

This un-edited manuscript has been accepted for publication in Biophysical Journal and is freely available on BioFast at <http://www.biophysj.org>. The final copyedited version of the paper may be found at <http://www.biophysj.org>.

## Excitation energy transfer and charge separation in photosystem II membranes revisited

Koen Broess<sup>a</sup>, Gediminas Trinkunas<sup>a,b</sup>, Chantal D. van der Weij - de Wit<sup>c</sup>, Jan P. Dekker<sup>c</sup>, Arie van Hoek<sup>ad</sup>, and Herbert van Amerongen<sup>ad</sup>.

<sup>a</sup>Wageningen University, Laboratory of Biophysics, PO Box 8128, 6700 ET, Wageningen, The Netherlands

<sup>b</sup>Institute of Physics, Vilnius 02300, Lithuania

<sup>c</sup>Department of Biophysics, Division of Physics and Astronomy, Faculty of Sciences, Vrije Universiteit, De Boelelaan 1081, 1081 HV Amsterdam, The Netherlands

<sup>d</sup>MicroSpectroscopy Centre, Wageningen University, 6703 HA Wageningen, The Netherlands

### Abstract

We have performed time-resolved fluorescence measurements on photosystem II (PSII) containing membranes (BBY particles) from spinach with open reaction centers. The decay kinetics can be fitted with two main decay components with an average decay time of 150 ps. Comparison with recent kinetic exciton annihilation data on the major light-harvesting complex of PSII (LHCII) suggests that excitation diffusion within the antenna contributes significantly to the overall charge separation time in PSII, which disagrees with previously proposed trap-limited models. In order to establish to which extent excitation diffusion contributes to the overall charge separation time, we propose a simple coarse-grained method, based on the supramolecular organization of PSII and LHCII in grana membranes, to model the energy migration and charge separation processes in PSII simultaneously in a transparent way. All simulations have in common that the charge separation is fast and nearly irreversible, corresponding to a significant drop in free energy upon primary charge separation, and that in PSII membranes energy migration imposes a larger kinetic barrier for the overall process than primary charge separation.

Keywords: antenna, light-harvesting complexes, model, reaction center, time-resolved fluorescence.

### INTRODUCTION

Photosystem II (PSII) is a large supramolecular pigment-protein complex embedded in the thylakoid membranes of green plants, algae and cyanobacteria. It uses sunlight to split water into molecular oxygen, protons and electrons. PSII is conventionally subdivided into 1) a core consisting of light-harvesting complexes CP43 and CP47 and the reaction center (RC), where excitation energy is used to create a charge separation that is stabilized by secondary electron transfer processes, and 2) an outer antenna of chlorophyll (Chl) *a/b* binding proteins, containing the majority of the light-harvesting pigments. The latter proteins, of which the trimeric light-harvesting complex II (LHCII) is by far the most abundant, are not only required for the efficient absorption of light and the transfer of excitation energy to the RC under light-limiting conditions,

they also play essential roles in several regulation mechanisms of the photosynthesis process under light-saturating conditions, like state transitions and non-photochemical quenching (See e.g. Pascal et al)(1).

The overall quantum efficiency of the charge separation process depends on the relative rate constants of various processes: 1) excitation energy transfer (EET) from chlorophylls in the light-harvesting antenna to the chlorophylls in the RC that perform the charge separation (CS), 2) charge separation and charge recombination in the RC, 3) stabilization of the charge separation by secondary electron transfer, and 4) trivial relaxation or loss processes of the excited state: intersystem crossing, internal conversion and fluorescence.

It is important to know which of the above-mentioned processes determine the overall rate of charge separation in open, fully functional PSII (with an RC in which the secondary electron acceptor  $Q_A$  is oxidized). This knowledge is needed for a detailed understanding of the kinetics of regulation processes like non-photochemical quenching. For a long time it has been assumed by many authors that the charge-separation process in PSII is trap-limited, i.e. the excitation energy diffusion through the antenna to the RC is much faster than the overall charge separation time. Since the eighties the so-called exciton/radical-pair-equilibrium (ERPE) model (2,3) has been a popular way to interpret time-resolved and steady-state fluorescence data of PSII containing preparations. More recently, Klug and coworkers concluded from the study of a whole range of PSII containing preparations possessing different antenna sizes that the charge separation is indeed trap-limited (4). However, from singlet-singlet annihilation studies on LHCII trimers and aggregates it was concluded that the excitation diffusion within the outer antenna is relatively slow (5) and that charge separation in LHCII-containing PSII systems cannot be entirely trap-limited (6,7). Also Jennings and coworkers came to the same conclusion (8).

At the moment a large number of experimental data is available on the charge-separation kinetics of isolated PSII RC's and PSII core complexes (9,10). In PSII RC and CP47-RC preparations (which contain 6 and 22 chlorophylls, respectively, and 2 pheophytins, but do not contain the secondary electron acceptor  $Q_A$ ) the kinetics were strongly multi-exponential. They could be explained by three reversible radical pair states, of which the first is nearly isoenergetic with the singlet-excited state of the primary electron donor (P680\*), in combination with the absence of severe kinetic limitation for the excitation energy transfer between CP47 and the RC (11). PSII core complexes (with 35 chlorophylls and 2 pheophytins) do contain  $Q_A$ , and in open centers (with  $Q_A$  oxidized) the decay kinetics are dominated by a major phase in the 30-60 ps time range and a minor phase of about 200 ps (12-14). The energy difference between the first radical pair state and P680\* is probably much larger than in PSII RC and CP47-RC preparations (14).

It is unknown to which extent these systems give kinetics compatible with more native-like systems like chloroplasts, thylakoid membranes and PSII membranes (the so-called BBY preparations). Most studies on entire chloroplasts or thylakoid membranes suggested average values for the trapping time in PSII in the range from ~300 to ~500 ps (15-17). However, fast PSI fluorescence may partly mask faster PSII decay processes for these preparations. Moreover, unconnected light-harvesting complexes may be present in the stroma lamellae, which can further complicate the identification of the PSII fluorescence (18).

PSII grana membranes (BBY preparations) do not contain PSI or stroma lamellae, but do contain a much larger antenna than PSII core particles. Due to the presence of trimeric and monomeric Chl *a/b* containing complexes, these membranes contain about 150 Chls *a* per PSII, about 4 times more than isolated PSII core particles (19). The kinetics in these membranes were described by a single lifetime of about 210 ps (20) or with a major lifetime of 140 ps and a minor lifetime of 330 ps (12). A number of other studies revealed slower kinetics, which can be explained by a ‘contamination’ of closed centers (with Q<sub>A</sub> single or double reduced).

In this paper we present new time-resolved fluorescence data on BBY preparations and propose a coarse-grained model in which previously reported antenna and RC kinetics can easily be incorporated, allowing a comparison with the obtained fluorescence kinetics of PSII in grana membranes. To this end we make use of available knowledge about the supramolecular organization of PSII (19). The results reveal a number of essential differences in primary charge separation in isolated RC’s, PSII cores and PSII membranes, and stress that diffusion of the excitation energy in the membranes contributes significantly to the overall charge separation kinetics. The presented framework will facilitate new studies that are directed at the contributions of individual complexes to the overall kinetics by using mutant preparations with altered PSII composition or organization.

## **MATERIALS AND METHODS**

### **Sample preparation**

PSII membranes (BBY particles) were prepared according to Berthold et al. (21) from fresh spinach leaves. An analysis by diode-array-assisted gel filtration chromatography, performed as described previously (22), showed that the preparations contained at most 1-2% of PSI.

### **Time-correlated single photon counting**

Steady-state fluorescence spectra were measured with a Fluorolog-3.22 (SPEX Industries, USA) at room temperature. Time-correlated single photon counting (TCSPC) measurements were performed at magic angle (54.7°) polarization as described previously (23). The BBY particles were diluted to an OD of 0.08 per cm in a buffer of 20 mM Hepes pH 7.5, 15 mM NaCl and 5 mM MgCl<sub>2</sub>. The repetition rate of excitation pulses was 3.8 MHz and the excitation wavelength was 430 nm. Pulse energies of sub-pJ were used with pulse duration of 0.2 ps and spot diameter of 1 mm. The samples were placed in a 3.5 mL and 10 mm light path fused silica cuvet and stirred in a temperature controlled (20 °C) sample holder. In combination with the low intensities of excitation this guaranteed that close to 100% of the reaction centers stayed open (see also results) and significant build-up of triplet states was avoided. The full-width at half maximum (fwhm) of the system response function was 60 ps with a resolution of 2.51 ps per channel. The dynamic instrumental response function of the setup was obtained from pinacyanol in methanol with a lifetime of 10 ps. The following interference filters were used for detection: 671, 679, 688, 693, 701, 707, 713, and 724 nm (Balzers, Liechtenstein model B40). Data analysis was performed using a home built computer program (24,25). A fast component (~5 ps) was needed

in most cases to fit the time range around the rising edge of the excitation pulse but this component is not relevant for this study and is omitted in the further modelling.

### **Synchroscan streak camera**

For the streak-camera measurements the BBY particles were diluted to an OD of 0.7 per cm in a buffer of 20 mM BisTris pH 6.5 and 5 mM MgCl<sub>2</sub>. Ferricyanide (1 mM) was added to keep the reaction centers open. 400 nm excitation pulses of ~100 fs were generated using a Ti:sapphire laser (VITESSE, Coherent St. Clara, CA) with a regenerative amplifier (REGA, Coherent). The repetition rate was 150 kHz, and the pulse energy was 1 nJ. The excitation light was focused with a 15 cm focal length lens, resulting in a focal diameter of 150 μm in the sample. To refresh the sample between the excitation pulses, the sample was placed into a 2 mm thick spinning cell of 10 cm diameter, rotating at a speed of 20 Hz. The fluorescence was detected in a direction at 90° from the excitation beam through a detection polarizer at magic angle, an orange sharp cut-off filter glass, a Chromex 250IS spectrograph and a Hamamatsu C 5680 synchroscan streak camera. The streak images were recorded with a cooled, Hamamatsu C4880 CCD camera. The FWHM of the overall time response of the experiment was 5 ps. Global analysis was applied, using a model with a number of parallel decaying compartments, which yields decay associated spectra (DAS) (26).

