

# Emerging technologies for biobased aromatics

FBR Inspiration day, 10 april 2014

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# Renewable feed stocks; what is driving the demand?

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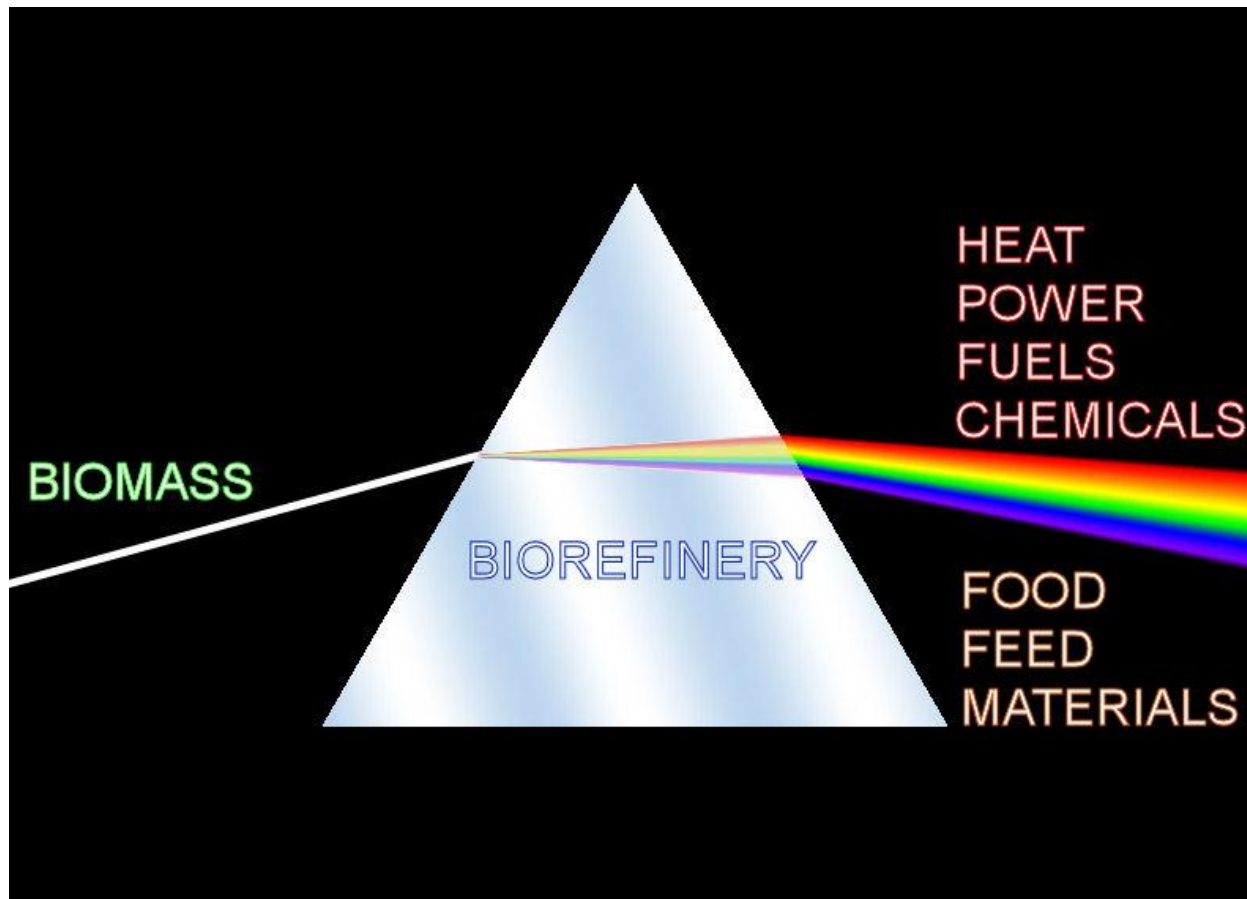
- General driving forces for renewables/biobased chemicals:
  - Declining reserves of easily accessible fossil feed stocks
  - Societal need to become more independent with regard to feedstock supply
  - Potential for GHG savings
  - Potential for unique properties
  - Economical potential for rural regions
  - Potential for more environmentally friendly products

# Biobased Chemicals Programme WUR/FBR

- Consists of over 40 projects in which biobased chemicals are one of the dominant aims
  - Confidential bilateral projects with international multinationals as well as SME' s.
  - Public private sponsored projects
  - Comprises 3 focus areas:
    - **Carbohydrate based chemicals:**
      - *Furan platform*
      - *Isohexide platform*
      - Sugar biotechnology platform
    - Lignin based chemicals
    - Vegetable oil and algae oil based chemicals
- Internationally cooperates with numerous universities and institutes
- Internally intensively cooperates with the biorefinery and biobased materials programme
- Approximately 30 coworkers involved



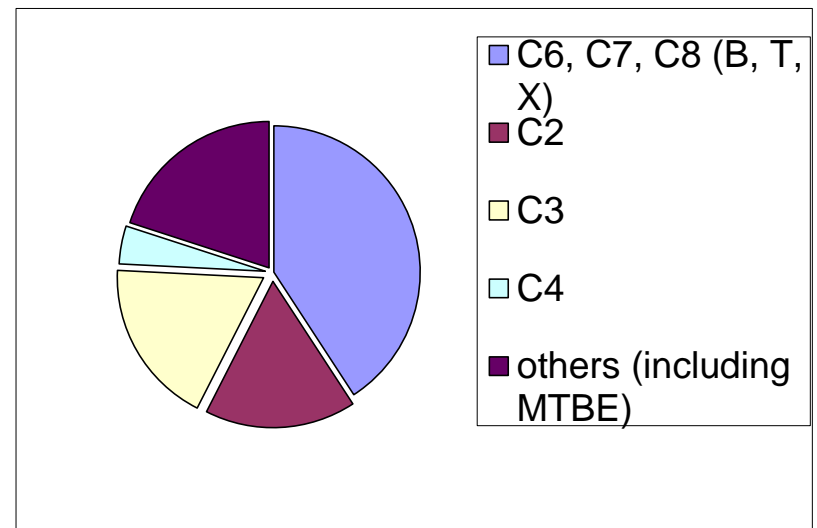
# Existing and future biorefineries will refine biomass into a spectrum of products



