Micro-structuring as a tool to control water binding of dairy-protein systems

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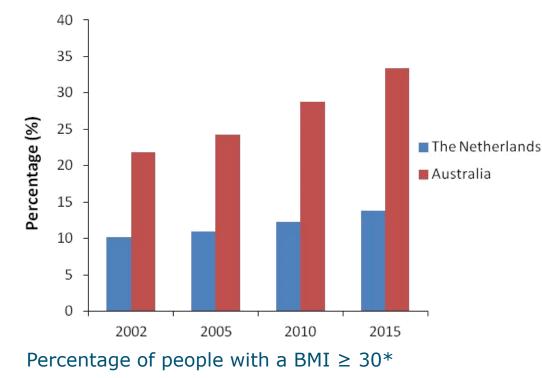


Overview presentation

- The need for low-caloric products
- The creation of superabsorbing particles
- Changing swellability MPs
- Conclusions



An increase in health awareness asks for the production of low-caloric foods







- Increase in people with overweight
- Increase in people that are health conscious

\rightarrow production of low caloric products

*WHO, 2014

Highly swellable protein structures are thought to make a difference

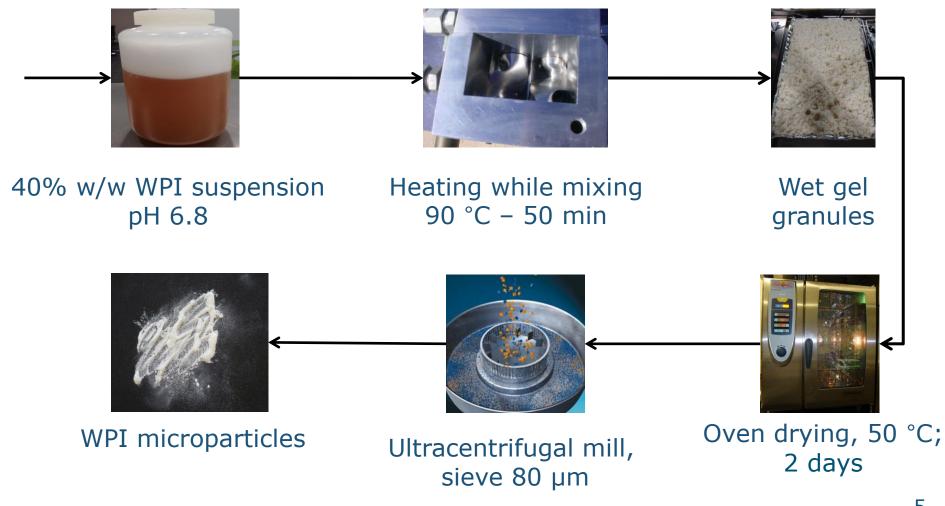


Problem: the addition of water can lead to softer products and/or syneresis

<u>Solution</u>

Creating protein structures which can absorb and tightly bind a relatively high amount of water

Swellable whey microparticles can be created by heat-induced gelation



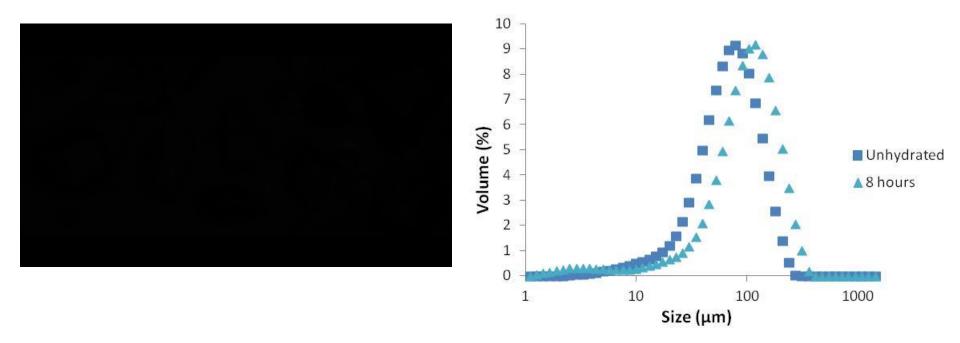
Properties of native WPI and WPI MPs

Native WPI	WPI microparticles
Contains undenatured proteins [*]	Complete*
~ 8 nm**	~ 70 µm
~ 96% w/w***	~ 93% w/w*
1.1 g/cm ^{3***}	1.3 g/cm ^{3*}
	Contains undenatured proteins* ~ 8 nm** ~ 96% w/w***