## **RESULTS AND DISCUSSION**

### **Time-resolved fluorescence measurements**

In Fig. 1 a typical TCSPC decay curve for PSII grana membranes with open RC and Chl *a* excitation (430 nm) is shown. To obtain a good multi-exponential fit, at least 4 decay times are needed. The contribution of a 2.9 ns component is very small (less than 0.5%) and is probably due to very small amounts of PSII with closed RC's, free Chl and/or detached pigment-protein complexes. Most of the decay can be described by two major components and a minor one: 77 ps (41%), 206 ps (56%) and 540 ps (3%). The excitation intensity was low enough to avoid excitation annihilation (singlet-singlet or singlet-triplet) or accumulation of closed RC's. Increasing the excitation intensity with a factor of 10 led to identical decay kinetics, whereas an increase with a factor of 1000 led to substantially longer decay times because of the closure of RC's (data not shown).

Decay curves were measured at different detection wavelengths and the decay times were very similar in all cases. The result of a global analysis of all decay curves is given in Fig. 2, showing decay-associated spectra (DAS). At all detection wavelengths the two longest decay times are nearly absent. The fitted decay times are 80 ps and 212 ps for the two major components. The contribution from a 633 ps component is small and the 2.9 ns component has negligible amplitude. The DAS are dominated by a main fluorescence band peaking between 680 and 690 nm and show small vibronic bands above 700 nm. The average lifetime of 150 ps is significantly shorter than previously estimated values for chloroplasts and thylakoid membranes (300 to 500 ps), but it is closer to the values obtained for BBY by Schilstra et al. (20) and Van Mieghem et al. (12). To determine whether processes are present that are faster than the time resolution of the

photon counting setup, the experiments were also performed with a streak-camera with 3 ps time resolution. The results are shown in Fig. 3. The data are rather similar; the decay is dominated by two components with lifetimes 81 ps (60%) and 258 ps (40%). No short-lived component was resolved. Note that the times are similar but not identical to those obtained with the TCSPC measurements. This is mainly due to some variability in the samples. However, this variability is irrelevant for the main conclusions drawn in this paper and leave some uncertainty in the presented parameters. Note that the difference in excitation wavelength can also cause some variability but it is less outspoken (work in progress).

## Modelling of the fluorescence kinetics

The overall average charge separation time  $\tau$  can be considered as the sum of two times, the first passage time or migration time  $\tau_{\text{mig}}$ , representing the average time that it takes for an excitation created somewhere in PSII to reach the RC (primary donor), and the trapping time  $\tau_{\text{trap}}$  (7) (27), page 23-27. The trapping time is the product of the intrinsic charge separation time  $\tau_{\text{ICS}}$  (when the excitation is located on the primary donor) and the probability that the excitation is located on the primary donor after Boltzmann equilibration of the excitation over PSII. In a system with  $N$  isoenergetic pigments this would mean that  $\tau_{\text{trap}} = N \tau_{\text{ICS}}$ . Note that  $\tau_{\text{mig}}$  can be split into an equilibration time in the antenna and a transfer-to-the-trap time (7) (27), page 23-27(7) (27), page 23-27.. but this approach is not needed here.

First we introduce a simple basic model to describe the overall CS kinetics in PSII in terms of the CS kinetics in the RC and EET in the antenna complexes. Thereafter, we show how the results are affected when the model is extended. In Fig. 4 we show the dimeric supercomplex of PSII (28) that is used for our coarse-grained modelling. It is a basic unit that can be further associated in different ways to form larger organization patterns (19,29). Besides two RC's it contains 2 CP47, 2 CP43, 2 CP24, 2 CP26, 2 CP29 monomers and 4 LHCII trimers. We define a hopping rate  $k_h$  for transfer between all neighbouring monomeric complexes and/or subunits that are connected via a bar in Fig 4. It is worth mentioning that also EET between monomeric LHCII subunits in the trimer is modelled in this way. The reason that we take the same hopping rate in all cases is the fact that all outer antenna complexes are rather homologous and that energy transfer is largely determined by the transfer within the complexes (see also below). The situation may be different for hopping from CP43 or CP47 to the RC and this case will be discussed separately. Forward and backward rates between complexes have been adjusted by rescaling the single hopping rate in accordance with the differences in the Chl  $a$  numbers (see Appendix for the details and figure 4 for the number of Chl  $a$  molecules). The outer antenna complexes all transfer their excitations to the RC via CP47 or CP43. Excitations can leave the RC again into the antenna.

It should be noted that we also examined the effect of increasing the number of (connected) supercomplexes, based on the various models for megacomplexes (dimeric supercomplexes) that have been detected thus far (19). However, no essential differences were observed. Therefore it is sufficient to consider the basic unit with only two RC's. All complexes are taken to be isoenergetic (30).

At first we assume that irreversible charge separation takes place in the RC, which in our definition consists of the 6 central chlorins in the RC, with a rate  $k_{CS}$ . Note that this is different from the intrinsic CS rate  $k_{iCS}$ . The 2 peripheral Chls in the RC are assumed to be part of the antenna system and one of them is assigned to CP47 and the other to CP43. This is justified because the distance of these peripheral chlorophylls to the nearest chlorophylls in CP47 or CP43 is shorter than to the nearest central chlorin in the RC (31). In the simplest (but non-realistic) case of 6 isoenergetic central chlorins in the RC with primary CS occurring from one pigment,  $k_{iCS}$  would be equal to  $6 k_{CS}$ .

Fig. 4 shows 2 LHCII trimers per RC but it is known that on average 4 LHCII trimers are present per RC (7). The other two trimers can be in a different membrane layer organized in such a way that they can still transfer the excitation energy to the RC's (29), but they can also be located close to a PSII-LHCII supercomplex in the same layer, in particular in membranes without ordered arrays of PSII. Because it is unknown how the extra 2 LHCII's are connected to the RC it is only possible to guess their contribution to the overall trapping time. We consider two extreme cases. If these four LHCII trimers per PSII would be in the same membrane layer as the RC, the overall  $\tau_{mig}$  would become close to 160 ps, as was concluded from singlet-singlet annihilation (5). The only assumption in that case is that the connectivity between the additional light-harvesting complexes and the others is the same as between the ones that were already present. The important point is that the migration time increases. The value of 160 ps is approximately equal to the observed average lifetime for BBY preparations, which would imply that the charge separation is nearly diffusion limited. Although we cannot rule out this possibility, it seems highly unlikely. We will return to this point later.

As another extreme case we assume that a regular 2-dimensional lattice with 100 sites (the approximate number of Chl *a* per RC in the supercomplex shown in Fig. 4) is extended to a regular 3-dimensional lattice with 148 sites (2 extra trimers) with the same hopping rates. This reduces  $\tau_{mig}$  with approximately 10% (Ref. (27), page 406). The same hopping rates may not be realistic, but energy transfer between membranes in a grana stack will very likely occur within the excited state lifetime (32,33). On the other hand  $\tau_{trap}$  increases with 48% because the equilibrium distribution of excited states is shifted further towards the antenna. For the purpose of this paper it is not necessary to discuss explicitly all the different possible organizations. They will be discussed implicitly by considering different combinations of the hopping rates and charge-separation rates.

The overall fluorescence decay (reflecting decay of excited-state population) can now be calculated for the model system in Fig. 4 for any initial excitation distribution (see Appendix). It can be compared to the fluorescence kinetics of PSII membranes with open centers (Figs. 1-3). In other words we reconstruct the experimental decay by including only the dominating components and the minor component of 633 ps (Fig. 2). We assume an initial distribution between the various complexes which is proportional to the number of Chls *a* per complex. Fig. 5 shows the reconstructed decay and the best fit of the above model over the time range 0-700 ps for the TCSPC data. This simple model provides a good description of the kinetics. The fitted hopping rate is  $(17 \text{ ps})^{-1}$  and the charge separation rate is  $(1.2 \text{ ps})^{-1}$ . It should be noted that the experimentally observed non-exponentiality in this case is not modelled because it is explicitly assumed that charge separation is irreversible. We will show the effect of including charge recombination below. In this way non-exponentiality is introduced. However, it is also possible

that the non-exponentiality is due to some structural heterogeneity and in this case one might expect to obtain a distribution of trapping time and the fitted value should be considered to be an average trapping time.

Before we discuss more realistic models and the uniqueness of the fit, it is worthwhile to look at the consequences of these rates. The hopping rate  $(17 \text{ ps})^{-1}$  is rather slow and corresponds to a value of  $\tau_{\text{mig}}$  of 130 ps (see Appendix for method of calculation). The charge-separation rate  $k_{\text{CS}}$  of  $(1.2 \text{ ps})^{-1}$  is the effective rate for the whole RC, i.e. the central 6 chlorins. If primary CS occurs from one specific Chl, then  $k_{\text{ICS}} = (1.2 \text{ ps})^{-1} / 6 = (0.2 \text{ ps})^{-1}$  in the case of isoenergetic pigments. This would mean that  $\tau_{\text{trap}} = \tau_{\text{ICS}} \times N = 0.2 \text{ ps} \times 100 = 20 \text{ ps}$  if an organization as in Fig. 4 is considered, or  $\tau_{\text{trap}} = 0.2 \text{ ps} \times 150 = 30 \text{ ps}$  if the Chl *a* content in PSII membranes is considered. Clearly, in this case the overall trapping time is dominated by the migration time. The streak-camera data were modelled in the same way and led to  $k_h = (17.5 \text{ ps})^{-1}$  and  $k_{\text{CS}} = (0.4 \text{ ps})^{-1}$ . The observed differences in the fluorescence lifetime can easily be explained by some variability in the preparations. At this point it is not useful to discuss the differences in fitting results because the fitting outcome is not unique (see below).

The given rates do not uniquely describe the data within the context of the above model. In Fig. 6 we show different combinations of  $k_h$  and  $k_{\text{CS}}$  that lead to a reasonable description of the TCSPC data. The results were obtained as follows: We chose a particular value for  $\tau_h (= k_h^{-1})$  and looked for the best fit of  $\tau_{\text{CS}} (= k_{\text{CS}}^{-1})$ . Varying for instance  $\tau_h$  from 10 to 20 ps leads to fits for which the quality is rather similar (see Fig. 6),  $\tau_{\text{mig}}$  varies from 77 to 150 ps whereas the charge separation time varies from 4.3 to 0 ps. A slower migration towards the RC requires a faster charge separation in order to obtain the same experimentally observed decay rates. It is clear that different combinations of hopping and charge separation times can explain the observed kinetics. Given the approximate nature of the modelling, no strong conclusions can be drawn from the differences between the simulated and the experimental curves.