# Usage of bulk (platform) chemicals

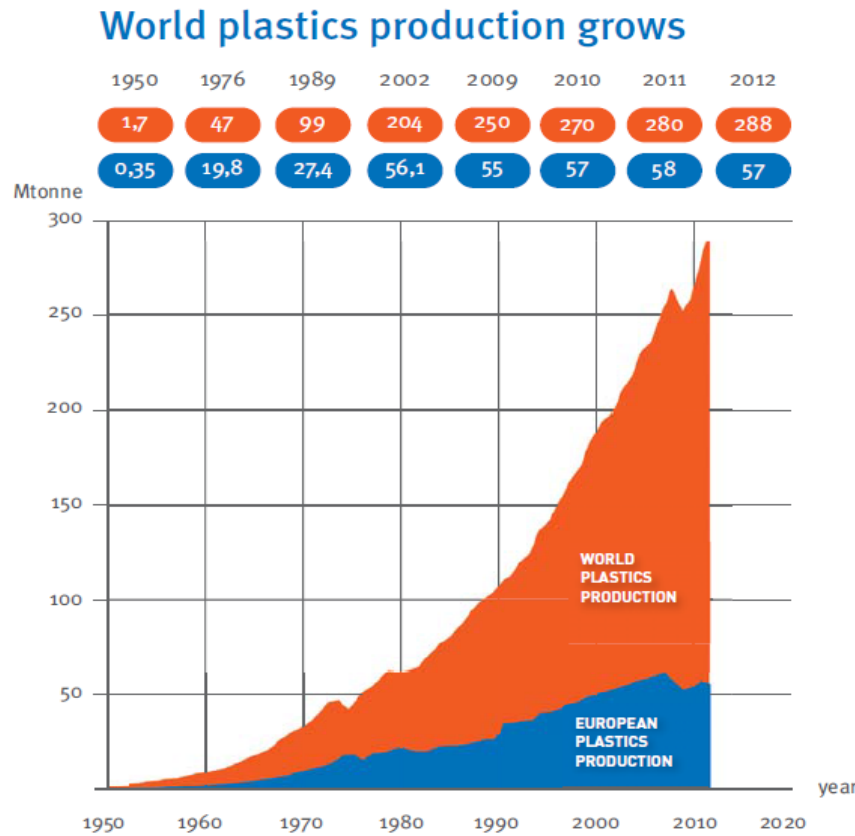
- Bulk chemicals are used as:
  - Solvents
  - Starting components for soaps, lubricants, additives (low molecular weight components)
  - Mostly as building blocks for polymers (high molecular weight components)
- Building blocks can be either aliphatic (flexible) or aromatic (rigid) nature

*Up to 40% of basic chemicals produced in Port of Rotterdam is of aromatic nature*



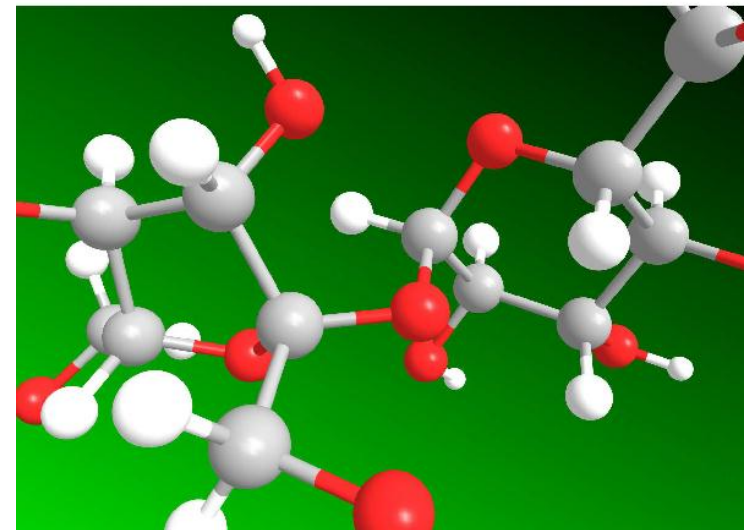
# World production of plastics

- Recent finished desk study requested by Dutch Ministry of Economic Affairs. Emphasis on biobased building blocks for polymer applications



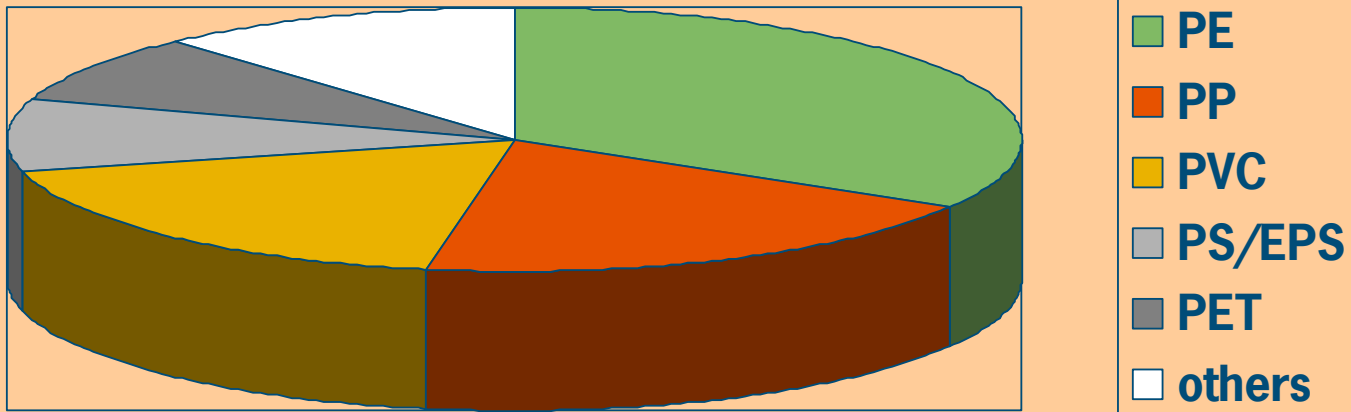
## Green building blocks for biobased plastics

PAULIEN HARMSSEN AND MARTIJN HACKMANN



# Major thermoplastic polymer materials

## Consumption of thermoplastics in Western Europe



Total about 50.000.000 tonnes in 2014

About one third of all (more high end)  
Polymers comprise aromatic building blocks

# Increasing relevance biobased aromatics

**The availability of hydrocarbons from fossil resources is affected by the shale gas boom**

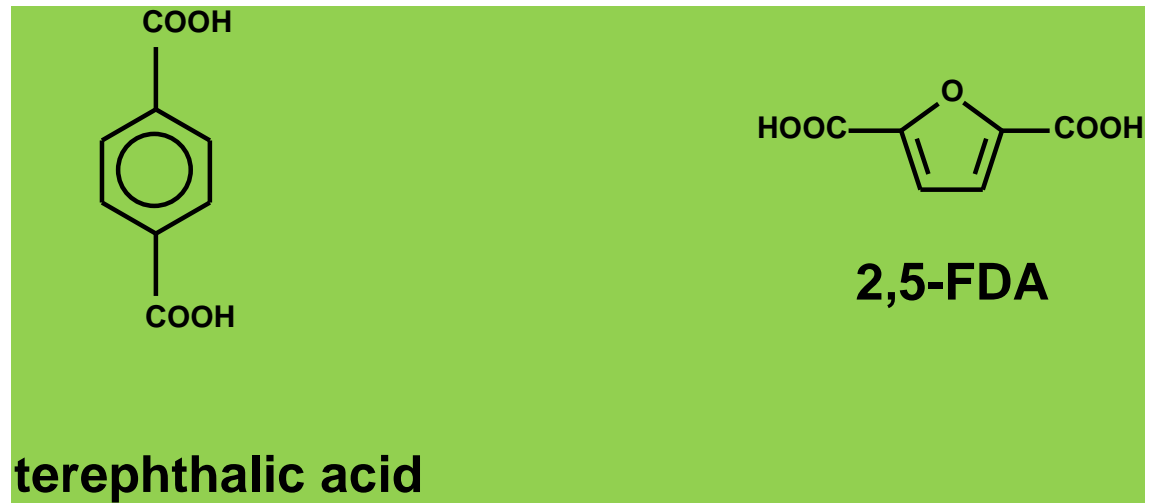
	Availability fossil
Ethylene	↑
Propylene	↓
Butadiene	↓
Aromatics	↓
Methane	↑
Funktionalised C3-molecules	↓

**Is this the chance for bio-based production?**



## Furandicarboxylic acid (2,5 FDA or FDCA) vs. terephthalic acid

- Terephthalic acid used to produce e.g. PET (bottle, fleece) or e.g. Aramid fibres
- Terephthalic acid and its isomers are used in up to 70 million tonnes/year quantities