* Purwanti *et al.*, 2013
** Ju and Kilara, 1998

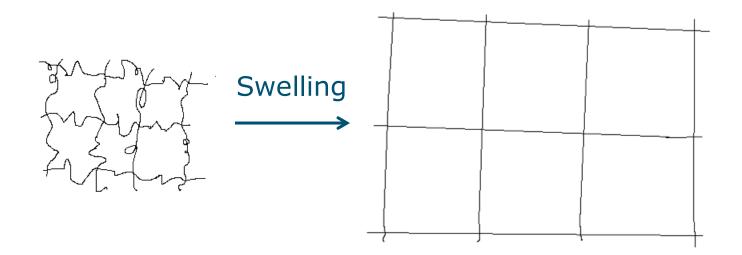
*** Purwanti *et al.*, 2012

Whey microparticles can swell



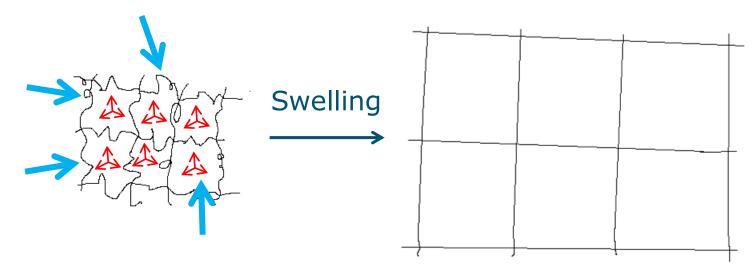
Whey microparticles can swell in water

How to increase swellability?



The Flory-Rehner equation describes swelling

Flory-Rehner equation



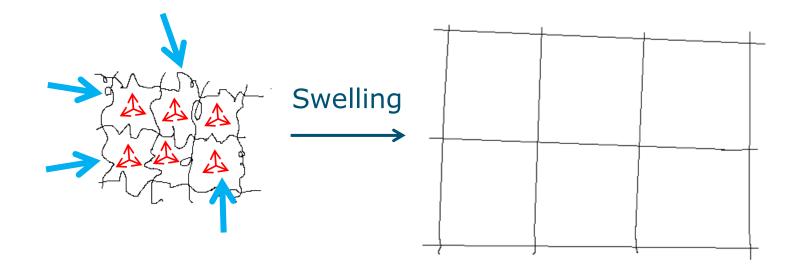
Swelling occurs when

- The protein and solvent want to mix
- The protein does not dissolve \rightarrow network
- The network can expand

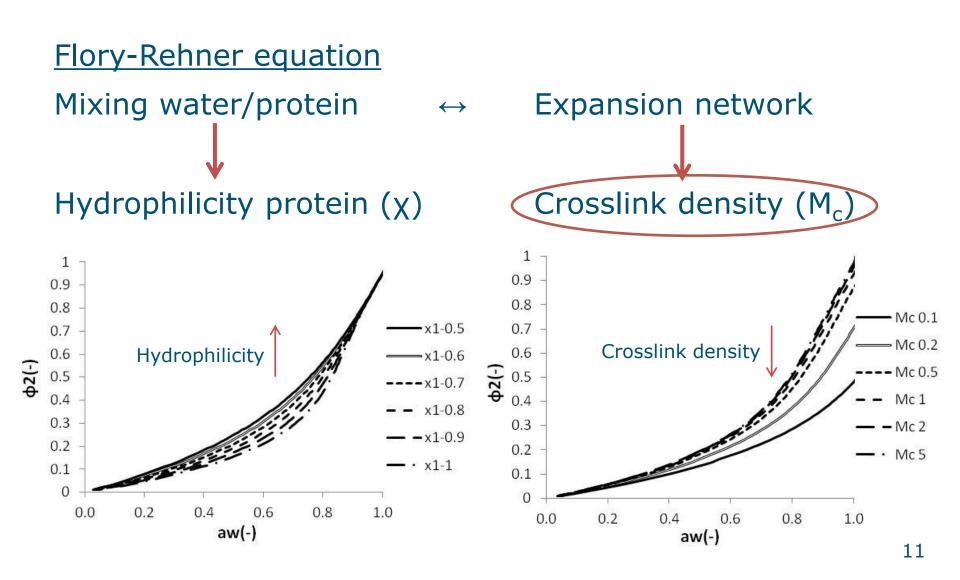
Networks swell until equilibrium is reached