### **Modulating excitation energy transfer from CP47 or CP43 to the RC**

In the above model we assumed that CP47 can transfer energy to two different RC's (see Fig. 4). It is not entirely clear from the crystal structure whether this really is the case. Therefore, we also considered the case that CP47 can transfer to only one RC. Then a hopping time of 15.2 ps is obtained for the best fit and a charge separation time of 0.23 ps. Because there are less routes for reaching the RC, one needs to speed up the transfer and charge separation process in order to arrive at a good fit. The migration time is 147 ps, i.e. the contribution from the migration time remains dominant. This illustrates the fact that the outer antenna determines to a large extent the total migration time. However, in this case the charge separation becomes unrealistically fast and in the following we consider the situation that CP47 is connected to two RC's.

Although it has been argued that energy transfer from CP47 or CP43 to the RC is relatively slow (7,34), we also consider a rather extreme case in which this transfer time is 3 times shorter than the general hopping time. This would be in agreement with measurements on RC and CP47-RC preparations, which indicated that the energy transfer between CP47 and RC is not rate-limiting (11) and that the connecting chlorophylls of CP47 and CP43 are optimally oriented for fast

energy transfer (35). The best fit now requires a value of  $\tau_h$  of 24.8 ps and  $\tau_{\text{mig}}$  is 100 ps, meaning that the migration time is still dominant.

### Reversible charge separation

Above we made the assumption that the charge separation is irreversible. Although it leads to a satisfactory description of the data, it contrasts with the general opinion that substantial charge recombination occurs. Therefore, we extended our model by including recombination and a second charge-separated state. It is not required for the fitting to specify the nature of such a second charge-separated state but it might for instance be the reduced  $Q_A$  in combination with the oxidized primary donor. The electron back-transfer rate ( $k_{bCS}$ ) to the primary donor is related to the intrinsic charge-separation rate from this primary donor via the detailed balanced relation  $k_{bCS} / k_{iCS} = e^{-\Delta G/kT}$  where  $\Delta G$  is the drop in free energy upon primary charge separation,  $k$  is the Boltzmann constant and  $T$  is the absolute temperature. The rate and time constant of secondary charge separation are called  $k_{RP}$  and  $\tau_{RP}$ , respectively. The data can now be fitted in different ways, depending on the starting values of the different fitting parameters. Two fits are shown in Fig. 7. The dashed line corresponds to a slow hopping time (17 ps). In this case the times for primary and secondary charge separation are 1.24 and 13.3 ps, respectively and  $\Delta G = 2380 \text{ cm}^{-1}$ . The solid line is a fit with an extremely fast hopping time (1.3 ps). The times for primary and secondary charge separation are 6.6 and 168 ps, respectively and  $\Delta G = 890 \text{ cm}^{-1}$ . The crucial point is that although the rates of hopping and secondary charge separation cannot be separately estimated from these fits, fast primary charge separation in combination with a large drop in free energy is needed to describe the data.

### Comparison with charge separation in isolated reaction centres and core complexes

We make a comparison with models for charge separation that have been presented in literature based on measurements on isolated PSII RC and core complexes. We inspect what happens to the calculated trapping kinetics for BBY particles when charge separation in the PSII RC is described according to these models. We restrict ourselves to the most recent ones that can be directly incorporated into the above framework.

The model of (open) PSII core complexes from Vassiliev et al. (13) is given in Fig. 8 (model 3). It is characterized by many fast electron transfer steps, the first one being the intrinsic charge separation rate from the presumed lowest exciton state of the “special pair” or two accessory chlorophylls. This model is incorporated in our initial description above, i.e. instead of a unidirectional charge separation step in the RC with  $\tau_{CS} = 1.2 \text{ ps}$  we use the charge separation scheme from Fig. 8c. The first step in the latter scheme is slowed down by a factor of 6/4 because CS can take place from 4 out of 6 chlorophylls. The results are given in Fig. 9b (thick, solid line). The hopping time is taken to be 17 ps, i.e. the time that we found in the best fit of the first model. The simulated kinetics show a slightly faster initial decay and a larger contribution from a slow decay component when compared to the experimentally observed decay for BBY preparations. We inspected how we could bring the model into accordance with the BBY data by keeping everything the same except the first rate of charge separation and the corresponding change in free energy. It was possible to obtain a very good fit (not shown) by changing  $\tau_{CS}$  from 0.15 ps into 0.75 ps and  $\Delta G$  from  $-464 \text{ cm}^{-1}$  into  $-826 \text{ cm}^{-1}$  respectively, with a hopping time of



17 ps. The charge separation is slowed down in order to match the initial part of the decay curve and the drop in free energy is increasing, making the charge separation less reversible, leading to smaller contributions from slow components. It is exactly this lack of slow component in the BBY data that requires a large drop in free energy upon fast charge separation. The presence of two additional trimers per RC will shift the equilibrium further towards the excited states, leading to even more fluorescence at longer times in the modelled curve, i.e. to a greater discrepancy.

Alternatively, we tried to fit their PSII core model to the BBY data by optimizing the hopping rate. The result can also be seen in Fig. 9b (thick, dashed line) and the fitted hopping time in this case is 13.4 ps. The fit is better than in the first case above, but the decay remains too fast at early times and the contribution of the slow component is too large. So independent of the details of the model, the large drop in free energy upon initial charge separation appears to be essential to describe the BBY data. It is unclear whether the core preparations contain a fraction of complexes in which the lifetimes are too long, or whether the drop in free energy in (cyanobacterial) core particles is indeed less pronounced than in BBY particles. The latter possibility could arise from a different ligation of the pheophytin that serves as electron acceptor. The residue that is involved in a H-bond with pheophytin is a Gln in cyanobacteria and a Glu in higher plant PSII, which can give rise to a shift of the redox potential of the pheophytin by about 30 meV (36-38). It should be noted that the presence of two additional trimers per RC will shift the equilibrium further towards the excited states, leading to even more fluorescence at longer times in the modelled curve, i.e. to a greater discrepancy.

Very recently, new data on isolated cores were obtained and a different model was proposed by Miloslavina et al.(39). The model assumes ultrafast energy transfer from CP47 and CP43 to the RC, and the authors conclude that the kinetics are trap-limited in these complexes and that charge separation can be described by a scheme that includes reversible charge separation to several radical pair states (Fig. 8d). When we incorporate this scheme for charge separation into our model (hopping time 17 ps) the resulting kinetics are far too slow (Fig. 9b, thin, solid line). Even when we assume the excitation energy transfer throughout the antenna to be infinitely fast, the resulting kinetics are still too slow (Fig. 9b, thin, dashed line). To improve the fit, like before, a larger drop in free energy is needed ( $648\text{ cm}^{-1}$  instead of  $294\text{ cm}^{-1}$ ) and the charge separation time should decrease: 2.1 ps instead of 5.9 ps. It should be noted that the scheme of Miloslavina et al.(39) does not include the fitting of a 111 ps component (amplitude 10%) that was observed in their experiments. Incorporating this component in the model would further increase the discrepancy.

Other models have been proposed that were based on the measurements on isolated PSII RC complexes. Two recent ones are represented in Figs. 8a and 8b. They cannot directly be compared directly to the BBY results because the isolated RC's do not contain the electron acceptor  $Q_A$ . Therefore, we tested the hypothesis that the initial charge separation kinetics/energetics in isolated RC's are the same as in the open BBY's. We used the models as presented in 8a and 8b up to state RP2, whereas electron transfer to  $Q_A$  was modelled by an irreversible decay of state RP2 with rate constant  $(200\text{ ps})^{-1}$ . With a hopping time of 17 ps the modelled kinetics are far too slow for both models (Fig. 9a, solid lines). Even when the hopping is assumed to be infinitely fast, the simulated kinetics are still much slower. Again, the agreement between the BBY data and the RC model can only be improved by using a fast charge

separation in combination with a large drop in free energy. However, a fast charge separation and a large drop in free energy are not in agreement with the measured fluorescence kinetics of isolated PSII RC complexes. Note that the inclusion of back transfer of an electron from the  $Q_A$  would only increase the discrepancy.

One might compare the RC in isolated PSII RC's (without  $Q_A$ ) with closed RC's in  $Q_A$ -containing PSII preparations. It was already observed many years ago (40) that closed RC's show considerably slower fluorescence kinetics than open RC's, which could be modelled by a 6-fold slower rate constant for charge separation and a  $400\text{ cm}^{-1}$  higher energy of the primary radical pair (3). The slowing down of the primary charge separation reaction was explained by electrostatic repulsion due to the negative charge on  $Q_A$ , but other authors suggested that the charge on  $Q_A$  has a minor effect on the energy level of the primary radical pair (4,14). The relative importance of electrostatic repulsion for slowing down charge separation was demonstrated by Van Mieghem et al. (12), who found a considerable difference in charge separation kinetics between centers with singly and doubly reduced  $Q_A$  in PSII membranes (where the fluorescence kinetics and integrated emission yield in centers with doubly reduced  $Q_A$  was just in between those with oxidized and singly reduced  $Q_A$ ) but not in PSII core particles. In conclusion, the RC in isolated PSII RC's has indeed a closer resemblance to closed RC's in  $Q_A$ -containing PSII complexes, already at the level of primary charge separation, despite the absence of a reduced  $Q_A$ .

### **Contribution of the migration time to the overall trapping time**

It was recently suggested that the overall charge separation process cannot be entirely trap-limited in grana membranes (41). From singlet-singlet annihilation studies (5) on LHCII trimers and aggregates it was apparent that the spatial equilibration time per trimer is several tens of ps. A value of 48 ps was determined for trimers whereas this number was approximately 32 ps per trimer in lamellar LHCII aggregates. It was argued (6) that the latter time might be faster because excitations have the tendency to be located at the outside of the trimer, thereby facilitating energy transfer and thus annihilation in aggregates. The fact that the annihilation in trimers is slower than in aggregates indicates that it is not limited by hopping between different complexes but by relatively slow transfer within the complexes, in agreement with pump-probe and photon-echo data (6,42).