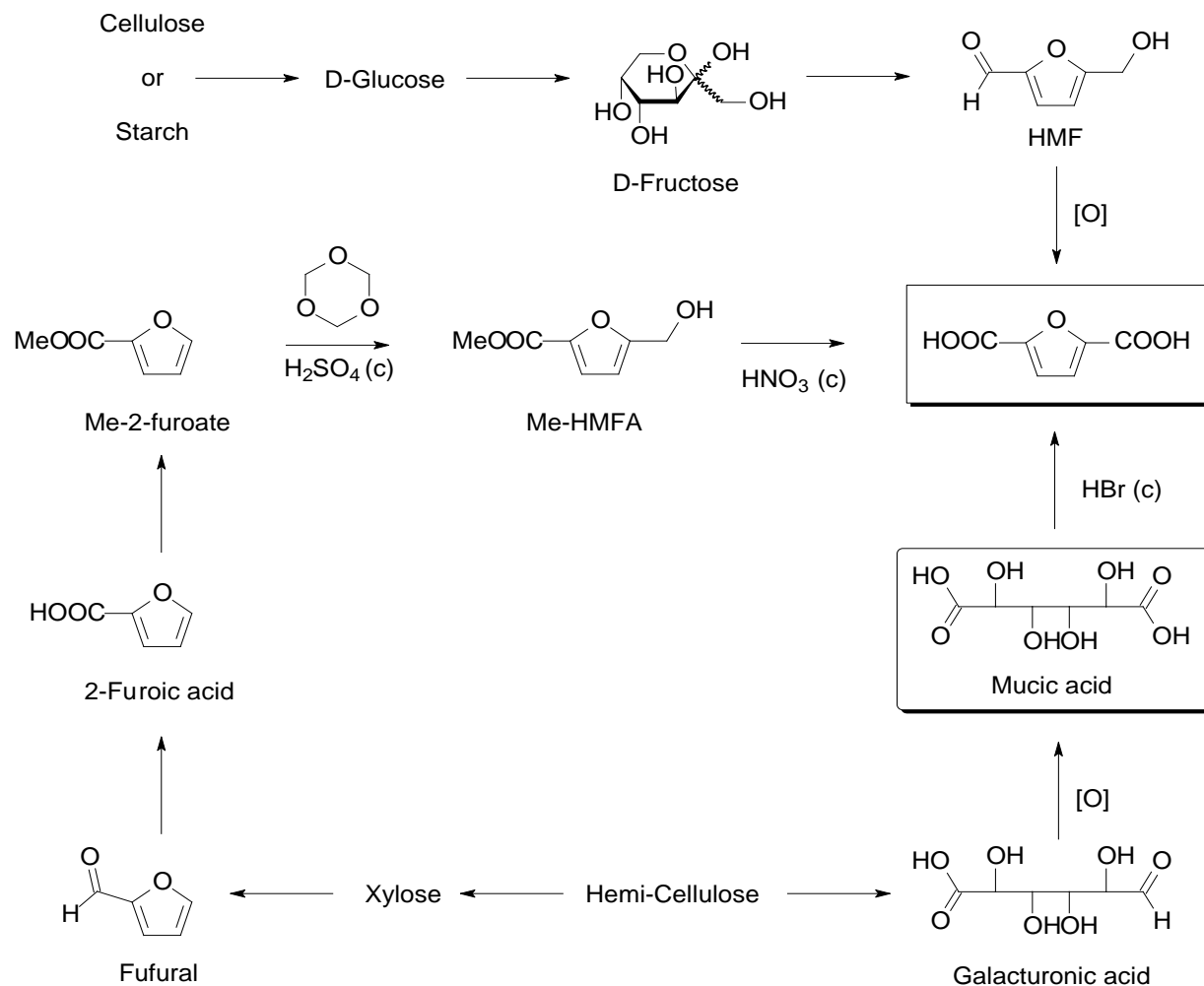
- Furan dicarboxylic acid could be a bio based alternative to terephthalic acid
- Should we go for the real stuff (drop-in) or its look alike (FDCA)?

# FDCA related research

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- Numerous knowledge institutes and companies in recent years have become involved in FDCA related research;
  - ADM, Ava Biochem, Eastman, DSM, Nuplex, Perstorp, BASF, Braskem,....
- The Netherlands is a major player in FDCA related research:
  - Avantium
  - Corbion/Bird Engineering
  - Cosun
- FBR has over 10 projects dedicated to the development of FDCA and project based upon FDCA
- Potential market is that huge that presumably more than one technology will be commercialised

# Potential Routes to 2,5 FDCA

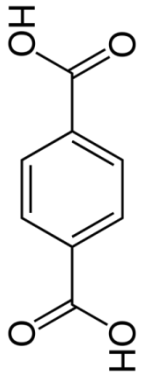


# FDCA versus PTA; Avantium YXY technology



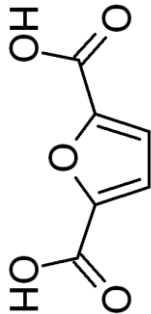
## PTA

- Oil-based
- Building block for PET
- 50 million ton per year
- Today's price: €1200/ton
- Price drivers:
  - Oil price
  - Supply/demand



## FDCA

- Bio-based
- Building block for PEF
- Potential market > 100 million ton
- Price at commercial scale:  
<€ 1000 per ton
- Price drivers:
  - Carbohydrate price
  - Economy of scale



# Compete on Performance

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*Courtesy Dr. Ed de Jong, Avantium*

A clear plastic bottle is tilted, pouring water into a glass. The bottle has a blue cap with a red ring. The water is captured mid-pour, creating a dynamic splash in the glass. The background is a soft, out-of-focus light blue.

## **PEF has the potential to beat PET**

- Better or similar properties compared to PET
- Based on carbohydrates instead of oil
- Recyclable

## **PEF has great barrier properties**

- O<sub>2</sub> barrier > 6 times better than PET
- CO<sub>2</sub> barrier > 2 times better than PET
- H<sub>2</sub>O barrier > 2 times better than PET

# EU Splash project; algae as an alternative source for FDCA precursors

- Why algae:
- To convert CO<sub>2</sub> (e.g. from the atmosphere of capture in industrial processes) into high added-value products and biofuels
- Alleviate food versus fuel conflicts
- May become particularly advantageous for regions with limited biomass availability and land unusable for agriculture



Figure 4 AlgaePARC. The opening of this facility took place in June 2011. There are 4 different production systems of 24 m<sup>2</sup> each and 3 of 2.5 m<sup>2</sup> each.