Flory-Rehner equation

Mixing water/protein ↔ Expansion network



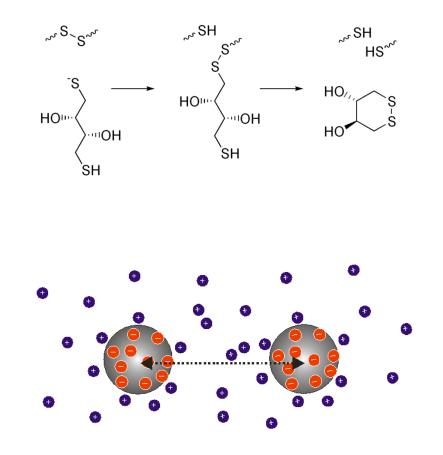
Changing the swellability of MPs by their hydrophilicity and/or crosslink density



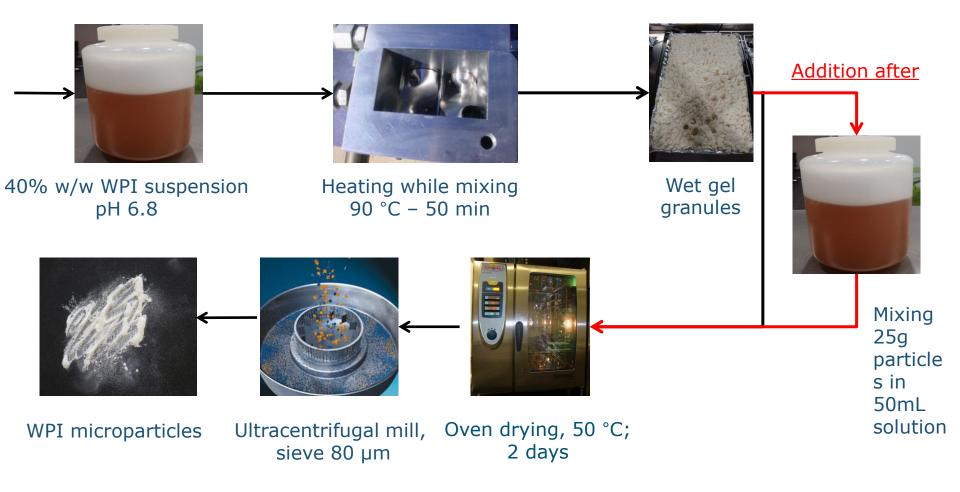
Changing the crosslink density of MPs with DTT and changing the pH

Dithiothreitol (DTT)

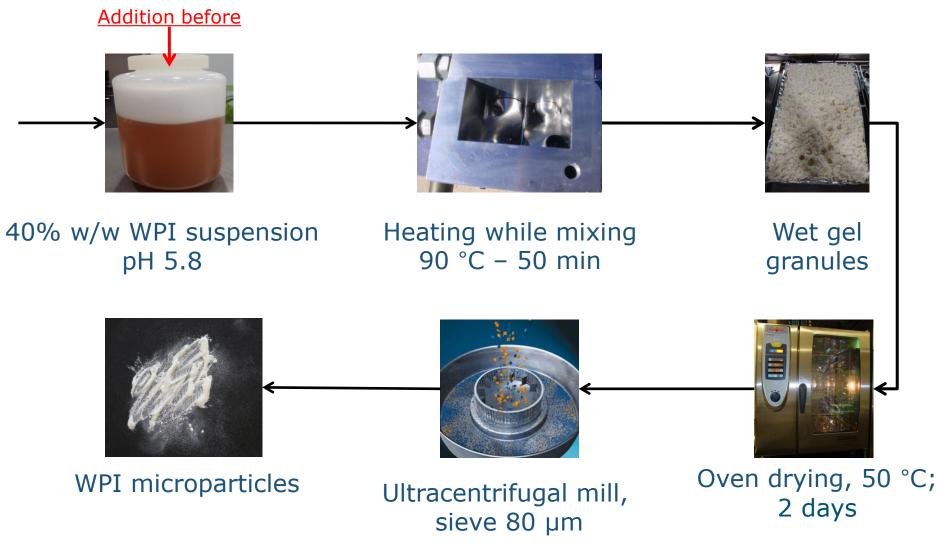
- Reducing agent
- Breaks disulphide bonds
- Decreasing pH to value closer to IEP
 - Reduction amount of charges → Proteins can come closer to eachother



