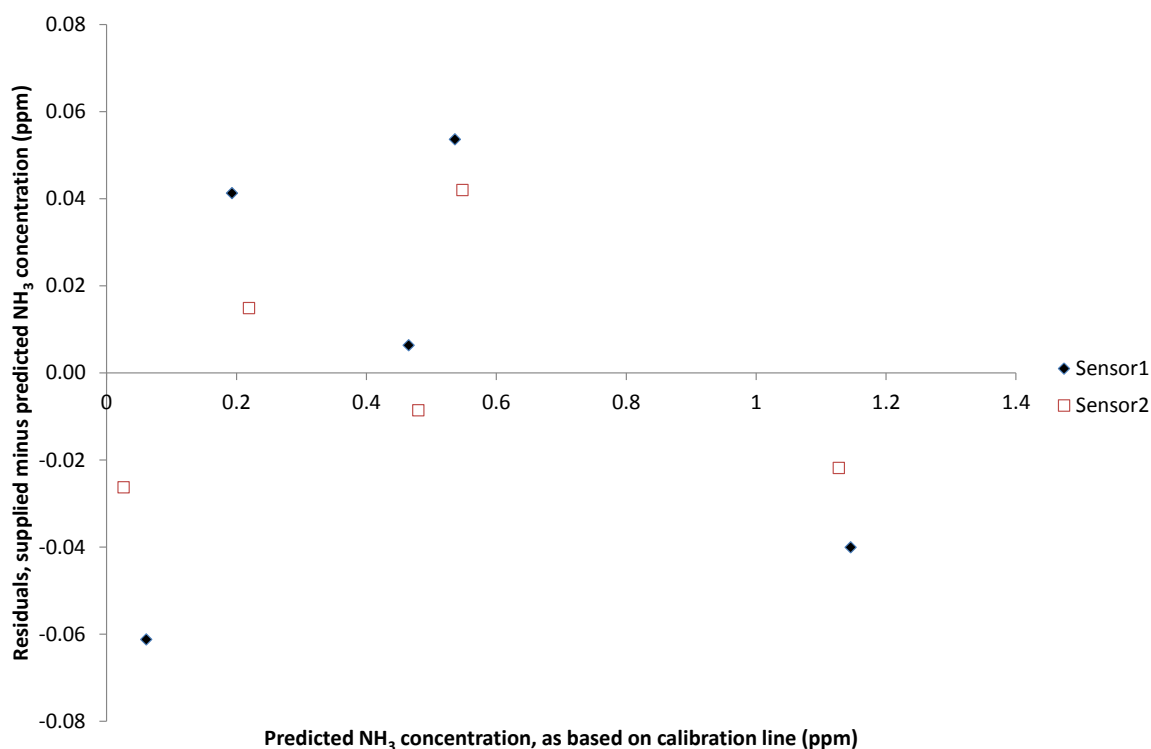
$$\text{Measured [NH}_3\text{] (ppm)} = \text{Supplied [NH}_3\text{] (ppm)} \times 1.0585 + 0.0921$$

[R<sup>2</sup> = 0.9997; RSD = 0.076]

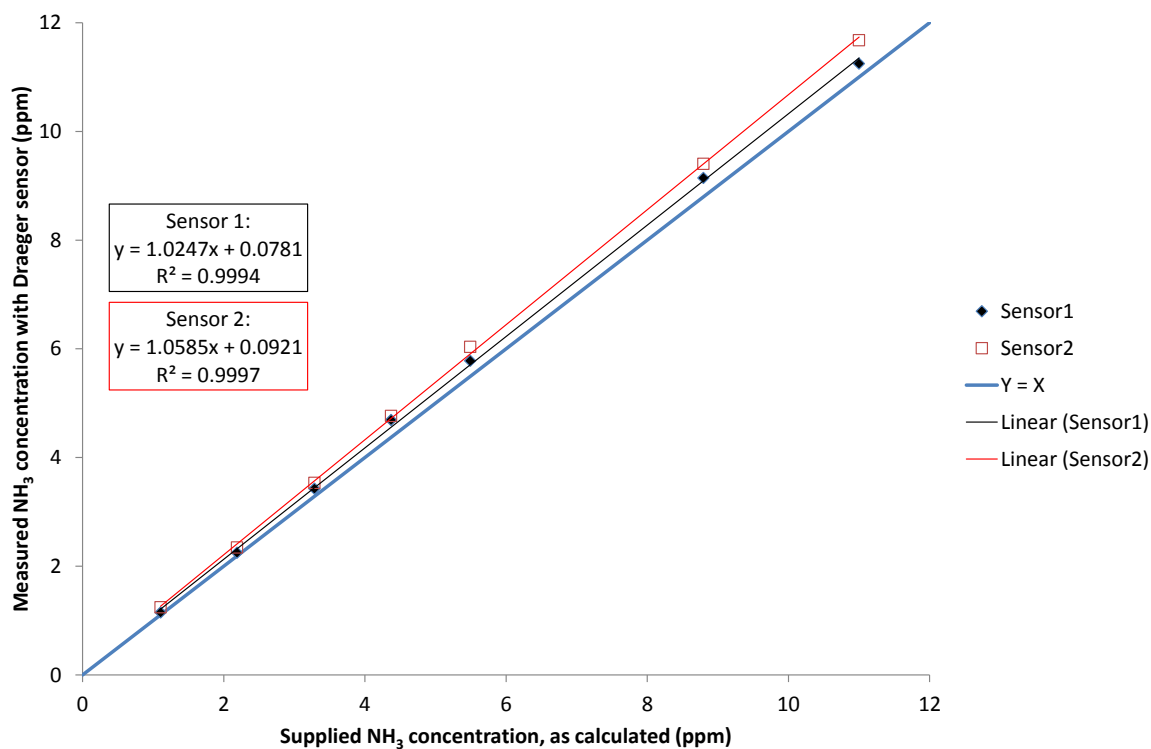
In this way for measurements in the range of 0 - 1 ppm the first regression line can be used for calibration, while in the range of 1 - 11 ppm the second regression line can be used.