# Valorising sugarbeet pulp

- Within EU annually approximately 25-30.10<sup>6</sup> tonnes of sugarbeet pulp (SBP) are being produced
- SBP is a byproduct of sugar refining (agricultural residue)
- Cosun produces annually about 1 million tonnes of SBP (25% dry matter)
- SBP contains a mixture of components not suitable as human food
- Cosun is interested in valorising SBP towards non-food applications
- Cosun teams up with Wageningen UR/FBR and other partners in valorising SBP

Table Composition sugarbeet pulp (in %)	
ash	10
protein	8
fat	1
cellulose	25
pectinic sugars	23
arabinanes	24
other sugars	24
lignin	5
	100

# Biorefining of sugar beet pulp

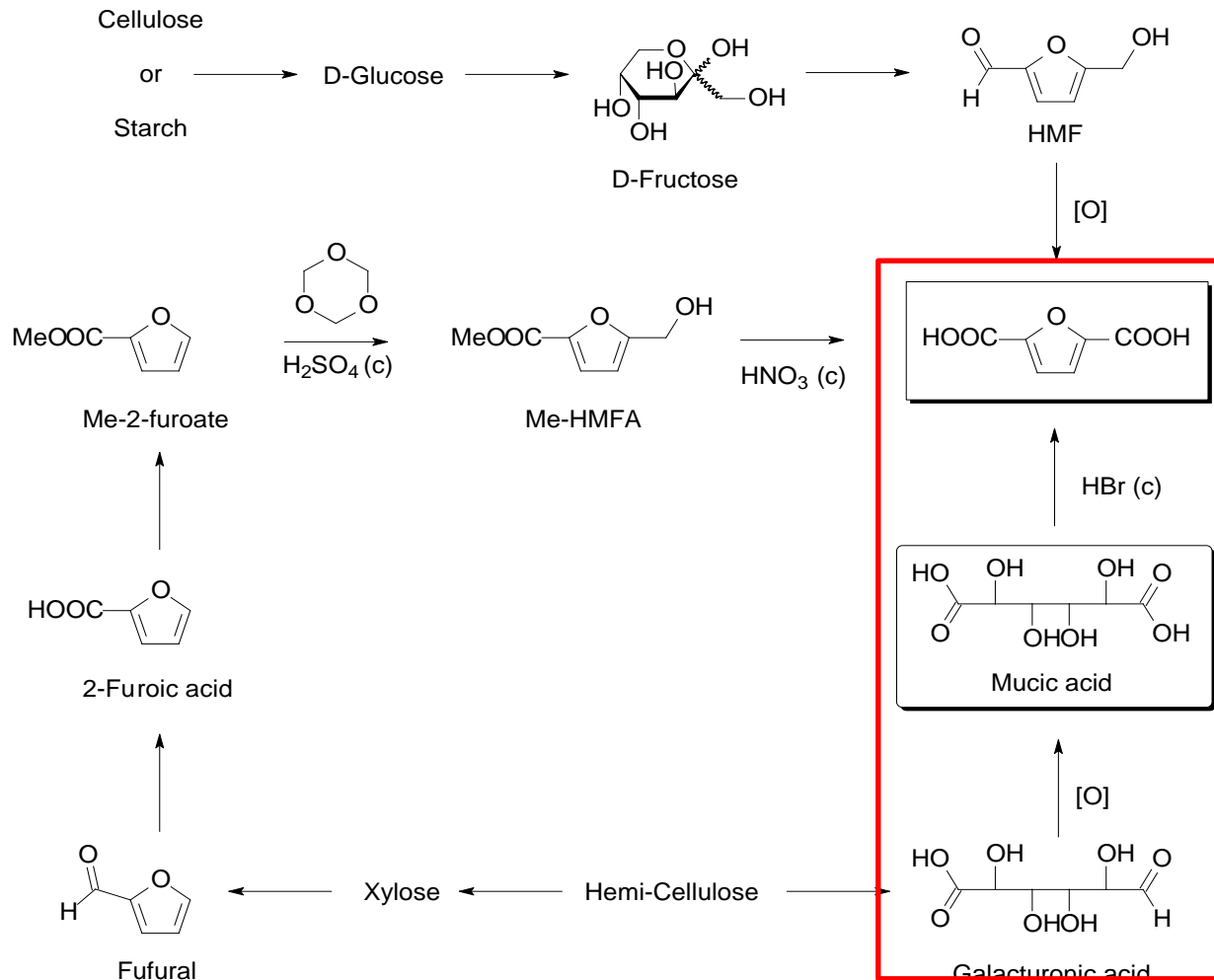


- Number of projects awarded within Topsector Agri&Food
- Overall aims of the projects
  - Continue developing biorefinery process for isolating monomeric sugars from sugarbeet pulp
  - Develop new catalytic technologies to produce high purity building blocks for high performance polymers based on sugarbeet pulp as raw material
  - Develop new chemical technologies to produce these high performance polymers
  - Investigate the properties of these new materials, and compare to conventional materials
  - Obtain insight into the technical and economical viability of bio-based polymers



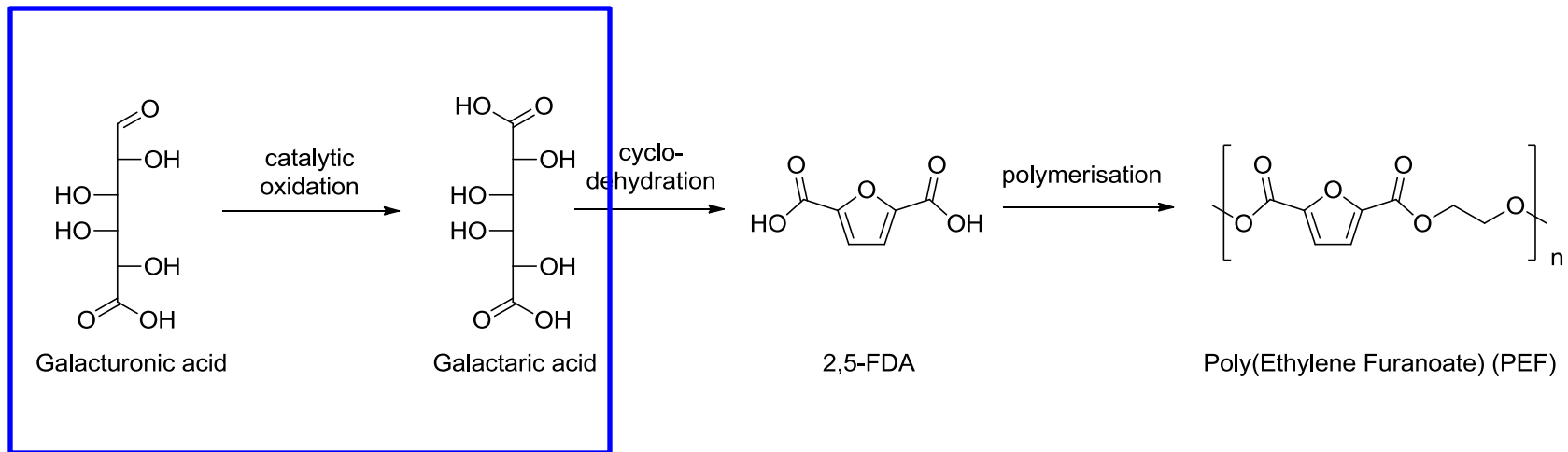


# Potential Routes to 2,5 FDCA



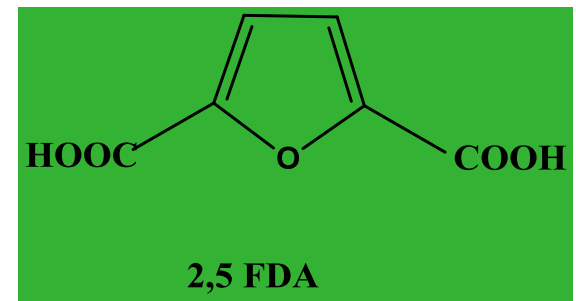
- Focus on galacturonic acid route
- Conversion of mucic acid to 2,5-FDCA already extensively explored
- In the project focus has been on oxidation of galacturonic acid (and other uronic acids and aldoses)