Making of MPs with DTT



Making of MPs at pH 5.8

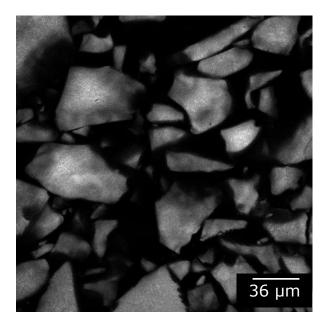


Measuring the WBC

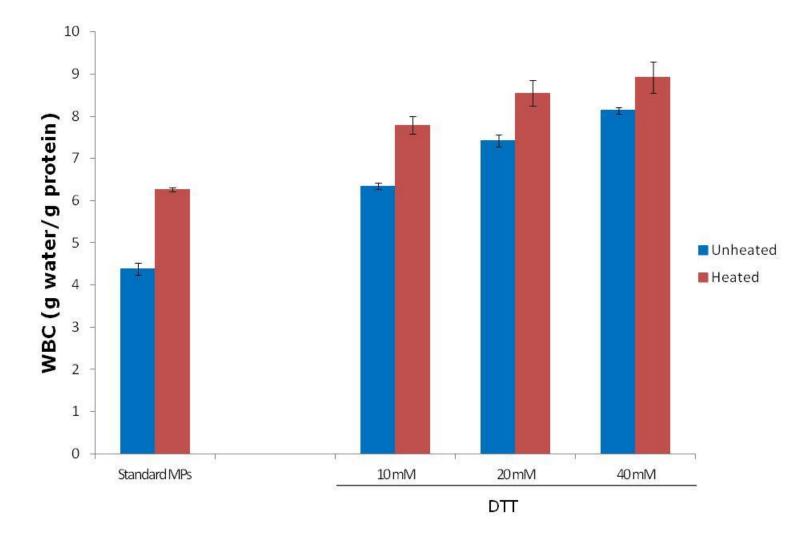
- Mixing a 10% dispersion of MPs + MQ-water
- Heating for 30 min. at 90°C)
- Centrifuging at 3000 rpm for 20 min (25°C)
- Dry pellet for 24 h at 105°C





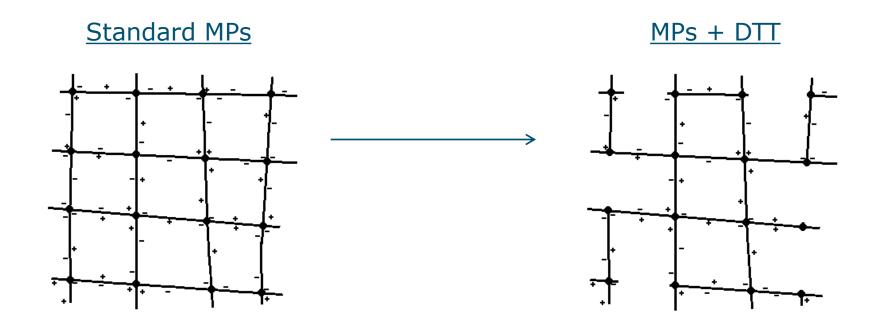


Decreasing the crosslink density with DTT



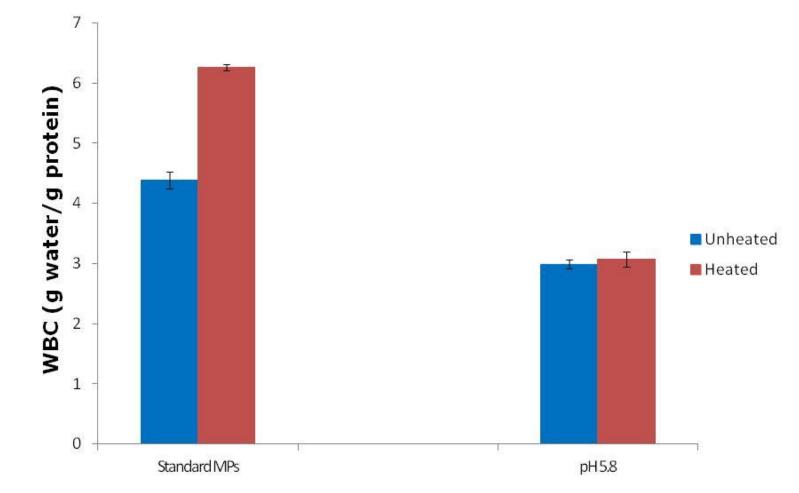
↑ concentration DTT \rightarrow ↑ WHC

Possible reactions MPs with DTT



Reduced crosslink density \rightarrow decreased elasticity \rightarrow increased swellability