Therefore, the contribution of the time of transfer in or between LHCII trimers to the overall migration is approximately equal to the number of trimers per RC multiplied by the equilibration time per trimer, provided that they are in "intimate contact" within one plane. The supercomplex in Fig. 4 contains 2 LHCII trimers per RC which contribute each  $\sim 32$  ps to the  $\tau_{\text{mig}}$  (5). Moreover, CP24, CP26, and CP29 each show high homology to an LHCII monomer, and together they add another  $\sim 32$  ps. For CP47 and CP43 these numbers are less well known, but they are probably faster. The overall migration time would thus be around 100 ps, which constitutes a large fraction of the overall trapping time. This number would even be larger when the 2 "missing trimers" are located in the same plane, but if they would be in a different layer this value might be slightly smaller (see above). In the simulations that we showed above it was found that a hopping time of 17 ps leads to a total migration time of 130 ps. To arrive at a migration time of 100 ps, the hopping time has to be decreased proportionally, i.e. from 17 ps to

(100/130)x 17 ps = 13 ps. At the moment it is uncertain to what extent the excitation migration times determined for isolated LHCII trimers and lamellar aggregates are directly applicable to the BBY preparations. The organization of the complexes will have some influence, although it was argued above that migration is to a large extent determined by migration within the individual complexes. Also the details of the annihilation process from which the migration times were determined have some influence. This issue will be addressed in a future study.

Of course, the proposed modelling procedure for BBY is approximate. However, it provides an easy way to incorporate existing knowledge and models for individual complexes and despite remaining uncertainties it is demonstrated that valuable conclusions can be drawn about both the excitation energy transfer and the charge separation. The exact contribution of excitation diffusion (migration time) to the overall charge separation remains somewhat uncertain, which results in uncertainty in  $\tau_{CS}$ . However, the relation and consequences are transparent and can easily be extracted. It is also clear that charge separation should be rather fast and is accompanied with a large drop in free energy. This contrasts with existing models for primary charge separation in isolated PSII RC's without quinone and in PSII RC's with quinone as present in core preparations.

Possible future experiments include preferential excitation of different pigments to study the effect on the overall kinetics. Mutants are available that are lacking specific pigment-protein complexes and the kinetics can be measured and modelled. Moreover, the effect on the fluorescence kinetics by introducing quenchers in different positions can be predicted and tested in case of the occurrence of nonphotochemical quenching. As such, the proposed method offers a way to study PSII performance as a whole in a directed way, which hopefully contributes to a gradual improvement of the knowledge about PSII functioning.

## Acknowledgements

This work is part of the research programme of the 'Stichting voor Fundamenteel Onderzoek der Materie (FOM)', which is financially supported by the 'Nederlandse Organisatie voor Wetenschappelijk Onderzoek (NWO)'. K.B and C.D.W.W. were both supported by FOM. J.P.D. acknowledges support from the European Union (Grant MRTN-CT-2003-505069, Intro2). GT acknowledges support (visitor grant) from the 'Nederlandse Organisatie voor Wetenschappelijk Onderzoek (NWO)'. The authors thank Henny van Roon for help with the biochemical analysis of the samples. The authors thank Dr. Roberta Croce for critically reading the manuscript.

## References

1. Pascal, A.A., Z. Liu, K. Broess, B. van Oort, H. van Amerongen, C. Wang, P. Horton, B. Robert, W. Chang, and A. Ruban. 2005. Molecular basis of photoprotection and control of photosynthetic light-harvesting. *Nature* 436:134-137.

2. van Grondelle, R. 1985. Excitation-energy transfer, trapping and annihilation in photosynthetic systems. *Biochim. Biophys. Acta* 811:147-195.
3. Schatz, G.H., H. Brock, and A.R. Holzwarth. 1988. Kinetic and energetic model for the primary processes in photosystem II. *Biophys. J* 54:397-195.
4. Barter, L.M.C., M. Bianchetti, C. Jeans, M.J. Schilstra, B. Hankamer, B.A. Dinner, J. Barber, and J.R. Durrant. 2001. Relationship between excitation energy transfer, trapping, and antenna size in photosystem II. *Biochemistry* 40:4026-4034.
5. Barzda, V., V. Gulbinas, R. Kananavicius, V. Cervinskis, H. Van Amerongen, R. Van Grondelle, and L. Valkunas. 2001. Singlet-singlet annihilation kinetics in aggregates and trimers of LHCII. *Biophys. J.* 80:2409-2421.
6. van Amerongen, H., and R. van Grondelle. 2001. Understanding the energy transfer function of LHCII, the major light-harvesting complex of green plants. *J. Phys. Chem. B* 105:604-617.
7. van Amerongen, H., and J.P. Dekker. 2003. Light-Harvesting in Photosystem II. *In Light-Harvesting Antennas in Photosynthesis*. Green BR, Parson WW, editors. Kluwer Academic Publishers. 219-251.
8. Jennings, R., G. Elli, F.M. Garlaschi, S. Santabarnara, and G. Zucchelli. 2000. Selective quenching of the fluorescence of core chlorophyll-protein complexes by photochemistry indicates that Photosystem II is partly diffusion limited. *Photosynth. Res* 66:225-233.
9. Dekker, J.P., and R. van Grondelle. 2000. Primary charge separation in photosystem II. *Photosynth. Res* 63:195-208.
10. Dinner, B.A., and F. Rappaport. 2002. Structure, dynamics, and energetics of the primary photochemistry of photosystem II of oxygenic photosynthesis. *Ann. Rev. Plant Biol* 53:551-560.
11. Andrizhiyevskaya, E.G., D. Frolov, R. van Grondelle, and J.P. Dekker. 2004. On the role of the CP47 core antenna in the energy transfer and trapping dynamics of photosystem II. *Phys. Chem. Chem. Phys* 6:4810-4819.
12. van Mieghem, F.J.E., G.F.W. Searle, A.W. Rutherford, and T.J. Schaafsma. 1992. The influence of the double reduction of QA on the fluorescence decay kinetics of photosystem II. *Biochim. Biophys. Acta* 1100:98-206.
13. Vassiliev, S., C.-I. Lee, G.W. Brudvig, and D. Bruce. 2002. Structure-based kinetic modelling of excited-state transfer and trapping in histidine-tagged PSII core complexes from *Synechocystis*. *Biochemistry* 41:12236-12243.
14. Andrizhiyevskaya, E.G. 2005. Energy transfer and trapping in photosynthetic complexes with variable size. Vrije Universiteit Amsterdam.
15. Roelofs, T.A., C.-I. Lee, and A.R. Holzwarth. 1992. Global target analysis of picosecond chlorophyll fluorescence kinetics from pea chloroplasts – a new approach to the characterization of the primary processes in photosystem II alpha-units and beta-units. *Biophys. J* 61:1147-1163.
16. Gilmore, A.M., T.L. Hazlett, P.G. Debrunner, and Govindjee. 1996. Photosystem II chlorophyll a fluorescence lifetimes and intensity are independent of the antenna size differences between barley wild-type and chlorina mutants: Photochemical quenching and xanthophyll cycle-dependent nonphotochemical quenching of fluorescence. *Photosynth. Res* 48:171-187.

17. Vasil'ev, S., S. Wiebe, and D. Bruce. 1998. Non-photochemical quenching of chlorophyll fluorescence in photosynthesis. 5-hydroxy-1,4-naphthoquinone in spinach thylakoids as a model for antenna based quenching mechanisms. *Biochim. Biophys. Acta* 1363:147-156.
18. Zhang, S., and H.V. Scheller. 2004. Light-harvesting Complex II Binds to Several Small Subunits of Photosystem I. *The Journal of Biological Chemistry* 279(5):3180-3187.
19. Dekker, J.P., and E.J. Boekema. 2005. Supramolecular organization of thylakoid membrane proteins in green plants. *Biochim. Biophys. Acta* 1706:12-39.
20. Schilstra, M.J., J. Nield, W. Dörner, B. Hankamer, M. Carradus, L.M.C. Barter, J. Barber, and D.R. Klug. 1999. Similarity between electron donor side reactions in the solubilized Photosystem II-LHC II supercomplex and Photosystem-II-containing membranes. *Photosynth. Res* 60:191-198.
21. Berthold, D.A., G.T. Babcock, and C.F. Yocum. 1981. A highly-resolved, oxygen-evolving photosystem II preparation from spinach thylakoid membranes. *FEBS Lett* 134:231-234.
22. van Roon, H., F.L. van Breemen, F.L. De Weerd, J.P. Dekker, and E.J. Boekema. 2000. Solubilization of green plant thylakoid membranes with n-dodecyl- $\alpha$ ,D-maltoside. Implications for the structural organization of the photosystem II, photosystem I, ATP synthase and cytochrome b6/f complexes. *Photosynth. Res* 64:155-166.
23. Somsen, O.J.G., A. van Hoek, and H. van Amerongen. 2005. Fluorescence quenching of 2-aminopurine in dinucleotides. *Chem. Phys. Lett* 402:61-65.
24. Digris, A.V., V.V. Skakoun, E.G. Novikov, A. van Hoek, A. Claiborne, and A.J.W.G. Visser. 1999. Thermal stability of a flavoprotein assessed from associative analysis of polarized time-resolved fluorescence spectroscopy. *Eur. Biophys. J.* 28:526-531.
25. Novikov, E.G., A. van Hoek, A.J.W.G. Visser, and J.W. Hofstraat. 1999. Linear algorithms for stretched exponential decay analysis. *Opt. Commun.* 166:189-198.
26. van Stokkum, I.H.M., D.S. Larsen, and R. van Grondelle. 2004. Global and target analysis of time-resolved spectra. *Biochim. Biophys. Acta* 1657:82-104.
27. van Amerongen, H., L. Valkunas, and R. van Grondelle. 2000. Photosynthetic excitons. World Scientific Publishing Co. Pte. Ltd, Singapore.
28. Boekema, E.J., H. van Roon, F. Calkoen, R. Bassi, and J.P. Dekker. 1999. Multiple types of association of photosystem II and its light-harvesting antenna in partially solubilized photosystem II membranes. *Biochemistry* 38:2233-2239.
29. Boekema, E.J., F.L. van Breemen, H. van Roon, and J.P. Dekker. 2000. Arrangement of photosystem II supercomplexes in crystalline macrodomains within the thylakoid membranes of green plants. *J. Mol. Biol* 301:1123-1133.
30. Jennings, R.C., R. Bassi, F.M. Garlaschi, P. Dainese, and G. Zucchelli. 1993. Distribution of the chlorophyll spectral forms in the chlorophyll-protein complexes of photosystem II antenna. *Biochemistry* 32:3203-3210.
31. Loll, B., J. Kern, W. Saenger, A. Zouni, and J. Biesiadka. 2005. Towards complete cofactor arrangement in the 3.0 Å resolution structure of photosystem II. *Nature* 438:1040-1044.
32. Leibl, W., J. Breton, J. Deprez, and H.-W. Trissl. 1989. Photoelectric study on the kinetics of trapping and charge stabilization in oriented PS-II membranes. *Photosynth. Res* 22:257-275.