# Oxidation of galacturonic acid



# Second generation FDCA based upon SBP

- Key; catalytic oxidation of pectin sugars
  - Mild conditions (room temperature)
  - High conversions (>99%)
  - High selectivity (>97%)
  - Short reaction times (<3h)
  - Environmentally benign:
    - Air can be used for the oxidation
- Patent filed:
  - *Catalytic oxidation of uronic acids to aldaric acids*
  - *Subsequent conversion aldaric acid into 2,5 FDCA*



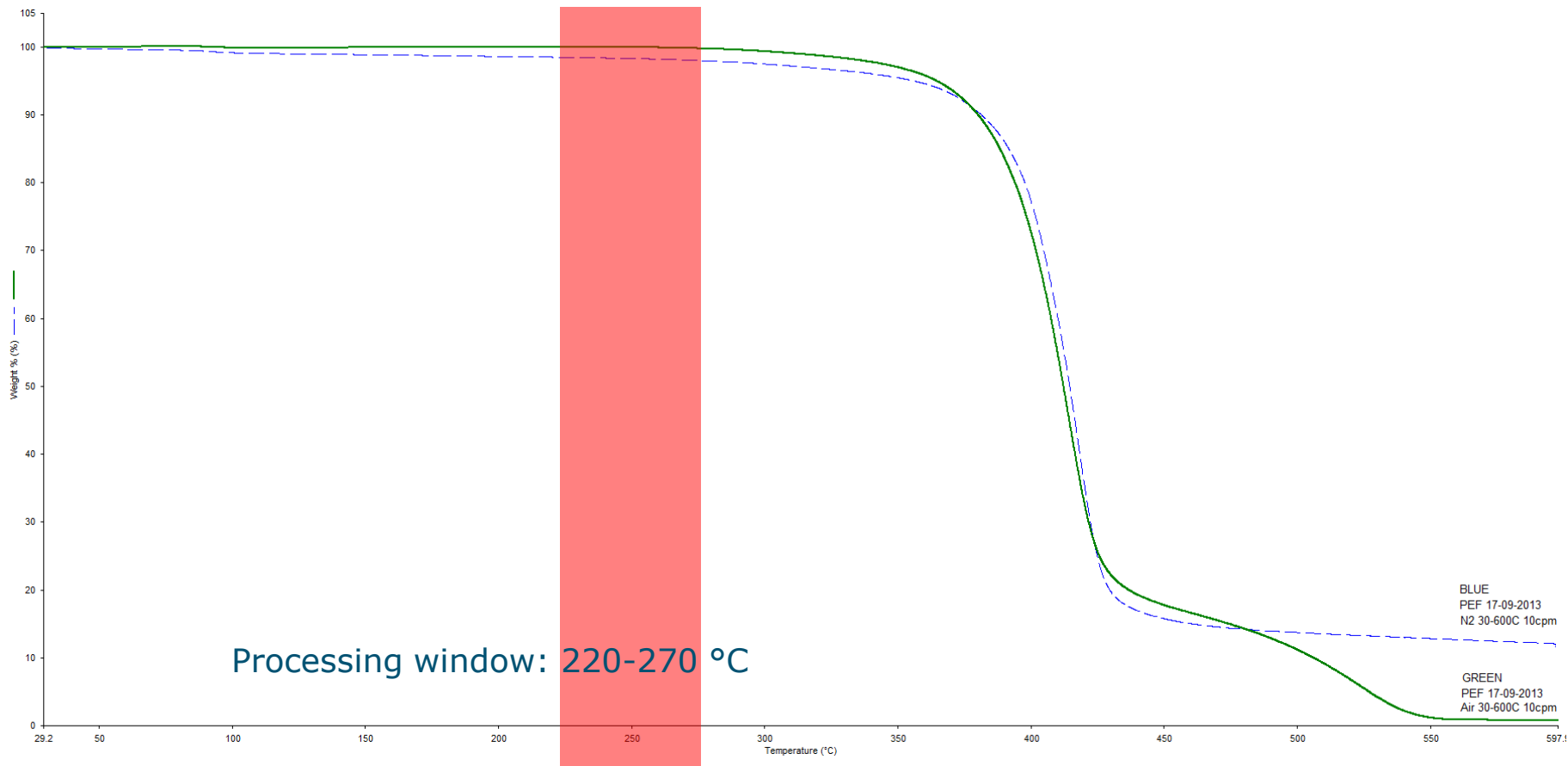
# PEF synthesis and properties

- Small scale synthesis of PEF
  - Glass, flat-flange reactor
  - 50-100 g scale
  - 2-stage melt polymerisation
    - 165°C, 16h
    - 225°C, 4h ambient pressure; 1 mbar, 2h;  $1 \cdot 10^{-1}$  mbar, 2h.
  - Purification: crude product dissolved in  $\text{CDCl}_3/\text{TFA}$  (6:1) and precipitated from MeOH (1,000 %
- 2 Liter polymerisation reactor installed



# PEF synthesis and properties

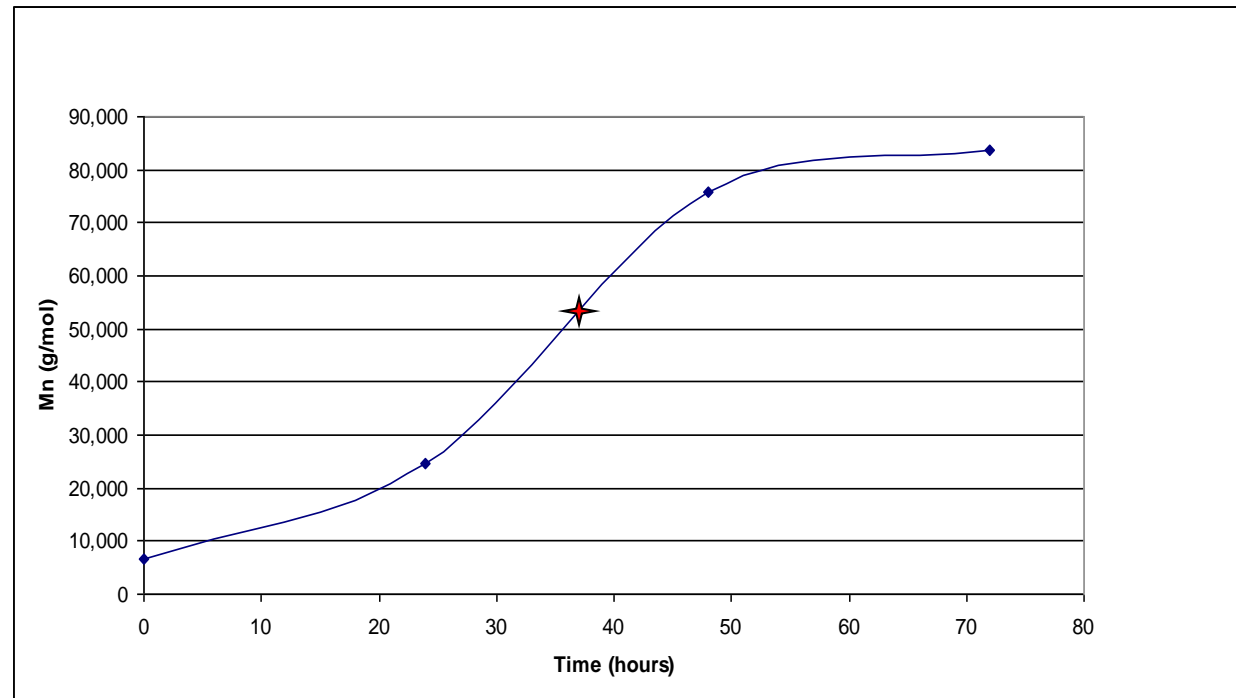
- TGA analysis of PEF sample; thermal stability (N<sub>2</sub>, air)



# Results (MW after solid state polymerization)



Rotary evaporator adapted for solid state polymerization



Molecular weight increment of PEF during SSP at 185°C over time. The red star shows the molecular weight of bottle grade PET .

