

# Ni<sub>2</sub>P as catalyst in oleic acid HDO: Influence of different carbon based supports

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#### Introduction

- Biomass is a potential source for making liquid fuel;
- Vegetable oils have C-chains similar in length to diesel;
- Oxygen content in vegetable oil too high for direct use as diesel
- Deoxygenation is needed;
- Metal phosphides are promising catalysts for deoxygenation;
- Carbon supports are stable under the relevant deoxygenation conditions;
- Role of the type of carbon support is unknown.

# Objective

Investigate the role of the carbon support (Activated carbon, carbon nanofibers, carbon covered alumina) in carbon supported Ni-phosphide catalysts for the deoxygenation of oleic acid.

#### **Methods**

Catalysts were synthesized via TPR from room temperature to 650 °C under  $H_2$  flow (heating rate 1 °C min<sup>-1</sup>).



#### **TPD to visualize electronic effects**

- Ni<sub>2</sub>P/CCA binds CO more strongly than Ni<sub>2</sub>P/CNF and Ni<sub>2</sub>P/AC;
- The stronger CO binds to the catalyst the higher the C17 yield.



Reactions were carried out in a trickle bed reactor at 350 °C and 30 bar  $H_2$  with dodecane as solvent and tetradecane as standard pattern.

Catalysts were diluted in SiC in a ratio of 1:1 (wt/wt).

Figure 1. Scheme of reaction.

# **Catalytic activity**



**Figure 3.** CO TPD profile for catalysts Ni<sub>2</sub>P/AC, Ni<sub>2</sub>P/CNF and Ni<sub>2</sub>P/CCA.

### **Catalyst properties**

Support	S <sub>B.E.T.</sub> (m <sup>2</sup> g <sup>-1</sup> )
Activated carbon (AC)	823
Carbon covered alumina (CCA)	193
Carbon nanofibers (CNF)	180

Crystallite diameter (nm)
32
32
19

• Metal loading, 30 wt% in all cases;

• Lower density of Ni sites for  $Ni_2P/AC$  compared to  $Ni_2P/CNF$  and  $Ni_2P/CCA$  (same loading and different surface area of supports).

#### **Conclusions/summary**

**Figure 2.** Product distribution of deoxygenation of oleic acid at 350 °C and 30 bar  $H_2$  over Ni<sub>2</sub>P/AC, Ni<sub>2</sub>P/CNF and Ni<sub>2</sub>P/CCA

- At conversion > 90% the C17/C18 product ratio increased in the order  $Ni_2P/AC < Ni_2P/CNF < Ni_2P/CCA$ ;
- Over Ni<sub>2</sub>P/CCA the decarbonylation and decarboxylation prevails;
- Over  $Ni_2P/AC$  hydrodeoxygenation is more significant (higher C18 yield).
- Support influences reaction pathway;
- Stronger interaction between CO and catalyst enables higher C17 yield – decarbonylation and decarboxylation pathway (Ni<sub>2</sub>P/CCA);
- Lower density of Ni sites enables higher C18 yield hydrodeoxygenation pathway (Ni<sub>2</sub>P/AC).

#### Acknowledgements

CNPq – National Counsel of Technological and Scientific Development - Brazil



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