33. Kirchhoff, H., M. Borinski, S. Lehnert, L.F. Chi, and C. Büchel. 2004. Transversal and lateral exciton energy transfer in grana thylakoids of spinach. *Biochemistry* 43:14508-14516.
34. Vasil'ev, S., P. Orth, A. Zouni, T.G. Owens, and D. Bruce. 2001. Excited-state dynamics in photosystem II: Insights from the x-ray crystal structure. *Proc. Natl. Acad. Sci. USA* 98:6802-6807.
35. Vasil'ev, S., and D. Bruce. 2004. Optimization and evolution of light-harvesting in photosynthesis: the role of antenna chlorophyll conserved between photosystem II and photosystem I. *Plant Cell* 16:3059-3068.
36. Merry, S.A.P., P.J. Nixon, L.M.C. Barter, M. Schilstra, G. Porter, J. Barber, J.R. Durrant, and D.R. Klug. 1998. Modulation of Quantum Yield of Primary Radical Pair Formation in Photosystem II by Site-Directed Mutagenesis Affecting Radical Cations and Anions. *Biochemistry* 37:17439-17447.
37. Rappaport, F., M. Guergova-Kuras, P.J. Nixon, B.A. Diner, and J. Lavergne. 2002. Kinetics and Pathways of Charge Recombination in Photosystem II. *Biochemistry* 41:8518-8527.
38. Cuni, A., L. Xiong, R. Sayre, F. Rappaport, and J. Lavergne. 2004. Modification of the pheophytin midpoint potential in photosystem II: Modulation of the quantum yield of charge separation and of charge recombination pathways. *Phys.Chem.Chem.Phys.* 6:4825-4831.
39. Miloslavina, Y., M. Szczepaniak, M.G. Müller, J. Sander, M. Nowaczyk, M. Rögner, and A.R. Holzwarth. 2006. Charge Separation Kinetics in Intact Photosystem II Core Particles Is Trap-Limited. A Picosecond Fluorescence Study. *Biochemistry* 45:2436-2442.
40. Schatz, G.H., H. Brock, and A.R. Holzwarth. 1987. Picosecond kinetics of fluorescence and absorbancy changes in photosystem II particles excited at low photon density. *Proc. Natl. Acad. Sci. USA* 84:8414-8418.
41. Engelmann, E.C.M., G. Zucchelli, F.M. Garlaschi, A.P. Casazza, and R.C. Jennings. 2005. The effect of outer antenna complexes on the photochemical trapping rate in barley thylakoid photosystem II. *Biochim. Biophys. Acta* 1706:276-286.
42. Novoderezhkin, V.I., M.A. Palacios, H. van Amerongen, and R. van Grondelle. 2005. Excitation Dynamics in the LHClI Complex of Higher Plants: Modeling Based on the 2.72 Å Crystal Structure. *J. Phys. Chem. B* 109:10493-10504.
43. Groot, M.L., N.P. Pawlowicz, L.J.G.W. van Wilderen, J. Breton, I.H.M. van Stokkum, and R. van Grondelle. 2005. Initial donor and acceptor in isolated photosystem II reaction centers identified with femtosecond mid-IR spectroscopy. *Proc. Natl. Acad. Sci. USA* 102:13087-13092.

## Figure Legends

Figure 1. Room temperature fluorescence decay curve (measured with TCSPC) for open BBY preparations together with a fit. The sample was excited at 430 nm and fluorescence was detected at 693 nm. The decay times and their relative amplitude are 77 ps (41%), 206 ps (56%) and 540 ps (3%). Also shown are the residuals (difference between data and fit)

Figure 2. Decay associated fluorescence spectra (measured with TCSPC) of BBY preparations at room temperature. The sample was excited at 430 nm.

Figure 3. Decay associated fluorescence spectra (measured with streak-camera) of BBY preparations at room temperature. The sample was excited at 400 nm.

Figure 4. Membrane organization of PSII that is used for our coarse-grained modelling. Besides two RC's (D1/D2) (2 Phe *a* and 6 Chl *a* per RC) this dimeric supercomplex contains 1 CP47 (16 Chl *a*), 1 CP43 (13 Chl *a*), 1 CP24 ( 5 Chl *a* and 5 Chl *b*), 1 CP26 (6 Chl *a* and 3 Chl *b*), 1 CP29 (6 Chl *a* and 2 Chl *b*) monomer and 2 LHCII ( 8 Chl *a* and 6 Chl *b* per LHCII) trimers per RC. LHCII trimers are represented by 4-5-6, 7-8-9, 16-17-18, 19-20-21. Also indicated are added putative energy transfer links (short thick bars) between the light-harvesting pigment-protein complexes.

Figure 5. Reconstructed BBY fluorescence decay (dots) using the three main decay components (80, 212 and 639 ps) and the best fit (line) assuming irreversible charge separation (see text) over the time range 0-700 ps. The fitted hopping rate is  $(17 \text{ ps})^{-1}$  and the charge separation rate is  $(1.2 \text{ ps})^{-1}$ .

Figure 6. Different combinations (circles) of  $k_h$  and  $k_{CS}$  that lead to the best description of the BBY decay kinetics, assuming irreversible charge separation (see text). The numbers were obtained as follows: We chose a particular value for  $\tau_h (= k_h^{-1})$  and looked for the best fit of  $\tau_{CS} (= k_{CS}^{-1})$ . Indicated are also the difference between the model and the experimental BBY curve defined as sum of least squares of the deviates (diamonds). The squares indicate the fraction of the trapping time that is due to migration at a particular value for the hopping time. The arrows indicate which vertical axis corresponds to which curve.

Figure 7. The BBY kinetics (dots) are fitted with reversible charge separation into a primary charge separated state and subsequent irreversible charge separation into a secondary charge separated state. The solid line is a fit with a slow hopping time (17 ps). In this case the times for primary and secondary charge separation are 1.24 and 13.3 ps, respectively and  $\Delta G = 2380 \text{ cm}^{-1}$ . The dashed line is a fit with an extremely fast hopping time (1.3 ps). The times for primary and secondary charge separation are now 6.6 and 168 ps, respectively and  $\Delta G = 890 \text{ cm}^{-1}$ .

Figure 8. Models for charge separation in the PSII RC taken from literature. These models are based on measurements on isolated RC's, Fig. 8a model 1 (11) and Fig. 8b model 2 (43) and on measurements on PSII cores, Fig. 8c model 3 (13) and Fig. 8d model 4 (39).

Figure 9a. Solid circles represent reconstructed experimental fluorescence kinetics (TCSPC) of BBY (see text for details). Solid and dashed line represent simulated decay curves, using the 2 RC models of figures 8a and 8b. Electron transfer to the quinone is implemented by assuming irreversible transfer from RP2 with a rate constant of  $(200 \text{ ps})^{-1}$ . The thick lines (1) refer to model 8a, the thin ones (2) to model 8b. For the solid lines a hopping time of 17 ps is taken and an infinitely fast hopping time for the dashed ones.

Figure 9b. Solid circles represent reconstructed experimental fluorescence kinetics (TCSPC) of BBY (see text for details). Solid and dashed line represent simulated decay curves, using the 2 core models of figures 8c and 8d. The thick lines (3) refer to model 8c, the thin ones (4) to model 8d. For the solid lines a hopping time of 17 ps is taken. For the dashed lines a hopping time of 13.4 ps is taken for case 3 (thick dashed) and 0 ps for case 4 (thin dashed).