# PEF synthesis and characterisation

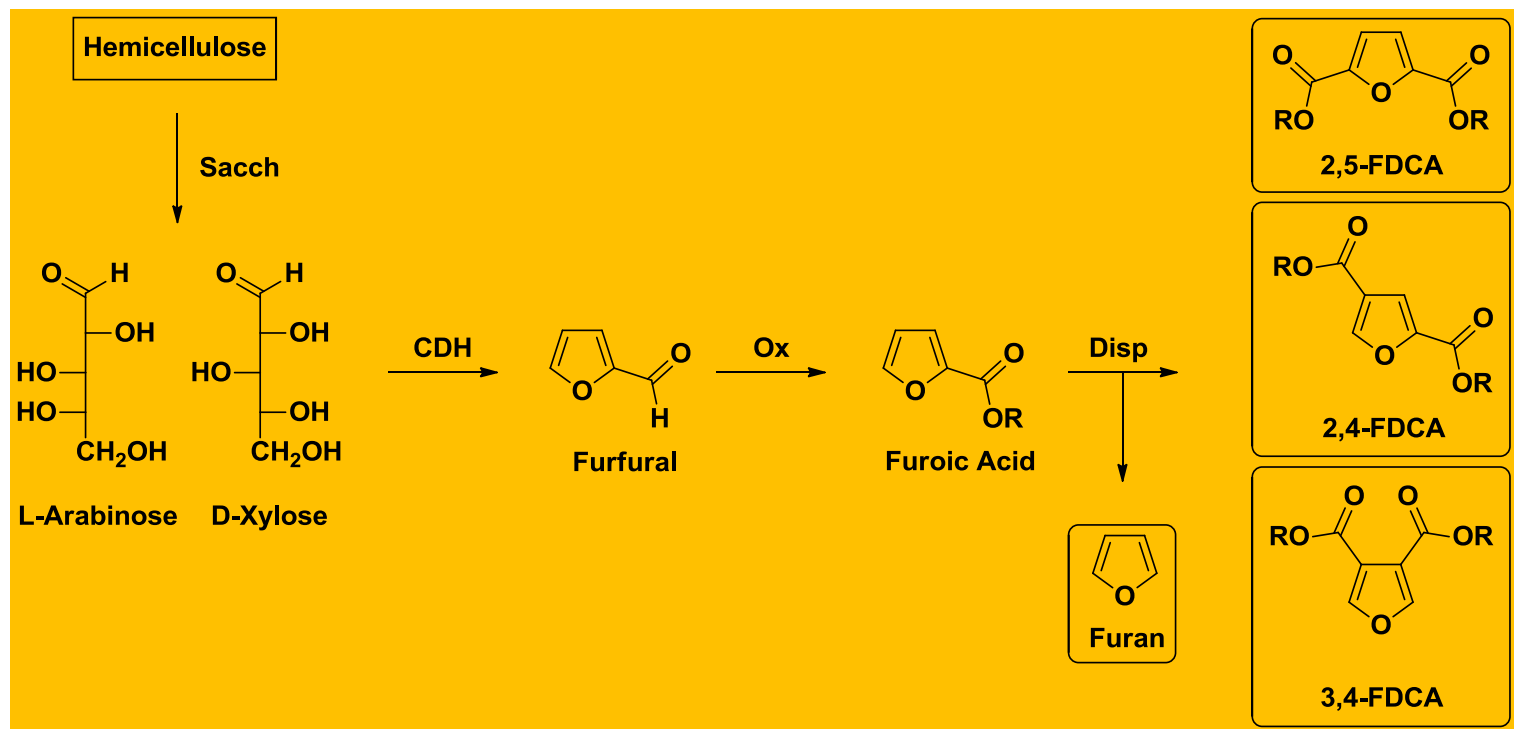
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## Conclusions

- High molecular weight PEF can be prepared from FDCA derived from sugarbeet pulp
- Galactaric acid derived FDCA is more easy to purify than HMF derived FDCA; beneficial in polymerisation reactions
- Molecular weight allows for potential application in e.g. bottles
- Further processing and characterisation (e.g. gas barrier properties) ongoing

# Furan dicarboxylic acids based upon hemicellulose

- Hemicellulose from agricultural side streams (e.g. sugarbeet pulp, wheat bran, bagasse) have huge potential to serve as source for the production of furandicarboxylic acids via C5 sugars



Thiyagarajan, S.; Pukin, A.; van Haveren, J.; Lutz, M.; van Es, D. S. *RSC Advances* **2013**, 3, 15678.