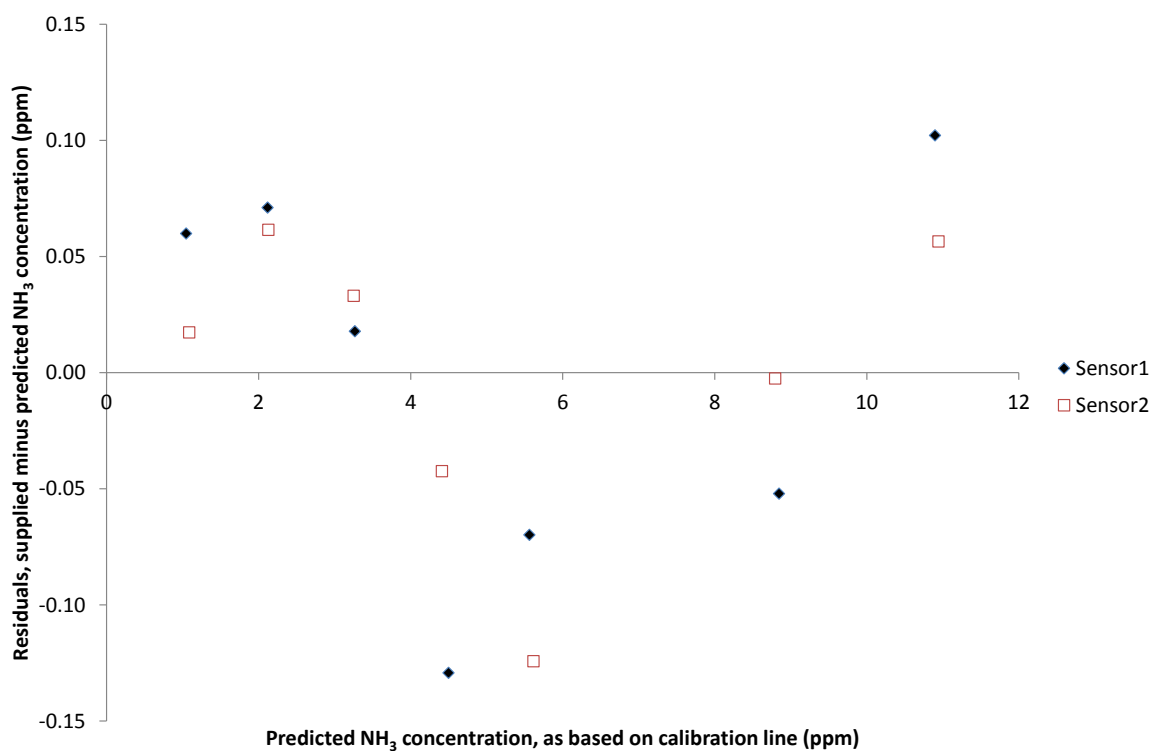
**Figure 5a** Measured  $\text{NH}_3$  concentration versus supplied  $\text{NH}_3$  concentration for both Draeger sensors, range of 0 - 1 ppm (RV = 3-5%).



**Figure 5b** Residual values (i.e. supplied minus predicted  $\text{NH}_3$  concentration) for both Draeger sensors, range of 0 - 1 ppm (RV = 3-5%).



**Figure 6a** Measured  $\text{NH}_3$  concentration versus supplied  $\text{NH}_3$  concentration for both Draeger sensors, range of 1 - 11 ppm (RV = 3-5%).



**Figure 6b** Residual values (i.e. measured  $\text{NH}_3$  concentration minus predicted  $\text{NH}_3$  concentration) for both Draeger sensors, range of 1 - 11 ppm (RV = 3-5%).



Comparison of the residual values in Figure 4 (whole range) with Figure 5b (0 - 1 ppm) and Figure 6b (1 - 11 ppm) shows that the residual values decrease when a different calibration curve is used for the low and the high range concentrations. This means that more accurate measurements can be done.

## 3.2 Experiment 2: Influence of relative humidity

In Table 5 the results of experiment 2 are shown, after stabilization of the output signal of the sensors.