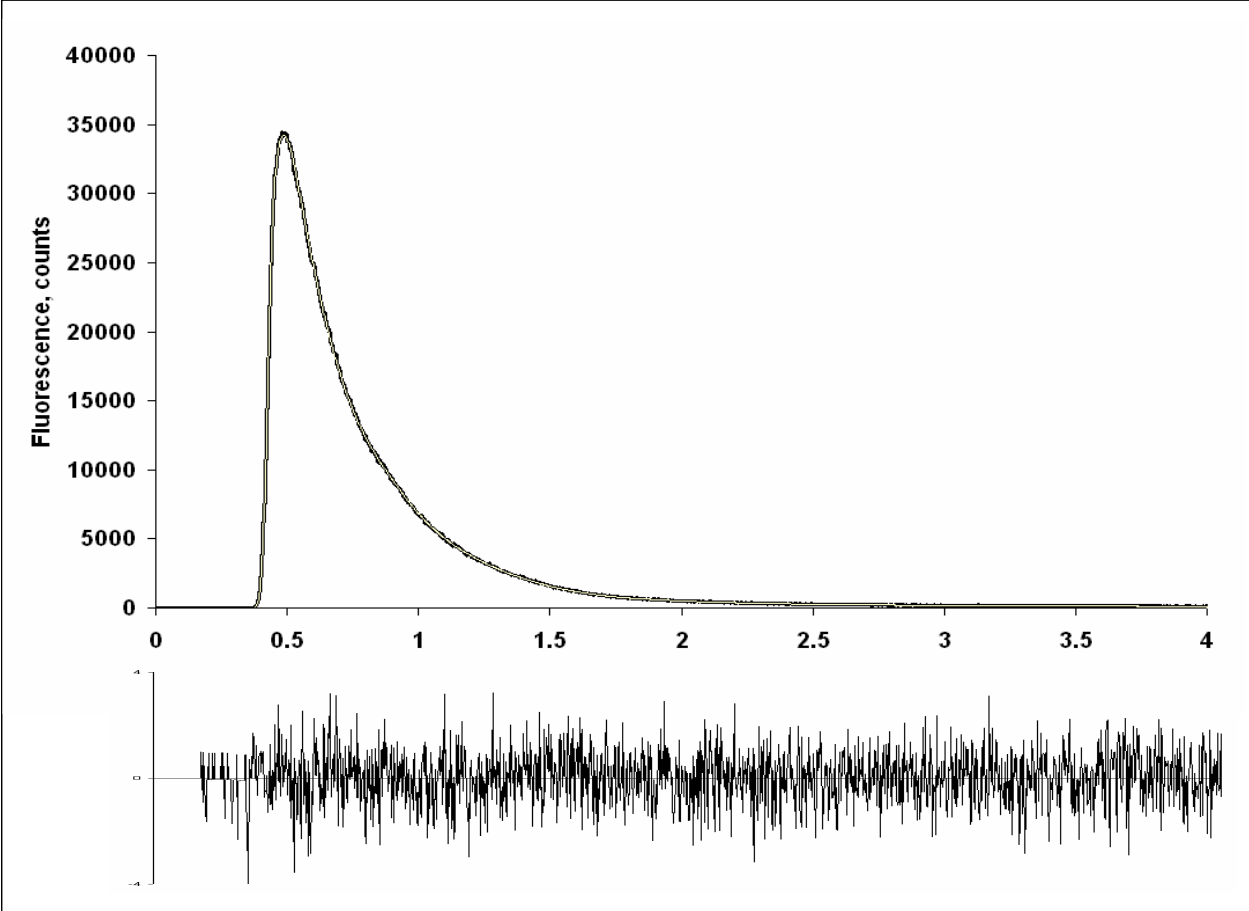


Figure 1

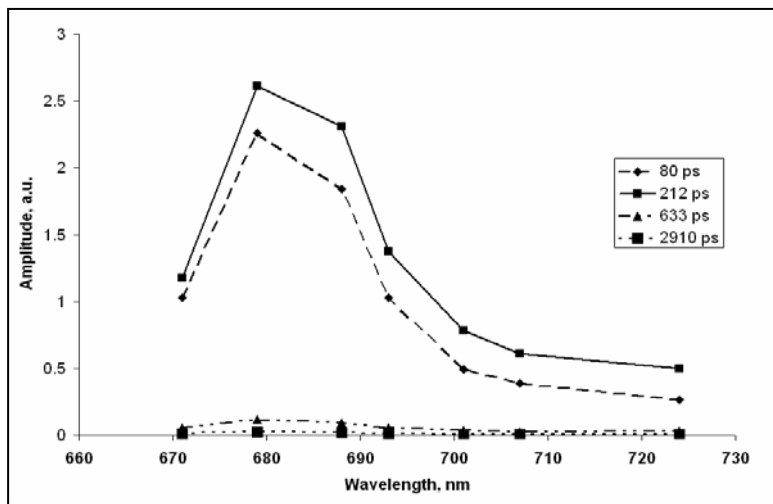


Figure 2

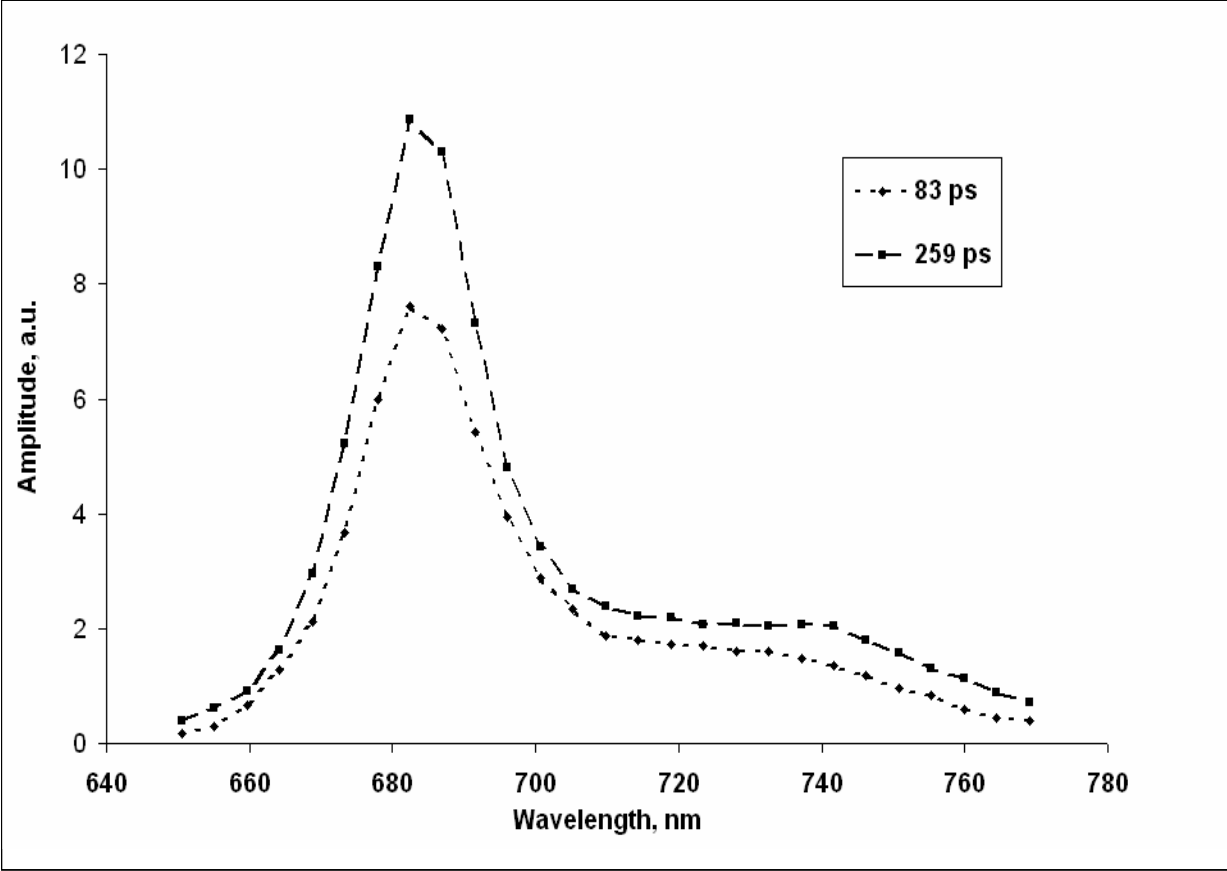


Figure 3