Increasing the crosslink density by changing the pH to 5.8



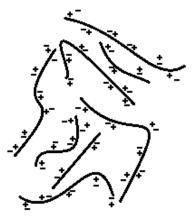
MPs made at pH 5.8 $\rightarrow \downarrow$ WBC

Possible reactions at pH 5.8

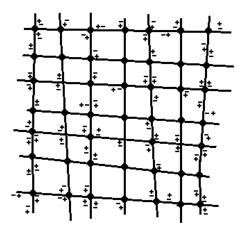




<u>WPI</u> dispersion at pH 5.8



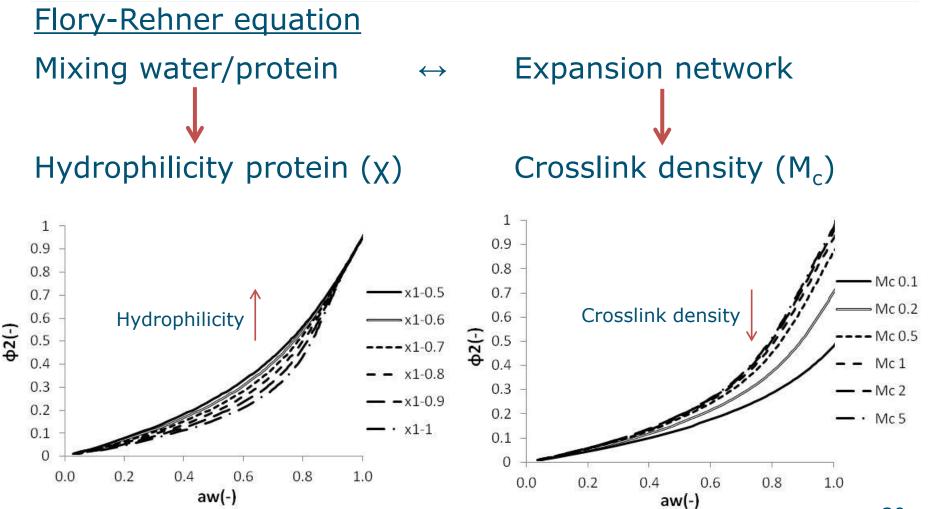




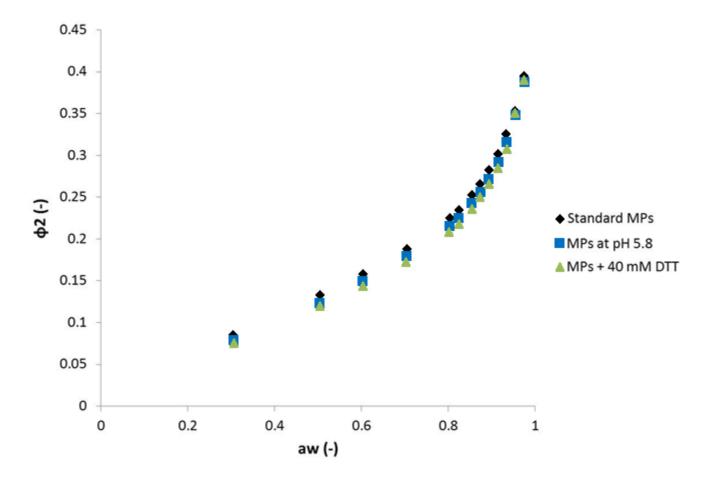
Whey protein chains are charged Charges proteins are screened Increased crosslink density in MP network

Increased crosslink density \rightarrow increased elasticity \rightarrow decreased swellability

Changing the swellability of MPs by their hydrophilicity and/or crosslink density



Isotherms standard and modified MPs



Isotherms are the same for standard and modified MPs

Conclusions



- MPs are structures that have a potential to be used in low-caloric cheese
- The crosslink density could be used to change the WBC of the MPs
 - DTT decreased the crosslink density and increased the WBC
 - Lowering the pH to values close to IEP increased the crosslink density and decreased the WBC

Thank you for your attention