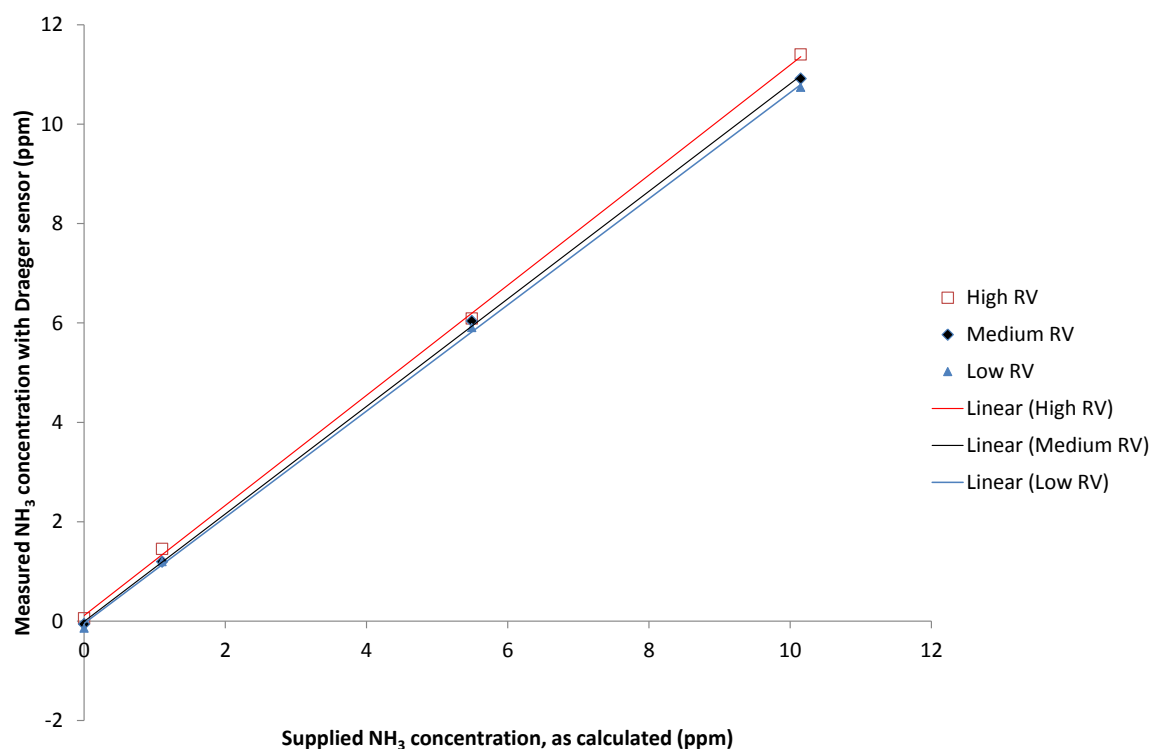
Table 5

*Results of NH<sub>3</sub> measurements with electrochemical diffusion sensors at different humidity levels (temperature: 23-25°C); n.a. = not applicable.*

Relative humidity (%)	Supplied NH <sub>3</sub> concentration (ppm)	Measured NH <sub>3</sub> concentration (ppm)	
		Sensor #1	Sensor #2
3-5 (low)	0	-0.10	-0.18
	1.11	1.15	1.24
	5.49	5.78	6.04
	10.2	10.6	10.9
40-50 (medium)	0	-0.103	-0.020
	1.11	1.15	1.24
	5.49	5.95	6.12
	10.2	10.7	11.1
70-80 (high)	0	0.045	0.068
	1.11	1.41	1.49
	5.49	5.98	6.20
	10.2	11.1	11.7

In Figure 7 the average measurement values of the sensors are plotted against the supplied NH<sub>3</sub> concentration for all RV ranges, together with a linear regression line. The regression line for low RV (3-5%) is the calibration line that was presented already in section 3.1. Figure 7 clearly shows that the sensors show higher readings at higher humidity levels, as the lines for medium RV (black) and high RV (red) are above the line for low RV (blue).

Furthermore, in this experiment it took more time for the sensors to stabilize than in Experiment 1, in some cases up to 25 minutes; in Experiment 1 the time needed for stabilization was only about 3 minutes. This delay is caused by the fact that by connecting the bottle for humidification, the gas volume of the measurement system is largely increased. As a result, the time that is needed to reach the desired concentration in the air present at the sensors increases (as it is no perfectly mixed system), thus ending up in a slower responses to concentration changes.



**Figure 7** Measured NH<sub>3</sub> concentration versus supplied NH<sub>3</sub> concentration (average of both Draeger sensors) at different humidity levels.

### 3.3 Comparability of sensors

To test the comparability of the results the two different NH<sub>3</sub> sensors were exposed simultaneously to the same polluted air, and the response of the devices to the provided concentration was registered, as described. This test was repeated at different ammonia concentration levels and different relative humidity levels. Then, the dataset with the paired results from the devices was analysed for outliers by using the box plot method with three times the interquartile range (IQR) as measure for extreme outliers:

$$\text{IQR} = Q3 - Q1$$

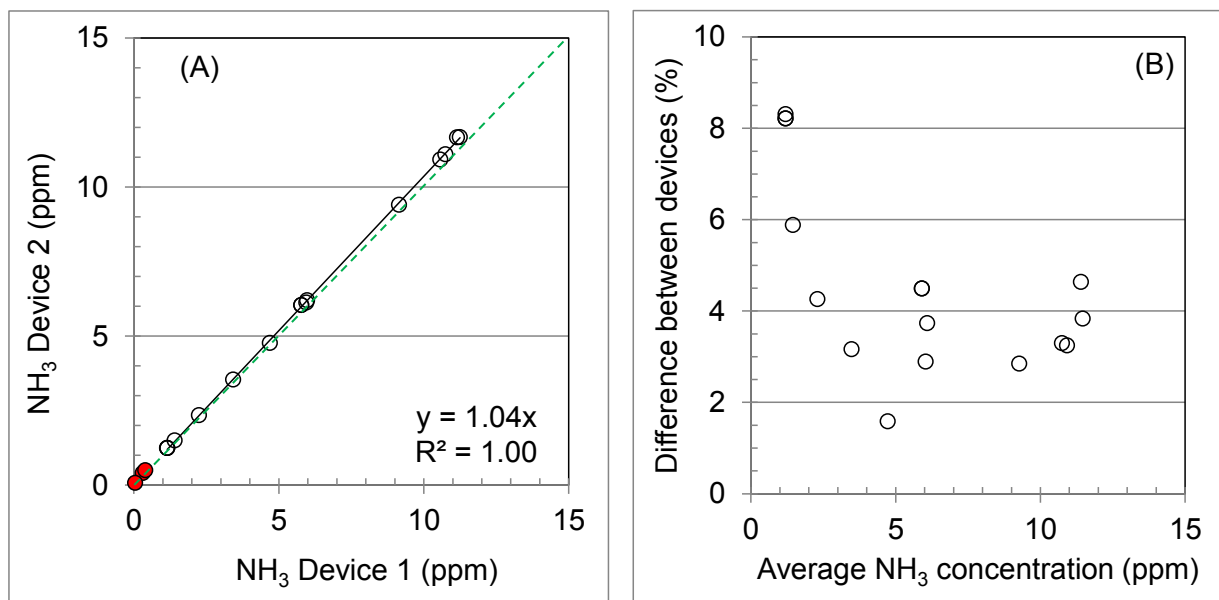
Q1: lower (first) quartile (25%)