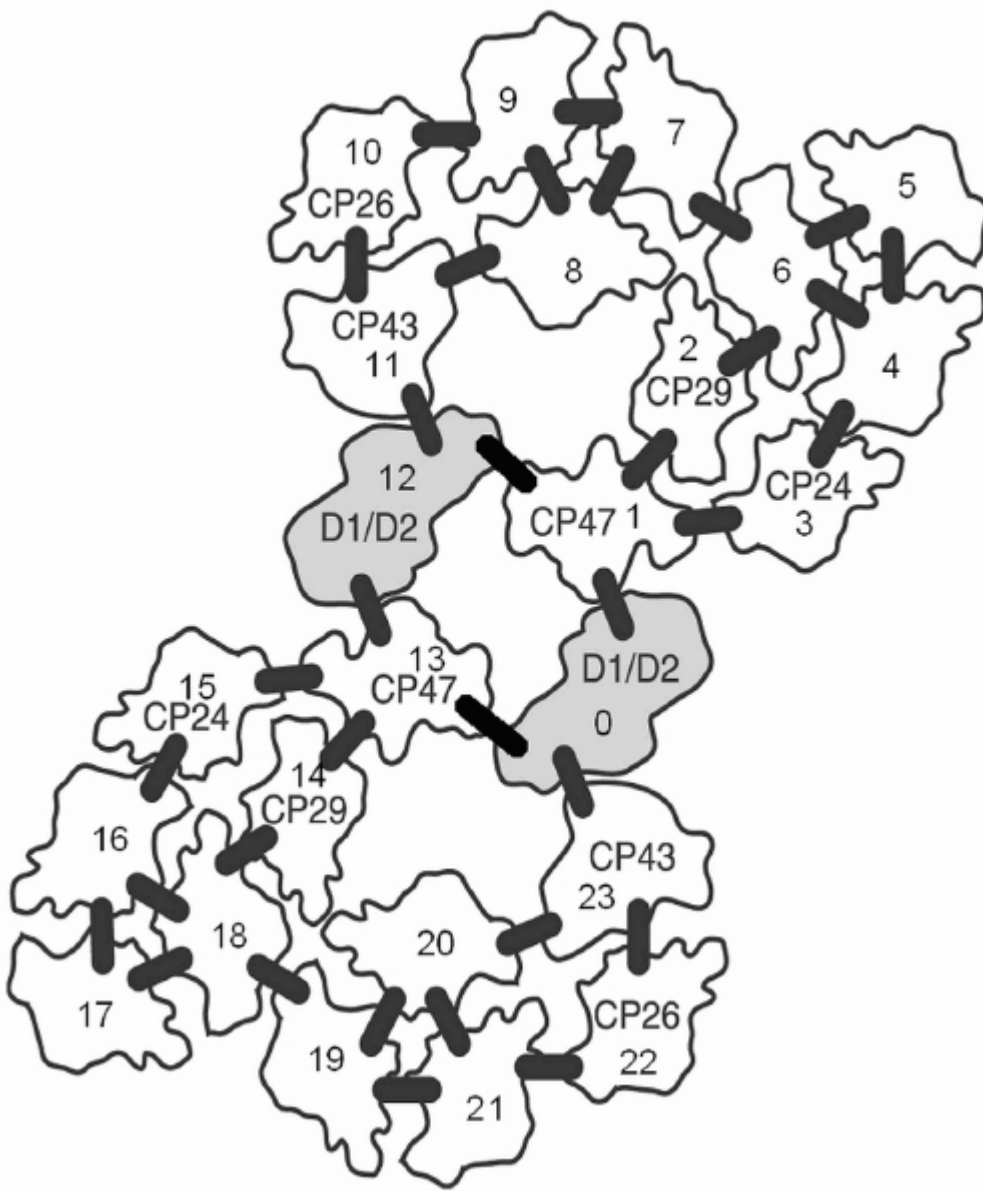


Figure 4

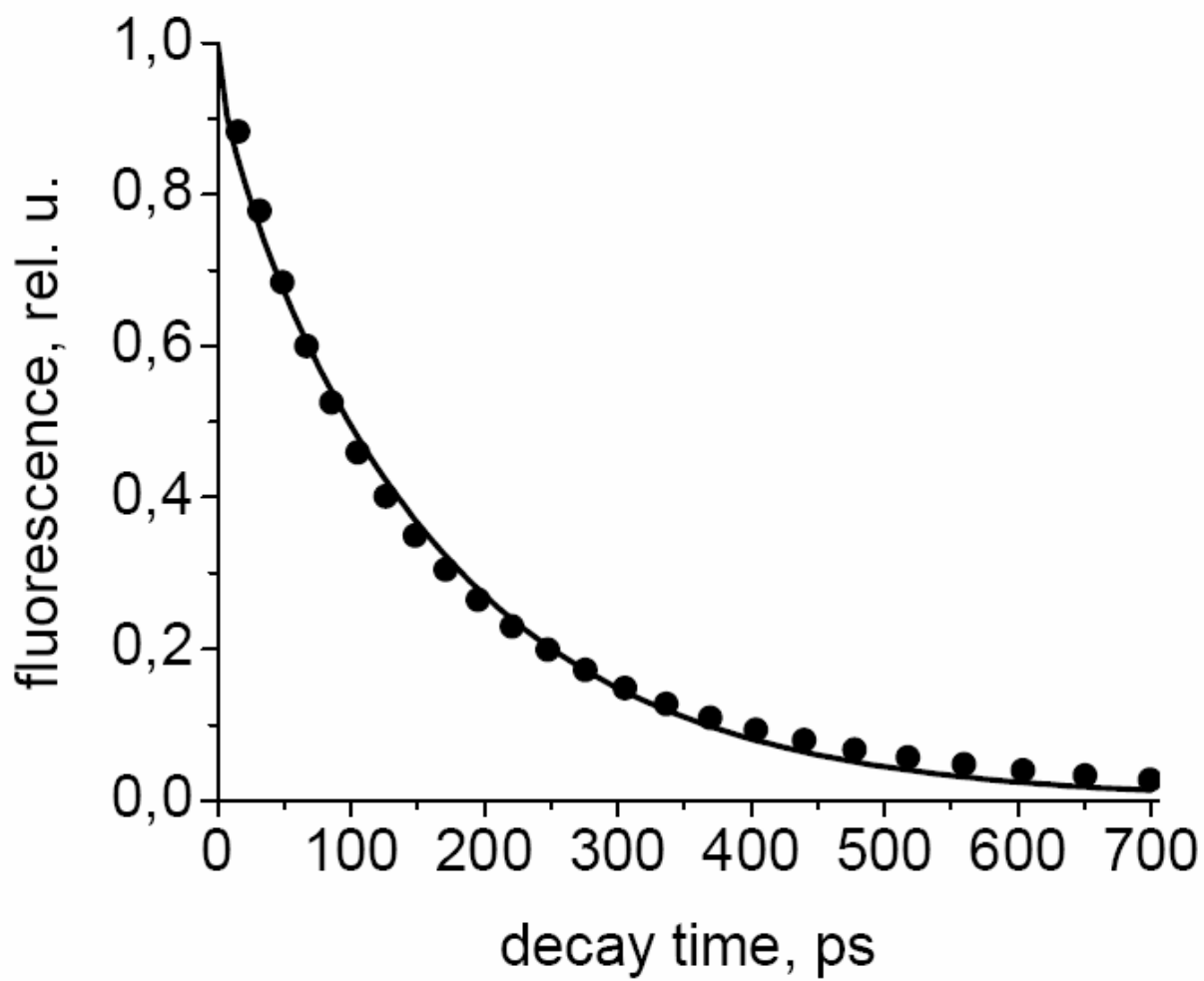


Figure 5

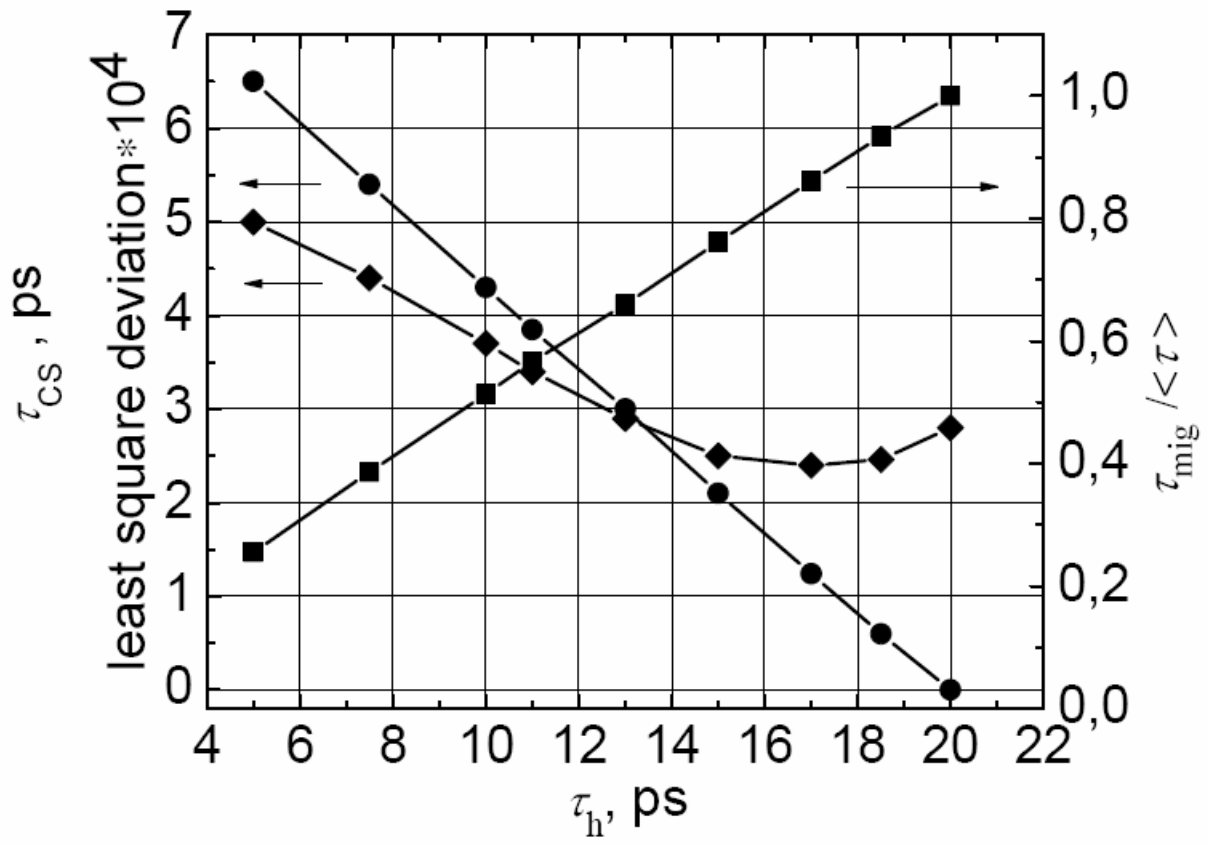


Figure 6

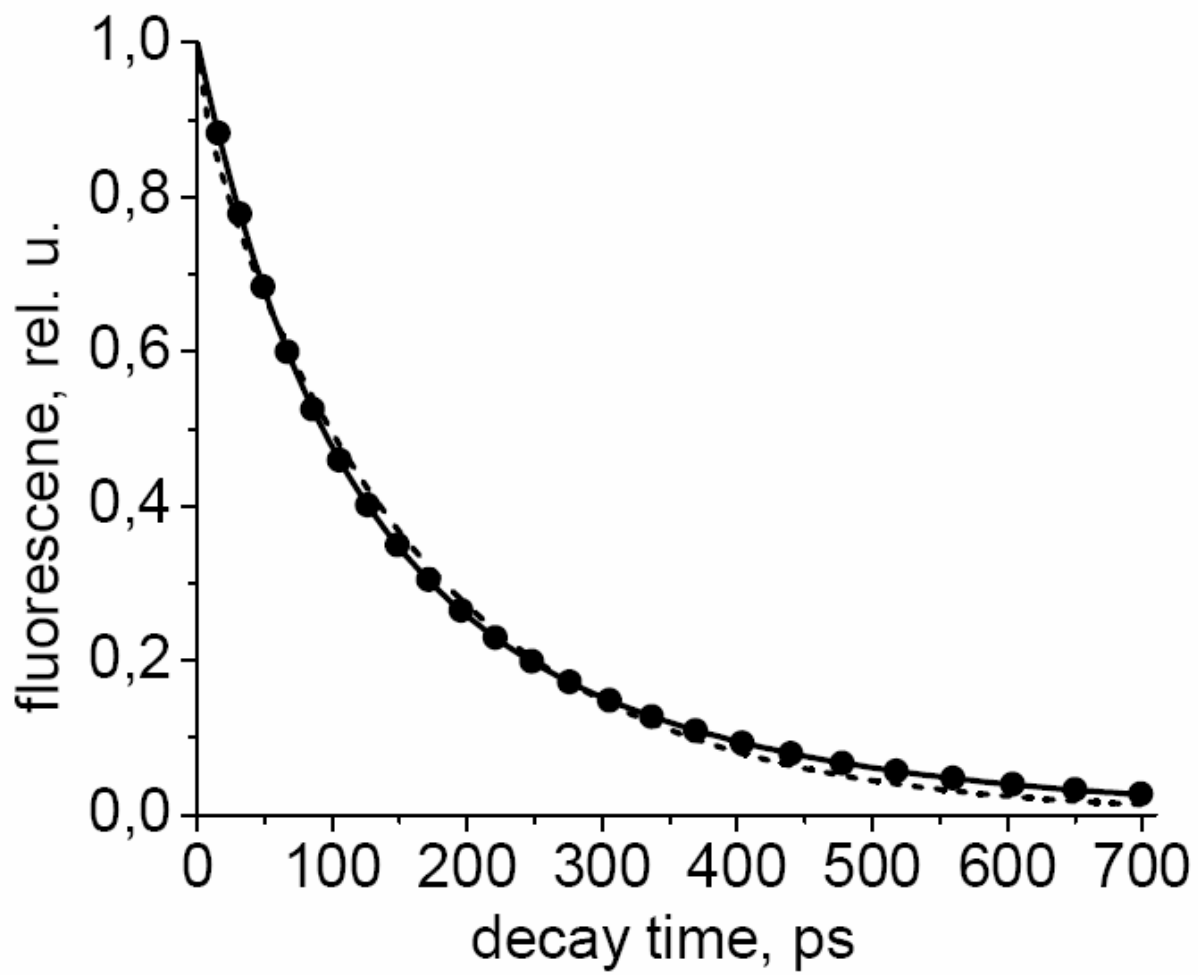