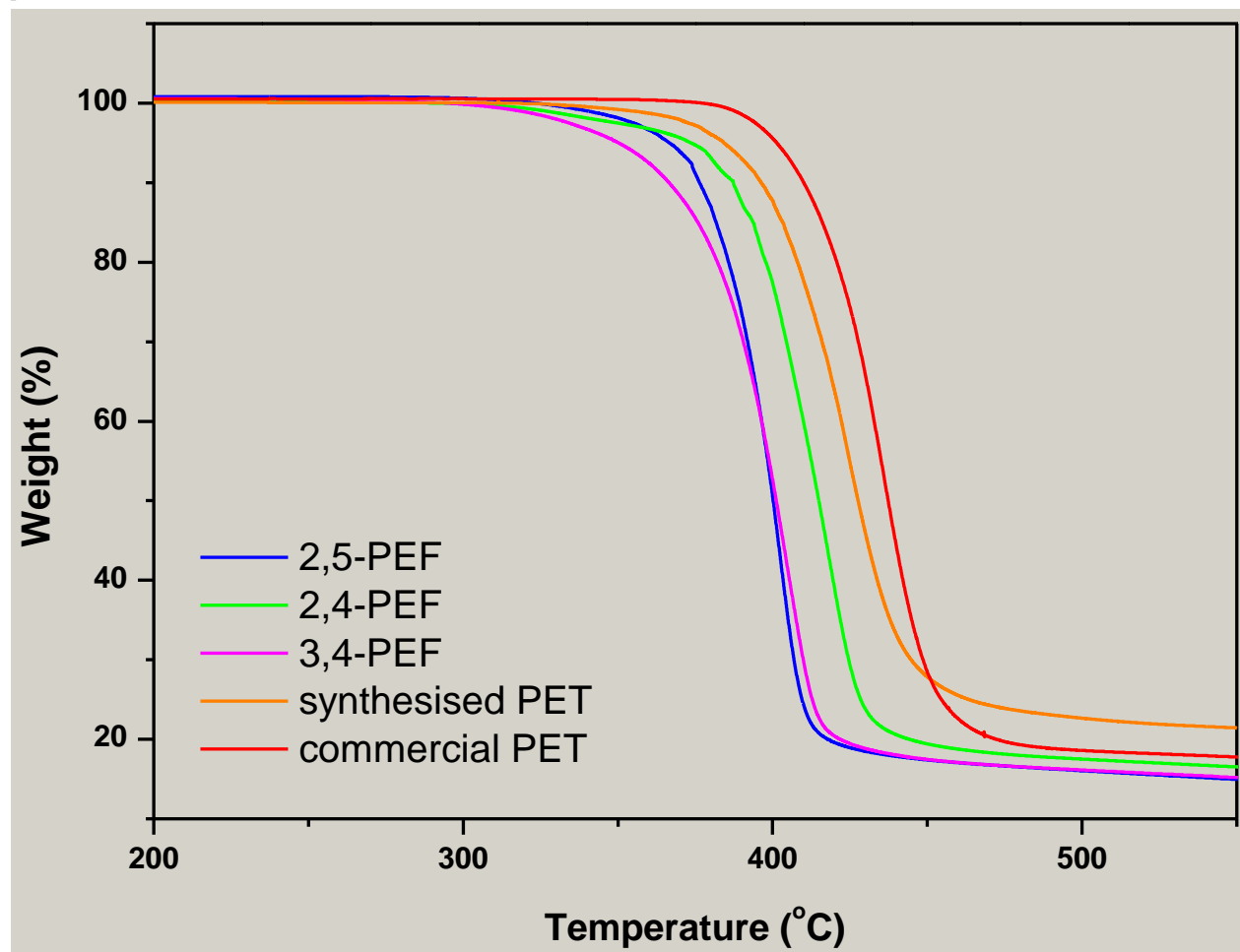


# Polymer results with 2,4 FDCA

## ■ PEF TGA analysis

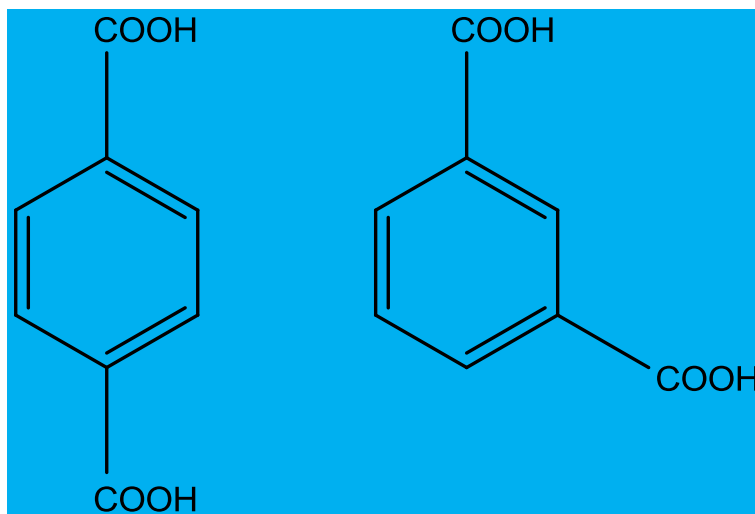
N<sub>2</sub>

10°C/min

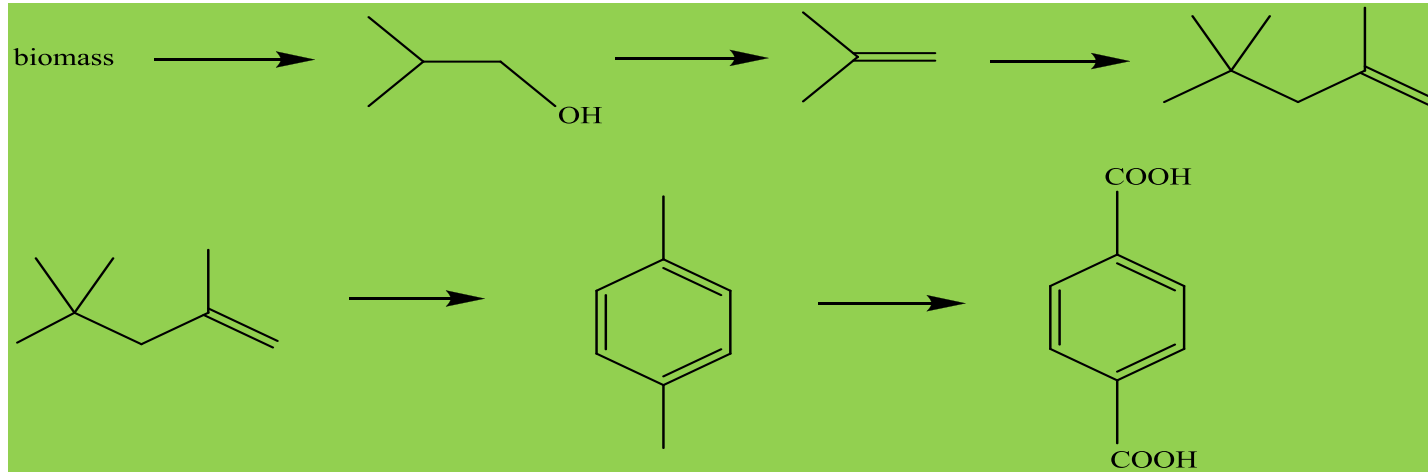


# Going for the drop-in approach....

- How to derive biobased terephthalic acid or isophthalic acid?



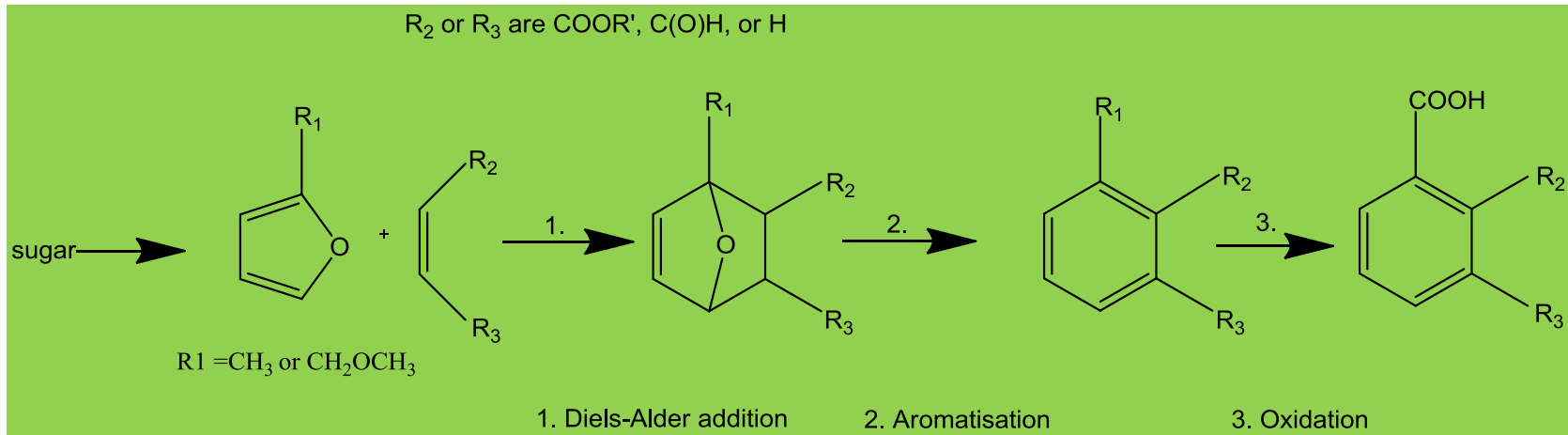
# Terephthalic acid via biotechnologically produced isobutanol



- Gevo Inc. promotes biotechnologically produced isobutanol as a platform chemical toward e.g. PTA; several patent applications on genetically modified organisms for production of isopropanol and isobutanol
- Conclusions Isobutanol platform Rotterdam; Current technology not yet mature; more research needed!