Q2: median (second quartile)

Q3: upper (third) quartile (75%)

Outlier if: value > Q3+3\*IQR OR value < Q1-3\*IQR

This test shows that devices are giving not comparable results at the low concentration levels, i.e. measured concentrations  $\leq 0.5$  ppm (supplied concentration:  $\leq 0.59$  ppm), as all results in this range were found to be outliers. For measured concentrations at and above 1.2 ppm (supplied concentration  $\geq 1.11$  ppm), no outliers were found. Figure 8a also shows a (small) systematic difference (4%) between both devices. The observed difference between devices is independent of the provided NH<sub>3</sub> concentration level (Figure 8b).



**Figure 8** Comparability of results between devices. (A) Absolute measured values; red dots are the points considered as outliers in the box plot method; (B) Difference between devices vs. average  $\text{NH}_3$  concentration.

---

## 4 Conclusions and recommendations

Two Draeger Polytron 8000 instruments with FL-6813260 sensor for NH<sub>3</sub> measurement were tested in short-term laboratory experiments in the range from 0 to 11.0 ppm. The results showed that the response of the sensor was not fully rectilinear in this range, and that calibration with a linear model is best applied separately at the 0 – 1 ppm range and the 1-11 ppm range.

The mean residual standard deviation of the 0-1 ppm calibration line amounted 0.056 ppm. Estimating an uncertainty interval, as an approximation, on two times the standard deviation this would result in an uncertainty interval of  $\pm 0.11$  ppm. Expressed as a relative error of the median value of 0.5 ppm in this range, the relative error here would amount 23%. At values in the 1-11 ppm range the residual standard deviation amounted 0.086 ppm, resulting in an uncertainty interval of 0.17 ppm, and a relative error at the median value of 6 ppm of 2.8%. In both intervals we assume that no systematic errors are present because bias is corrected on basis of the regression lines. Our test setup was not suited to evaluate drift from the calibration lines, hence estimated uncertainties represent a best case scenario.

Nevertheless the results give the impression that in the 1-11 ppm range the accuracy is better than expected from the specifications of the manufacturer (Specified measurement uncertainty:  $\leq \pm 5\%$  of measured value with a minimum of  $\leq \pm 1.5$  ppm, where the greater value applies). In the 0-1 ppm range the measurement uncertainty amounted  $\pm 0.11$  ppm, being considerably lower than the specified minimum of  $\leq \pm 1.5$  ppm.


Also, the influence of humidity was tested. The measured NH<sub>3</sub> concentration slightly increased at higher humidity levels.

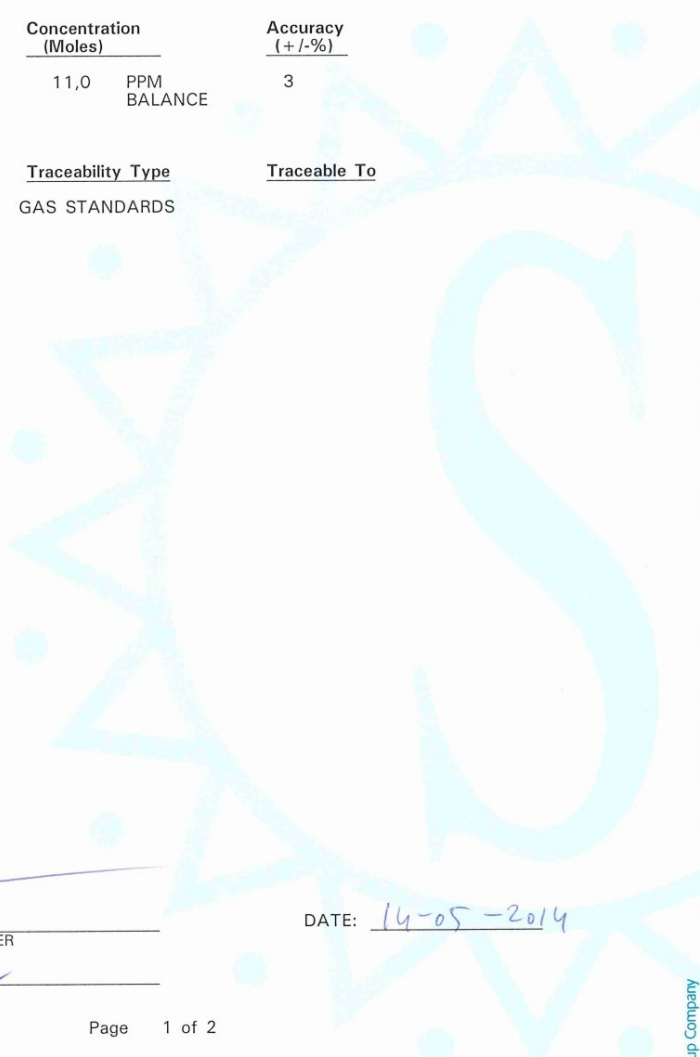
It is concluded from the laboratory test that this Draeger sensor is a promising device for ammonia measurements in and around animal houses, and is suitable for further evaluation under practical conditions. The measurement uncertainty found in the calibration tests was lower than specified by the manufacturer. In the 1-11 ppm range the measurement accuracy is sufficient to meet requirements for emission monitoring. However higher accuracy levels are needed in the 0-1 ppm range, which range is especially of interest in naturally ventilated cattle barns. For application in practice it is important to monitor the sensor performance and the stability of calibration lines on long term under barn conditions and also its performance at higher concentration levels.