Figure 7

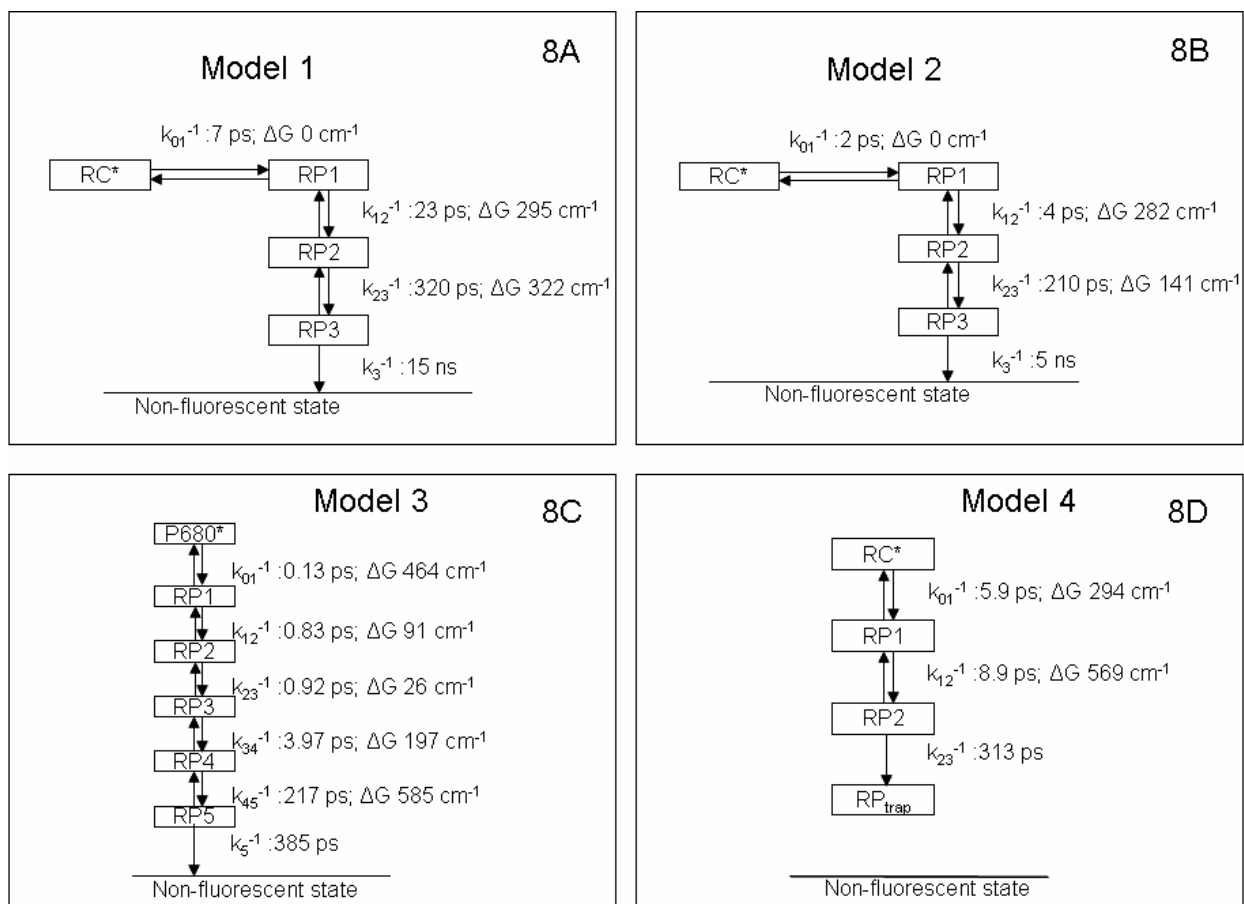


Figure 8



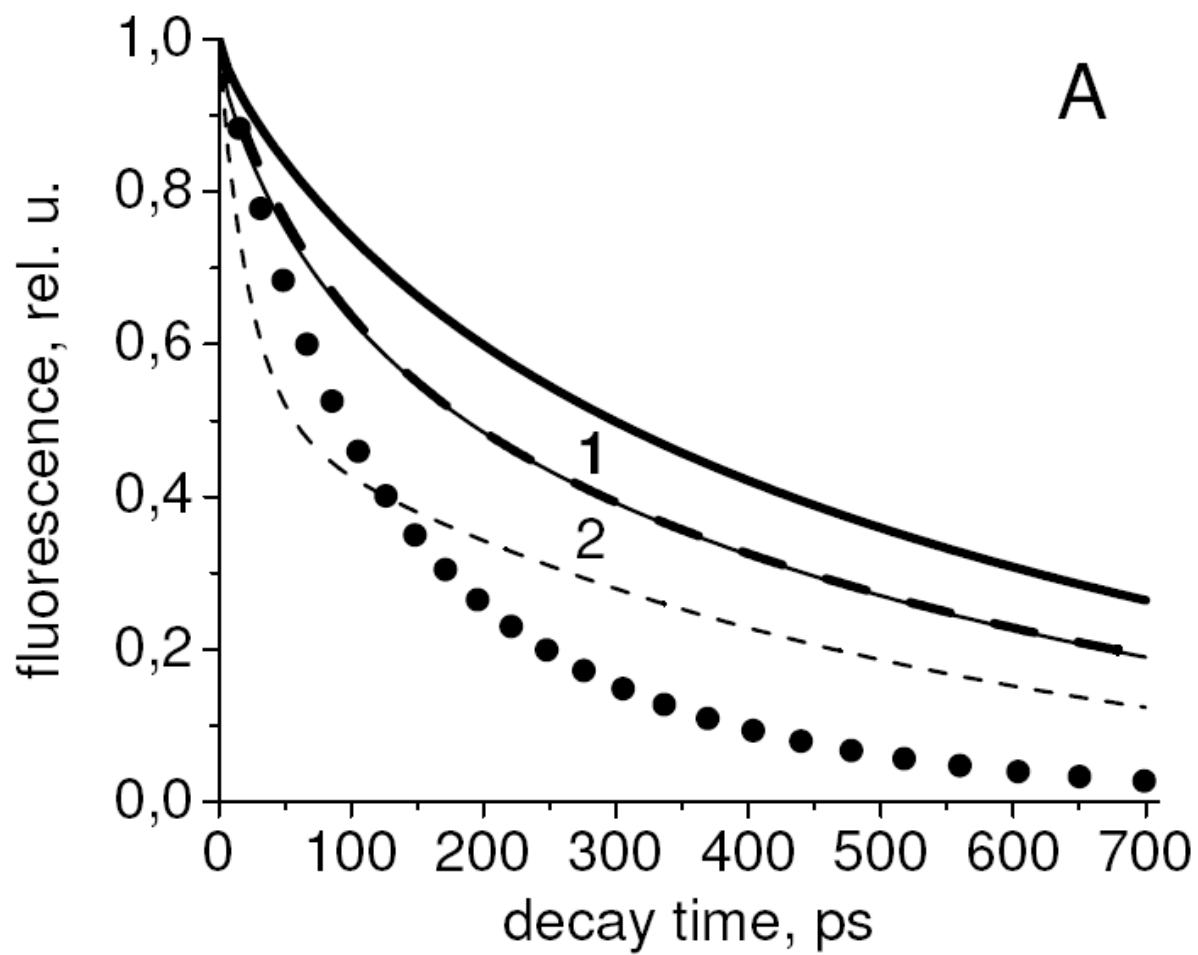


Figure 9a

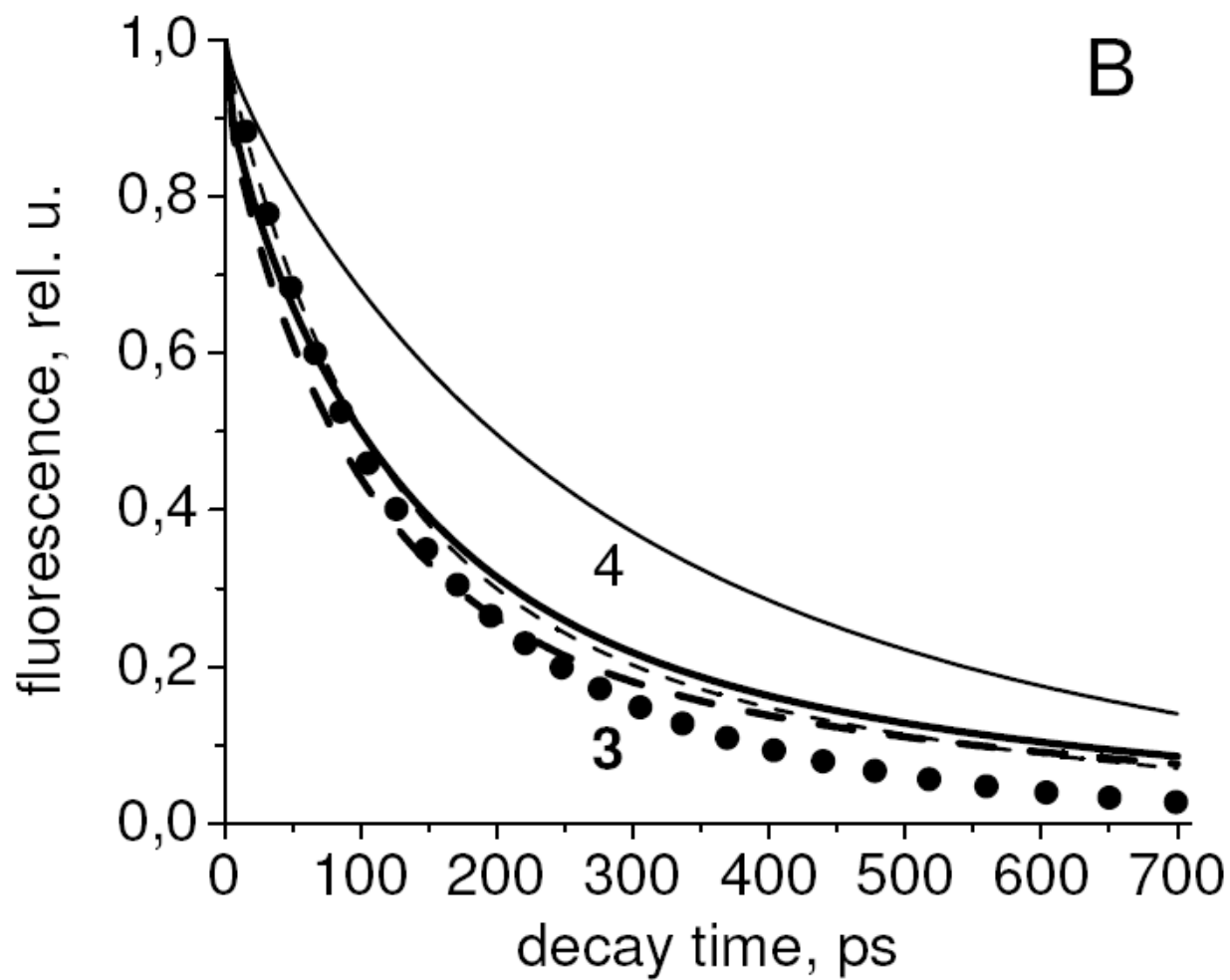


Figure 9b