# Biobased terephthalic acid

- Diels –Alder addition of furans with dienophiles is a powerful tool to generate aromatic drop in chemicals from biomass



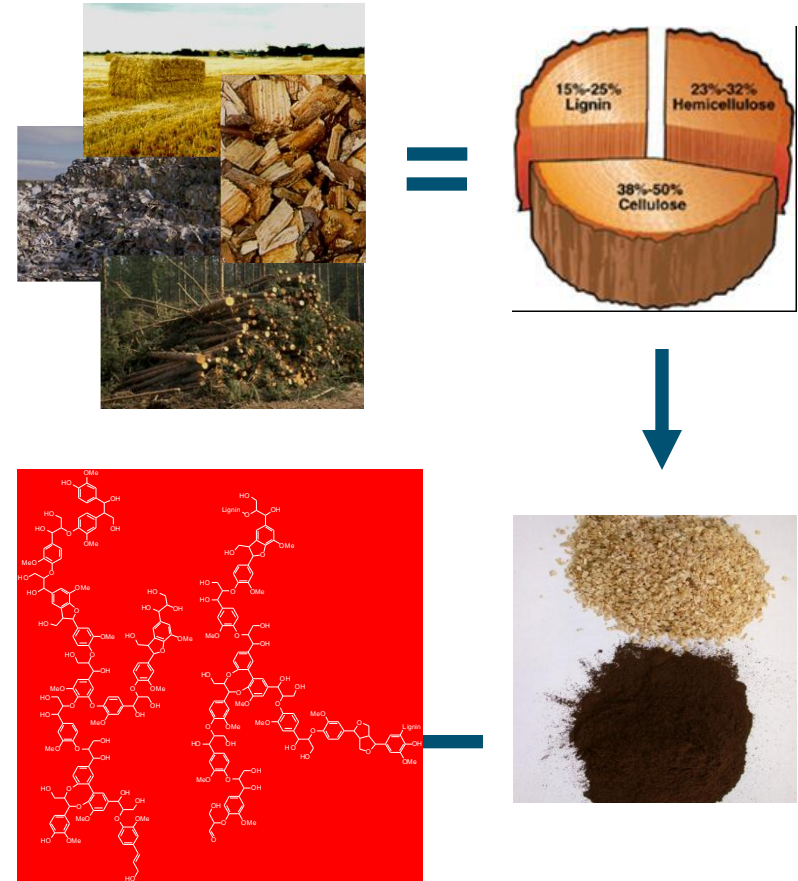
- Reaction of substituted furans with acrylates gives access to isophthalic acid (comonomer in PET), reaction with maleic acid anhydride results in hemimellitic acid (potential comonomer in coating formulations).
- Patent application to FBR and Avantium

# Lignin as a source for biobased aromatics

Lignin is found in plants and trees and is a rest stream from:

1. Established Pulp and Paper industry (50 M t/y, 1 M t/y products)
2. Novel Biorefinery industry (>20 M t/y)

- Major aromatic resource in Biobased Economy
- Complicated aromatic structure
- Complexity and heterogeneity has limited its use in materials and for chemicals production



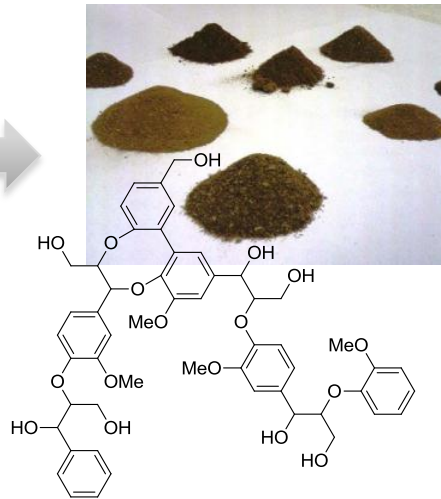
*softwood lignin structure Brunow 2001)*

# Hydrothermal depolymerization of lignin

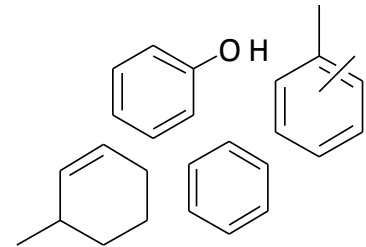
Biomass



Biorefinery lignins  
(commercial soda non-wood lignin)



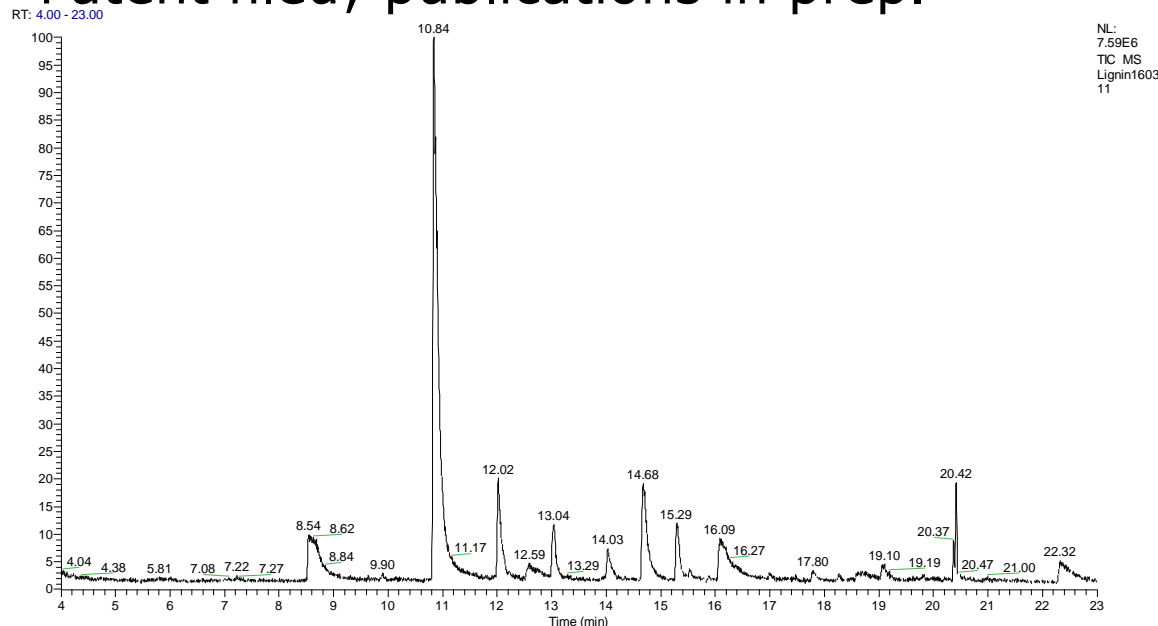
Aromatics and phenolics



- Turn lignin into high value aromatics (BTX) and building blocks (phenol)
- Thermal decomposition of lignin (pyrolysis, super critical fluids) often accompanied by unwanted re-condensation of monomeric phenol species
- Approach: Selective catalytic hydrothermal depolymerisation  
Prevent re-condensation & no external use of hydrogen

# Hydrothermal depolymerization of lignin

- Limited set of monomeric phenols (9-12) obtained in 20 mass% yield
- Literature: 20-30 identifiable components, 10% yield
- 50% 1 component
- Patent filed; publications in prep.



# Conclusions

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- Biomass offers excellent opportunities for the synthesis of biobased chemicals, including aromatic chemicals
- Both “drop-in” as well as biobased chemicals can be derived from biomass
- Biobased chemicals with a unique structure can result in product with unique properties
- Both lignin as well as carbohydrates offer huge opportunities for the synthesis of biobased aromatics



# Acknowledgements

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