Therefore it is recommended to test the Draeger sensor in a field study. A field study could include:

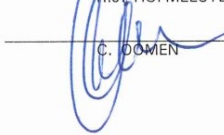
- Measurements inside a natural ventilated dairy house, combined with CO<sub>2</sub> measurements for estimation of the ventilation rate;
- Measurements of the ammonia concentration levels inside a pig or poultry house;
- Measurements of the ammonia removal efficiency of a scrubber operated at a pig house.

# Appendix A - Certificate of 11.0 ppm NH<sub>3</sub> gas cylinder

		<i>CUSTOM CLASS</i>	
Takkebijsters 48 4817 BL, Breda, The Netherlands Phone: +31(0)76-5711 828 Fax: +31(0)76-5713 267			
<b>CERTIFICATE OF ACCURACY: Custom Class Calibration Standard</b>			
<b>Product Information</b> Project No.: 20-97567-002 Item No.: 20020000150Z50L P.O. No.: WUR728325  Cylinder Number: 5902857 Cylinder Size: 50L Certification Date: 05May2014 Expiration Date: 04May2015		<b>Customer</b> WAGENINGEN UR, ANIMAL SCIENCES GROUP T.A.V. JOHAN PLOEGAERT VEEHOUDERIJ BV VIJFDE POLDER 1 6708 WC WAGENINGEN NEDERLAND	
<b>CERTIFIED CONCENTRATION</b>			
<u>Component Name</u>	<u>Concentration (Moles)</u>	<u>Accuracy (+/-%)</u>	
AMMONIA NITROGEN	11,0 PPM BALANCE	3	
<b>TRACEABILITY</b>			
<u>Description</u>	<u>Traceability Type</u>	<u>Traceable To</u>	
ANALYTICAL TRACEABILITY	GAS STANDARDS		



APPROVED BY:   
R.J. HOFMEESTER

SUPERVISOR:   
C. DOEMEN

DATE: 14-05-2014

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## SPECIFICATIONS

Component Name	Requested Concentration (Moles)		Gravimetric Concentration (Moles)	Analyzed Concentration (Moles)		Blend Tolerance Result (+/- %)	Blend Process Accuracy Result (+/- %)	Analytical Accuracy Result (+/- %)	Interlocking Result (+/- %)
AMMONIA NITROGEN	10,	PPM	N/A	11,0	PPM	10,0	N/A	3,00	N/A

## TRACEABILITY

### Analytical Traceability

Type/SRM No.	Expiration Date	Cylinder Number	Concentration	Component
SRS TSTD	05May2014		0,000	AMMONIA

## PHYSICAL PROPERTIES


Cylinder Size: 50L  
Pressure: 150 BAR  
Valve Connection: DIN-5  
Expiration Date: 04May2015  
Min. Cyl. Pressure: 5 BAR  
Min. Temperature: -10 °C

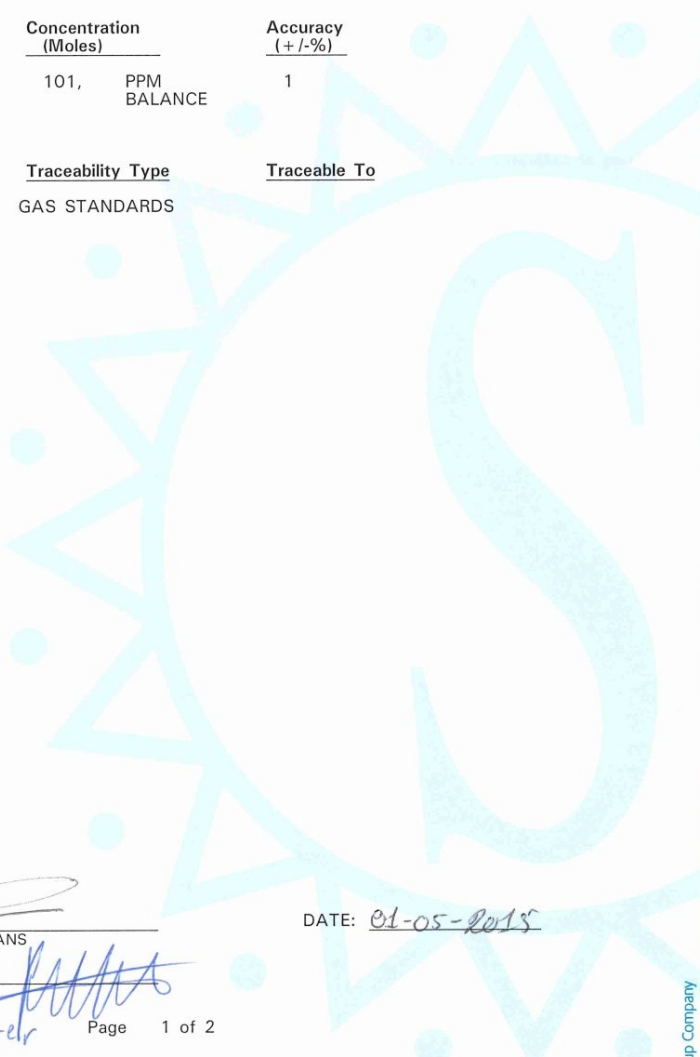
## SPECIAL HANDLING INSTRUCTIONS


Do not use or store cylinder at or below the stated dew point temperature. Possible condensation of heavier components could result. In the event the cylinder has been exposed to temperatures at or below the dew point, place cylinder in heated area for 24 hours and then roll cylinder for 15 minutes to re-mix.

Use of calibration standards at or below dew point temperature may result in calibration error.

## Appendix B - Certificate of 101 ppm NH<sub>3</sub> gas cylinder

	<b>Scott</b> specialty gases	<b>CUSTOM CLASS</b>
Takkebijsters 48 4817 BL, Breda, The Netherlands Phone: +31(0)76-5711 828 Fax: +31(0)76-5713 267		
<b>CERTIFICATE OF ACCURACY: Custom Class Calibration Standard</b>		
<b>Product Information</b> Project No.: 20-01957-002 Item No.: 20020000150Z50L P.O. No.: WUR837436  Cylinder Number: 5902919 Cylinder Size: 50L Certification Date: 30Apr2015 Expiration Date: 29Apr2018	<b>Customer</b> WAGENINGEN UR, ANIMAL SCIENCES GROUP ATTN: JOHAN PLOEGAERT VEEHOUDERIJ BV, ZODIAC GEBOUW 122, DE ELST 1 6708 WD WAGENINGEN NEDERLAND	
<b>CERTIFIED CONCENTRATION</b>		
<b>Component Name</b>	<b>Concentration (Moles)</b>	<b>Accuracy (+/-%)</b>
AMMONIA NITROGEN	101, PPM BALANCE	1
<b>TRACEABILITY</b>		
<b>Description</b>	<b>Traceability Type</b>	<b>Traceable To</b>
ANALYTICAL TRACEABILITY	GAS STANDARDS	



APPROVED BY:   
H.P.A. AKKERMANS

SUPERVISOR:   
R.J. Hofmeester

DATE: 01-05-2015

Page 1 of 2

An Air Liquide Group Company

## SPECIFICATIONS

<u>Component Name</u>	<u>Requested Concentration (Moles)</u>		<u>Gravimetric Concentration (Moles)</u>	<u>Analyzed Concentration (Moles)</u>		<u>Blend Tolerance Result (+/- %)</u>	<u>Blend Process Accuracy Result (+/- %)</u>	<u>Analytical Accuracy Result (+/- %)</u>	<u>Interlocking Result (+/- %)</u>
AMMONIA NITROGEN	100,	PPM	N/A	101,	PPM	1,0	N/A	1,00	N/A

## TRACEABILITY

### Analytical Traceability

<u>Type/SRM No.</u>	<u>Expiration Date</u>	<u>Cylinder Number</u>	<u>Concentration</u>	<u>Component</u>
SRS TSTD	30Apr2015		0,000	

## PHYSICAL PROPERTIES

Cylinder Size: 50L      Pressure: 150 BAR      Valve Connection: DIN 5  
Expiration Date: 29Apr2018

Min. Cyl. Pressure: 5 BAR

## SPECIAL HANDLING INSTRUCTIONS

If a dew point is listed, do not use or store cylinder at or below the stated dew point temperature. Possible condensation of heavier components could result. In the event the cylinder has been exposed to temperatures at or below the dew point, place the cylinder in heated area for 24 hours and then roll cylinder for 15 minutes to remix.

Use of calibration standards at or below the stated dew point temperature may result in calibration error.



## Appendix C - Certificates of Signal 821 diluter

**2015:**



SIGNAL GROUP LIMITED  
12 DOMAN ROAD, CAMBERLEY  
SURREY, ENGLAND GU15 3DF  
TELEPHONE: +44 (0)1276 682841 / 4 (4 LINES)  
FAX: +44 (0)1276 691302  
E-MAIL: INSTRUMENTS@SIGNAL-GROUP.COM  
SITE: WWW.SIGNAL-GROUP.COM

### CERTIFICATE OF CALIBRATION

CONSIGNEE Wageningen

ORDER NUMBER \_\_\_\_\_

W/O NUMBER \_\_\_\_\_

DATE OF CALIBRATION 01/12/2015

DESCRIPTION OF GOODS	SERIAL NO.	REMARKS
821S gas divider	4789	

### CALIBRATION RESULTS

DIVIDER SETTING	ACTUAL RESULT
100.00%	100.00
90.00%	90.00
80.00%	79.92
70.00%	69.93
60.00%	60.04
50.00%	50.06
40.00%	39.90
30.00%	30.02
20.00%	20.08
10.00%	10.12
0.00%	0.00

### CALIBRATION GAS DETAILS

CALIBRATION GAS TYPE	SUPPLIER	CERTIFICATE No.	TRACEABILITY
1000ppm C3H8/air	BOC	1433495	NPL

SIGNAL AUDIT PROCEDURE NUMBER TPB00101

UNCERTAINTY OF MEASUREMENT +/- 0.2% F.S.D.

Signal Group certifies that the materials and processes used in the calibration of the unit detailed above hereon unless otherwise stated conform in all respects to Signal Group Ltd procedures and specifications

SIGNED Roy Kinslow

**CERTIFICATE OF CALIBRATION**

STANDARDS HOUSE,  
12 DOMAN ROAD, CAMBERLEY  
SURREY, GU15 3DF, ENGLAND.  
TELEPHONE: +44(0)1276 682841/6  
FACSIMILE: +44(0)1276 691302  
EMAIL: [instruments@signal-group.com](mailto:instruments@signal-group.com)  
Web: [www.signal-group.com](http://www.signal-group.com)

Everything for the Emissions Engineer

CONSIGNEE Agrotechnology & Food Innovations

ORDER NUMBER \_\_\_\_\_

W/O NUMBER \_\_\_\_\_

DATE OF CALIBRATION 14/10/05

DESCRIPTION OF GOODS	SERIAL No	REMARKS
8215 CAS Divide.	4789	

**CALIBRATION RESULTS**

DIVIDER SETTING	EXPECTED RESULT	ACTUAL RESULT
100.00%		100.00
90.00%		89.92
80.00%		79.82
70.00%		69.81
60.00%		59.85
50.00%		50.03
40.00%		39.85
30.00%		30.01
20.00%		20.00
10.00%		10.07
0.00%		0.0

**CALIBRATION GAS DETAILS**

CALIBRATION GAS TYPE	SUPPLIER	CERTIFICATE No.	TRACEABILITY
100% pure CO <sub>2</sub> gas	Messer	H114073	NPL

SIGNAL AUDIT PROCEDURE NUMBER TP800101

UNCERTAINTY OF MEASUREMENT +1-0.2% f.s.d.

Signal Group certifies that the materials and processes used in the calibration of the unit detailed above hereon unless otherwise stated conform in all respects to Signal Group Ltd procedures and specifications.

SIGNED \_\_\_\_\_



A DIVISION OF SIGNAL GROUP LTD. REGISTERED IN ENGLAND NO. 1369865



Nederlands Meetinstituut

347

## Certificaat

Nummer 316430-01  
blad 1 van 2

Aanvrager: IMAG - DLO  
Mansholtlaan 10 - 12  
Wageningen

Aangeboden: Een gas divider  
fabrikaat : Signal  
type : 821  
serie nummer : 4789  
bereik : 0 - 100%, in stappen van 10%

Wijze van onderzoek: De door de gas divider gemaakte verdunningen zijn onderzocht met een referentiemonitor welke gekalibreerd is met primaire standaard gasmengsels. Als nulgas is hoogzuivere stikstof aangeboden aan de gas divider. Als spangas is een mengsel gebruikt van  $(1001 \pm 10) \cdot 10^{-6}$  mol/mol stikstofmonoxide in stikstof. Bepaald is het werkelijke verdunningspercentage. De metingen werden uitgevoerd bij een omgevingstemperatuur van  $21 \pm 0,5^\circ \text{C}$ .

Datum van onderzoek: 6 tot en met 20 september 1996

Resultaat: De resultaten staan vermeld op blad 2 van 2.  
De totale meetonzekerheid is gebaseerd op twee maal de standaardafwijking (2s).

Herleidbaar- De bij de kalibratie gebruikte gasmengsels zijn herleidbaar naar de primaire  
heid: standaard gasmengsels stikstofmonoxide in stikstof.

Datum en ondertekening: 24 september 1996  
NMI Van Swinden Laboratorium B.V.

R. M. Wessel  
Afdeling Chemie

$\text{N}_2$  &  $1000 \text{ ppm NO in N}_2$



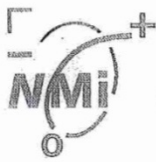
Nederlands Meetinstituut  
Postbus 654  
2600 AR Delft (NL)  
Schoemakerstraat 97  
2628 VK Delft  
Telefoon (015) 269 15 00  
Telefax (015) 261 29 71

Nederlands Meetinstituut N.V. (ingeschreven bij de  
Kamer van Koophandel Delft onder nummer 28701)

Werkmaatschappijen:  
NMI Certin B.V. (KvK nr. 33418)  
NMI Van Swinden Laboratorium B.V. (KvK nr. 28703)  
NMI IJkwezen B.V. (KvK nr. 28700)  
NMI International B.V. (KvK nr. 39176)

Dit certificaat wordt verstrekt onder het voorbehoud dat  
generel aansprakelijkheid wordt aanvaard en dat  
aanvrager vrijwaring geeft voor elke aansprakelijkheid  
jegens derden.

Reproductie van het volledige certificaat is toegestaan.  
Gedeelten van dit certificaat mogen slechts worden  
gereproduceerd na verkregen schriftelijke toestemming.



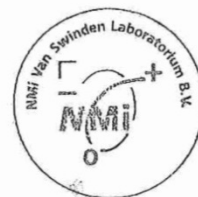
Nederlands Meetinstituut

Nummer 316430-01  
blad 2 van 2



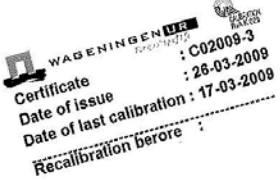

Resultaat: Hieronder wordt het resultaat van de kalibratie weergegeven.

Meetbereik 0-100% in stappen van 10%.

Werkelijk verdunnings- percentage (%)	ingesteld verdunnings- percentage (%)
10,05 ± 0,10	10
19,91 ± 0,12	20
29,86 ± 0,12	30
39,73 ± 0,18	40
49,93 ± 0,20	50
60,04 ± 0,25	60
69,88 ± 0,25	70
79,95 ± 0,25	80
89,99 ± 0,25	90



## Appendix D - Certificate of EPM 797-430 diluter

 <b>WAGENINGEN UR</b> For quality of life		
<b>Kalibratie-certificaat</b> CO2009-3		
<b>Aanvrager:</b>	ASG Veehouderij B.V. Postbus 65 8200 AB Lelystad	
<b>Aangeboden:</b>	Laboratoriumverdunner	
Fabrikaat	EPM	
Type	797-430	
Serienummer	E 77	
<b>Wijze van onderzoek:</b>	Het onderzoek werd uitgevoerd met CO als tracergas. De CO-concentraties werden gemeten met een gekalibreerde GFC CO-analyser. De verdunningen werden berekend uit de in en uitgaande CO-gasconcentraties van het verdunningstoestel.	
<b>Omgevingscondities</b>	Het onderzoek is uitgevoerd bij een ruimtetemperatuur van $(21 \pm 2) ^\circ\text{C}$ .	
<b>Datum van onderzoek:</b>	17 maart 2009.	
<b>Resultaat:</b>	De resultaten en meetonzekerheden staan vermeld op blad 2.	
<b>Onzekerheid:</b>	De gerapporteerde onzekerheid is gebaseerd op de standaardonzekerheid vermenigvuldigd met een dekkingsfactor $k = 2$ , welke overeenkomt met een betrouwbaarheidsinterval van ongeveer 95%. De standaardonzekerheid is bepaald volgens EA-4/02.	
<b>Herleidbaarheid:</b>	De metingen zijn uitgevoerd met meetmiddelen waarvan de herleidbaarheid naar (inter)nationale standaarden is aangetoond ten overstaan van de Raad voor Accreditatie.	
<b>Datum en ondertekening:</b>	Wageningen, 26 maart 2009  Ing. J.P.M. Ploegaert Coördinator Geurlaboratorium	
 Certificate : CO2009-3 Date of issue : 26-03-2009 Date of last calibration : 17-03-2009 Recalibration before : 		
<div style="display: flex; justify-content: space-between;"><div>Versie 01-05-2007</div><div><p>Wageningen UR ASG Veehouderij B.V. Cluster MHE Wageningen Bornsesteeg 59 6708 PD Wageningen Telefoon +31 (0) 317 487276 Telefax +31 (0) 317 475347</p><p>Reproductie van het volledige certificaat is toegestaan. Gedeelten mogen slechts worden gereproduceerd na schriftelijke toestemming van ASG Veehouderij.</p><p>Dit certificaat wordt verstrekt onder de voorwaarde dat ASG Veehouderij generlei aansprakelijkheid aanvaardt.</p></div></div>		
1/2		



Kalibratie-certificaat  
C02009-3

Instelling	Verdunning		Onzekerheid (k=2)	Nauwkeurigheid	Stabiliteit %
	[ ]	[ ]	[ ]	[ ]	[ ]
2.6	10.0	+/-	0.03	0.008	0.24
3.0	11.0	+/-	0.01	0.008	0.28
4.0	13.5	+/-	0.06	0.005	0.22
5.0	16.0	+/-	0.08	0.004	0.19
6.0	18.7	+/-	0.15	0.003	0.23

**Opmerkingen:** Metingen conform NEN-EN 13725:2003. Uitgevoerd door JP.  
De laboratoriumverdunner werd met de hand bediend.

Versie 01-05-2007

Wageningen UR  
ASG Veehouderij B.V.  
Cluster MHE Wageningen  
Bornsesteeg 59  
6708 PD Wageningen  
Telefoon +31 (0) 317 487276  
Telefax +31 (0) 317 475347

Reproductie van het volledige certificaat is toegestaan. Gedeelten mogen slechts worden gereproduceerd na schriftelijke toestemming van ASG Veehouderij.

Dit certificaat wordt verstrekt onder de voorwaarde dat ASG Veehouderij generlei aansprakelijkheid aanvaardt.

2/2

To explore  
the potential  
of nature to  
improve the  
quality of life



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Wageningen UR Livestock Research  
P.O. Box 338  
6700 AH Wageningen  
The Netherlands  
T +31 (0)317 48 39 53  
E [info.livestockresearch@wur.nl](mailto:info.livestockresearch@wur.nl)  
[www.wageningenUR.nl/livestockresearch](http://www.wageningenUR.nl/livestockresearch)

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Together with our clients, we integrate scientific know-how and practical experience to develop livestock concepts for the 21st century. With our expertise on innovative livestock systems, nutrition, welfare, genetics and environmental impact of livestock farming and our state-of-the art research facilities, such as Dairy Campus and Swine Innovation Centre Sterksel, we support our customers to find solutions for current and future challenges.

The mission of Wageningen UR (University & Research centre) is 'To explore the potential of nature to improve the quality of life'. Within Wageningen UR, nine specialised research institutes of the DLO Foundation have joined forces with Wageningen University to help answer the most important questions in the domain of healthy food and living environment. With approximately 30 locations, 6,000 members of staff and 10,000 students, Wageningen UR is one of the leading organisations in its domain worldwide. The integral approach to problems and the cooperation between the various disciplines are at the heart of the unique Wageningen Approach.

